

Summary of Doctoral Thesis

"Heterogeneous Catalysis for CO₂ Valorization: Catalyst Studies and Infrared Spectroscopy."

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1. Introduction

Carbon dioxide (CO₂) valorization through catalytic hydrogenation can simultaneously mitigate emissions and produce fuels and chemical intermediates when renewable hydrogen (H₂) is available. In this context, CO₂ hydrogenation to methane represents a key Power-to-Gas route, enabling compatibility with natural-gas infrastructures and, potentially, acting as an energy carrier for hydrogen. CO₂ hydrogenation to methanol is an alternative pathway that yields a versatile platform molecule for chemicals and fuels. This thesis investigates heterogeneous catalysts for both routes and uses *in situ* and *operando* infrared (IR) spectroscopy as a central tool to connect catalyst formulation, surface chemistry, and performance.

2. CO₂-to-methane

For CO₂-to-methane, the target product is directly compatible with the existing natural-gas value chain and can support renewable-energy storage. Methanation is thermodynamically favored but kinetically limited and therefore requires active catalysts, typically based on Ni or Ru and operated at moderate-to-high temperatures; under these conditions, the reverse water-gas shift (RWGS) remains the main competing pathway, producing CO that can also participate in subsequent hydrogenation steps. These aspects motivate a detailed investigation of operating parameters and of the catalyst/support interplay to maximize CH₄ selectivity while limiting RWGS. So, a benchmark 1 wt.% Ru/Al₂O₃ catalyst was first tested in fixed-bed reactors to quantify how temperature, feed composition, and space velocity affect activity, selectivity, and competition with RWGS. Subsequently, a library of low-Ru catalysts supported on different oxides was investigated by combining catalytic evaluation with physicochemical and morphological characterization (BET, XRD, STEM-EDX, and temperature-programmed analyses) to rationalize structure-performance relationships. Composition screening highlighted the decisive role of the support and identified low-loading Ru/TiO₂ as a particularly promising formulation (1).

Operando DRIFTS and complementary transmission FTIR were employed to follow surface species under working feeds and to support mechanistic interpretation. On Ru/TiO₂, carbonate and bicarbonate species dominate at low temperature, while a weak band around 1991 cm⁻¹, attributed to linearly adsorbed CO, appears under reaction mixtures. Switching to more, reducing feeds increases CO₂ conversion above 90% while maintaining CH₄ selectivity above 99%, and combining IR and online mass-spectrometric evidence indicates only a minor RWGS contribution at the highest temperatures.

3. OPL-aware DRIFTS methodology

Because diffuse-reflectance spectra are intrinsically sensitive to optical artifacts, a methodological part of the thesis develops an optical-pathlength (OPL) assessment for DRIFTS using CaCO₃ as an internal standard, tracking characteristic CaCO₃ bands (e.g., near 1795 and 2510 cm⁻¹). The study shows that temperature ramps and, especially, redox activation can markedly modify the effective sampling depth as catalysts become more absorbing upon reduction. A practical workflow is proposed to monitor and compensate OPL variations during activation and reaction, enabling robust comparison of *operando* spectra across pretreatments and materials (2, 3).

4. CO₂-to-methanol and zeolite investigations

For CO₂-to-methanol, the thesis builds on the industrial relevance of methanol as a versatile feedstock and on the fact that catalyst selectivity is governed by the balance between CO₂ activation, hydrogenation kinetics, and competing pathways (e.g., CO formation via RWGS and dehydration routes leading to DME). In addition to conventional Cu-based catalysts, indium oxide has emerged as a high-performing material for CO₂-to-methanol, emphasizing the role of oxide redox chemistry and oxygen vacancies in stabilizing CO₂-derived intermediates (4). For In₂O₃-based ZrO₂-supported systems, hydrogen activation sites have been identified as a key descriptor for methanol synthesis under CO₂ hydrogenation conditions (5).

Catalyst development and benchmarking were therefore combined with physicochemical characterization (BET, XRD, STEM-EDX, TPR/TPD) and *operando* transmission FTIR in a high-temperature/high-pressure cell. A catalyst library including Cu-Ce oxides (also prepared via exsolution), Cu/ZrO₂ with different Cu loadings, In₂O₃/ZrO₂, and Cu-In₂O₃/ZrO₂ were tested under pressurized conditions (up to 23 bar, 200–300 °C), highlighting trade-offs between activity and methanol selectivity. Cu-rich Cu/ZrO₂ maximized the CO₂ conversion rate, whereas Cu-In₂O₃/ZrO₂ improved methanol selectivity and stability relative to In₂O₃/ZrO₂. For In₂O₃/ZrO₂, *operando* FTIR supports the formation of CO₂-derived intermediates and underlines the need to control pretreatment severity to avoid over-reduction toward inactive metallic indium. These investigations complement earlier work on In-Cu binary oxides for CO₂ hydrogenation (6).

Operando IR spectroscopy provides additional interpretative handles for selectivity and by-product formation. In methanol adsorption experiments, methoxy species form readily on Cu/ZrO₂, while under CO₂ hydrogenation conditions, carbonate and formate signatures dominate, consistent with a formate-related reaction network. The spectroscopic trends, interpreted together with catalytic testing, help rationalize the emergence of by-products such as CO and DME and clarify how catalyst composition and pretreatment influence surface coverage under working conditions.

Finally, the thesis establishes a quantitative transmission FTIR protocol based on CD₃CN adsorption to compare Brønsted and Lewis acidity in ZSM-5 zeolites. The resulting workflow provides a reproducible mapping of acid-site distributions that is directly relevant to downstream methanol-valorization chemistry (e.g., acid-catalyzed upgrading steps, including DME-related pathways) and supports the broader goal of designing integrated concepts for CO₂ conversion and upgrading.

5. Conclusion

Overall, the thesis combines catalyst structure-performance studies with IR-based methodological advances that strengthen mechanistic interpretation under working environments. The integrated approach enables rational comparison among catalyst formulations for CO₂ methanation and CO₂-to-methanol under representative conditions, while also improving the reliability of *operando* DRIFTS analyses when optical properties evolve during activation and reaction.

6. References

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