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Doctoral Dissertation
Doctoral Program in Electrical, Electronics and Communications Engineering
(36th cycle)

Modeling and Simulation Strategies for Advancing Molecular FCN Pioneering Design Rules Toward Implementation from Molecules to Devices

By

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2024

Declaration

I hereby declare that the contents and organization of this dissertation constitute my own original work and do not compromise in any way the rights of third parties, including those relating to the security of personal data.

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2024

* This dissertation is presented in partial fulfillment of the requirements for **Ph.D. degree** in the Graduate School of Politecnico di Torino (ScuDo).

To the true spirit warriors

Acknowledgements

I would like to thank the constellation of kindred spirits, whose orbits intertwined with mine over the past years, igniting sparks of inspiration that illuminated the winding road of this odyssey.

Thanks to my unwavering support system, whose belief in me has been the compass guiding this journey. You stood by me through the toughest times, offering consistent relief that kept me moving forward, always reminding me of the greater purpose and guiding me through the obstacles, ensuring I never lost sight of the journey ahead and behind. You taught me how to be serene in accepting things I cannot change, to be courageous in changing the things I can, and to be wise in knowing the difference. Your encouragement is etched into the very fabric of this dissertation, and this achievement is as much yours as it is mine.

Abstract

Molecular Field-Coupled Nanocomputing (molFCN) represents a paradigm shift in computational technology, exploiting the unique properties of molecules to revolutionize information processing. In this paradigm, information is encoded in the charge distribution on suitable molecules, and propagation is sustained by Coulomb's force through electrostatic coupling. Logic elaboration is obtained by positioning molecules according to specific layouts. Being a current-less paradigm, molFCN offers extremely low-power and high-frequency operations. Compared to other implementations, having molecules as unit elements offers the advantage of working at room temperature. Despite these noteworthy advantages, obtaining a working prototype requires effort in multiple research areas, slowing down the development of the technology over the last two decades.

This thesis aims to facilitate the future development of experimental validations and a working prototype. It explores different characteristics of the technology with the goal of defining design rules and properties for molecules and circuits, driving future research on the topic.

The proposed goal is addressed by decomposing the problem into subsequent abstraction levels, each one based on the previous finding, and hiding the non-relevant details as a matryoshka. Each abstraction level requires its own theory and simulation tools.

The first layer is composed of molecules, being the unit elements. This work analyzes various candidates and defines a set of parameters identifying the behavior to be provided to chemists to drive ad-hoc syntheses. Adopting DFT simulations and the MoSQuiTo methodology, this work also analyzes a case study to investigate the direct effects of multiple molecular modifications on its electrostatic behavior.

Based on the analyses performed in the previous layer, the second step treats molecules as simplified electrostatic devices composed of a set of point charges.

This modeling enables the study of molecule ensembles through the application of electrostatic fundamentals and the SCERPA simulator. The work analyzes the fundamental interactions found in molFCN devices using different electrostatic molecular footprints to test their impact. Consequently, the design of tailored molecular transcharacteristics enables the successful realization of new functions, such as a fully electrostatic solution for in-plane information crossover.

As an outer layer, this work proposes a methodology to model entire logic gates and develops an automatic tool named BBchar to create a library comprising the input/output behavior of fundamental gates and interconnections. The possibility to simulate circuits composed of multiple gates without focusing on the internal behavior of each of them enables a strong simulation speed-up, moving from hours or days down to a few milliseconds. Moreover, the proposed methodology is designed to consider the actual physical behavior of the technology, resulting in a 99.99% accuracy of the proposed model.

The final point of the present work proposes a modification of the paradigm that aims to simplify the reading of the information at the laboratory level and on-circuit data conversion. The paradigm has been named bend-boosted molFCN to highlight the property of including molecule bending in the charge-encoded information. Through DFT and dynamic simulation, this work demonstrates its correct working principle, and the offered read-out capabilities.

Overall, this thesis contributes to the advancement of molFCN technology by providing insights into molecular behavior, developing efficient simulation methodologies and tools, and proposing innovative device architectures that simplify data sensing and conversion. It lays the groundwork for future research in this field and offers a pathway toward the unprecedented realization of a working prototype.

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List of Acronyms

AC Aggregated Charge

AIMD Ab Initio Molecular Dynamics

B3LYP Becke3 Lee–Yang–Parr

BBchar Block-Based characterization

BDL Binary Dot Logic

BF Bistable Factor

CAD Computer-Aided Design

CHELPG CHarges from ELectrostatic Potentials using a Grid

CMOS Complementary Metal-Oxide-Semiconductor

CUT Cell Under Test

DB Dangling Bonds

DFT Density Functional Theory

DrC Driver Cell

DSOA Device Safe-Operating Area

DUT Device Under Test

EBL Electron Beam Lithography

EDA Electronic Design Automation

ESP ElectroStatic Potential

FCN Field-Coupled Nanocomputing

FEM Finite Element Method

FIB Focused Ion-Beam

H-Si Hydrogen-terminated Silicon

HF Hartree-Fock

iNML In-plane Nano Magnetic Logic

LUT Look-Up Table

MBPT Many-Body Perturbation Theory

MD Molecular Dynamics

molFCN molecular Field-Coupled Nanocomputing

MoSQuiTo Molecular Simulator Quantum-dot cellular automata Torino

MUT Molecule Under Test

MV Majority Voter

NEGF Non-Equilibrium Green's Functions

QCA Quantum-dot Cellular Automata

RT-TDDFT Real-Time Time-Dependent Density Functional Theory

SAM Self-Assembled Monolayers

SCERPA Self-Consistent Electrostatic Potential Algorithm

SET Single-Electron tunneling Transistor

SiDB Silicon Dangling Bonds

STM Scanning Tunnelling Microscope

TJ Tunnel Junction

TP Transmission Pathways

UHV Ultra-High Vacuum

UKS Unrestricted Kohn-Shann

VACT V_{in} -AC transcharacteristic

VHDL VLSI Hardware Description Language

Preface

The research presented in this thesis explores the fascinating world of molecular Field-Coupled Nanocomputing (molFCN), a field positioned at the intersection of molecular electronics, nanotechnology, and computational science. This endeavor has been a labor of passion, curiosity, and intellectual pursuit, and it is with great pleasure that I present the culmination of my research efforts in this document. Over the course of my doctoral studies, I have delved deep into the exploration of molecular-scale devices and their potential applications and integration in computing.

The primary objective of this research is to investigate the potential of molFCN as a promising paradigm for next-generation electronics. By leveraging the unique properties of molecules, nanoscale devices, and ultra-low power consumption, this study aims to reduce the distance from a prototypal validation of the technology. The major challenge is tightly linked with the need for expertise in various technical areas, such as molecular synthesis, 2D deposition, and nanofabrication. The ultimate goal is to develop a framework that enables simulations at different abstraction levels, providing valuable insights to experts in low-level technology requirements, and offering engineers a tool to work with the technology without necessarily needing to understand the underlying physics.

The research presented in this thesis adopts a multidisciplinary approach, drawing on principles from computational chemistry, molecular electronics, computational modelling, and algorithm development. Investigations are complemented by computational simulations, theoretical modeling, and data analysis to explore underlying mechanisms and validate theoretical predictions.

It is important to note that while this research represents a significant step forward in advancing our understanding of molFCN, it is not without limitations. Challenges such as device scalability, fabrication reproducibility, and reliability remain areas of

ongoing research and development. Additionally, the scope of this study is focused on specific architectures and ideal practical implementations.

In conclusion, I am sincerely grateful for the opportunity to embark on this intellectual adventure and to contribute to the ongoing pursuit of knowledge in the field of molecular field-coupled nanocomputing. It is my hope that this work will inspire further exploration and collaboration among different research entities, fostering innovation for a transformative future.

Outline of Chapters

This thesis is organized into the following chapters:

Chapter 1 - Introduction and context The chapter provides an overview of various field-coupled nanocomputing implementations, discussing their fundamental concepts, advantages, and current challenges. It concludes by outlining the motivation for the present work and explaining the justification behind choosing molecular implementations over other alternatives.

Chapter 2 - Molecules design principles This chapter explores the crucial role of molecules in molFCN, detailing how their composition and arrangement shape device behavior. It examines theoretical methods, structural characteristics, and key molecules in the field. Finally, it concludes with a detailed case study analyzing the impact of modifications on device performance metrics.

Chapter 3 - Molecules electrostatics and environment This chapter explores the significance of intra-molecular interactions and electrostatic footprints in molFCN, focusing on how molecular charge distributions influence information propagation. By analyzing different molecular species and their electrostatic behaviors, the chapter investigates their impact on logic computation and explores strategies for enhancing circuit capabilities through the integration of diverse molecules. Additionally, it discusses the feasibility of achieving coplanar crossover interconnections using purely electrostatic solutions.

Chapter 4 - Device modeling The chapter outlines a methodology for developing physically accurate models of entire devices using a hierarchical approach, validated against Self-Consistent Electrostatic Potential Algorithm (SCERPA)

simulations for accuracy. It provides a tool to handle devices easily, facilitating investigations into the computational potential and scope of the technology.

Chapter 5 - Interface with the external world This chapter highlights the challenges of experimental validation due to difficulties in measuring molecule charges and introduces a novel bend-boosted molFCN paradigm, demonstrating its feasibility through Density Functional Theory (DFT) simulations. Additionally, it discusses compatibility with existing charge-based molFCN, paving the way for experimental validation and integration with CMOS technology.

Chapter 6 - Conclusion and future perspective The conclusion chapter summarizes the key findings and contributions of the research, highlighting the advancements made in understanding and modeling molecular field-coupled nanocomputing. It reflects on the challenges encountered, the implications of the proposed solutions, and the potential future directions for further exploration and development in the field.

Chapter 1

Introduction and context

Since its establishment, modern digital electronics has experienced a remarkable evolution thanks to the use of Complementary Metal-Oxide-Semiconductor (CMOS) technology, which has dominated the semiconductor industry for several decades. In the past two decades, the limitations of CMOS technology have become increasingly evident. The continuous demand for higher performance, lower power consumption, and more complex functionalities poses significant challenges for CMOS. These challenges include power dissipation, scaling limits, and the increasing complexity of manufacturing processes. Consequently, there is a growing need to explore alternative solutions that can push the boundaries of what is achievable in digital electronics. In response to these challenges, researchers and industry leaders are actively exploring a diverse range of options, from novel materials like graphene and carbon nanotube [1, 2] to different computation approaches [3, 4].

In the last two decades, two main approaches have been identified, namely “More than Moore” and “Beyond CMOS” [5]. The “More than Moore” concept acknowledges that further advancements in electronic technology cannot solely rely on shrinking transistor sizes as predicted by Moore’s Law [6]. Instead, it emphasizes diversifying and integrating new functionalities into microelectronics, including sensors, power management, and embedded components [5]. On the other hand, the “Beyond CMOS” movement explores alternative technologies to CMOS, which is approaching its physical scaling limits. These novel technologies enclose a broad spectrum of approaches, such as DNA nanocomputing [7], spintronics [8], and photonic computing [9], offering the potential for enhanced performance, energy

efficiency, and functionality. Under the definition of “Beyond CMOS” solutions, the Field-Coupled Nanocomputing (FCN) emerges [10], offering the possibility to perform computation in a transistor-less environment and not exploiting charge transfer, preferring the local interaction of some fields.

The following sections discuss the four different technological implementations of the FCN paradigm, introducing the fundamental concepts of information encoding and propagation. The implementations are compared, and advantages and current challenges are presented. The chapter ends with the motivation of the present work.

1.1 Field-coupled nanocomputing implementations

The FCN paradigm exploits an innovative concept for computing at the nanoscale and finds a realization in different technologies. The basic idea is to use a field-based control, e.g., electric or magnetic, to drive the behavior of nanoscale components. This innovative approach aims to overcome some of the limitations of traditional semiconductor-based computing, such as power consumption and scalability [10]. Indeed, one of the primary advantages of FCN is its potential for significantly reduced power consumption because of the use of coupling fields instead of an electric current. Nevertheless, each specific implementation comes with its own advantages and challenges.

The following paragraphs discuss the leading implementations of the FCN paradigm, which are the metallic Quantum-dot Cellular Automata (QCA), the semi-conductive QCA, the In-plane Nano Magnetic Logic (iNML), and the molecular Field-Coupled Nanocomputing (molFCN).

1.1.1 Metallic quantum dots

The principal approach investigated for the paradigm realization is called QCA [11]. A fundamental QCA unit comprises a square arrangement of four quantum dots interacting by tunneling. Fig. 1.1a illustrates a QCA cell with the four sites highlighted, where t indicates the tunneling energy between adjacent sites [12]. When two additional electrons are introduced into the unit, their Coulombic repulsion forces them to occupy dots on opposite corners. Fig. 1.1b shows the two resulting

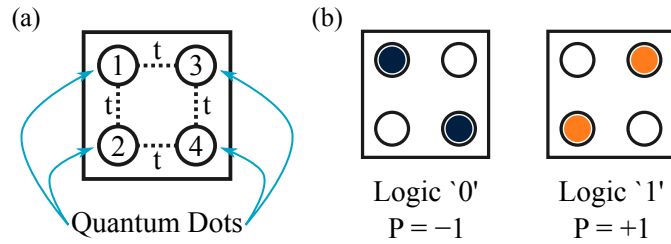


Fig. 1.1 Quantum-dot Cellular Automata (QCA). (a) Unit cell composed of four quantum dots interacting by tunneling, where numbers identify the position of the site in the cell and t is the tunneling energy linking two sites. (b) Resulting state configurations obtained by adding two electrons to the cell, represented as colored dots. The two distinct cells encode the logic values '0' and '1' and are associated with a cell polarization value of $P = -1$ and $P = +1$.

distinct state configurations. The two states are energetically equivalent and can be designated as logic '0' and '1'. A useful quantity describing the charge arrangement of the cell is the so-called cell polarization P , defined as

$$P = \frac{(\rho_2 + \rho_3) - (\rho_1 + \rho_4)}{\rho_1 + \rho_2 + \rho_3 + \rho_4} \quad (1.1)$$

where ρ_i is the electronic charge of the dot i . The cases corresponding to the two logic values are associated with $P = -1$ and $P = 1$ as reported in Fig. 1.1b. Besides, for an isolated cell, the ground state is an equal combination of these two states and, therefore, has a net polarization of zero.

Cells interaction: information propagation and elaboration

In a system of several cells, it is assumed that electrons can tunnel between the sites within the same cell, whereas electrons are prevented from tunneling between distinct cells. This restriction is achieved by maintaining a greater distance between dots in different cells compared to the spacing within the same cell, given that the probability of tunneling decays exponentially with distance. So, separate cells interact through the coupling of Coulomb interactions. Considering a system of two cells like the one in Fig. 1.2a, if we fix the polarization of the cell on the left, it is possible to predict the charge distribution of the cell on the right [11]. Ideally, the minimum energy configuration is the one for which the two cells encode the same information [11], as shown in Fig. 1.2b for both logic '0' and logic '1'. The steepness of the two-cell

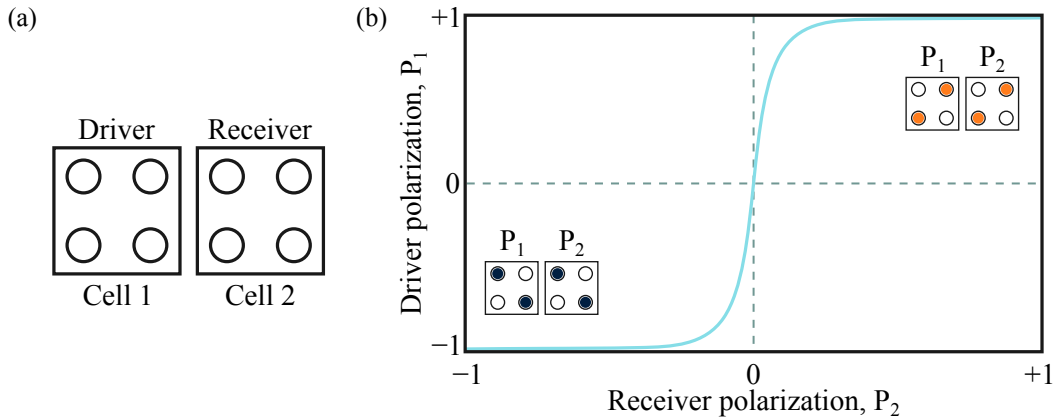


Fig. 1.2 Quantum-dot Cellular Automata (QCA) two-cells system. (a) Scheme and nomenclature for a system composed of two adjacent QCA cells. (b) Cell-to-cell response showing the variation of the receiver polarization versus the driver polarization.

polarization curve mainly depends on the tunneling energy between the dots and the Coulomb energy for electrons on different sites [11].

The same principle of information copying is at the basis of information propagation [13]. By placing several cells in a row, the value forced at the beginning propagates through all the cells through electrostatic coupling. Fig. 1.3a illustrates the ground state for a wire with a logic ‘0’ forced as input, and for a wire propagating a logic ‘1’. From a logical point of view, the wire just described can be seen as a sequence of drivers copying the input information. By rotating the cell 45° , nearby cells assume opposite configurations, thus creating a wire of inverting cells, as shown in Fig. 1.3b [13].

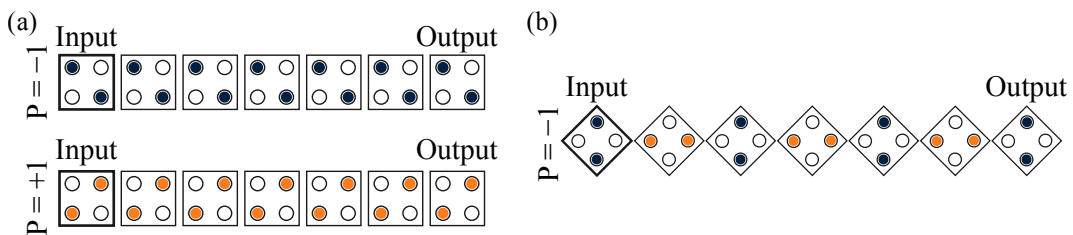


Fig. 1.3 Quantum-dot Cellular Automata (QCA) information propagation. (a) Propagation through electrostatic coupling of adjacent cells for the two logic values. (b) Inverting propagation through electrostatic coupling of cells rotated by 45° .

Since the interaction is based on electrostatic interaction, the outcome of a cell depends on the superposition of all the nearby cells. Considering the arrangement of Fig. 1.4a, the polarization of the computing cell depends on the polarization of

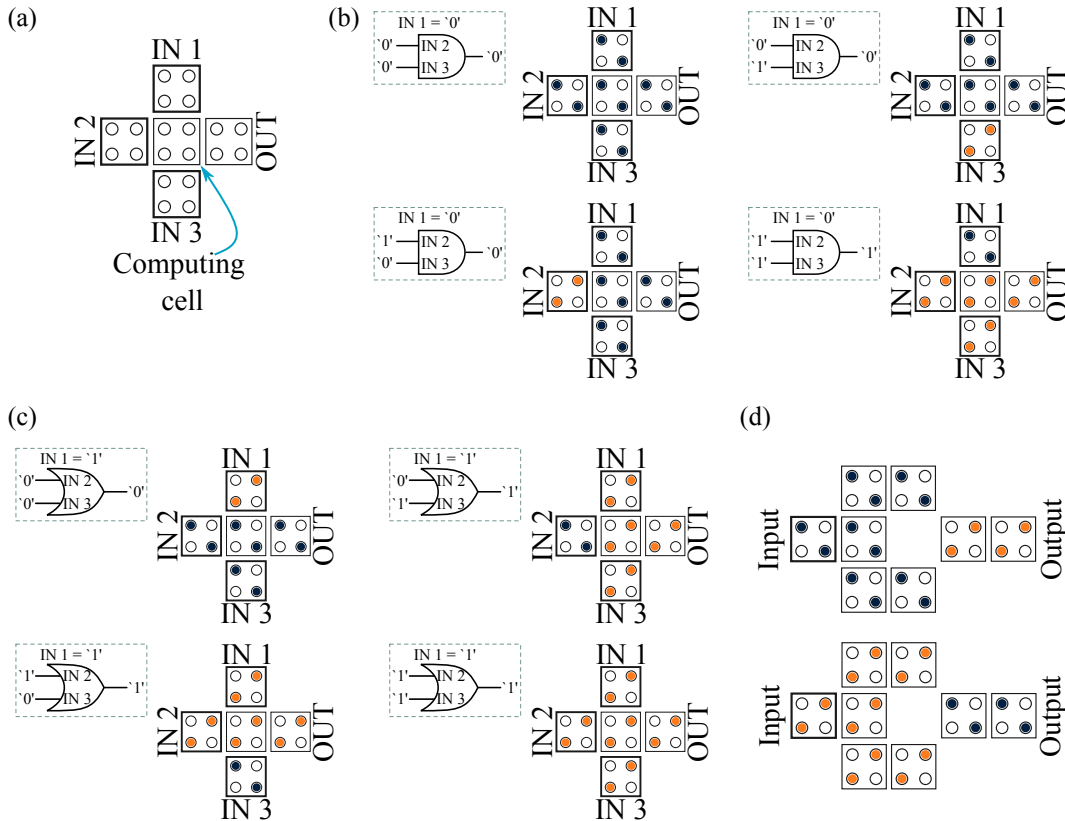


Fig. 1.4 Quantum-dot Cellular Automata (QCA) fundamental gates. (a) Majority voter (MV) layout with cells name highlighted. (b) Charge configuration of the MV with IN1 fixed to '0' implementing a two input AND gate between IN2 and IN3. (c) Charge configuration of the MV with IN1 fixed to '1' implementing a two input OR gate between IN2 and IN3. (d) Inverter layout and charge distribution for the two possible input values.

all three input cells [14]. Specifically, this layout implements the majority voting function, whose truth table is Table 1.1. In the Majority Voter (MV) gate, the polarization of the computing cell equals the same assumed by most of the input cells. The same layout can also work as a programmable gate implementing an AND or an OR gate between the remaining inputs [13, 15], as observable in Fig. 1.4b and reported in Table 1.1. Fig. 1.4b shows the charge configuration of a MV where IN1 has been fixed to implement the AND operation between the remaining inputs. Similarly, Fig. 1.4c shows the charge configuration of a MV where IN1 has been fixed to implement the logic OR. The union of the configurations shown in Fig. 1.4b and Fig. 1.4c represents the possible charge distribution in a generic 3-input MV.

Playing with the cell arrangement makes it possible to obtain other fundamental logic operations, such as the inverter shown in Fig. 1.4d for both the logic configu-

Table 1.1 Truth table of a three-input majority voter gate. Note that the top half of the table corresponds to an AND gate between IN2 and IN3, keeping IN1 fixed, whereas the bottom half identifies an OR gate in the same conditions.

IN 1	IN 2	IN 3	OUT
0	0	0	0
0	0	1	0
0	1	0	0
0	1	1	1
1	0	0	0
1	0	1	1
1	1	0	1
1	1	1	1

rations. In this case, the laying principles are the vertical copying mechanism and the diagonal interaction. Indeed, cells are equal on each side, making it possible to replicate the information in any direction. On the contrary, diagonal interaction behaves similarly to rotated cell propagation, generating an inversion on the output cell.

The clocking necessity

We mentioned that the two logic states correspond with two energy minima, meaning that information is potentially stable even without a power supply. If we want to change the state encoded by a cell, an external force is needed to reduce the energy barrier existing between the two states. To address this problem, six-dot cells and the application of an external electric field have been proposed [16, 17].

Fig. 1.5a shows the six-dot unit cell in the two logic states and the added state used to erase the information before switching to the new logic. The third created state is named ‘NULL’ to highlight that it does not encode any information. The external field is called *clock field*, and is a signal that enables the quasi-adiabatic switching between the possible cell configurations. The clock field provides energy to the system cyclically to enable electrons to move and varies smoothly enough for the system to stay close to its equilibrium ground state configuration during the switching [18]. Fig. 1.5b shows the accepted trapezoidal clock waveform and the name of the phase sequence. Fig. 1.5b also shows the equivalent potential profile

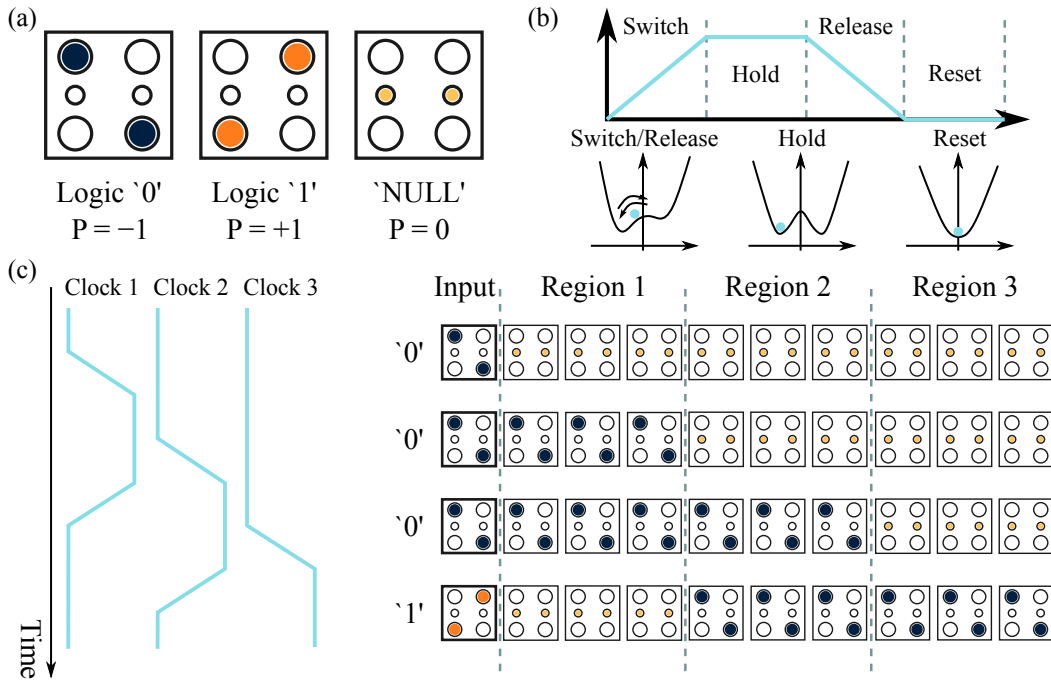


Fig. 1.5 Clocked Quantum-dot Cellular Automata (QCA). (a) Unit cells composed of six quantum dots and state configurations obtained by adding two electrons to the cell. (b) Typical period of the trapezoidal clock signal highlighting the name of each phase and energy-coordinate scheme for each of them showing the schematic potential profile. (c) Propagation example in a clocked wire divided into three regions subjected to out-of-phase clock signals. The sequential activation of the regions guides the propagation direction.

in the various phases versus the coordinate, and the blue circle is the particle in the system [18].

In addition, the electrostatic interaction fundamental for propagating information is not directional, meaning that a driving cell influences all the surrounding cells, losing the concept of the direction of propagation. By dividing the circuit into zones to which we provide out-of-phase clocking signals, we can force the information to follow specific paths to guarantee the synchronous operation of QCA devices. Fig. 1.5c shows an example of a wire divided into three regions. After each time step, information is guided from the input to the output thanks to the sequential activation of the clocking scheme [19].

Clocking in QCA introduces a unique advantage concerning energy efficiency. QCA devices only consume energy during state transitions, minimizing overall power consumption and potentially addressing the growing concerns associated with energy efficiency in modern computing [18].

Fabrication experiments

The fabrication of metal dots involves techniques employed to produce arrays of dots on a substrate. The precise placement and spacing of these dots are critical to the functioning of the QCA system, as they define the logic states and interactions [20]. Few experiments have been carried out using aluminum dots and aluminum-oxide tunnel junctions, demonstrating the possibility of switching cells [21, 22].

Metal dots are typically in the order of a few tens of nanometers [23, 18] and offer several benefits, including a comparatively high fabrication yield and consistent electrical behavior [10]. This consistency implies that the energy needed to introduce each subsequent electron into the dot remains relatively constant, even as numerous electrons are added. Such uniform behavior simplifies the process of loading the QCA cell with the precise number of electrons and facilitates biasing the cell to ensure that the two polarizations possess equivalent energy levels [24].

Regarding the clock signal generation, it is accepted to use metal rods beneath the circuit level to generate the potential associated with the clocking signal [17].

While these fabrication and modeling procedures are considered feasible, the main challenges are represented by scaling and the necessity of operating at cryogenic temperatures [25, 15].

1.1.2 Semiconductive quantum dots

In semiconductive QCA, the quantum dots themselves are made from semiconductor materials, such as silicon or gallium arsenide [26]. This choice of material offers several benefits, including compatibility with existing semiconductor fabrication processes and the potential for integration with current electronic devices [27]. By utilizing semiconductor quantum dots, semiconductive QCA systems seek to overcome some of the challenges associated with purely quantum-based QCA, such as scalability or operation at extremely low temperatures [27].

Silicon Dangling Bonds (SiDB) is a FCN implementation exploiting the properties of passivated silicon surfaces and manipulating unpaired electrons on these surfaces [28]. SiDB can be implemented following the QCA paradigm [29], even if the most studied implementation is referred to as Binary Dot Logic (BDL) [30].

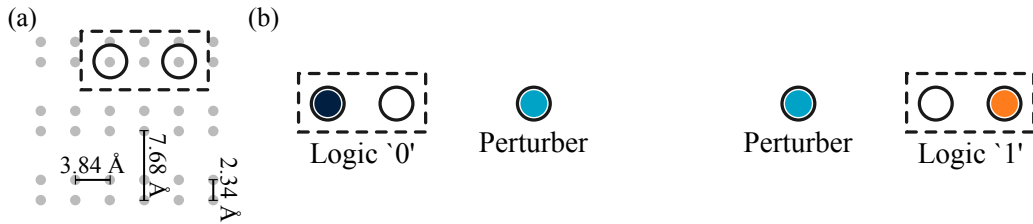


Fig. 1.6 Silicon Dangling Bonds (SiDB) in the binary dot logic. (a) Schematic representation of a pair of dangling bonds (DBs) on a Hydrogen-terminated Silicon (H-Si). (b) Resulting charge configurations obtained by placing a third DB near the pair. The two distinct cells encode the logic values '0' and '1', and colored dots are associated with the DB occupation by the electron.

In the SiDB technology, quantum dots are realized through dangling bonds in Hydrogen-terminated Silicon (H-Si), pursuing the ultimate miniaturization in technology. SiDB on H-Si can exhibit three charge states [31]: positive, neutral, and negative. The charge state of a Dangling Bonds (DB) is determined by the number of electrons it holds. Specifically, positive charge states (+) hold zero electrons, neutral charge states (0) occur when the DB holds one electron, and negative charge states (-) are associated with the dangling bond holding two electrons. By placing two DB away by two lattice sites, a DB pair is formed, as shown in Fig. 1.6a [31]. It has been experimentally demonstrated that at specific applied bias, the pair has one net negative charge and, from an electrostatic point of view, it behaves as an isolated DB presenting fluctuations between left and right dots [30, 31].

Introducing a third DB to the pair demonstrates a clear contrast between DBs. As Fig. 1.6b shows, the charge is biased to stay on one DB in the pair, creating a binary charge configuration that we can associate with logic '0' and logic '1'. The isolated DB added to the system to force a logic state is typically called *perturber*.

A line composed of multiple paired SiDB forms a binary wire, schematically shown in Fig. 1.7a. Experimentally, it has been demonstrated that the wire states can be manipulated by placing a perturber at one end of the line or the other [30]. Also, the wire becomes self-polarized without any external electrostatic input if the perturber is missing.

In BDL, the easiest device to design is the OR gate, which has been also demonstrated experimentally [30]. Fig. 1.7b illustrates the two-input OR layout and the four possible combinations. A complete library of working devices has been proposed and validated through simulations in [32]. Even SiDB circuits need the presence of a

clock that helps the synchronization and information guidance. So far, no complete studies have addressed the influence of an external field on the physical behavior of DB or the maximum clock region dimension that must be satisfied to guarantee the correct functioning [33]. Anyway, a tool based on Finite Element Method (FEM) simulations has been developed in [34] to evaluate the clock potential generated by metal rods similar to those proposed in [17].

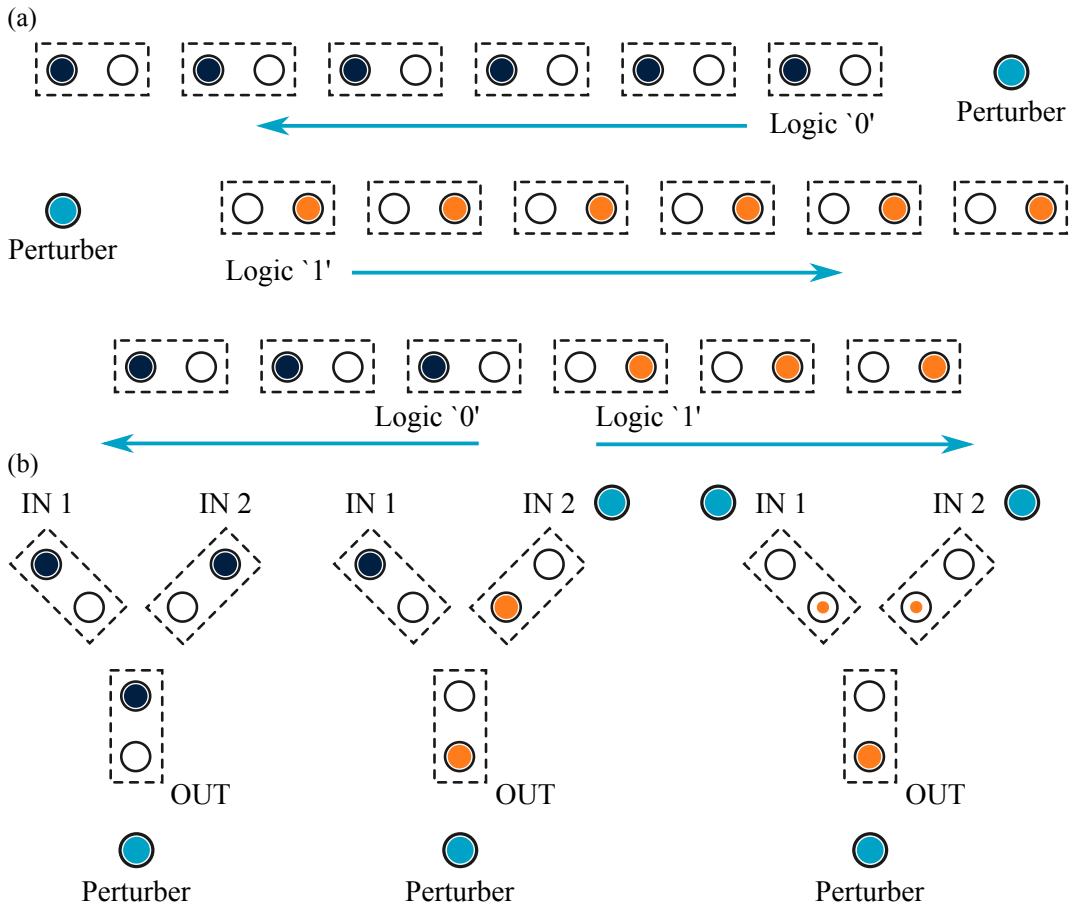


Fig. 1.7 Silicon Dangling Bonds (SiDB) propagation and fundamental gate. (a) Binary wire in different situations: with a perturber on each side of the wire, and in the case of self-polarization. (b) Combinations of a two-input OR gate.

Besides the experimental effort made to assess the technology [30, 29], a lot of investigation on the power of the implementation comes from the simulation of circuits and the possibility of automating the design phase through Computer-Aided Design (CAD) tools [35, 36]. The main challenges are related to the development of

input/output interfaces that will permit the removal of the perturbers, which are now fundamental for the functioning of circuits.

1.1.3 In-plane Nano Magnetic Logic

In-plane Nano Magnetic Logic (iNML) technology represents another implementation of FCN, exploiting single-domain nanomagnets as the holders of binary information [37]. As Fig. 1.8a shows, these nanomagnets are characterized by stable magnetizations parallel or antiparallel to the easy (longest) axis and encode logic '1' and '0' states. The technology capitalizes on the shape anisotropy of rectangular magnets, with typical dimensions of a few tens of nanometers [38]. If magnets are correctly designed, magnetization vectors are stable, meaning there's no standby power consumption, and magnets serve as both memory and logic elements.

A magnetic wire is constructed by cascading an even number of nanomagnets. Information propagates through the magnetic coupling of neighboring magnets. Fig. 1.8b shows that if magnets are juxtaposed facing the easy axis, then the coupling is antiferromagnetic. On the contrary, if magnets are aligned facing the hard (shortest) axis, information is coupled ferromagnetically.

If we change the input information by changing the magnetization vector of the initial magnet, a cascading effect is expected, causing all the subsequent magnets in the wire chain to flip in a domino-like manner [39]. Nevertheless, the energy barrier associated with the shape of the magnets that guarantee the information stability is too elevated to be overcome by the dipole-dipole energy provided by the input magnet only [40]. As shown in Fig. 1.8c, an external clock field becomes necessary to induce a 90° rotation of the magnetization vector along the short axis, facilitating information propagation in iNML [41]. However, thermal noise limits the number of magnets that can be cascaded above a clocked zone, typically to 4 or 6 [42, 38]. Various clocking schemes have been proposed to ensure data flow direction and correct information propagation [19].

The potential for constructing magnetic circuits is related to the possibility of aligning magnets in various ways. The resulting iNML gates, depicted in Fig. 1.8d, form the basic building blocks, which have been proved through simulations and experimental validation [43, 38]. The MV is efficiently designed even in this implementation. Nevertheless, it is possible to build two-input AND and OR gates

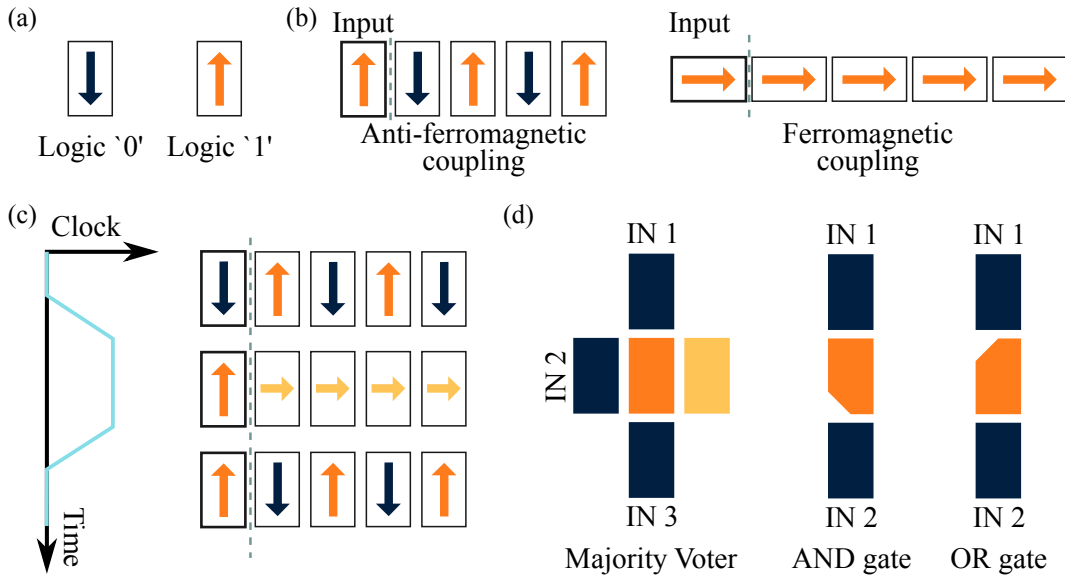


Fig. 1.8 In-plane Nano Magnetic Logic (iNML) logic, propagation, and fundamental gates. (a) Elementary cell composed of a rectangular magnet highlighting the magnetization vector. (b) Information propagation through antiferromagnetic and ferromagnetic coupling. (c) Effect of applying a clocking signal on the magnets of a wire, resetting the information before changing the input value. (d) iNML fundamental gates: majority voter, two-input AND gate, and two-input OR gate.

without using a MV by exploiting magnets with a cut on one angle. The cut favors the magnetization vector pointing toward the cut itself [44]. Since the horizontal propagation is anti-ferromagnetic, an odd number of magnets following the input will make an inversion of the information, and it's possible to balance length by adjusting the width of the magnets forming the inverter.

Standard tools like low-level and high-level simulators are used to study iNML devices at different abstraction layers. Micromagnetic simulators like oommf [45] and mumax³ [46] are employed for accurate magnetic behavior simulations. The limitations of low-level simulators, such as being time-consuming for large architectures, and the drawbacks of high-level simulators, which result in losing physical information, led to the development of ToPoliNano [47]. This tool follows a top-down approach, allowing the description of circuits in VLSI Hardware Description Language (VHDL), automatic generation of optimized layouts, fast and accurate simulations, consideration of fabrication process faults, and flexibility for extension to other emerging technologies.

1.1.4 Molecular implementation

Another way to implement FCN is to use the molecule to implement a QCA cell or, more often, a half cell where redox-active molecular sites replace general dots [48]. Fig. 1.9a shows the thiolated-carbazole linked bisferrocenes molecule (bisferrocene for short) that we use throughout this work for the simulations [49]. The coordinates of the molecule are reported in Appendix A. The molecule comprises two ferrocenes connected to a carbazole in a ‘V’ shape. The carbazole is itself linked to a thiol, forming a ‘Y’ shaped molecule. The functional groups of the molecule are usually referred to as *dots*. The two ferrocenes, i.e., dot 1 and dot 2, are also named logic or active dots. Similarly, the union of dot 3 and dot 4 is called reset dot.

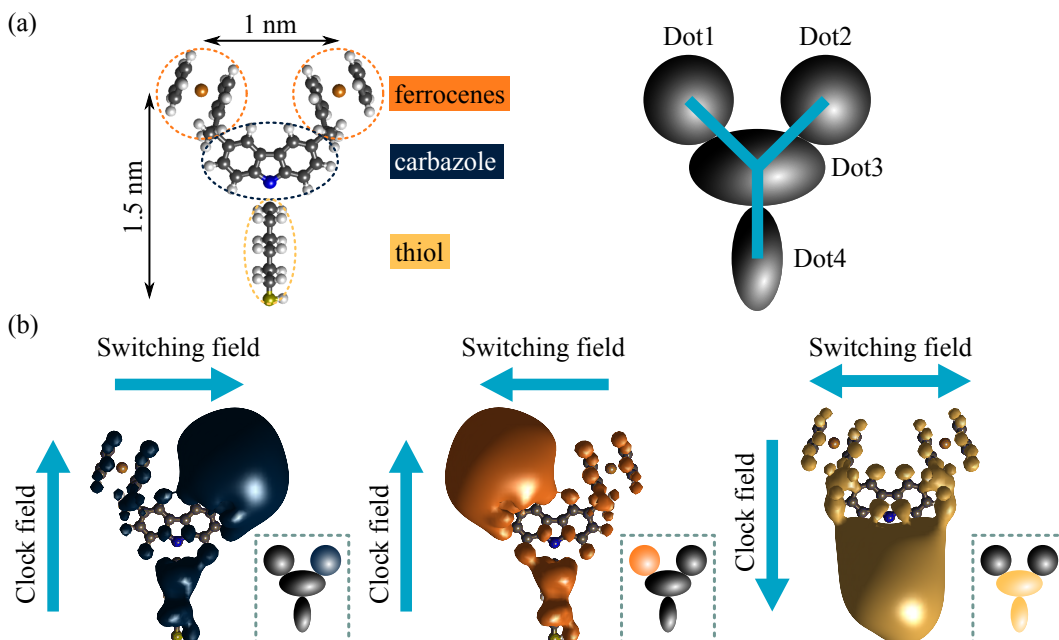


Fig. 1.9 Molecular Field-Coupled Nanocomputing (molFCN). (a) Geometry of the thiolated-carbazole linked bisferrocenes molecule, highlighting the functional groups and the extension. The figure on the right shows a schematic representation of the molecule. (b) The application of electric fields across the molecule changes the electronic distribution, represented as a colored cloud. Combining the clock field and the switching field generates three different charge configurations. Atoms in figure are colored as follows: H = white, C = charcoal, N = blue, S = yellow, Fe = orange.

Applying electric fields across the molecule moves the charge distribution in different molecule regions. Fig. 1.9b shows the isopotential surface on the bisferrocene in different situations, corresponding to three distinctive charge localization. In particular, the clock field is applied along the longitudinal axes and can move

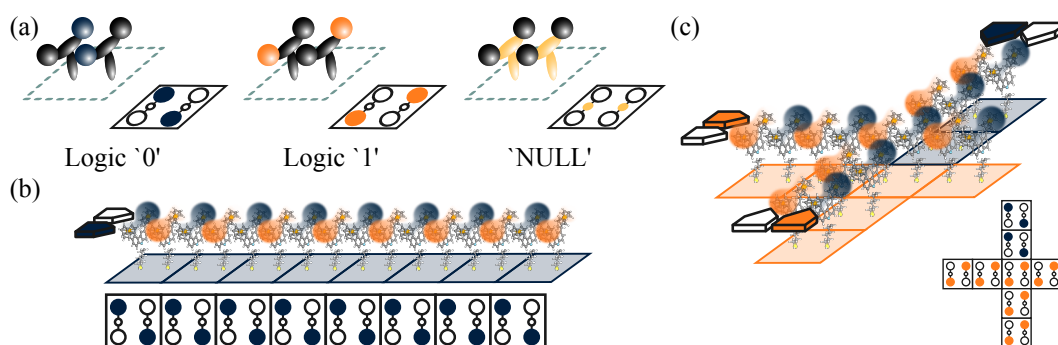


Fig. 1.10 Molecular Field-Coupled Nanocomputing (molFCN) information encoding and propagation. (a) Molecular cell composed of two juxtaposed molecules encoding the logic states '0', '1', and 'NULL'. The molecular cells are linkable to 6-dots QCA cells. (b) Molecular wire composed of 8 cells propagating a logic '0'. (c) Majority voter with the top input equal to '0' and the other inputs encoding a logic '1'.

the charge between the ferrocene and the carbazole/thiol group [50]. When the clock field moves the charge in the ferrocenes, the switching field applied across the molecule transversal axis forces the charge on one of them.

Different charge configurations are possible by placing two of those molecules side-by-side and depending on the applied clock field. When the clock field favors the presence of the charge in the ferrocenes (*hold state*), the charge clouds on the molecules maximize the distance for Coulombic repulsion, and two energy-equivalent configurations are possible. Fig. 1.10a shows the schematic molecular cells composed of two molecules in the mentioned configurations associated with logic '0' and '1'. Fig. 1.10a also shows the molecular cell when the clock favors the presence of the charge in dot 3 and dot 4. In that case, the molecular cell is associated with the 'NULL' state, where no information is encoded. Fig. 1.10a also illustrates that the molecular cells can be associated with a 6-dots QCA cell.

In molecular FCN (molFCN), the information propagation principle is the same as general QCA: the localized charge in a cell influences nearby cells, and the clock signal drives the information along the desired path. Fig. 1.10b shows a molFCN wire of 8 cells propagating a logic '0' and the schematic QCA view that simplifies the representation. In molFCN, fundamental gates are designed by playing with cell displacement, and the way to obtain AND and OR gates is by using the MV, shown in Fig. 1.10c when the top input equals '0' and the other two encode a logic '1'.

1.2 Molecular field-coupled nanocomputing

Using molecules as unit elements provides some unique advantages. Typical molecules proposed for molFCN are nanometric sized, and devices are built with few of them, allowing for highly compact and densely packed circuits [51]. The inherent nature of QCA and molFCN, that is, utilizing charge configurations and Coulombic interactions, allows for energy-efficient computation [52, 53]. Information propagates without charge transfer, requiring only the switching within each cell, whose energy has been studied and is of the order of 10^{-20} J [53]. To have an idea, the switch for a transistor requires a minimum energy of the orders of 10^{-18} J, so two orders of magnitude bigger [54]. Then, the Coulomb force between neighboring cells generates an interaction, letting molFCN work at room temperature [55, 56].

Alongside molecules being complex systems, it is not likely to treat several molecules as they are in a molFCN circuit. We need to define ad-hoc models and make specific considerations. Two main approaches exist in the literature to describe a molecule in a FCN environment: one focused on a theoretical quantum-mechanic description and the other based on data extrapolation from computational chemistry simulations. The following paragraphs describe the main concepts of the two approaches.

1.2.1 Theoretical approach: three-state approximation

The theoretical model has been introduced in [51] and describes a two-dot QCA half-cell using a double-well system. The cell binary information is encoded by the localization of a mobile charge in one of two wells. The entire four-dot cell can be viewed as a pair of half-cells with alternating polarizations. This model is named two-state approximation and describes electron transfer dynamics in a two-dimensional Hilbert space. The electron transfer in a cell is described in the context of interaction with a neighboring molecule, emphasizing Coulomb interactions. The work in [51] outlines a simplified electrostatic model for intermolecular interactions, considering two neighboring molecules with four dots forming a square, as in the general QCA paradigm. The cell–cell response function is used as a tool to evaluate molFCN functionality in relation to molecular structure, incorporating structural elements such as the electron transfer matrix element and the distance between redox centers [51].

The three-state approximation is a natural evolution of the model including an effective description for a clocked molFCN cell [57]. In this model, the Hamiltonian of a cell is constructed from basis vectors corresponding to the charge configurations of three logic states: the null state and two completely polarized states. Cell-to-cell interactions are characterized by the intercellular Hartree approximation, treating Coulombic interactions with a mean-field approach [57].

For molFCN half-cells represented by three stable charge configurations, the strategy involves conducting high-level *ab initio* calculations on single molecules to determine energies corresponding to the ‘0’, ‘1’, and ‘NULL’ states. The 3×3 model Hamiltonian constructed with the three-state approximation is given by

$$H = \begin{bmatrix} E_0 - e\epsilon d/2 & -\gamma & 0 \\ -\gamma & E_0 + E_c & -\gamma \\ 0 & -\gamma & E_0 + e\epsilon d/2 \end{bmatrix} \quad (1.2)$$

where E_0 is the energy of each state, E_c is the energy provided by the clock, ϵ is the switching field from neighboring molecules, and d and γ are the effective distance and tunneling energy between central and edge dots [58]. The Hamiltonian’s two zeros correspond to the spatial separation between the two edge dots.

A tool named QCADesigner has been developed following this theoretical description for the simulation of molFCN, and QCA in general, circuits [59, 60].

1.2.2 Simulative approach: from *ab initio* to electrical devices

An alternative method for investigating molFCN employs *ab initio* simulations. While this serves as a reference and initial step, crucial for validating the system under examination, it comes with certain limitations. Firstly, these simulations demand substantial computational resources and extensive memory. Additionally, outcomes like molecular orbitals and atomic charge distribution are challenging to interpret from a device-level perspective. Controlling biasing conditions, such as input electric field, clock electric field, or technological parameters, proves complex. In response to these challenges, an approach named Molecular Simulator Quantum-dot cellular automata Torino (MoSQuiTo) has been introduced [61]. This approach, organized into three stages, involves analyzing the molecule through *ab initio* simulations, post-processing the obtained results to generate new figures of merit and device

characteristics, and finally implementing a higher-level algorithm to simulate the circuit and interactions among its elements.

First stage

The depiction of molecules involves using the Z-matrix or the coordinates, serving as the starting point for each *ab initio* simulation. Subsequently, biasing conditions are applied to consider elements such as the electric field generated by an input driver, a clock signal, or the presence of other molecules simulated through point charges [61]. The orientation of each field is specified, and the magnitude is systematically varied across a relevant range. The outcomes typically derived from these simulations include potential surfaces and the distribution of charges associated with each atom.

Second stage

The second stage is employed to analyze the information obtained in the preceding step and formulate novel metrics that characterize the molecule at a higher level of abstraction, sufficient to enclose the characteristics functional for FCN computation [62]. By using Fig. 1.11 as a reference, the principal metrics are the following [62].

Aggregated Charge (AC) is calculated as the sum of all the charges belonging to a functional group or a dot and is centered on a point of interest. Rather than collecting information on atomic charges, having information related to each dot is simpler.

Input voltage V_{in} is the integral of the electric field computed along the segment that connects the active dots (Dot 1 and Dot 2) of the Molecule Under Test (MUT).

Voltage at the receiver V_{out} is the integral of the electric field generated by the MUT, computed using the AC, on the segment that connects the receiver active dots when the receiver is placed at a distance $d = w$, where w is the distance between the MUT active dots. The same electric field that is related to the information encoding is usually named *switching field*.

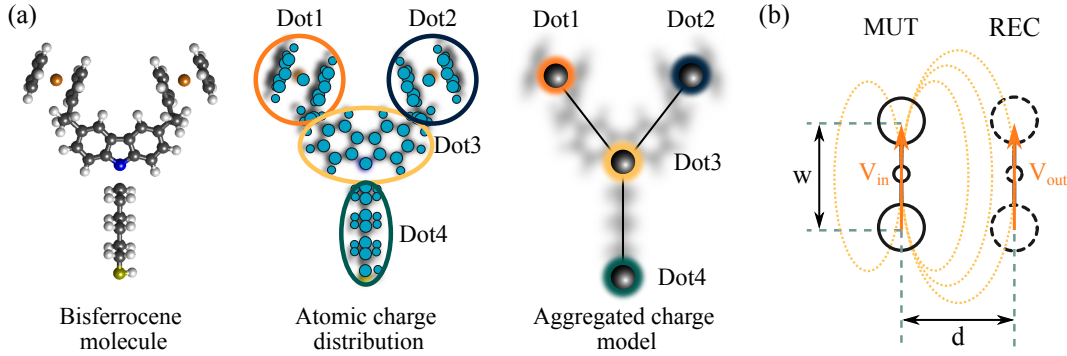


Fig. 1.11 Fundamental figures of merit defined in the Molecular Simulator Quantum-dot cellular automata Torino (MoSQuiTo) methodology. (a) Schematic representation of the aggregated charges obtained by the sum of the atomic charges for each dot. (b) Definition of the input voltage on the Molecule Under Test (MUT) and the voltage at the receiver (REC) evaluated at a distance $d = w$. Atoms in figure are colored as follows: H = white, C = charcoal, N = blue, S = yellow, Fe = orange.

By their associations, we define the V_{in} - V_{out} transcharacteristic (VVT) relating the output voltage generated on a dummy receiver molecule to the input voltage of a driving molecule, and the V_{in} -AC transcharacteristic (VACT) relating the AC of a molecule to a specific input voltage and a clock field.

The behavior of a molFCN cell can be linked to the general QCA paradigm by combining the QCA polarization expressed in equation (1.1) and the molecule AC. Fig. 1.12a shows the molecular cell composed by two molecules $M1$ and $M2$, whose logic dot charges are $Q_{M1,1}$ and $Q_{M1,2}$, $Q_{M2,1}$ and $Q_{M2,2}$, respectively. The polarization of the cell can be defined as in equation (1.3), according to the QCA paradigm.

$$P = \frac{(Q_{M1,2} - Q_{M1,1}) - (Q_{M2,2} - Q_{M2,1})}{Q_{M1,1} + Q_{M1,2} + Q_{M2,1} + Q_{M2,2}} \quad (1.3)$$

In particular, considering a maximum switching AC equal to 1 a.u. for each molecule, the cells with polarizations $P = -1$ and $P = +1$ are those propagating logic values '0' and '1', respectively. Fig. 1.12b shows these two simplistic cases encoding the two logic states, with the entire charge of each molecule on one of the two logic dots.

Third stage

In this stage, all the necessary elements for formulating an algorithm to simulate interaction have been established, giving rise to Self-Consistent ElectRostatic Potential

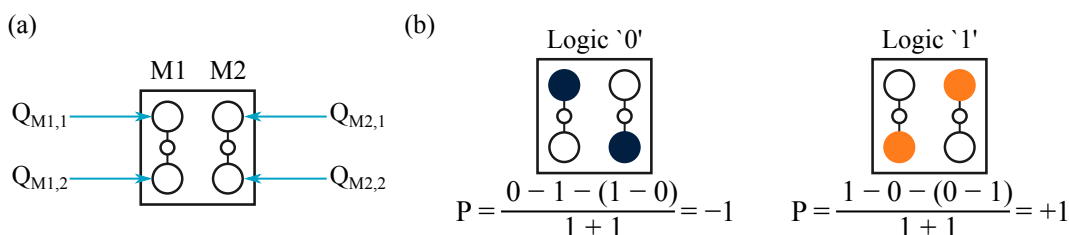


Fig. 1.12 Definition of the cell polarization for molFCN to associate it with the QCA paradigm. (a) Molecular cell composed of two molecules, $M1$ and $M2$, with a highlight on the names of logical dot charges. (b) Association of two cell polarization values to the logic states.

Algorithm (SCERPA) [63, 64]. SCERPA is an iterative algorithm designed to assess information propagation in molFCN circuits. In a molecular system, each molecule generates and responds to electric fields induced by other molecules or electrodes, defining different system states. Given the circuit description and information obtained from previous MoSQuiTo stages, SCERPA determines the relative effects of all the molecules in a self-consistent procedure. Upon reaching convergence, SCERPA provides information on the charge distribution of each molecule in the circuit for a specific clock/input combination.

1.2.3 Fabrication ideas

Most of the work in the last years focused on theoretical analysis and simulations. Unlike the other FCN implementations, an experimental demonstration of propagation or elaboration in molFCN is still missing. From the technological point of view, some works proposed molecules that hold the properties described for molFCN [49, 65, 66]. Regarding the possibility of fabricating a device, few studies proposed using a substrate onto which molecules attach employing a tile [49, 67, 68].

The primary challenge arises from molecules typically having dimensions in the range of 1–2 nm. Consequently, using lithography-based technology to achieve wires with comparable dimensions takes time and effort [69]. Additionally, regardless of size, precision in constructing angles is especially crucial for a cross structure like that of the MV. Once the nanopatterned substrate is defined, one valid approach proposed for molecule deposition involves using Self-Assembled Monolayers (SAM) [70–72]. It comes by itself that to obtain regular SAM, the nanopatterned substrate must be

precisely built. Any imprecision at the atomic level of the substrate may reflect in shifted, tilted, rotated, or misaligned molecules [67, 61].

Managing the generation of electric fields essential for computation, including the clock field and the switching field of input molecules, introduces another complexity requiring additional wires. The clock electrodes can be buried under the molecule deposition level [68] or on the edge of a trench [73]. Achieving precision at the level of few molecules is challenging. The dimension of the electrodes determines the number of molecules affected by the generated field, and inside this region, propagation must be guaranteed. Information write-in is achieved with the same concept, forcing an electric field on the input molecules that behave as molecules in a nanogap [67].

1.3 Motivations of the work in the FCN panorama

Besides the concepts behind each implementation being the same, each implementation comes with its own advantages and challenges to overcome. In all the implementations, the power consumption was reduced thanks to the principle of no-current flowing to propagate information, which is the red thread of the paradigm.

First of all, the importance of the clock has already emerged as a character helping information switching and propagation along the desired paths. The general concept of the clock is to vary the potential barrier of unit elements, such as molecules or magnets, to erase the information, then by releasing the clock, depending on the surrounding fields, the unit element switches to a specific logic state.

The switching dynamic also depends on the working temperature since the presence of thermal noise can superpose to signal fields, possibly inducing an error. The metal QCA is restricted to work at cryogenic temperatures to prevent information loss [25]. Those temperatures reduce the impact of thermal noise, but the power consumption to keep the system at such low temperatures increases. In SiDB, circuits present errors at a few tens of kelvin, depending on the circuit [74]. Magnetic coupling in iNML is strong enough to work at room temperature [37, 40]. However, the presence of high thermal noise forces designers to shrink the maximum length of clock regions to a few magnets, implying a fine control of the clock network [75]. Even in molFCN thermal noise induces oscillation in the charge occupation of the

molecule dots [55, 56]. At room temperature, oscillations of the aggregated charge stay around the equilibrium point, and their amplitude is not sufficient to induce an encoding error [56].

The minimum dimension for information encoding is small in all the implementations. Indeed molFCN and SiDB unit elements extend below 1 nm [49, 30], whereas iNML and metal QCA need few tens of nanometer for the fundamental cell [40, 18].

Reduced minimum dimensions combined with the need for a few unit elements to build logic gates result in highly dense circuits. However, the small dimensions and the need for punctual control of the clock system make the fabrication of prototypes challenging. Except for molFCN, the other implementations have been experimentally validated. The fabrication of metal QCA exploits a top-down approach and requires a few process steps, using standard techniques such as Electron Beam Lithography (EBL), double layer resist development, and metal evaporation and oxidation [18]. Magnets in iNML can be defined through Focused Ion-Beam (FIB) lithography, a technique that can also be used for local manipulation of dot magnetic properties [38]. Currently, SiDB are created on top of a hydrogen-passivated silicon surface, using an artifact-free tip of a Scanning Tunnelling Microscope (STM) in a Ultra-High Vacuum (UHV) chamber at cryogenic temperature [30].

Writing and reading the information are typically two distinct problems in all the implementations. Writing data means applying a controlled field locally in a specific region of a circuit, not very different from the application of the clock field. The write-in system in iNML can be a magnetic Tunnel Junction (TJ) or a spin valve [38]. Metal QCA are metal TJs that can be driven by an external voltage applied through a coupling capacitance [18]. Molecules are sensitive to electric fields that can be applied by nanoelectrodes, as the ends of a nanogap with the molecule inside the gap itself [67, 76], or be light-induced [77]. A CMOS driver/inverter configuration can be used to drive the signal for inputs, as well as for cells with fixed polarization [78]. For SiDB, up to now, information is written using directly the STM tip, by creating single DBs that hold an electron, generating the electric field used as input [30]. Using a fixed DB enables the laboratory-level prototype validation, yet investigations are missing on a controllable runtime solution without the need for a STM tip.

Reading the information in iNML for its integration into transistor-based systems needs a magnetoelectric interface. The magnetic field encoding the information is used to induce a magnetization of a magnetic TJ, and then the magnetoresistance

phenomenon enables to read the information [38]. Detecting the position of the electron in metal QCA is a critical operation. The natural implementation proposed is the use of a Single-Electron tunneling Transistor (SET) placed next to the dot we are interested in sensing [79]. As for information writing in SiDB, even the output reading is performed through scanning microscopy [30]. This technique enables validation at the laboratory level but can't be used as a charge-sensing technique in the integration with transistor-based systems. Scanning microscopy techniques are also compatible with measuring information stored in molecules as a charge distribution [80, 81]. One particular challenge for the measurement is the ability to build a precise prototype.

FCN is an active research and development area, and each implementation still faces challenges. Among them, molFCN offers a unique combination of both room-temperature operation and miniaturization. The price to pay is the complexity of the fabrication that involves several actors: chemists for molecule syntheses and deposition, technology experts for the creation of nanopatterned hosting structures, and physicists for the measurement step. All these actors must cooperate in all the stages, and still the required techniques are very advanced in all those aspects.

Given the difficulty of conducting experiments, simulations become crucial to establish the behavior of systems, helping the prediction of their functioning, and driving focused experiments. Knowing in advance what to expect from a molecule or how a molecular system interacts with the surroundings becomes crucial to understanding how such systems could be realized technologically. Simulations are used to justify potential investments at the experimental level, demonstrating specific advantages over one implementation or another. The present work aims to provide considerations and design rules for molFCN systems. By starting from the investigation and the analysis of single molecules, the work focuses on the electrostatic interaction at the device level, especially describing how this can be used as a design parameter. Then, the work defines device modeling rules that can be used by CAD and Electronic Design Automation (EDA) tools to simulate large systems. The ability to design large circuits, proving them advantageous to other technologies, motivates the interest in the experimental validation of the technology. Finally, this work proposes a couple of alternatives for information reading again intending to facilitate the experimental proof.

Chapter 2

Molecules design principles

The role of molecules in the context of molecular Field-Coupled Nanocomputing (molFCN) is multifaceted; their composition dictates the electronic structure and quantum characteristics of the molecular cell, thereby shaping the behavior and functionality of the molFCN devices.

The selection of appropriate molecules becomes a critical aspect of molFCN design, as it directly influences the electrostatic effects that underlie the computational processes. The tailored arrangement of molecules for controlled manipulation of charge states offers a means to control and process information at the nanoscale. Furthermore, the engineering of molecule properties introduces the potential for tunability and customization, allowing researchers to fine-tune the performance characteristics of molFCN devices to meet specific computational requirements. This nuanced understanding of the molecular feature not only contributes to the optimization of molFCN technology but also opens avenues for exploring novel applications in quantum information processing and beyond, such as neural computation.

The next sections discuss molecules for molFCN, starting from the theoretical methods that can be used to study them and some core features that are of interest for molFCN applications. Then, a list of structural characteristics a molecule must have to be suitable for information propagation, before briefly describing the most studied molecules from the literature. Finally, the chapter analyzes in detail a case of study. We start with a molecule, and then various parts are modified to validate their effect on the valuable figures of merit for the molFCN.

2.1 Methods for studying molecules

Ab initio calculations and Density Functional Theory (DFT) are both approaches used in quantum mechanics to model the electronic structure and properties of molecules [82]. However, they differ in their underlying principles and computational methods.

Ab initio simulations represent a powerful computational approach for materials science and quantum chemistry, providing an accurate and detailed understanding of the properties and behavior of matter from first principles. Unlike empirical methods that rely on experimental data, *ab initio* simulations directly use quantum mechanical principles to predict the electronic structure and properties of molecules, starting with fundamental physical equations, one for all the Schrödinger equation

$$H\Psi = E\Psi \quad (2.1)$$

and solving them numerically to predict material properties.

Since this equation has an exact solution only for the hydrogen atom, some mathematical approximations are needed for any other molecules or systems: the nature of the necessary approximations determines the level of the calculation. *Ab initio* computations are based mainly on the laws of quantum mechanics and a set of physical constants (the speed of light, masses and charges of electrons and nuclei, Planck's constant).

One of the key strengths of *ab initio* simulations is their versatility, allowing researchers to investigate a variety of materials, from simple elements to complex compounds. These simulations play a crucial role in designing and analyzing novel materials or molecules with tailored properties, being general and theoretically reliable. It's important to note that *ab initio* simulations can be computationally demanding, requiring substantial computational resources. Nevertheless, advancements in computer power have significantly mitigated these drawbacks. The increase in computational speed over the years has transformed *ab initio* calculations from being considered impractical for molecules with more than 20 electrons to now routinely handling molecules with 100 electrons, demonstrating the remarkable progress in computational capabilities.

Unlike traditional *ab initio* methods, DFT doesn't attempt to directly solve the complex many-body Schrödinger equation. Instead, it focuses on the electron density, a key factor in determining the behavior of a quantum system. It aims to find the electronic density that minimizes the total energy of the system.

The core foundation of DFT lies in the Hohenberg-Kohn theorems [83], which assert that the ground-state electron density uniquely determines the potential energy of a system and, consequently, all other observables [82]. This theoretical principle enables a shift from dealing with many interacting electrons to a problem involving a single electron moving in an effective potential. The Kohn-Sham equations, a set of auxiliary non-interacting electron equations, are introduced to emulate the behavior of the real interacting system [83]. The challenge then becomes finding the exchange-correlation functional, a term accounting for the elusive electron-electron interactions.

It's important to note that the accuracy of DFT predictions is influenced by the choice of functionals and basis sets since both play a central role in approximating and describing the electronic structure of molecules and materials.

A functional in the context of quantum chemistry refers to a mathematical expression that associates a corresponding total energy with each possible electron density [82]. The exchange-correlation functional encapsulates the intricate electron-electron interactions that are challenging to account for explicitly. Various forms of exchange-correlation functionals exist, each designed to strike a balance between computational efficiency and accuracy in predicting molecular properties. Researchers continually refine and develop new functionals to improve the overall performance of DFT across a diverse range of systems and interactions.

Basis sets are sets of mathematical functions used to represent the spatial distribution of electrons within a molecule [82]. They serve as a foundation for expressing molecular orbitals, which are solutions to the Schrödinger equation. The choice of basis set significantly impacts the accuracy of quantum chemical calculations. Generally, larger basis sets with more functions provide a more accurate representation of electronic structure, but they come at the cost of increased computational complexity. Basis sets can be categorized into several types, such as minimal, double-zeta, and triple-zeta, each offering a trade-off between accuracy and computational efficiency.

DFT provides an excellent framework for accurate partial charge calculations as it accounts for the electronic correlation effects that influence the charge distribution

within a molecule. The choice of the exchange-correlation functional is crucial in ensuring the reliability of the results. Hybrid functionals, which combine Hartree-Fock exchange with DFT exchange-correlation functionals, are often employed for improved accuracy in describing charge distribution.

2.1.1 Partial charges calculation

As mentioned in subsection 1.2.2, DFT is used to obtain the data set for figures of merit extraction. The main interesting property is the charge distribution inside a molecule in response to external perturbations provided by electrodes or other molecules. The way to obtain such information is the use of partial charges.

DFT provides an excellent framework for accurate partial charge calculations as it accounts for the electronic correlation effects that influence the charge distribution within a molecule. The choice of the exchange-correlation functional is crucial in ensuring the reliability of the results. Hybrid functionals, which combine Hartree-Fock exchange with DFT exchange-correlation functionals, are often employed for improved accuracy in describing charge distribution.

Partial charges are a concept in chemistry that arise when atoms within a molecule share electrons unevenly [84]. In a covalent bond, where two atoms share electrons, the electronegativity of each atom determines how strongly it attracts the shared electrons. Electronegativity is a measure of an atom's ability to attract and hold onto electrons.

When two atoms with different electronegativities form a covalent bond, the electrons are not shared equally. The more electronegative atom attracts the electrons more strongly, leading to an uneven distribution of electron density. As a result, the atom with higher electronegativity acquires a partial negative charge (δ^-) because it has a greater share of the electron density, while the atom with lower electronegativity acquires a partial positive charge (δ^+).

Partial charge calculation is a critical aspect of understanding the electronic distribution within molecules, and DFT serves as a powerful tool for carrying out such calculations. In the context of DFT, partial charges are used to quantify the electron density distribution across different atoms within a molecular system. In DFT, the electron density is represented as a continuous distribution rather than discrete electron positions. Different methods have been proposed to calculate partial

charges or net atomic charges. Mulliken, Löwdin, and ElectroStatic Potential (ESP) derived charges are different methods used to calculate partial charges on atoms within a molecule. These methods provide different approaches to estimating how electrons are distributed among atoms in a molecular system.

The concept of Mulliken charges is rooted in the idea that the electron density around an atom is related to its electronegativity and, therefore, influences the charge distribution within a molecule [85, 86]. The calculation of Mulliken charges involves determining the average electron density of an atom within a molecule and then comparing it to the electron density of the isolated atom in its ground state. The difference between these values provides an estimate of the partial charge on the atom.

Mathematically, Mulliken charges (Q) are often expressed as follows:

$$Q_i = Z_i - \frac{1}{2} \sum_j P_{ij} S_{ij} \quad (2.2)$$

where Q_i is the Mulliken charge on atom i , P_{ij} is the off-diagonal element of the density matrix, and S_{ij} is the overlap matrix element. The summation is over all atoms j in the molecule.

While Mulliken charges are relatively straightforward to compute, they have some limitations. They can be sensitive to the choice of the basis set used in quantum mechanical calculations, and they may not always provide physically meaningful values, especially in cases where electron delocalization and resonance effects play a significant role [87].

Löwdin charges represent another approach to estimating partial charges on atoms within a molecule. Löwdin charges are derived from the density matrix and aim to provide a more physically meaningful description of the electron distribution than Mulliken charges [88]. Calculating Löwdin charges involves diagonalizing the overlap matrix (S) between atomic orbitals to obtain a set of orthogonalized orbitals. The density matrix elements (P) are then transformed to this new basis, and the trace of the resulting matrix provides the Löwdin charges for each atom [89].

Löwdin charges are considered to be more physically meaningful than Mulliken charge schemes. Nevertheless, even Löwdin charges can eventually become unstable with large basis sets [87].

ESP derived charges consider the electrostatic potential surrounding a molecule [90]. Among ESP-derived charge methods, the CHarges from ELelectrostatic Potentials using a Grid (CHELPG) scheme is noteworthy for its effectiveness. The CHELPG method employs quantum chemical calculations to generate the molecular ESP. This potential is determined by solving the Poisson equation, considering the charge distribution of the atoms within the molecular system. The molecular ESP is then used to derive partial charges on individual atoms.

The fundamental idea behind CHELPG is to fit point charges onto the molecular surface in a way that reproduces the calculated electrostatic potential as closely as possible. The point charges are adjusted iteratively until the molecular ESP is well-matched. The resulting set of point charges represents the partial charges on each atom in the molecule.

The CHELPG method is widely used in quantum chemistry and molecular modeling because it provides accurate and physically meaningful partial charges. It accounts for the effects of nearby atoms and the overall molecular environment, making it especially valuable for studying molecular interactions and other properties that involve charge distribution. ESP-derived charges, particularly those obtained using the CHELPG method, are significantly more systematic and predictable behavior than the methods based on an analysis of the wave function or the electron density topology.

In summary, while Mulliken and Löwdin charges are older and more straightforward methods, ESP charges are generally more accurate and robust for calculating partial charges in molecular systems [91]. Researchers often choose the method based on the specific requirements of their studies and the level of accuracy needed for their calculations.

2.1.2 Molecular polarizability

Molecular polarizability is a fundamental concept in chemistry and physics that describes the ability of a molecule to respond to an external electric field by inducing a temporary dipole moment [92, 93]. This phenomenon arises from the redistribution of electrons within the molecule in response to the applied electric field.

Molecular polarizability is typically represented by the symbol α and quantifies how easily the electron cloud of a molecule can be distorted. Higher polarizability

values indicate that the molecule's electron cloud is more easily deformed, suggesting greater flexibility in its electronic structure. The molecular polarizability (α) is mathematically described by the equation:

$$\alpha = \frac{\mu}{E} \quad (2.3)$$

where μ is the induced dipole moment and E is the magnitude of the applied electric field.

The magnitude of the molecular polarizability depends on several factors, including the size and shape of the molecule, the distribution of electron density within the molecule, and the nature of the chemical bonds present. Generally, larger and more elongated molecules tend to have higher polarizability values due to the increased flexibility of their electron clouds.

The ORCA package offers three methods to calculate polarizability: analytic calculation, numeric differentiation of dipole moments, and fully numeric second derivatives.

In analytic calculation, the polarizability tensor is derived analytically from the electronic structure of the molecule using quantum mechanical methods such as Hartree-Fock (HF) theory, DFT, or Many-Body Perturbation Theory (MBPT). This method involves solving the electronic Schrödinger equation to obtain wavefunctions and energies, from which analytical expressions for the polarizability tensor components are derived.

With the numeric differentiation of dipole moments approach, the polarizability tensor is calculated numerically by perturbing the molecule's electronic structure with an external electric field and computing the resulting change in the dipole moment. The polarizability tensor components are obtained by numerically differentiating the dipole moment with respect to the electric field strength along different directions.

The fully numeric second derivatives method involves numerically computing the second derivatives of the molecular electronic energy with respect to atomic displacements induced by an external electric field. The polarizability tensor components are obtained from these second derivatives using finite difference or numerical differentiation methods.

Analytic calculation offers accuracy and efficiency for small to medium-sized molecules, while numeric methods provide flexibility and applicability to larger or

more complex systems at the cost of increased computational resources and potential numerical errors. In the following sections, the polarizability is calculated using analytic calculations.

2.2 Molecule requirements from a structural standpoint

Molecules for the molFCN technology need to have some characteristics to ensure the correct information treatment and the feasibility of a prototype. The requirements are strictly related to both the geometry of the system and the electrostatic behavior of the molecule. This section focuses more on aspects related to the space occupation, whereas chapter 3 deals in detail with the electrostatics of the molecules.

In the thesis, the molecules are treated as static entities to simplify the computational models and focus on fundamental interactions and behaviors. However, it is crucial to acknowledge that molecules are inherently dynamic, undergoing various vibrational modes and conformational changes. These dynamics can significantly influence the electrostatic interactions that are pivotal for molFCN. Recent studies have preliminarily investigated the impact of vibrations due to different working temperatures using Ab Initio Molecular Dynamics (AIMD) [56]. A recent work has also started investigating the dynamic behavior of molecules for molFCN using Real-Time Time-Dependent Density Functional Theory (RT-TDDFT) [94]. These types of simulations provide information to eventually improve the model of single molecules and are also compatible with the modular Molecular Simulator Quantum-dot cellular automata Torino (MoSQuiTo) methodology.

2.2.1 The bistability specifications

A vital aspect of molFCN is the requirement for bistable propagation [95, 66, 96]. Ideally, we have bistable propagation if adjacent molecules have the charge localized in the opposite part of the molecule and the amount of localized charge does not tend to zero going onward the wire/device.

Fig. 2.1(a) shows the charge along a wire and the associated dipole moment on a perfect bistable propagation: the charge and the dipole moment oscillate between

two constant values that do not vary along the wire. If the propagation is not bistable, information vanishes and gets lost after a few molecules. Ultimately, bistability depends on the molecule characteristics, and a parameter called Bistable Factor (BF) has been proposed to relate the possibility of having bistable propagation to the molecule features [96]. Considering the system shown in Fig. 2.1(b), composed of the driver molecule (DR) and the Molecule Under Test (MUT), the BF is defined as

$$BF_0(d, w, \alpha) = \frac{\alpha}{2\pi\epsilon_0 w^2} \left(\frac{1}{d} - \frac{1}{\sqrt{d^2 + w^2}} \right) \quad (2.4)$$

where d is the distance between the driver and the MUT (i.e., the intermolecular distance), w is the distance between the center of the groups involved in the logic encoding in the MUT (i.e., the molecular width), and α is the molecule polarizability.

In a system composed of N molecules, the bistable factor is computed by the following equation

$$BF_N = \sum_{i=1}^N (-1)^{\binom{d_i}{d} + 1} BF_0(d_i, w, \alpha) \quad (2.5)$$

Bistable propagation occurs when $BF \geq 1$. However, if the BF surpasses a certain threshold, molecules polarize more rapidly, increasing the likelihood of errors during propagation. For instance, in the presence of thermal noise while working at room temperature, a small surrounding disturbance would be enough to polarize the molecule with a random value. This leads to the need for controlling the clock regions' dimensions: the larger the clock zones, the later the domino effect of the driving signal would cross the region, as already demonstrated for In-plane Nano Magnetic Logic (iNML) [75].

Here, we analyze the BF while varying the molecule width w and the polarizability α to determine the molecular safe-operating area. In the images displayed in the right column of Fig. 2.1c, the colored regions represent the molecular safe-operating area, with the gradient indicating the degree of BF satisfaction. Conversely, by adjusting the intermolecular distance d , a technology parameter, we delineate the technological safe-operating area, as illustrated in the images in the left column of Fig. 2.1c.

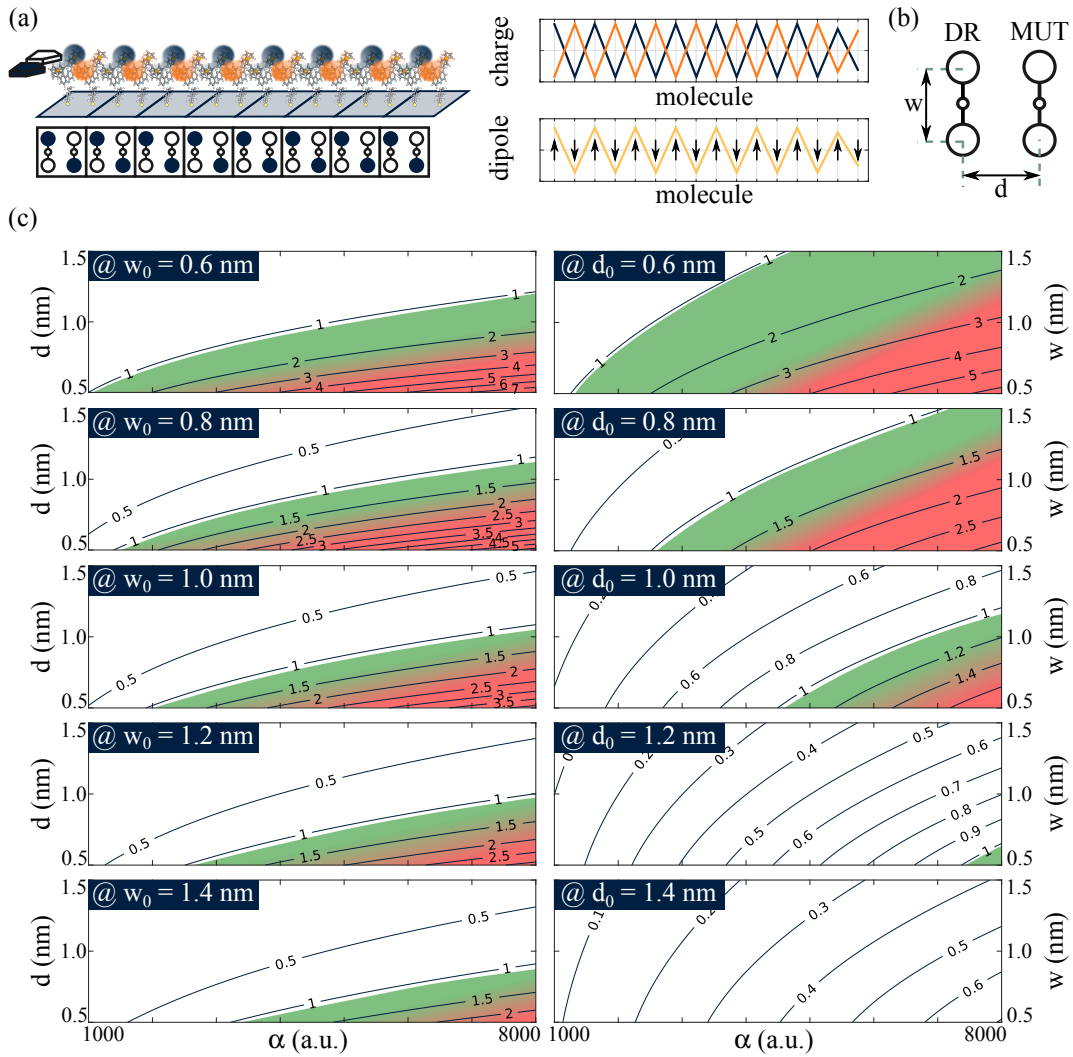


Fig. 2.1 Bistable propagation. (a) Charge and dipole along a wire in the case of bistable propagation. (b) System composed by the driver molecule and the molecule under test for calculating BF_0 . (c) Bistable factor varying the polarizability α , the molecular width w , and the intermolecular distance d .

2.2.2 Molecules in space

Using the bisferrocene molecule, Fig. 2.2a defines the geometrical reference used throughout this document. Specifically, information propagates along the z-axis, the molecule stands along the x-axis, and the electric fields associated with the logic information are along the y-axis.

The geometrical structure of molecules affects the global electrostatic behavior and the physical occupation. Both aspects are crucial since the technology is based

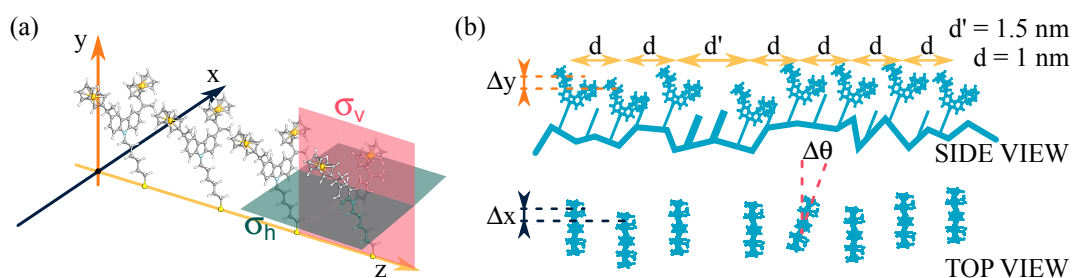


Fig. 2.2 Space organization of molecules. (a) Geometrical reference system and orientation of the molecule symmetry planes in space. (b) Possible process defects in a molFCN wire.

on electrostatic field coupling, and the logic devices are obtained through particular dispositions of the molecules.

Generally, along the x-axis, the molecule should be small enough to guarantee bistability, and the molecule should have a null net dipole so as not to unbalance the dipole moment linked to the information. Ideally, the molecule geometry should be Y-shaped or V-shaped, meaning that the molecule has a vertical symmetry σ_v and no horizontal symmetry σ_h . Fig. 2.2a shows the symmetry planes on the bisferrocene in the reference system. It should be imagined that the molecule aggregates its charge distribution on the three points of the 'Y' shape. In this way, the clock field can affect the charge distribution and provide the information reset ('NULL' state). Finally, the molecule should be synthesizable to implement an actual circuit.

Since molFCN is a technology based on the relative position of molecules and the electrostatics is crucial for the correct functioning, being able to build a device precisely helps the validation of a prototype. Molecules in a device must follow a specific pattern and orientation, and Fig. 2.2b shows that in an actual structure, we must consider process defects [97, 61]. Also, the molecule should not lose the electrostatic properties when deposited on the substrate. Besides, the substrate itself may be exploited to satisfy the requested properties, for example, obtaining charge localization using localized point defects [98].

Aiming to lighten the constraints on the nanometric layouts, which pose a significant challenge in molFCN technology, one solution is to workaround the necessity for nanometric patterning of molecular devices by exploiting molecules that organize in uniform Self-Assembled Monolayers (SAM) structures. Specifically, SAM can be considered a matrix of uniform blocks composed of several molecules capable of independently storing information and being activated [99]. Activating these

blocks in a controlled sequence enables the configuration of the SAM for mapping fundamental logic gates (such as majority voters and inverters) and routing elements.

2.3 Molecules from the literature

Researchers explored various organic and inorganic compounds to identify promising candidates for molFCN implementation. The choice of molecules is intricately linked to the desired functionalities of the device, such as logic operations and information storage. The goal is to harness the principles of quantum mechanics to create robust and scalable molecular systems. Several candidates have been proposed for both cases over the past 20 years. The geometry of all the following atoms is reported in Appendix A.

Among the proposed molecules to implement half cell, the $(\text{NH}_3)_5\text{Ru}(\text{pyrazine})\text{Ru}(\text{NH}_3)_5$ has been deeply studied and is typically called Creutz-Taube ion [100]. Fig. 2.3a shows the geometry of the Creutz-Taube ion, studied in several works by Lieberman and Bernstein between 2000 and 2010, with an experimental goal [101–103] for both deposition and nanopatterning in molFCN. The Creutz-Taube ion has been used as a starting point to obtain mixed-valence complexes with different properties by changing the bridge between the two $(\text{NH}_3)_5\text{Ru}$ groups [104, 95].

Another interesting idea regards the possibility of exploiting the largely-studied fullerenes [105]. Fig. 2.3b shows a double-cage fluorinated fullerenes that form ordered arrays on Ag(100) surfaces [106, 107]. Even single fullerenes encapsulating single atoms or diatomic molecules show a sensitivity to an externally applied field, simplifying the global structure of the system [108].

These molecules model only two dots of the cell, and the effect of the clock field is not considered. So more recent studies investigate molecules that present three functional groups to encode also the ‘null’ state. One example is the $7\text{-Fc}^+ - 8\text{-Fc} - 7,8\text{-nido-[C}_2\text{B}_9\text{H}_{10}]$, called diferrocenyl carborane, and shown in Fig. 2.3c [65]. The molecule has been synthesized to incorporate several features useful for a switching molecule for molFCN, in particular, the presence of the nido-carborane employed as an internal counterion. Authors in [109] investigate encoding and transmission using the diferrocenyl carborane embedded within the channels of

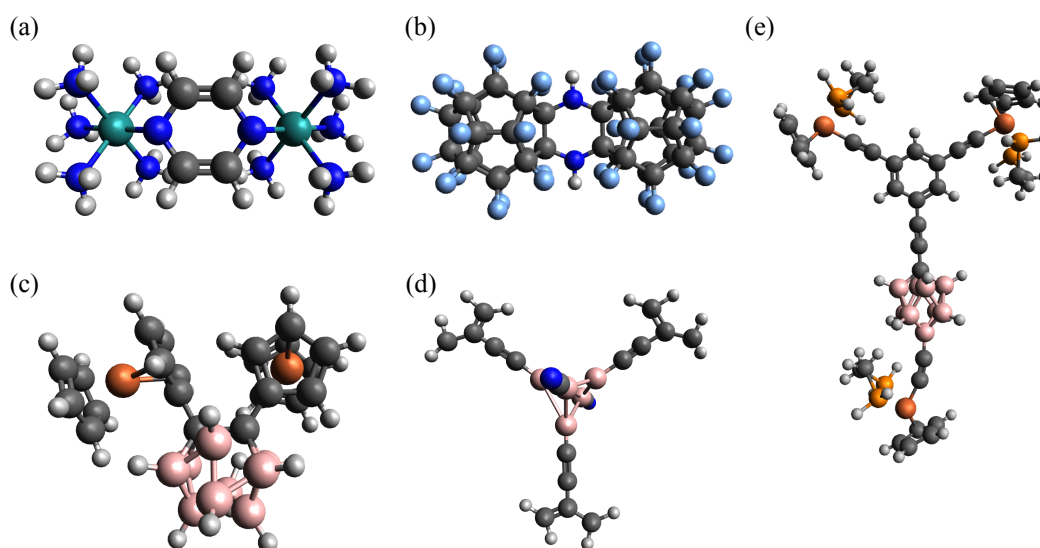


Fig. 2.3 Geometry of some molecules proposed in the literature for molFCN shown in a ball-and-stick diagram. (a) Creutz-Taube ion. (b) Double-cage fluorinated fullerenes. (c) Diferrocenyl carborane. (d) Zwitterionic three-dot molecule with di-anionic pentaborate counterion. (e) Three-dot 'Y'-shaped functionalized zwitterionic neutral closo-carborane. Atoms in figure are colored as follows: H = white, C = charcoal, N = blue, B = pink, Fe = orange, Ru = teal, F = sky blue, P = tangerine.

a microporous matrix. This approach allows molecules to self-assemble into a row inside the material's pores, forming an encased wire that benefits from the stability and protection provided by the crystalline environment.

Other works proposed molecules with boron cages as internal counterions. The three-dot 'Y'-shaped functionalized zwitterionic neutral closo-carborane model shown in Fig.2.3e uses a carborane core attached to three iron centers hosting the positive charge in the logic states [66]. Similarly, Fig. 2.3d shows the zwitterionic three-dot molecule with di-anionic pentaborate counterion where the functionalized cage is connected to simpler allyl groups for the three logic states [110].

Molecules presented so far have 2 or 3 dots, meaning they implement only half a cell. Researchers also proposed molecules implementing a whole molFCN cell with the four logic dots. Many of them use iron and cobalt atoms as the principal actors of the functional groups, inside metallocene structures [111–113] or not [114–116].

Besides the several proposals, there is still a lack of homogeneity among the studies, and only a few of them calculate the charge distribution using the ESP-derived partial charges more suitable for the molFCN application.

2.4 A case of study: Bis(trimethylsilyl) sulfide

From an application standpoint, the ultimate goal is to obtain a set of possible molecules with different properties that can be used in different contexts depending on the need. This section addresses this huge goal by analyzing a case-of-study of a molecule, where different parts of the molecule are substituted step-by-step to understand better how each functional group acts on the molecule transcharacteristic. All the following calculations are performed using the ORCA package [117, 118]. The coordinates of each molecule are reported in Appendix A.

The starting molecule is the bis(trimethylsilyl) sulfide, consisting of a central sulfur atom bonded to two silicon atoms. Each silicon atom is attached to three methyl groups CH_3 , giving the molecule the chemical formula $((\text{CH}_3)_3\text{Si})_2\text{S}$. Fig. 2.4a displays the geometry of the bis(trimethylsilyl) sulfide in a ball-and-stick diagram, obtained from the geometry optimization of the isolated molecule with the Unrestricted Kohn-Shann (UKS) method using CAM-B3LYP functional with def2-TZVP basis set [119]. Grimme D3 correction is applied [120, 121]. The charge distribution is analyzed by exploiting the ESP-derived CHELPG paradigm. The ORCA tool computes the electrostatic potential produced by the molecule and is shown in Fig. 2.4b, yielding a series of atom-centered charges through fitting [90]. Fig. 2.4c shows the ESP charges on each atom, presenting maximum values on the Si atoms and a minimum value on the S atom.

From the DFT simulation, we also extracted the polarizability tensor reported below in equation (2.6) in atomic units. The atomic unit for polarizability is expressed in terms of the atomic unit of length cubed per atomic unit of electric field squared, denoted as a_0^3/E_h , where a_0 is the Bohr radius ($\approx 0.5291 \times 10^{-10}$ m) and E_h is the Hartree energy ($\approx 4.3591 \times 10^{-18}$ J).

$$\alpha = \begin{pmatrix} 167.20344 & -0.32374 & 3.95419 \\ -0.32374 & 132.07003 & 0.66045 \\ 3.95419 & 0.66045 & 131.31831 \end{pmatrix} \quad (2.6)$$

The diagonalized tensor is expressed in equation (2.7), whereas the isotropic value is $\alpha_{iso} = 143.53059$ a.u..

$$\alpha_{diag} = \begin{bmatrix} 130.56878 & 132.38724 & 167.63575 \end{bmatrix} \quad (2.7)$$

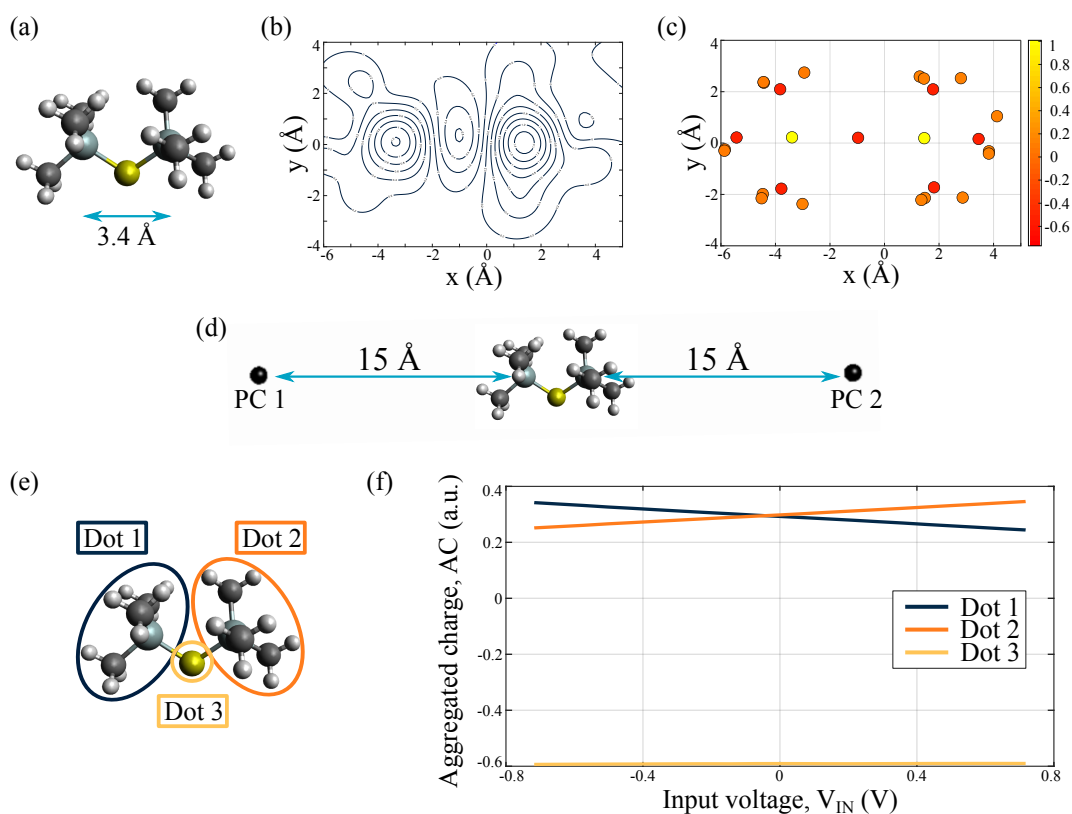


Fig. 2.4 Results of the DFT analysis on the bis(trimethylsilyl) sulfide molecule. (a) 3D geometry shown in a ball-and-stick diagram. (b) Molecular electrostatic potential evaluated with DFT calculation, the solid blue lines are isopotential curves. (c) ESP charges on each atom of the molecule. (d) 3D geometry of the molecule shown in a ball-and-stick diagram with the point charges used to generate the electric field represented as black dots. (e) Atom groups for the calculation of the aggregated charges. (f) V_{in} -AC transcharacteristic (VACT) of the bis(trimethylsilyl) sulfide. Atoms in figure are colored as follows: H = white, C = charcoal, S = yellow, Si = gray.

Once the relaxed structure is obtained, the goal is to study it in the presence of an external field to obtain the charge distribution so that we have the electrostatic model of the molecule when surrounded by other elements generating an electric field. The molecule is simulated in the presence of constant electric fields generated by positioning point charges in the space. Fig. 2.4d shows a 3D representation of the system composed of the bis(trimethylsilyl) sulfide and the position of the point charges used to start the simulations.

The generated electric field is calculated on the geometrical center ($x = -1.0192 \text{ \AA}$, $y = 0.1898 \text{ \AA}$, $z = 2 \times 10^{-7} \text{ \AA}$) of the molecule as the sum of the contri-

butions of all the point charges as expressed by the following equation:

$$E(x, y, z) = \sum_{i=1}^{N_{pc}} \frac{1}{4\pi\epsilon_0} \frac{q \cdot pc_i}{r_i^2} \quad (2.8)$$

where N_{pc} is the total number of point charges, ϵ_0 is the vacuum permittivity, q is the electron charge, pc_i is the value of the i^{th} point charge, and r_i is the distance between the i^{th} point charge and the molecule geometrical center. Table 2.1 lists the electric field component along each axis for all the tested cases. The dominant values are along the x-axis, which is the direction of the switching field in a molFCN system.

Table 2.1 Values of the point charges and the relative electric field components for the simulation of the bis(trimethylsilyl) sulfide.

PC1 (a.u.)	PC2 (a.u.)	E_x (V nm ⁻¹)	E_y (V nm ⁻¹)	E_z (V nm ⁻¹)
2.0	-2.0	2.0526	-0.0393	0.0572
1.8	-1.8	1.8473	-0.0354	0.0515
1.6	-1.6	1.6421	-0.0314	0.0458
1.4	-1.4	1.4368	-0.0275	0.0401
1.2	-1.2	1.2316	-0.0236	0.0343
1.0	-1.0	1.0263	-0.0197	0.0286
0.8	-0.8	0.8210	-0.0157	0.0229
0.6	-0.6	0.6158	-0.0118	0.0172
0.4	-0.4	0.4105	-0.0079	0.0114
0.2	-0.2	0.2053	-0.0039	0.0057
0	0	0	0	0
-0.2	0.2	-0.2053	0.0039	-0.0057
-0.4	0.4	-0.4105	0.0079	-0.0114
-0.6	0.6	-0.6158	0.0118	-0.0172
-0.8	0.8	-0.8210	0.0157	-0.0229
-1.0	1.0	-1.0263	0.0197	-0.0286
-1.2	1.2	-1.2316	0.0236	-0.0343
-1.4	1.4	-1.4368	0.0275	-0.0401
-1.6	1.6	-1.6421	0.0314	-0.0458
-1.8	1.8	-1.8473	0.0354	-0.0515
-2.0	2.0	-2.0526	0.0393	-0.0572

For each calculation, we extract the atomic charge by fitting the electrostatic potential. Subsequently, these atomic charges are aggregated into the groups shown in Fig. 2.4e to derive the Aggregated Charge (AC) distribution. From the electric field it is possible to obtain also the input voltage for each simulation, so Fig. 2.4f shows the VACT of the bis(trimethylsilyl) sulfide. The VACT points out a big charge separation between the negative sulfur and the positive trimethylsilane groups. Moreover, the application of a constant electric field does not impact the charge of the sulfur, since it only moves the charge from on Si to the other. The moving charge is small but still coherent with the molFCN information encoding.

We then study the oxidized molecule, aiming to obtain a bigger charge separation between logic dots. When a molecule undergoes oxidation, it loses electrons and becomes positively charged (a cation). This process can change the molecular structure and electronic properties, potentially leading to increased charge mobility and delocalization. We repeat the same analyses performed on the neutral molecule and Fig. 2.5a shows the ESP charges on each atom. Again, Si atoms show the maximum positive charge aggregation, but the charge on the sulfur atom is less negative in this case. The variation in the charge of the S atom is also reflected in the VACT shown in Fig. 2.5b. The aggregated charge on *Dot 3* (i.e., the sulfur atom) moved to -0.2 a.u.. The relative behavior between *Dot 1* and *Dot 2* did not change, except for the maximum charge separation, which is higher in this case.

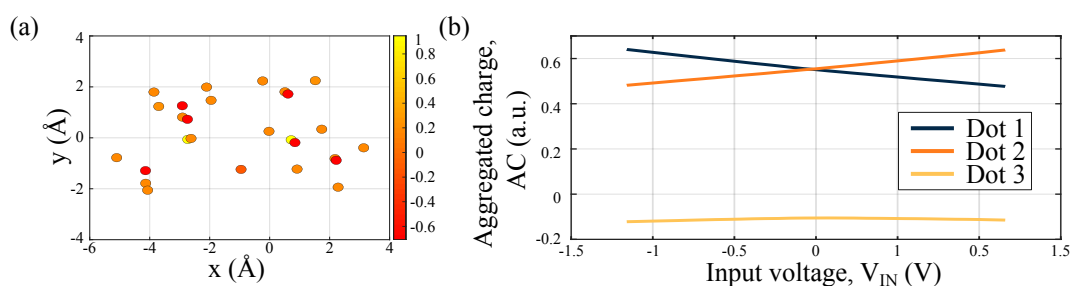


Fig. 2.5 Results of the DFT analysis on the oxidized bis(trimethylsilyl) sulfide molecule. (a) ESP charges on each atom of the molecule. (b) VACT of the bis(trimethylsilyl) sulfide.

The slightly increased charge separation is also reflected in the polarizability of the oxidized molecule, resulting in the following:

$$\alpha = \begin{pmatrix} 177.97479 & -2.99848 & 11.35569 \\ -2.99848 & 129.98007 & 0.44375 \\ 11.35569 & 0.44375 & 130.46194 \end{pmatrix} \quad (2.9)$$

The diagonalized tensor is expressed in equation (2.10), whereas the isotropic value is $\alpha_{iso} = 146.13893$ a.u..

$$\alpha_{diag} = \begin{bmatrix} 127.39440 & 130.31555 & 180.70685 \end{bmatrix} \quad (2.10)$$

2.4.1 Hexanedithiolbis(trimethylsilane)

The first modification increases the distance between the two trimethylsilane groups. By substituting the sulfur atom with the hexanedithiol, we obtain the molecule shown in Fig. 2.6a in a ball-and-stick diagram. The coordinates have been obtained from the geometry optimization of the isolated molecule with the UKS method using CAM-B3LYP functional with def2-TZVP basis set [119]. Grimme D3 correction is applied [120, 121]. The charge distribution is analyzed by exploiting the ESP-derived CHELPG paradigm. Following the same procedure as before, we use the ORCA tool to compute the electrostatic potential produced by the molecule, yielding a series of atom-centered charges through fitting [90]. Fig. 2.6b shows the ESP charges on each atom, again presenting maximum values on the Si atoms. This time, the connecting hexane does not hold any charge, whereas the S atoms hold a quite big negative charge. We then repeat the same analysis using the oxidized hexanedithiolbis(trimethylsilane). Oxidation removes an electron, making the molecule charge globally positive equal to 1 a.u.. Fig. 2.6(c) shows the ESP charges on each atom for the oxidized version of the molecule. Changes seem minimal, the biggest variation qualitatively observable is on the charge of both sulfur atoms becoming less negative.

The molecule polarizability is higher than the one obtained for the bis(trimethylsilyl) sulfide, resulting in the following:

$$\alpha = \begin{pmatrix} 307.57244 & 1.27015 & 3.76087 \\ 1.27015 & 206.68096 & 0.02147 \\ 3.76087 & 0.02147 & 195.49304 \end{pmatrix} \quad (2.11)$$

The diagonalized tensor is expressed in equation (2.12), whereas the isotropic value is $\alpha_{iso} = 236.58215$ a.u..

$$\alpha_{diag} = \begin{bmatrix} 195.36695 & 206.66503 & 307.71446 \end{bmatrix} \quad (2.12)$$

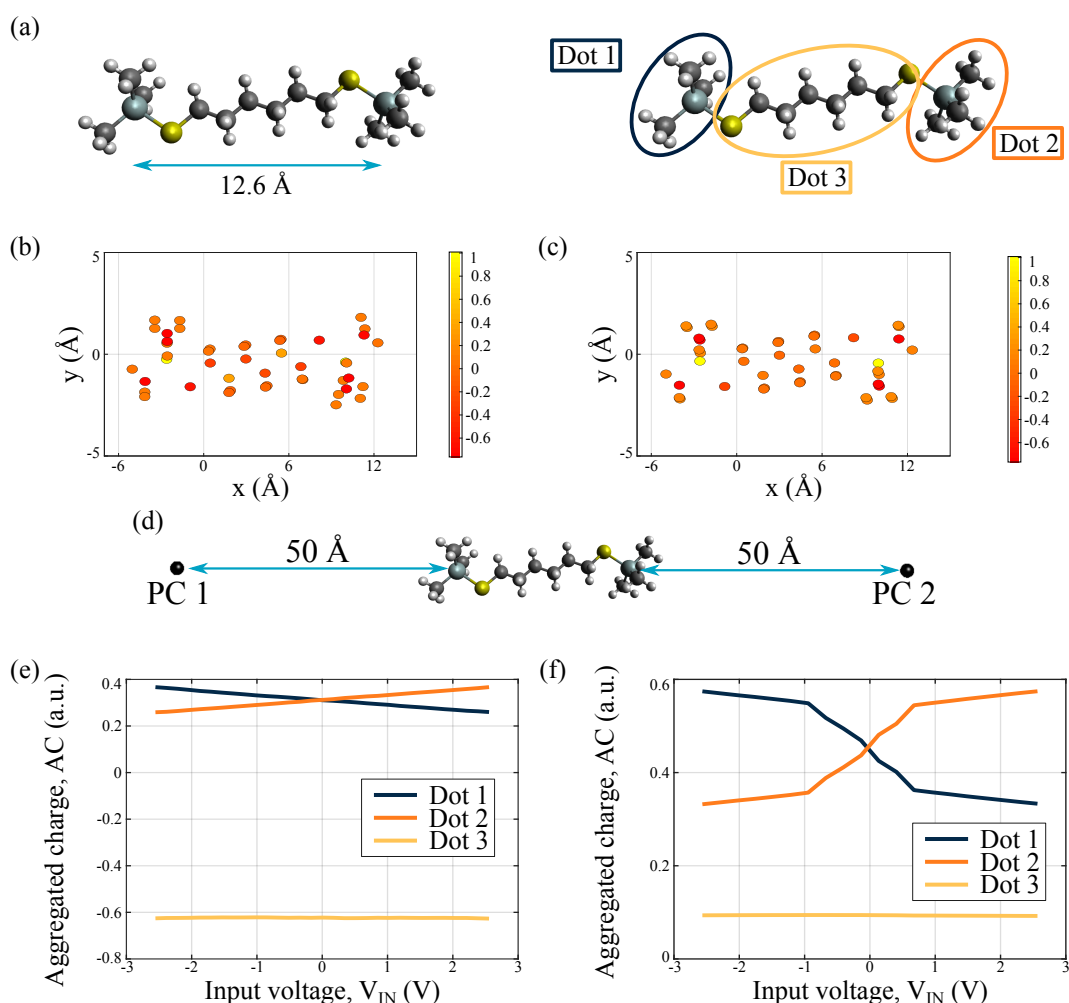


Fig. 2.6 Results of the DFT analysis on the hexanedithiolbis(trimethylsilane) molecule. (a) 3D geometry shown in a ball-and-stick diagram and the atom groups for the calculation of the aggregated charges. (b) ESP charges on each atom of the molecule. (c) ESP charges on each atom of the oxidized molecule. (d) 3D geometry of the molecule shown in a ball-and-stick diagram with the point charges used to generate the electric field represented as black dots. The scheme is not to scale. (e) VACT of the hexanedithiolbis(trimethylsilane). (f) VACT of the oxidized hexanedithiolbis(trimethylsilane). Atoms in figure are colored as follows: H = white, C = charcoal, S = yellow, Si = gray.

The molecule is then simulated in the presence of constant electric fields performed by positioning point charges in the space. Fig. 2.6d shows a 3D representation of the system composed of the hexanedithiolbis(trimethylsilane) and the position not to scale of the point charges used to start the simulations.

The generated electric field is calculated on the geometrical center ($x = 3.6846 \text{ \AA}$, $y = -0.3861 \text{ \AA}$, $z = -0.3126 \text{ \AA}$) of the molecule using equation (2.8). We extract

the atomic charge for each electric field and aggregate them into the groups shown in Fig. 2.6a to derive the AC distribution. Fig. 2.6e shows the VACT of the molecule. The VACT confirms that the presence of the hexane does not influence the charge distribution on the molecule.

Fig. 2.6f shows the VACT of the oxidized molecule, pointing out the different behavior. The charge on *Dot 3* is now positive and almost equal to 0.1 a.u.. The two trimethylsilane groups still exchange charge, but the charge separation is, in this case, higher. For voltages bigger than $V_{IN} = 1$ V in absolute value, the charge saturate to almost 0.3 a.u. and 0.6 a.u.. Since the charge separation is high, the information encoding for a molFCN system is stronger.

Oxidation, in this case profoundly, changed the characteristic of the molecule, and this is reflected in the polarizability tensor:

$$\alpha = \begin{pmatrix} 288.35743 & -756.33267 & -15.38033 \\ -756.33267 & 6892.61999 & 134.48621 \\ -15.38033 & 134.48621 & 192.29457 \end{pmatrix} \quad (2.13)$$

The biggest value is associated with the axis connecting the two silicon atoms. The diagonalized tensor is expressed in equation (2.14), whereas the isotropic value is $\alpha_{iso} = 2457.75733$ a.u..

$$\alpha_{diag} = \begin{bmatrix} 189.59316 & 202.85022 & 6980.82862 \end{bmatrix} \quad (2.14)$$

2.4.2 Dodecanedithiolbis(trimethylsilane)

Since the presence of the hexane as a linking bridge does not influence the charge distribution on the molecule, we substitute it with a dodecane chain, thus pulling the trimethylsilane further apart. Fig. 2.7a shows the resulting molecule in a ball-and-stick diagram, and the coordinates have been obtained from the geometry optimization of the isolated molecule with the UKS method using CAM-B3LYP functional with def2-TZVP basis set [119]. Grimme D3 correction is applied [120, 121]. The charge distribution is analyzed by exploiting the ESP-derived CHELPG paradigm, resulting in the ESP charges shown in Fig. 2.7b. Fig. 2.7c shows instead the ESP charges of the oxidized molecule dodecanedithiolbis(trimethylsilane). Considerations are very similar to the previous molecule. Hence, the silicon atoms present a

positive charge, whereas the sulfur and carbons in the methyl groups hold a negative charge. Again, the oxidation process impacts the most the charge on both S atoms and the dodecane chain.

Enlarging the distance between the two silicon atoms makes the polarizability grow, as expressed in the tensor:

$$\alpha = \begin{pmatrix} 408.80190 & 14.83344 & 10.31219 \\ 14.83344 & 269.17121 & 2.20230 \\ 10.31219 & 2.20230 & 253.03961 \end{pmatrix} \quad (2.15)$$

The diagonalized tensor is expressed in equation (2.16), whereas the isotropic value is $\alpha_{iso} = 310.33758$ a.u..

$$\alpha_{diag} = [252.26346 \quad 267.69308 \quad 411.05619] \quad (2.16)$$

The molecule is then simulated in the presence of constant electric fields performed by positioning point charges in the space along the principal axis of the molecule, each at 50 Å from the Si atom on their side. The generated electric field is calculated on the geometrical center ($x = -0.4614$ Å, $y = -1.0219$ Å, $z = -0.3952$ Å) of the molecule using equation (2.8). We extract the atomic charge for each electric field and aggregate them into the groups shown in Fig. 2.7a to derive the AC distribution. Fig. 2.7d shows the VACT of the molecule. The VACT is almost identical to the one obtained for the hexanedithiolbis(trimethylsilane), demonstrating that the length of the connecting chain does not influence, acting as a neutral actor.

Fig. 2.7e shows the VACT of the oxidized molecule, pointing out the different behavior. The charge on *Dot 3* is now positive and almost equal to 0.1 a.u.. The two trimethylsilane groups still exchange charge, but the charge separation is, in this case, higher. The charge on *Dot 1* and *Dot 2* is not symmetrical to the INPUT voltage, as seen from the different V_{IN} values where the slope changes. For voltages bigger than $V_{IN} = 1.5$ V in absolute value, the charge saturates to almost 0.3 a.u. and 0.6 a.u., as in the case of the oxidized hexanedithiolbis(trimethylsilane).

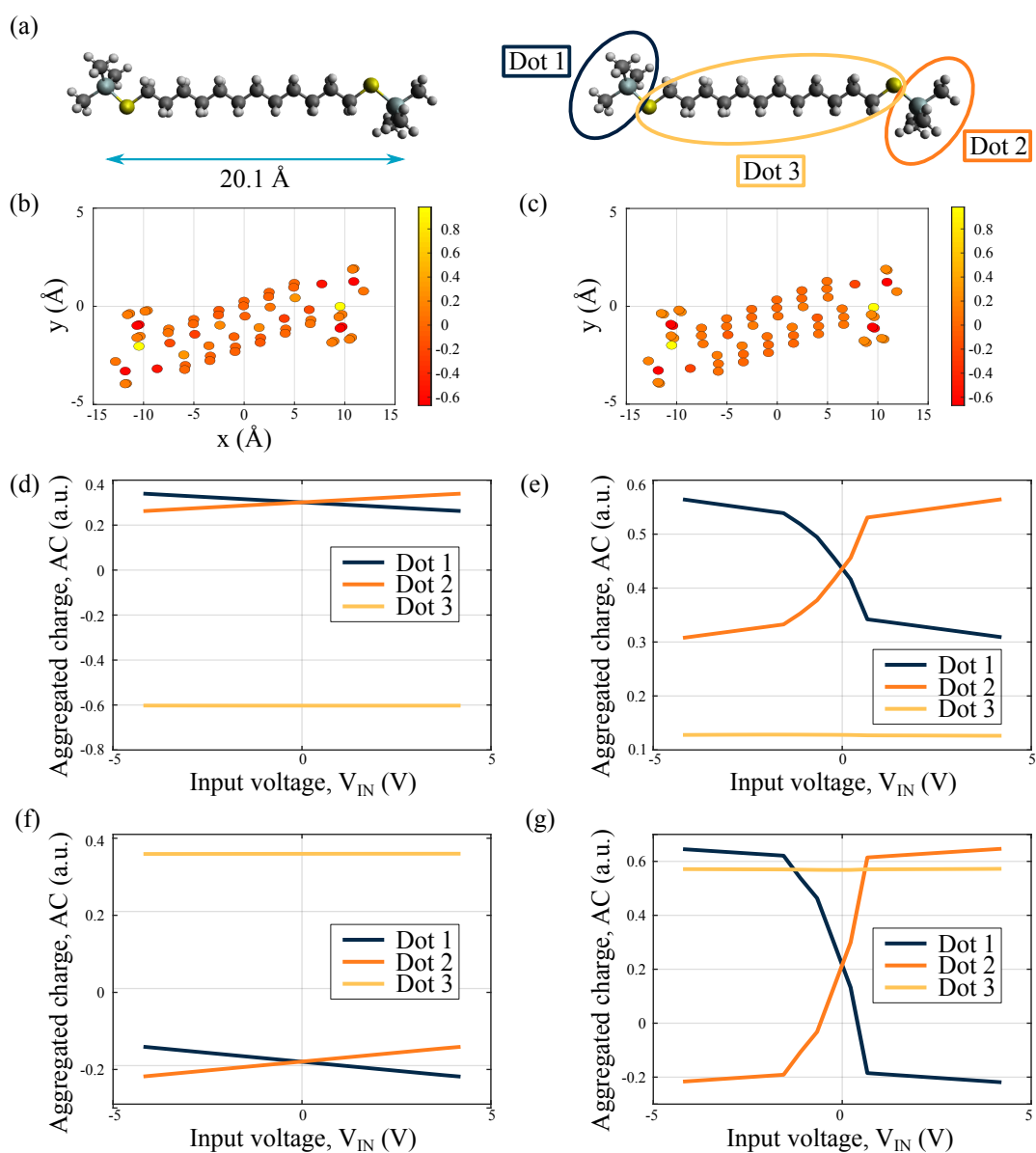


Fig. 2.7 Results of the DFT analysis on the dodecanedithiolbis(trimethylsilane) molecule. (a) 3D geometry shown in a ball-and-stick diagram and the atom groups for the calculation of the aggregated charges. (b) Molecular electrostatic potential evaluated with DFT calculation, the solid blue lines are isopotential curves. (c) ESP charges on each atom of the molecule. (d) VACT of the dodecanedithiolbis(trimethylsilane). (e) VACT of the oxidized dodecanedithiolbis(trimethylsilane). (f) VACT of the molecule with sulfur atoms belonging to *Dot 1* and *Dot 2*, instead of *Dot 3*. (g) VACT of the oxidized dodecanedithiolbis(trimethylsilane) with sulfur atoms belonging to *Dot 1* and *Dot 2*, instead of *Dot 3*. Atoms in figure are colored as follows: H = white, C = charcoal, S = yellow, Si = gray.

Even in case oxidation has a strong impact on the polarizability, indeed the tensor becomes:

$$\alpha = \begin{pmatrix} 14915.74840 & 2481.11846 & 879.81036 \\ 2481.11846 & 683.72260 & 150.26135 \\ 879.81036 & 150.26135 & 298.92055 \end{pmatrix} \quad (2.17)$$

The diagonalized tensor is expressed in equation (2.18), whereas the isotropic value is $\alpha_{iso} = 5299.46385$ a.u..

$$\alpha_{diag} = \left[246.03661 \quad 263.67372 \quad 15388.68122 \right] \quad (2.18)$$

Since from the charge values shown in Fig. 2.7b and Fig. 2.7c, the sulfur atoms emerge as key actors, we group them with *Dot 1* and *Dot 2*. So, in this case, *Dot 3* is just composed of the dodecane chain. Fig. 2.7f and Fig. 2.7g show the VACT obtained for the neutral and the oxidized molecules, respectively. The behavior of the aggregated charge strongly changes since, in both cases, the charge on *Dot 3* became positive. The charge exchange between the logic dots still occurs, but the values differ. This result demonstrates the importance of defining the molecule dots because even a few atoms can completely change the perspective of global behavior.

2.4.3 Tridecanedithiolbis(trimethylsilane)

The dodecane instead of the hexane does not change the aggregated charges behavior too much, but an asymmetry between *Dot 1* and *Dot 2* shows up in the oxidized molecule. To remove the asymmetry, we substitute the dodecane with a tridecane, leading to the structure shown in Fig. 2.8a. The coordinates have been obtained from the geometry optimization of the isolated molecule with the UKS method using CAM-B3LYP functional with def2-TZVP basis set [119]. Grimme D3 correction is applied [120, 121]. Since the VACT asymmetry is present only in the oxidized molecule, we analyzed only the difference in that case. Fig. 2.8b shows the ESP charges of the oxidized tridecanedithiolbis(trimethylsilane). Considerations are very similar to the previous molecules. Hence, the silicon atoms present a positive charge, whereas the sulfur atoms and the carbons in the methyl groups hold a negative charge.

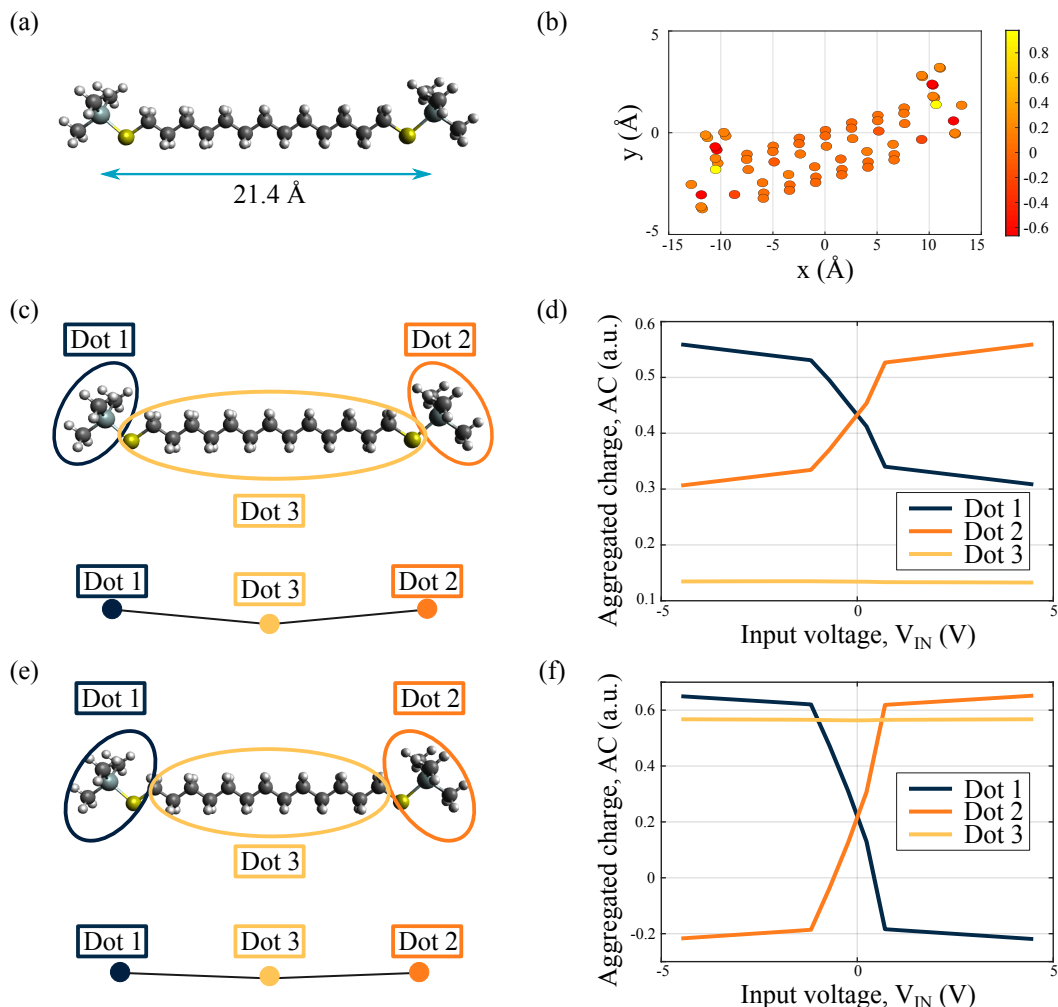


Fig. 2.8 Results of the DFT analysis on the tridecanedithiolbis(trimethylsilane) molecule. (a) 3D geometry shown in a ball-and-stick diagram. (b) ESP charges on each atom of the molecule. (c) Atom groups for the calculation of the aggregated charges and position of the point charges associated with the dots in the case in which S atoms belong to *Dot 3*. (d) VACT of the oxidized tridecanedithiolbis(trimethylsilane) with sulfur atoms belonging to *Dot 3*. (e) Atom groups for the calculation of the aggregated charges and position of the point charges associated with the dots in the case in which S atoms belong one to each logic dot. (f) VACT of the oxidized tridecanedithiolbis(trimethylsilane) with sulfur atoms belonging to *Dot 1* and *Dot 2*. Atoms in figure are colored as follows: H = white, C = charcoal, S = yellow, Si = gray.

Since the molecular width increases, also the polarizability increases, becoming

$$\alpha = \begin{pmatrix} 16851.23764 & 2496.05876 & 630.68429 \\ 2496.05876 & 649.99873 & 95.52290 \\ 630.68429 & 95.52290 & 279.48025 \end{pmatrix} \quad (2.19)$$

The diagonalized tensor is expressed in equation (2.20), whereas the isotropic value is $\alpha_{iso} = 5926.90554$ a.u..

$$\alpha_{diag} = \begin{bmatrix} 255.48929 & 274.17625 & 17251.05108 \end{bmatrix} \quad (2.20)$$

The molecule is then simulated in the presence of constant electric fields performed by positioning point charges in the space along the principal axis of the molecule, each at 50 Å from the Si atom on their side. The generated electric field is calculated on the geometrical center ($x = 0.1310$ Å, $y = -0.5863$ Å, $z = -0.4791$ Å) of the molecule using equation (2.8). By grouping the atoms as shown in Fig. 2.8c, that is, with the two sulfur atoms inside *Dot 3*, the resulting VACT is the one shown in Fig. 2.8d. The AC on *Dot 3* is still constant, and changing the field moves the charge between *Dot 1* and *Dot 2*. The curves for the logical dots are still not identical, but the global behavior improved its symmetry to V_{IN} , confirming that geometrical symmetry can imply electrostatic symmetry. Fig. 2.8c also shows the equivalent point charges system obtained by the group division that is eventually used in the Self-Consistent Electrostatic Potential Algorithm (SCERPA) simulator. The position of the point charges associated with the dots is calculated considering both geometry and electrostatic influence, as described by the following equation

$$p_{Dot_j}(x, y, z) = \frac{\sum_{i=1}^{N_j} p_i(x, y, z) \cdot |q_i|}{\sum_{i=1}^{N_j} |q_i|} \quad (2.21)$$

where p_{Dot_j} indicates the position of the dot j , N_j is the number of atoms belonging to *Dot_j*, p_i is the position of the i^{th} atom inside *Dot_j* and q_i is its ESP charge.

Fig. 2.8e shows atoms grouping when the two S move from *Dot 3* to logic dot and the corresponding position of the point charges associated with the dots. The position changes minimally, *Dot 1* and *Dot 2* are slightly closer, and the angle formed with *Dot 3* is a bit smaller. Besides the minor changes in the geometrical definition of the dots, the electrostatic behavior of the dot changes significantly, as reported in Fig. 2.8f by the VACT. Compared to the previous situation, *Dot 3* loses half an electron, bringing the AC to almost 0.6 a.u.. Logic dots exchange charge between -0.2 a.u. and 0.6 a.u., so the maximum charge separation between logic dots is equal to 0.8 a.u., much higher than the 0.25 a.u. charge separation shown in Fig. 2.8d.

In both the analyzed cases, the actual behavior of the molecule is obviously the same, as the data are extracted from the same simulations. What changed was only the application of the aggregated charges model from MoSQuiTo. Indeed, each model comes with its own scope, and one must pay attention to the implications of applying a model. In this case, the choice of which atoms belong to each dot is crucial, and not only must we consider the electrostatic behavior but also the shape and extension of the molecule in order not to lose the actual behavior of the molecule and have a realistic model of the system we are considering.

2.4.4 1,3,5benzenetrithiol linked bis(hexanedithiol) bis(trimethylsilane)

All the molecules presented so far present a good charge separation and can be divided into three functional groups, paying attention to where to place atoms. In all the molecules seen so far, *Dot 3* does not influence the charge localization for information encoding and keeps a constant charge independently from the applied field. Also, all the molecules analyzed so far are mainly extended along a single axis, not presenting a ‘V’ or ‘Y’ shape, as we discuss in section 1.1.4. This shape may affect the ability to reset the molecule by applying an electric field orthogonal to the direction in which the molecule lays out. Given the importance of the clock and the ability to reset the molecule discussed in section 1.3, we include a functional group in the middle of *Dot 3*, serving as a sort of joint for an eventual third group for clock handling.

The first group that we use is 1,3,5benzenetrithiol composed of a benzene ring with three thiols linked to it. Fig. 2.9a shows how the thiols are used as linking points to hexanes, bringing to the two trimethylsilane, leaving a free thiol for the connection of an additional dot for the reset. The coordinates have been obtained from the geometry optimization of the isolated molecule with the UKS method using CAM-B3LYP functional with def2-TZVP basis set [119]. Grimme D3 correction is applied [120, 121]. Focusing on the oxidized version of the molecule, Fig. 2.9b shows the ESP charges, demonstrating that the introduction of the new group does not change drastically the equilibrium charge configuration. The most positive charge is still located on the silicon atoms, sulfur atoms hold a significant negative charge,

and the connecting chain comprising the ring seems balanced with all the atoms having a slightly positive charge.

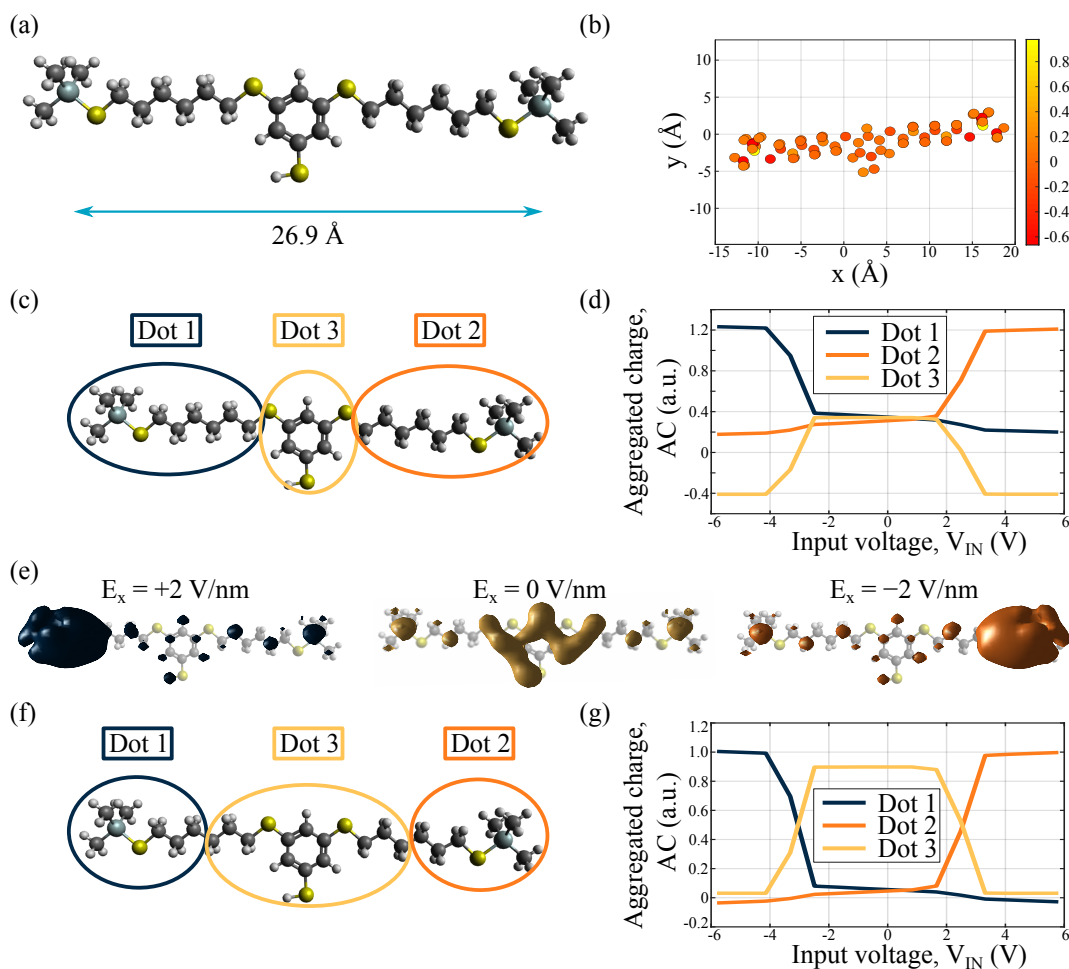


Fig. 2.9 Results of the DFT analysis on the oxidized 1,3,5benzenetriethiol linked bis(hexanedithiol) bis(trimethylsilane). (a) 3D geometry shown in a ball-and-stick diagram. (b) ESP charges on each atom of the molecule. (c) Atom groups for the calculation of the aggregated charges. (d) VACT of the molecule with only the 1,3,5benzenetriethiol belonging to *Dot 3*. (e) Isopotential surfaces for different electric field values showing where the charge localizes. (f) Atom groups for calculating the aggregated charges based on the analysis of the isopotential surfaces. (g) VACT of the molecule with dots chosen based on the actual charge localization on the molecule. Atoms in figure are colored as follows: H = white, C = charcoal, S = yellow, Si = gray.

The introduction of the benzene ring decreases the polarizability of the molecule, which is:

$$\alpha = \begin{pmatrix} 1086.18703 & 74.05931 & 20.51756 \\ 74.05931 & 498.60377 & 9.22278 \\ 20.51756 & 9.22278 & 306.30324 \end{pmatrix} \quad (2.22)$$

The diagonalized tensor is expressed in equation (2.23), whereas the isotropic value is $\alpha_{iso} = 630.36468$ a.u..

$$\alpha_{diag} = \begin{bmatrix} 305.47952 & 489.65150 & 1095.96303 \end{bmatrix} \quad (2.23)$$

By grouping atoms as shown in Fig. 2.9c, with *Dot 3* composed of only 1,3,5-benzenetriethiol, and *Dot 1* and *Dot 2* are both equally composed with the trimethylsilane and the hexanedithiol. Fig. 2.9d shows the resulting VACT. The behavior observed in the presence of an electric field is different from the previous ones. For small fields, the charge is equally distributed on the three dots. For electric fields that generate voltages bigger than 2 V in absolute value, the charge moves from *Dot 3* to either *Dot 1* or *Dot 2*, depending on the sign of the electric field.

To check if the VACT describes the actual behavior accurately, we analyze the isopotential surfaces for different electric field values to see where the charge localizes. Fig. 2.9e shows the isopotential surface for $E_x = 2 \text{ V nm}^{-1}$, $E_x = 0 \text{ V nm}^{-1}$, and $E_x = -2 \text{ V nm}^{-1}$ corresponding to the VACT points where the voltage is respectively $V_{IN} = -6 \text{ V}$, $V_{IN} = 0 \text{ V}$, and $V_{IN} = 6 \text{ V}$. When the field is applied, the isopotential surface encloses the trimethylsilane and half of the hexanedithiol chain. On the contrary, when no or a small field is applied, the charge localizes on the central ring and the other halves of the hexanedithiol, leaving the external parts of the molecule with almost no charge.

Since the VACT of Fig. 2.9d does not point out this behavior, we modify the dot definition to match the actual functional group of the molecule, obtaining the configuration shown in Fig. 2.9f. With the new definition of the dots, we get a different VACT, shown in Fig. 2.9g. In this case, we see that for small fields, the charge is constant and mainly localized on *Dot 3*, whereas with a big applied field, *Dot 3* releases almost all the charge to either *Dot 1* or *Dot 2*, depending on the value of the electric field.

2.4.5 Trithiocyanuric acid linked bis(hexanedithiol) bis(trimethylsilane)

Besides the peculiar behavior obtained using the 1,3,5benzenetrithiol as a joining element, we also investigate a different one. Specifically, using the trithiocyanuric acid as a linking group, we obtain the molecule shown in Fig. 2.10a. The trithiocyanuric acid is a hexagonal ring composed of alternating carbon and nitrogen atoms, and the sulfurs to which we want to attach the hexanedithiol are bonded to the C atoms of the ring. The coordinates have been obtained from the geometry optimization of the isolated molecule with the UKS method using CAM-B3LYP functional with def2-TZVP basis set [119]. Grimme D3 correction is applied [120, 121].

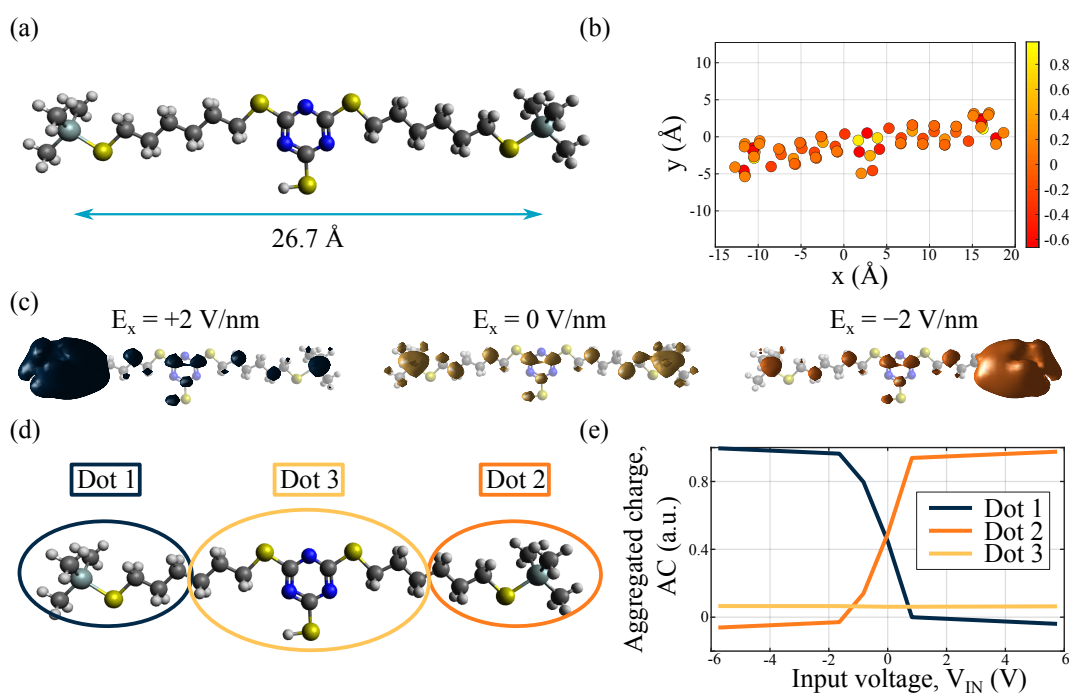


Fig. 2.10 Results of the DFT analysis on the oxidized trithiocyanuric acid linked bis(hexanedithiol) bis(trimethylsilane). (a) 3D geometry shown in a ball-and-stick diagram. (b) ESP charges on each atom of the molecule. (c) Isopotential surfaces for different electric field values showing where the charge localizes. (d) Atom groups for calculating the aggregated charges based on the analysis of the isopotential surfaces. (e) VACT of the molecule with dots chosen based on the actual charge localization on the molecule. Atoms in figure are colored as follows: H = white, C = charcoal, N = blue, S = yellow, Si = gray.

Even if the geometrical structure is similar to the previously analyzed case, the electrostatic behavior strongly changes. Fig. 2.10b shows the ESP charges of the

oxidized molecule at equilibrium. The Si atoms still keep a positive charge, and the two hexanedithiols hold a slightly positive charge. The trithiocyanuric acid charge alternates between a strong positive charge on the carbons and a strong negative charge on the nitrogen, especially on those linked to the two hexanedithiols through the S atoms.

The presence of the trithiocyanuric acid is also visible in the polarizability, which is much higher than the case in which the group is the benzenetrithiol

$$\alpha = \begin{pmatrix} 23550.06626 & 2978.05838 & 689.08133 \\ 2978.05838 & 742.56808 & 87.05502 \\ 689.08133 & 87.05502 & 295.61537 \end{pmatrix} \quad (2.24)$$

The diagonalized tensor is expressed in equation (2.25), whereas the isotropic value is $\alpha_{iso} = 8196.08324$ a.u..

$$\alpha_{diag} = \left[275.19988 \quad 360.14894 \quad 23952.90089 \right] \quad (2.25)$$

Fig. 2.10c shows the isopotential surface analyses for different electric field values. Similarly to the case with the 1,3,5benzenetrithiol, when an electric field is applied, the charge localizes on the trimethylsilane and half of the hexanedithiol on either left or right, depending on the sign of the electric field. When no field is present, a small positive charge is localized around both the silicon atoms and almost no charge is on the rest of the molecule.

Considering the charge localization described by the isopotential surface analysis, we group the atoms as shown in Fig. 2.10d. Fig. 2.10e shows the resulting VACT, confirming the constant slightly positive charge on *Dot 3*. With no applied field, the charge is balanced between *Dot 1* and *Dot 2*, which exchange charge in the presence of an electric field, reaching a maximum charge separation of almost 1 a.u..

2.4.6 Trithiocyanuric acid linked tris(hexanedithiol) bis(trimethylsilane)

The presence of the free thiol of the trithiocyanuric acid in the previous case allows for the connection of an additional hexanethiol to obtain a ‘T’ shape, theoretically helping the clock. The geometry of the resulting molecule is shown in Fig. 2.11a.

The choice of not attaching a third trimethylsilane group derives from the possibility of using the end of the hexanethiol to link the molecule to an eventual surface [122]. The coordinates have been obtained from the geometry optimization of the isolated molecule with the UKS method using CAM-B3LYP functional with def2-TZVP basis set [119]. Grimme D3 correction is applied [120, 121].

The introduction of the third hexanethiol slightly changes the polarizability, especially showing a reduced isotropic behavior:

$$\alpha = \begin{pmatrix} 25580.64594 & 5042.99662 & 670.19788 \\ 5042.99662 & 1518.71704 & 132.56929 \\ 670.19788 & 132.56929 & 377.85808 \end{pmatrix} \quad (2.26)$$

The diagonalized tensor is expressed in equation (2.27), whereas the isotropic value is $\alpha_{iso} = 9159.07369$ a.u..

$$\alpha_{diag} = \begin{bmatrix} 360.03482 & 504.56472 & 26612.62152 \end{bmatrix} \quad (2.27)$$

We group the atoms as in Fig. 2.11b and study the system with different applied electric fields. By using point charges, we generate an electric field in the clock direction equal to $E_y = 2 \text{ V nm}^{-1}$ to obtain the HOLD state and $E_y = -2 \text{ V nm}^{-1}$ to replicate the RESET state. For both situations, we superpose a switching field as in the previous cases along the axis connecting *Dot 1* and *Dot 2*.

Fig. 2.11c shows the isopotential surface when the clock field reproduces the HOLD state for the three switching fields $E_x = 2 \text{ V nm}^{-1}$, $E_x = 0 \text{ V nm}^{-1}$, and $E_x = -2 \text{ V nm}^{-1}$. The charge on *Dot 3* doesn't change with the switching field that moves the charge between the two regions with the silicon atoms. The introduction of the third hexanedithiol does not affect the global behavior of the charge, as emerges from the VACT shown in Fig. 2.11e. *Dot 3* holds a constant charge equal to -0.4 a.u. and logic dots are involved in the charge localization when E_x is different from zero.

Similarly, Fig. 2.11d shows the isopotential surfaces when the clock field reproduces the RESET state for the three switching fields $E_x = 2 \text{ V nm}^{-1}$, $E_x = 0 \text{ V nm}^{-1}$, and $E_x = -2 \text{ V nm}^{-1}$. The ideal situation would be to have the charge localized on *Dot 3*, independently from the applied E_x along the switching direction. As evinced by the isopotential surfaces, this is not the case since the charge localizes on the trimethylsilane when the switching field is different from zero. Fig. 2.11f shows the

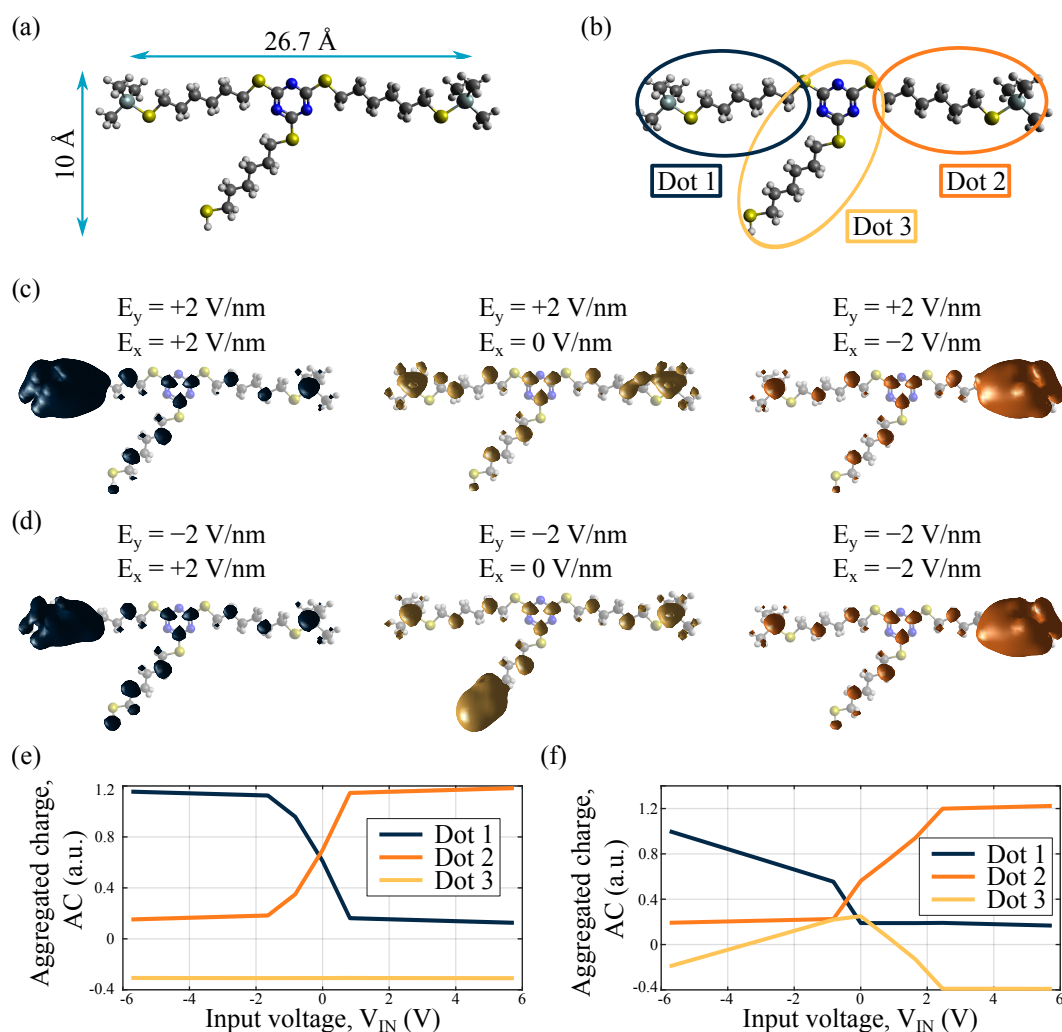


Fig. 2.11 Results of the DFT analysis on the oxidized trithiocyanuric acid linked tris(hexanedithiol) bis(trimethylsilane). (a) 3D geometry shown in a ball-and-stick diagram. (b) Atom groups for the calculation of the aggregated charges. (c) Isopotential surfaces when the clock field is $E_y = 2 \text{ V nm}^{-1}$ for different electric field values in the switching direction, showing where the charge localizes. (d) Isopotential surfaces when the clock field is $E_y = -2 \text{ V nm}^{-1}$ for different electric field values in the switching direction, showing where the charge localizes. (e) VACT of the molecule when the clock field is $E_y = 2 \text{ V nm}^{-1}$, so in the HOLD state. (f) VACT of the molecule when the clock field is $E_y = -2 \text{ V nm}^{-1}$, so in the RESET condition. Atoms in figure are colored as follows: H = white, C = charcoal, N = blue, S = yellow, Si = gray.

corresponding VACT, illustrating that the charge on *Dot 1* and *Dot 2* does switch, preventing the molecule from resetting the information. The differences observable between the case with positive and negative E_x derive from the lack of symmetry of

the molecule since the third hexanedithiol does not stand straight in the middle of the molecule.

The incremental theoretical results obtained on the presented case of study are promising for the electrostatic footprint shown and are an example of molecule engineering applied to molFCN. Further work, both theoretical and experimental, should address the method of providing a third dot able to hold the charge in the reset state while enabling surface attachment in ordered structures.

Chapter 3

Molecules electrostatics and environment

The analysis of the molecule in chapter 2 started to point out the centrality of intra-molecular interactions and the resulting electrostatic footprint in molecular Field-Coupled Nanocomputing (molFCN). The electrostatic footprint of molecules refers to the distribution and arrangement of electric charges within a molecular structure. Electrostatic interactions govern molecular behavior and thus play a pivotal role in determining how information propagates and can be manipulated.

Understanding how different charge distributions impact information treatment in molFCN is the first step to characterize intermolecular interactions between molecules. The ultimate goal is to reverse the process. Hence, starting from circuit requirements, design tailored molecules that accomplish the desired tasks [123].

Moreover, in this technology, circuits have traditionally utilized a single type of molecule, with logic functions achieved through tailored cell positioning. Integrating multiple types of molecules in the same layout augments circuit capabilities and introduces a fresh approach to conceptualizing molFCN. Specifically, by strategically incorporating different molecules with distinct electrostatic footprints into designated layout positions, it becomes feasible to emulate artificial neuron behavior using the Majority Voter (MV) layout [124].

The following sections consider three molecular species with different electrostatic behavior: neutral, oxidized, and zwitterionic. For each species, a reference geometry is fixed, and the electrostatic behavior is modeled by equation instead of

obtaining it by Density Functional Theory (DFT) simulations. The regularity of the molecular model enables the direct comparison of the effect of the pure electrostatic behavior in inter-molecular interactions. The fundamental cell-to-cell interactions are then analyzed, calculating the system energy for different charge configurations to obtain the energetically favored configurations. After that, we simulate a few elementary gates involving the fundamental interactions for all the considered molecular species. In this way, it is possible to detect the impact on logic computation of the electrostatic interaction of nearby molecules. The chapter continues with considering the electrostatic occupation of the whole device, which is a consequence of the crosstalk noise field. The last section discuss how it is possible to obtain a coplanar crossover interconnection exploiting a purely electrostatic solution.

3.1 Electrostatic molecular characteristics

Understanding the electrostatic characteristics of different molecular species is crucial for molFCN [125]. Researchers previously considered neutral, oxidized, and zwitterionic molecules as potential candidates.

Neutral molecules, lacking a net charge, constitute a significant portion of molecular species. Despite their lack of overall charge, neutral molecules exhibit intricate electrostatic characteristics arising from the distribution of electron densities within their structures.

Oxidized molecules are characterized by a positive charge resulting from the loss of electrons. The presence of a net positive charge imparts specific electrostatic behaviors to oxidized molecules, influencing their interaction with other molecules and their response to external stimuli. Oxidized molecules are the natural substitutes to realize a Quantum-dot Cellular Automata (QCA) cell. Indeed, a QCA cell is characterized by an excess of charge (electrons), as oxidized shows an excess of positive charge. Understanding the impact of the electrostatic profile of oxidized molecules in the fundamental interactions involved in molFCN is essential for understanding and designing circuits with tailored properties and functionalities.

It is known that circuits composed of oxidized molecules must be neutralized, for instance, with counterions or charged defects in the substrate, to prevent crosstalk effects [64]. Oxidized molecules with counterions behave similarly to zwitteri-

onic molecules regarding electrostatic properties, even if counterions disturb the electrostatic characteristics of intermolecular interactions [126, 127].

Zwitterionic molecules, featuring both positively and negatively charged functional groups within the same molecular structure, exhibit intriguing electrostatic characteristics. Despite the presence of localized charges, zwitterionic molecules maintain overall neutrality [128]. This unique electrostatic profile renders zwitterionic molecules versatile candidates, which have been intensely investigated in recent years [65, 66, 110, 126].

3.1.1 Reference structures and properties

To study the link between molecular electrostatics and information propagation or elaboration, we define a standard structure and conceptual characteristics for each molecule: neutral, oxidized, and zwitterionic. Working in terms of models allows for consistent observations with fewer geometry-dependent influences compared to using existent molecules.

Considering the typical dimensions of the proposed molecules, the conceptual molecule is ‘V’-shaped, composed of three dots. Fig. 3.1a shows the relative position of the three dots. For the sake of simplicity, we placed *Dot 1* and *Dot 2* at 1 Å from each other. *Dot 3* is positioned 5 Å away from the plane containing logic dots and along the axis of their connecting segment. The quantitative analyses presented in the following sections rely on the specified geometry, although the conclusions drawn remain applicable even with potential variations in molecular size.

The curve describing the electrostatic behavior of neutral, oxidized, and zwitterionic molecules, is the VACT. VACTs describe the behavior of aggregated charges in the presence of an electric field and are typically obtained by analyzing DFT-derived ElectroStatic Potential (ESP) charges. Since we are referring to a conceptual molecule, we must assume a particular behavior for the three molecular species under investigation. Considering the analysis of other molecules from the literature ([62]) and those in chapter 2, we choose equation (3.1) as a reference curve shape for the Aggregated Charge (AC) on a dot in the hold state expressed in atomic units. Each AC is denoted as Q_i , where i represents the i th dot of the molecule. The value V_{in} is

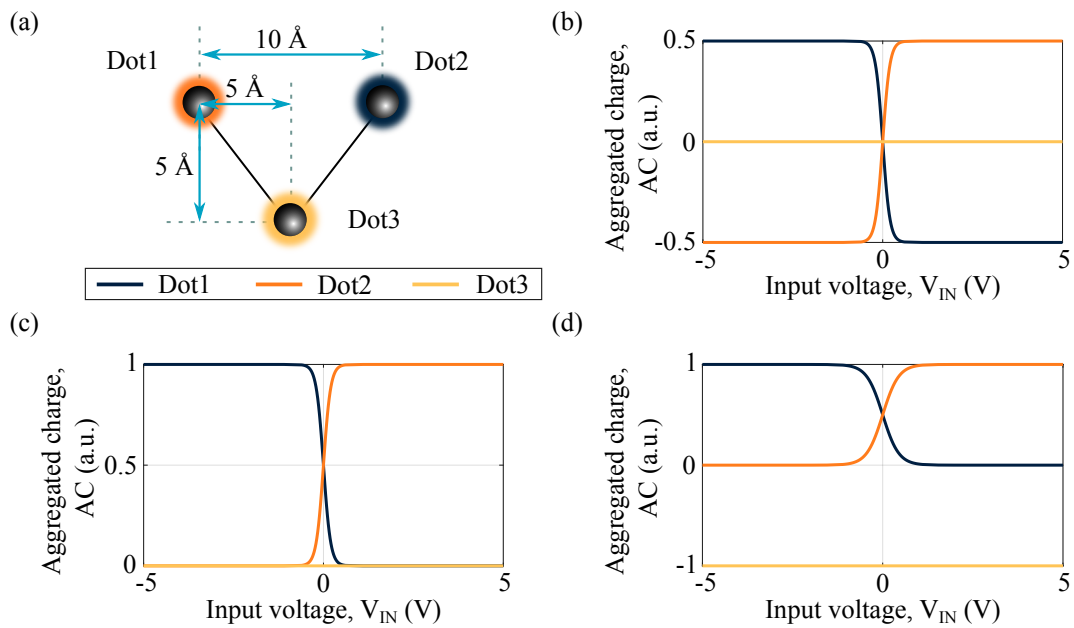


Fig. 3.1 Reference molecular structure and electrostatic properties of the conceptual molecules used to analyze the interactions. (a) Dot names and relative position. (b) V_{in} -AC transcharacteristic (VACT) defined for the neutral molecule in the hold state. (c) VACT defined for the oxidized molecule in the hold state. (d) VACT defined for the zwitterionic molecule in the hold state.

the input voltage expressed in volts.

$$Q_i = \frac{1}{1 + e^{10 \cdot V_{in}}} \quad (3.1)$$

Aiming to reduce the number of characters involved in the interactions, *Dot 3* always has a constant charge, and the value depends on the considered molecular species. Moreover, since we want to analyze the fundamental electrostatic interactions involved in propagation, we focus on the curves for the hold state, neglecting the reset condition.

Fig. 3.1b shows the VACT for the neutral molecule. The set of equations describing the AC is reported below in equation (3.2). The charge on *Dot 3* is chosen to be zero, so the charge on the logic dots is equal to the general curve shifted down by a

quantity equal to 0.5 a.u. to guarantee neutrality.

$$VACT_{ne} \rightarrow \begin{cases} Q_1 = \frac{1}{1+e^{10 \cdot V_{in}}} - 0.5 \\ Q_2 = \frac{1}{1+e^{-10 \cdot V_{in}}} - 0.5 \\ Q_3 = 0 \end{cases} \quad (3.2)$$

Fig. 3.1c shows the VACT for the oxidized molecule in the hold state. In this case, the total charge must be equal to 1 a.u. in all the points. Again, *Dot 3* does not hold any charge, and logic dots are the only ones involved in charge displacement. The equations describing the AC are reported below in equation (3.3).

$$VACT_{ox} \rightarrow \begin{cases} Q_1 = \frac{1}{1+e^{10 \cdot V_{in}}} \\ Q_2 = \frac{1}{1+e^{-10 \cdot V_{in}}} \\ Q_3 = 0 \end{cases} \quad (3.3)$$

Finally, Fig. 3.1d shows the VACT for the zwitterionic molecule in the hold state. As mentioned, they are neutral molecules presenting localized charge aggregation inside some functional groups. The behavior of zwitterionic molecules is thus in between those of neutral and oxidized species. We model its charge behavior as for oxidized molecules concerning the logic dots. However, this time *Dot 3* holds a negative charge equal to -1 a.u. to guarantee global charge neutrality and creates a vertical not-null dipole moment for $V_{in} = 0V$.

$$VACT_{zw} \rightarrow \begin{cases} Q_1 = \frac{1}{1+e^{10 \cdot V_{in}}} \\ Q_2 = \frac{1}{1+e^{-10 \cdot V_{in}}} \\ Q_3 = -1 \end{cases} \quad (3.4)$$

Following the exploited model, the ideal oxidized molecules with an external counterion have the same electrostatic behavior as zwitterionic molecules.

The fundamental interactions between molecular cells in molFCN circuits are four and are represented in Fig. 3.2. The horizontal and vertical interactions are involved in information propagation and are the laying blocks of gates like the MV. Diagonal interactions are the central elements of inversion.

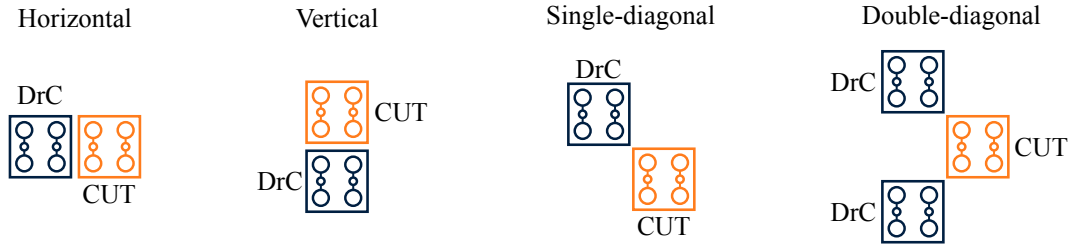


Fig. 3.2 Fundamental cell interaction involved in molFCN. The Driver Cell (DrC) is shown in blue, whereas the Cell Under Test (CUT) is shown in orange.

3.1.2 Molecule energy model

The electrostatic energy of a system of point charges refers to the potential energy associated with the arrangement of charged particles within the system. This energy arises from the electrostatic interactions between the charges, governed by Coulomb's law. Coulomb's law states that the force between two point charges is directly proportional to the product of their magnitudes and inversely proportional to the square of the distance between them. Mathematically, it is expressed as:

$$F = k \cdot \frac{q_1 q_2}{r^2} \quad (3.5)$$

where $k = 1/(4\pi\epsilon_0)$ is the Coulomb's constant with ϵ_0 being the vacuum permittivity, q_1 and q_2 are the magnitudes of the charges, and r is the distance between the charges.

The electrostatic potential energy (U) of a system of point charges can be calculated from the work done to assemble the charges from infinity, where they exert no force on each other, to their final configuration. For a system of N point charges, the total electrostatic potential energy is the sum of the potential energies of all pairs of charges. Mathematically, it is expressed as

$$U = \sum_{i=1}^{N-1} \sum_{j=i+1}^N k \cdot \frac{q_i q_j}{r_{i,j}} \quad (3.6)$$

where k is the Coulomb's constant, q_i and q_j are the magnitudes of the charges, and $r_{i,j}$ is the distance between the i^{th} and j^{th} charges

In the Molecular Simulator Quantum-dot cellular automata Torino (MoSQuiTo) methodology, molecules are properly replaced by a set of point charges with a charge equal to the AC. In a system composed of two molecules, like a molecular cell,

where the molecules are called *Mol a* and *Mol b*, the interaction energy is derived from equation (3.6), becoming:

$$W_{a,b} = \sum_{i \in a} \sum_{j \in b} k \cdot \frac{Q_i Q_j}{r_{i,j}} \quad (3.7)$$

where k is the Coulomb's constant, Q_i and Q_j are the values of the aggregated charges, and $r_{i,j}$ denotes the distance between the i^{th} of *Mol a* and the j^{th} of *Mol b* [129]. In a circuit composed of N molecules, the total energy is evaluated as:

$$W_{tot} = \sum_{i=1}^{N-1} \sum_{j=i+1}^N W_{i,j} \quad (3.8)$$

where $W_{i,j}$ is the interaction energy between the i^{th} and j^{th} molecules calculated as expressed in equation (3.7) [129].

Each molecule has a different set of AC and geometry, thus generating different interactions and energy. The analysis of the total energy of the system provides useful insight into the behavior the system has and can be used to predict the outcome.

In this section, we are interested in understanding the general implications of using different molecular species in the fundamental interactions depicted in Fig. 3.2. Except for the double-diagonal interaction system, the others are composed of two molecular cells (i.e., four molecules) named DrC and CUT. We can calculate their energy using equation (3.8), particularly when the cell encodes some logic values. Comparing the total energy of the system in different charge arrangements provides information regarding the minimum energy configurations that are expected to most likely appear.

We define W_{XY} as the total energy of the system when the DrC and the CUT encode logic values X and Y (i.e., '0' or '1'), respectively. Considering the cells as the unit elements, more in a QCA-like style, we need to remove the interaction between molecules of the same cells from the energy W_{XY} . In that case, the energy W_{XY} of the system becomes

$$W_{XY} = W_{tot} - W_{DrC}^X - W_{CUT}^Y \quad (3.9)$$

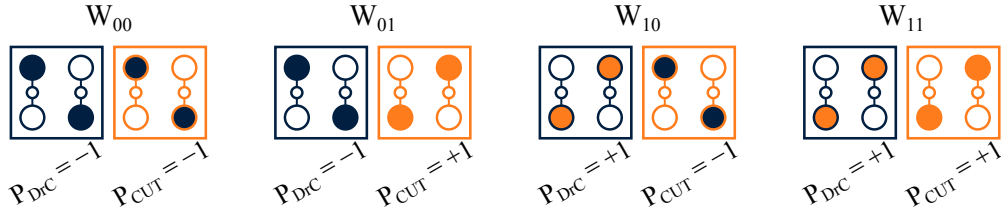


Fig. 3.3 Charge distribution in the cell for the four combinations of logic values in the horizontal interaction system.

where W_{tot} is the total energy of the whole DrC-CUT charge system, whereas W_{DrC}^X and W_{CUT}^Y are the electrostatic energies of the isolated DrC and CUT cells encoding logic values X and Y , respectively.

If we consider the cell as the unit element, thanks to the association between the logic value of a cell and its cell polarization defined in equation (1.3), we can refer W_{XY} to the DrC and CUT polarizations, P_{DrC} and P_{CUT} . Considering all the combinations of logic values for the two cells, we get:

- energy W_{00} when $P_{DrC} = -1$ and $P_{CUT} = -1$
- energy W_{01} when $P_{DrC} = -1$ and $P_{CUT} = +1$
- energy W_{10} when $P_{DrC} = +1$ and $P_{CUT} = -1$
- energy W_{11} when $P_{DrC} = +1$ and $P_{CUT} = +1$

Fig. 3.3 illustrates the charge distribution associated with the four cases in the system of DrC and CUT creating the horizontal interaction.

Another energy measurement value associated with the system energy is the so-called kink energy U_k [48]. Applied to the system of DrC and CUT we are considering, the kink energy measures the energy difference between the system with DrC and CUT encoding the same values (W_{00} or W_{11}) and the configuration encoding different logic values (W_{01} or W_{10}). The energy W_{XY} does not include the contribution of the interactions inside each single cell as expressed in equation (3.9), meaning that the U_k is a consequence of the interactions between separate cells only.

Referring the calculation to a specific DrC configuration, we obtain kink energies U_{k0} and U_{k1} according to the driver logic as:

$$U_{k0} = W_{00} - W_{01} \quad U_{k1} = W_{11} - W_{10} \quad (3.10)$$

In the literature, the values U_{k0} and U_{k1} are always assumed to be equal, they are never distinguished. However, in a molecular system, a difference might exist between the two values in some cases, as shown in the following. It is possible to define also the kink error Δ as the difference between the kink energies for the two possible values of the DrC cell:

$$\Delta = U_{k0} - U_{k1} \quad (3.11)$$

The kink error is directly related to the functional behavior of the interaction, measuring the dependence of the interaction on the values encoded by the DrC. If the kink error is null, the interaction between DrC and CUT is identical for the logic states, so we have a symmetric interaction. Contrarily, a non-null value of Δ implies a non-symmetric interaction, so the DrC encoding a logic '0' or a logic '1' generates different outcomes.

The discussion until now has been carried out considering a system composed of two cells, the DrC and the CUT. All the considerations remain valid if we have two DrCs as in the double-diagonal interaction.

3.1.3 Energy analysis on the fundamental interactions

This section calculates the energy of the fundamental cell-to-cell interactions presented in Fig. 3.2 for the three molecular species we are discussing. The intermolecular distance d has been chosen to equal the molecular width $w = 1$ nm.

Horizontal cell-to-cell interaction

Horizontal interaction is fundamental for propagation and elaboration. Two adjacent cells encode the same logic. Considering again the DrC and CUT system shown in Fig. 3.3, we expect that the minimum energy configurations are those with the CUT copying the information on DrC.

For all the molecular species, we evaluate the electrostatic energy of the system W_{XY} varying the value of the polarization of the two cells, spanning over the possible logic configurations W_{00} , W_{01} , W_{10} , and W_{11} . We expect the CUT to copy the information of the DrC. From an energy point of view, this is obtained if both the conditions expressed in equation (3.12) are satisfied. Table 3.1 lists the energy values

Table 3.1 Electrostatic energy values of the system obtained for the three molecular species in the case of horizontal cell-to-cell interaction.

Species	W_{00} eV	W_{01} eV	W_{10} eV	W_{11} eV	U_{k0} eV	U_{k1} eV	Δ
Neutral	-0.147	0.147	0.147	-0.147	-0.294	-0.294	0.000
Oxidized	2.913	3.208	3.208	2.913	-0.294	-0.294	0.000
Zwitterionic	0.099	0.394	0.394	0.099	-0.294	-0.294	0.000

obtained for the three species.

$$W_{11} < W_{10} \quad W_{00} < W_{01} \quad (3.12)$$

If we use neutral molecules, the system energy is negative when both cells encode the same logic value, i.e., W_{00} and W_{11} . If cells have the wrong configuration, the energy of the system W_{XY} is positive, confirming that the minimum energy configurations exploit correct non-inverting propagation, as expressed in equation (3.12). Fig. 3.4a shows a map of the system energy evaluated for all the polarization values between -1 and 1 , for both P_{DrC} and P_{CuT} . The two energy minima (black regions) are associated with the information being the same, whereas the inverting configurations correspond to the energy maxima (white regions).

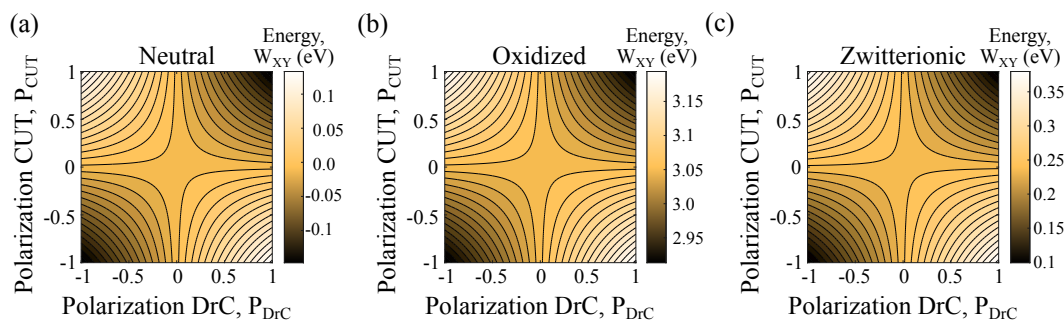


Fig. 3.4 Energy analysis for the horizontal cell-to-cell interaction for the different molecular species. (a) Energy of the two-cell system based on neutral molecules as a function of P_{DrC} and P_{CuT} . Each line represents a 0.01 eV energy increment. (b) Energy of the two-cell system based on oxidized molecules as a function of P_{DrC} and P_{CuT} . Each line represents a 0.01 eV energy increment. (c) Energy of the two-cell system based on zwitterionic molecules as a function of P_{DrC} and P_{CuT} . Each line represents a 0.01 eV energy increment.

The inequalities expressed in equation (3.12) also hold for oxidized and zwitterionic molecules. The positive charges in oxidized molecules induce purely repulsive interactions, resulting in positive energy values. In Fig. 3.4b, the energy map shows

for oxidized molecules the same trend obtained for neutral molecules, apart from the energy values, which are all positive. Conversely, Fig. 3.4c shows the energy map for the zwitterionic species, where values are again all positive. However, counterions mitigate the effects of oxidized molecules, leading to reduced energies.

Besides the different energy values, the behavior of the interaction is the same for all the species and independent from the logic encoded by the driver cell. We evaluate the kink energy according to equation (3.10) to confirm this observation. As reported in Table 3.1, the two kink energies are the same for the three species, leading to null kink errors Δ .

The logic encoded by the DrC does not affect the interaction for all molecular species. The adjacent horizontal cell-to-cell interaction is independent of the driver.

Vertical cell-to-cell interaction

Vertical interaction is fundamental for information propagation and is also present in gates such as the MV. In this case, the DrC and the CUT are vertically aligned, as Fig. 3.2 shows. Theoretically, in the QCA paradigm, there is no difference between this interaction and the horizontal adjacent one, given that the four logic dots are equally spaced and centered in the cell. For this reason, the energy analysis would give the same result obtained in the previous section.

However, two different situations happen by analyzing the field distribution in the horizontal and vertical interactions. Fig. 3.5 shows the two systems, highlighting the direction of the switching field on the CUT cell. If the DrC has only positive charges involved in the logic, as for oxidized molecules, then the generated radial field directly affects the adjacent CUT. As depicted in Fig. 3.5, the radial field is orthogonal to the switching field, which is unaffected, as the previous section demonstrated. Things change in the vertical interaction since, in this case, the two fields overlap, leading to possible unwanted charge configurations on the CUT. In the molecular implementation of QCA, the two molecules comprising the CUT are independent, contrarily to the general paradigm that permits only antipodal charge configurations. Hence, for a thorough interaction analysis, it is crucial to eliminate the assumption that the molecular cell only possesses diagonal charge distributions.

To highlight the difference with the horizontal interaction, we calculate the total energy of the two-cell system aligned vertically, with the DrC having a fixed logic



Fig. 3.5 Comparison between the direction of the switching field and the field generated by the DrC in adjacent horizontal and vertical cell-to-cell interaction.

value. Once the logic on the driver is fixed, we independently vary the charge distribution on the two molecules composing the CUT. Fig. 3.6a shows the case with $P_{DrC} = -1$ and the four limit configurations obtained by removing the constraint of having only antipodal charges. To identify the position of the charge in CUT, we use the normalized dipole moments $\widetilde{\mu}_1$ for molecule *Mol1* on the left of the cell and $\widetilde{\mu}_2$ for molecule *Mol2* on the right of the cell. Since vertical interaction should copy information from the DrC to the CUT, the minimum energy configuration must occur when $\widetilde{\mu}_1 = +1$ and $\widetilde{\mu}_2 = -1$ if $P_{DrC} = -1$.

At this point, we calculate the energy of the two-cell system linearly varying the dipole moment between -1 and 1 . Fig. 3.6b shows the energy map for the system implemented with neutral molecules. From the map emerges a minimum value for $\widetilde{\mu}_1 = +1$ and $\widetilde{\mu}_2 = -1$, thus respecting the condition for correct propagation. The total energy in the four limit configurations is also reported in Table 3.2. In this case, since both positive and negative charge aggregation are involved, the DrC field does not affect the switching field as much as to prevent correct propagation.

Fig. 3.6c shows the energy map for the system realized with oxidized molecules. The global minimum occurs for the condition with $\widetilde{\mu}_1 = \widetilde{\mu}_2 = +1$. That configuration corresponds to the charge aggregations being as far as possible from the DrC. The cause is to be found in the globally positive charge of oxidized molecules, generating the maximum influence of the DrC field over CUT.

Fig. 3.6d shows the energy map obtained for zwitterionic molecules. The logic dot behavior of zwitterionic molecules is the same as oxidized molecules, but they possess a negative counterion balancing the total charge of the molecule. The neutralization due to the presence of the counterion mitigates the impact of the DrC field, bringing the system back to the correct functioning, as the energy minimum coincides with $\widetilde{\mu}_1 = +1$ and $\widetilde{\mu}_2 = -1$.

Table 3.2 Electrostatic energy values of the system obtained for the three molecular species in vertical cell-to-cell interaction with a fixed DrC polarization. The subscript numbers refer to the normalized dipole moment of the two molecules composing the CUT. The values in parentheses are the energy without considering the electrostatic contribution of charges within the same molecule.

	Species	$W_{-1,1}$ eV	$W_{1,1}$ eV	$W_{-1,-1}$ eV	$W_{1,-1}$ eV
$P_{DrC} = -1$	Neutral	-1.715	-1.440	-1.440	-2.009
	Oxidized	5.244	4.757	6.280	4.950
	Zwitterionic	-7.538	-7.435	-7.092	-7.833
$P_{DrC} = +1$	Neutral	-2.009	-1.440	-1.440	-1.715
	Oxidized	4.950	4.757	6.280	5.244
	Zwitterionic	-7.833	-7.435	-7.092	-7.538

By repeating the analysis fixing $P_{DrC} = +1$ as shown in Fig. 3.6e, the results are perfectly opposite since the only thing changing is a mirroring of the charge distribution of the DrC cell. Fig. 3.6f,g,h show the energy map obtained varying the normalized dipole moments for neutral, oxidized, and zwitterionic molecules.

As the energy analysis demonstrates, vertical interaction is sensitive to the specific molecular species used to implement the system, which must be considered in simulations.

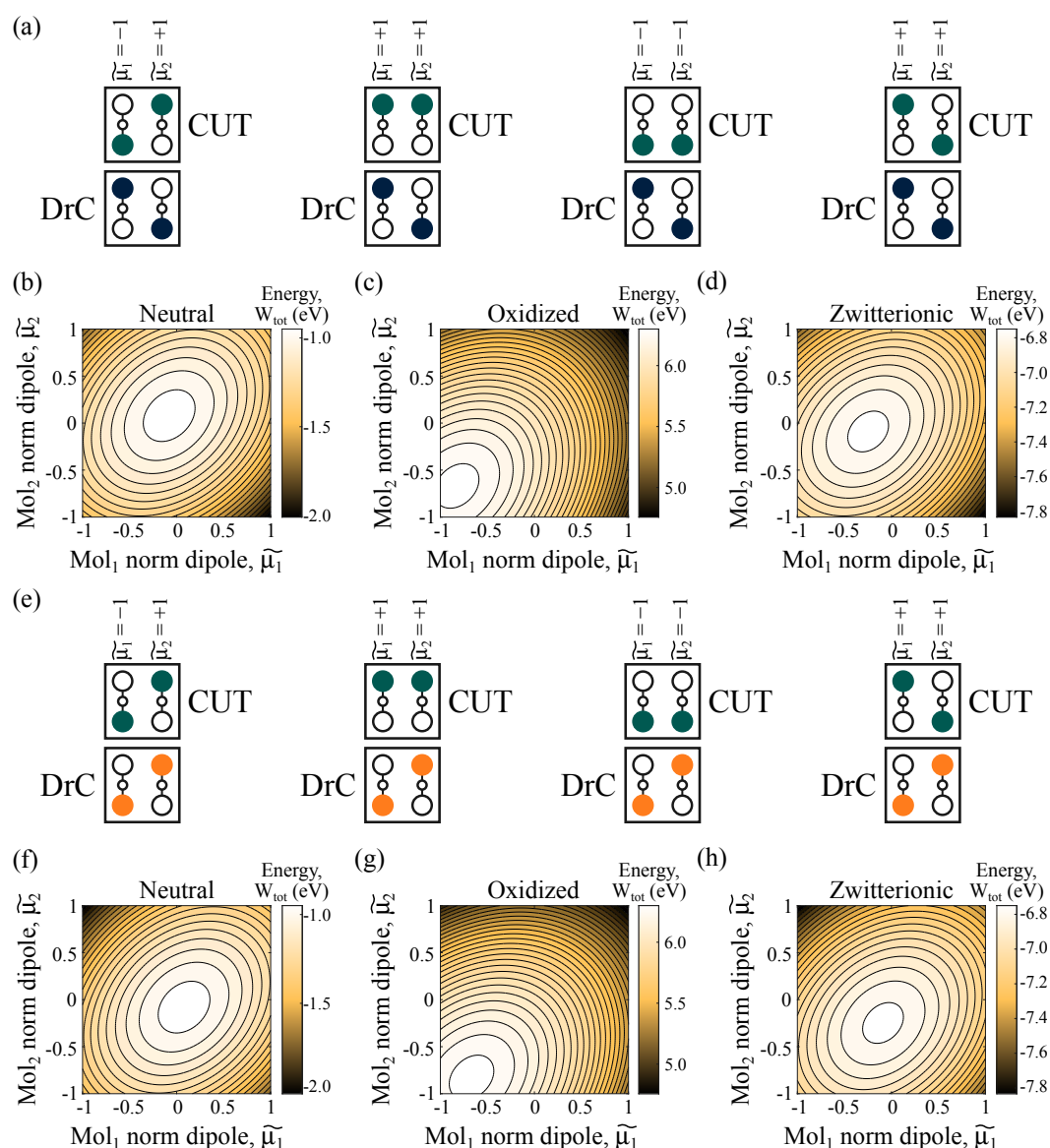


Fig. 3.6 Energy analysis for the vertical adjacent cell-to-cell interaction for the different molecular species. (a) The four limit configurations of the system when $P_{DrC} = -1$, showing the normalized dipole moments of the CUT molecules. (b) Energy of the two-cell system when $P_{DrC} = -1$ based on neutral molecules. Each line represents a 0.05 eV energy increment. (c) Energy of the two-cell system when $P_{DrC} = -1$ based on oxidized molecules. Each line represents a 0.05 eV energy increment. (d) Energy of the two-cell system when $P_{DrC} = -1$ based on zwitterionic molecules. Each line represents a 0.05 eV energy increment. (e) The four limit configurations of the system when $P_{DrC} = +1$, showing the normalized dipole moments of the CUT molecules. (f) Energy of the two-cell system when $P_{DrC} = +1$ based on neutral molecules. Each line represents a 0.05 eV energy increment. (g) Energy of the two-cell system when $P_{DrC} = +1$ based on oxidized molecules. Each line represents a 0.05 eV energy increment. (h) Energy of the two-cell system when $P_{DrC} = +1$ based on zwitterionic molecules. Each line represents a 0.05 eV energy increment.

Single-diagonal cell-to-cell interaction

Alongside adjacent interaction to copy information, diagonal interaction is fundamental for the inverting operation. Typical QCA layouts exploit only two cells arranged diagonally to invert information.

Fig. 3.7a illustrates the cell displacement and four limit charge configurations obtained by combining the two logic values. The theoretical expected behavior for the general QCA paradigm is that CUT encodes the opposite value of the DrC because of the interaction of the dots nearest to the touching angle. This behavior is obtained when both conditions expressed in equation (3.13) are satisfied.

$$W_{11} > W_{10} \quad W_{00} > W_{01} \quad (3.13)$$

We calculate the total energy of the system W_{XY} linearly varying the DrC and CUT polarizations for all the considered species. Table 3.3 lists the energy values obtained for the three species for clarity. Using neutral molecules, the energy of the system is negative when both cells encode opposite logic values, i.e., W_{01} and W_{10} . If cells have the wrong configuration, the system energy W_{XY} is positive, confirming that the minimum energy configurations are those exploiting correct inverting propagation, as expressed in equation (3.13). Fig. 3.7b shows a map of the system energy evaluated for all the polarization values between -1 and 1 , for both P_{DrC} and P_{CUT} . The two energy minima (black regions) are associated with opposite information, whereas the inverting configurations correspond to the energy maxima (white regions). From the values listed in Table 3.3 for neutral molecules, we notice that the system is symmetric from an electrostatic energy point of view ($W_{01} = W_{10} = -W_{11} = -W_{00}$). The symmetry of the values gives equal kink energies ($U_{k0} = U_{k1}$), meaning that the energy is independent of the input logic, which corresponds to a null kink error ($\Delta_n e = 0$). The diagonal interaction with neutral molecules is symmetric and makes possible the inversion operation with a single-branch inverter.

Fig. 3.7c shows the energy map calculated for the two-cell system implemented with oxidized molecules. The map shows only one global minimum, corresponding to the configuration in which $P_{DrC} = P_{CUT} = +1$. If the DrC encodes instead a logic zero ($P_{DrC} = -1$), the minimum energy configuration for CUT is again the one encoding a logic '1'. This behavior implies that the output of the device doesn't change by changing the input. From the values listed in Table 3.3 emerges an

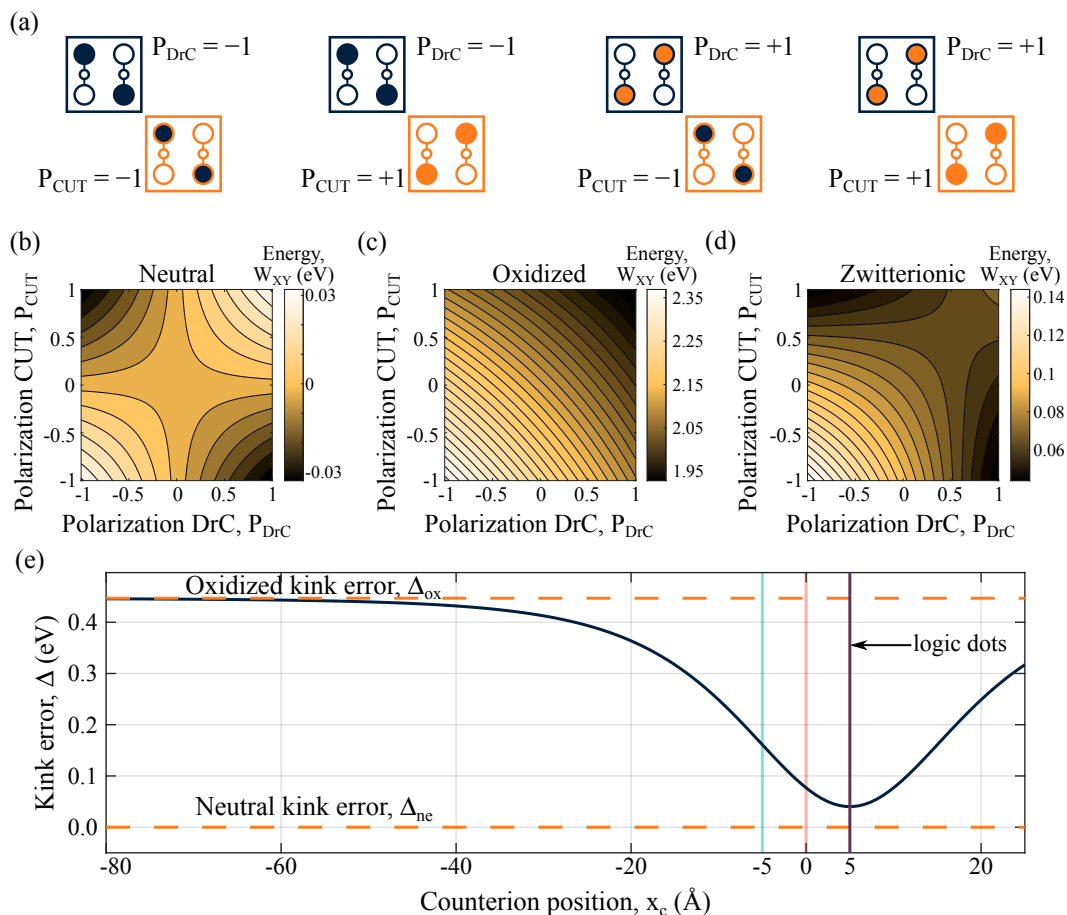


Fig. 3.7 Energy analysis for the single-diagonal cell-to-cell interaction for the different molecular species. (a) Charge distribution in the cell for the four combinations of logic values. (b) Energy of the two-cell system based on neutral molecules as a function of P_{DrC} and P_{CUT} . Each line represents a 0.005 eV energy increment. (c) Energy of the two-cell system based on oxidized molecules as a function of P_{DrC} and P_{CUT} . Each line represents a 0.015 eV energy increment. (d) Energy of the two-cell system based on zwitterionic molecules as a function of P_{DrC} and P_{CUT} . Each line represents a 0.005 eV energy increment. (e) Kink Error (Δ_{zw}) associated with the system composed of zwitterionic molecules as a function of the counterion distance from logic dots.

asymmetry of the electrostatic behavior, as $|W_{01}| \neq |W_{10}| \neq |W_{11}| \neq |W_{00}|$. In this case, the kink energies differ, and that's the reason why we must distinguish the kink energy for input encoding different logic values. When the system single-diagonal two-cell system is implemented with oxidized molecules, the interaction depends on the logic encoded by DrC, as confirmed by the non-null kink error $\Delta_{ox} = 0.477$ eV.

As we already said, zwitterionic molecules have an electrostatic behavior between neutral and oxidized molecules. This statement is confirmed by the energy map

Table 3.3 Electrostatic energy values of the system obtained for the three molecular species in the case of single-diagonal cell-to-cell interaction.

Species	W_{00} eV	W_{01} eV	W_{10} eV	W_{11} eV	U_{k0} eV	U_{k1} eV	Δ
Neutral	0.033	-0.033	-0.033	0.033	0.066	0.066	0.000
Oxidized	2.376	2.087	2.087	1.929	0.289	-0.158	0.447
Zwitterionic	0.148	0.044	0.044	0.071	0.104	0.027	0.077

shown in Fig. 3.7d, where two equivalent minima for the inverting conditions, thus satisfying equation (3.13). The two correctly positioned minima indicate that the interaction correctly inverts the information. However, electrostatic asymmetry is still present for the maxima in the device, yet the counterion is mitigating the effect of the localized positive charges, indeed $\Delta_{zw} < \Delta_{ox}$. The counterion is the key element that balances the electrostatic behavior, making the single-diagonal interaction work with zwitterionic molecules. For this reason, we analyze the value of the kink error Δ_{zw} varying the position of the counterion in the molecule (x_c). Fig. 3.7e shows the obtained curve. For all the values of x_c , the kink error stays in between the constant kink errors obtained for neutral and oxidized molecules, confirming the assumption of intermediate behavior. In the referenced system used, logic dots lay on the plane $x = 5 \text{ \AA}$, and in the simulation shown in Fig. 3.7d, the counterion position was $x_c = 0 \text{ \AA}$. If the counterion is far from the logic dots plane, the Δ_{zw} increases since the neutralizing electric field generated by the counterion reduces with the distance. The minimum kink error is indeed obtained when the counterion is placed on the same plane of logic dots ($x_c = 5 \text{ \AA}$), right in the middle of *Dot 1* and *Dot 2*. The analysis illustrates that single-diagonal inversion works only for well-engineered zwitterionic molecules, and again is fundamental to consider the specific electrostatic behavior to predict the outcome correctly.

Double-diagonal cell-to-cell interaction

The single-diagonal interaction is effective as logic inversion only for neutral and well-designed zwitterionic molecules. The interaction is non-symmetric for other species from an electrostatic point of view. The double-diagonal interaction helps to make the interaction electrostatically symmetric, favoring the success of the inversion and its stability.

Table 3.4 Electrostatic energy values of the system obtained for the three molecular species in the case of double-diagonal cell-to-cell interaction.

Species	W_{00} eV	W_{01} eV	W_{10} eV	W_{11} eV	U_{k0} eV	U_{k1} eV	Δ
Neutral	0.061	-0.070	-0.070	0.061	0.131	0.131	0.000
Oxidized	5.762	5.631	5.631	5.762	0.131	0.131	0.000
Zwitterionic	0.240	0.109	0.109	0.240	0.131	0.131	0.000

Fig. 3.8a shows the system in the four possible logic combinations. For the sake of simplicity, we force the polarization on the two DrCs to be the same. Then, by linearly varying P_{DrC} and P_{CUT} from ‘-1’ to ‘+1’, we calculate the system energy. Fig. 3.8b-d show the energy of the system when implemented with neutral, oxidized, and zwitterionic molecules, respectively. The three energy maps show the same profile, even if values are different for each species, that is, two minima located on the inverting values, satisfying the conditions expressed in equation (3.13). The energy values listed in Table 3.4 confirm that the system is symmetric from an energy standpoint for each species. The kink energy values U_{k0} and U_{k1} are equal, and the kink error is null for all the species, meaning that the behavior is independent of the input logic.

3.2 Electrostatics role in fundamental gates

The electrostatic analysis performed so far gives information regarding fundamental interactions and the behavior we can expect from devices that include those interactions. The simulation of circuits must consider the specific electrostatic footprint of the molecules used for the implementation since we demonstrated it impacts the final behavior of the system.

Currently, two tools are used to simulate molFCN circuits: Self-Consistent Electrostatic Potential Algorithm (SCERPA) and QCADesigner. The SCERPA algorithm treats each molecule as an electronic device with its primary state variable being the aggregated charge Q_i , determined by the molecule’s input voltage V_{in} . The input voltage for each molecule is determined by integrating the electric field produced by neighboring molecules along a path linking the two logic points. Consequently, the SCERPA algorithm iteratively assesses molecule input voltages and employs the

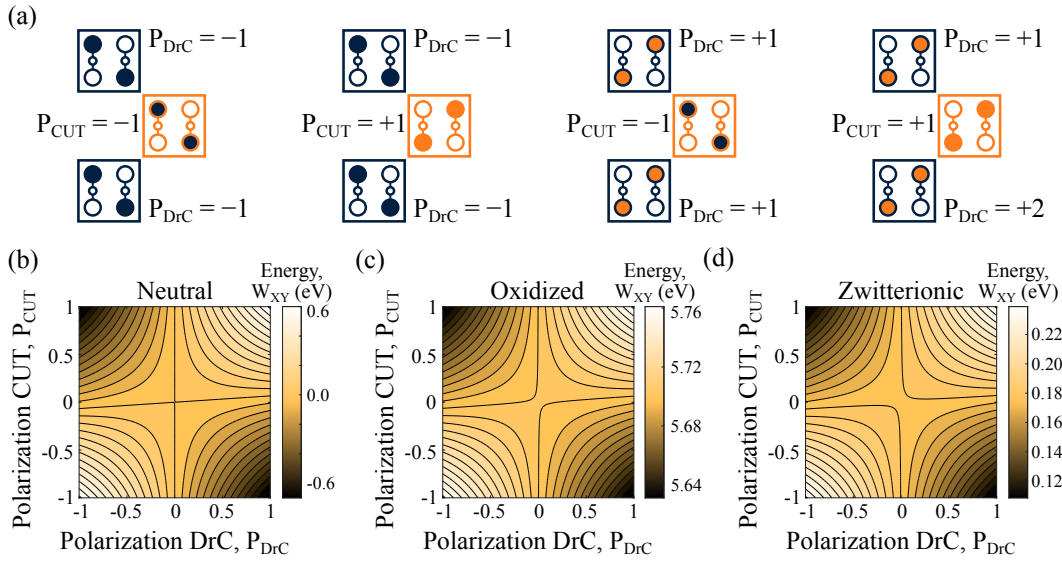


Fig. 3.8 Energy analysis for the double-diagonal cell-to-cell interaction for the different molecular species. (a) Charge distribution in the cell for the four combinations of logic values. (b) Energy of the system based on neutral molecules as a function of P_{DrC} and P_{CUT} . Each line represents a 0.005 eV energy increment. (c) Energy of the system based on oxidized molecules as a function of P_{DrC} and P_{CUT} . Each line represents a 0.005 eV energy increment. (d) Energy of the system based on zwitterionic molecules as a function of P_{DrC} and P_{CUT} . Each line represents a 0.005 eV energy increment.

VACT to compute the molecular charge [63, 64]. Since SCERPA uses the molecule VACT, it is implicit that it considers the electrostatics of the molecules.

QCADesigner is widely used in the literature to simulate QCA devices [59, 60]. This tool is intended for the general QCA paradigm and not specifically for molecular implementation. The QCADesigner tool mainly bases its calculations on kink energy, which is assumed to be input-independent. Thus it can correctly predict the behavior of circuits implemented with a few molecule types [125]. Since the energy analysis demonstrated that the kink energy, in some cases, depends on the encoded input, we perform the simulations using the SCERPA tool [130].¹

The following sections examine the SCERPA simulations of a few elementary gates to relate the molecular physics underlying fundamental interaction to the logical behavior of a circuit. This procedure underscores the significance of incorporating the effective behavior of molecules in simulation tools.

¹The tool is available online on GitHub at the following link <https://github.com/vlsi-nanocomputing/SCERPA/tree/v4.0.1>

All the circuits use an intermolecular distance of 1 nm and *Dot 3* of the zwitterionic molecule is fixed 5 Å far from the plane containing active dots, unless otherly specified.

3.2.1 Majority voter

The MV is a fundamental gate for designing logic since it provides AND and OR logic functions, as described in chapter 1. Fig. 3.9 shows the layout used for the electrostatic analysis, with the names of the three inputs and the output. All the cells belong to the same clock region. Since all the input branches have the same length, and the molecular width is equal to the intermolecular distance, the three input signals impact the output in the same way, so it does not need to be clocked independently to have a meaningful simulation.

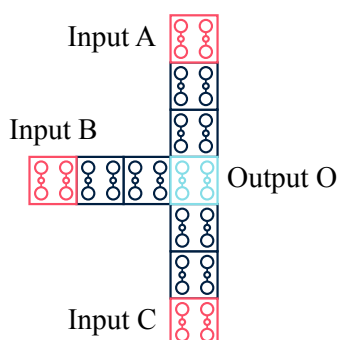


Fig. 3.9 Layout of the majority voter used for the electrostatic analysis and input/output definition. All the cells belong to the same clock region.

The MV exploits horizontal and vertical cell-to-cell interactions. The energy analyses on the two interactions suggest that the device should correctly work when implemented with neutral molecules. Fig. 3.10a shows, for the MV implemented with neutral molecules, the electrostatic potential evaluated with SCERPA on the plane 0.2 nm above the logic dot plane. White spots indicate aggregation of positive charges, whereas black dots are associated with negative charge aggregation. The figure shows all the possible eight input combinations, and the output is always correct, following the truth table commented in section 1.1.1.

Fig. 3.10b shows the results obtained with SCERPA for the gate implemented with oxidized molecules. The energy analysis shows that the vertical interaction is problematic since the charge on the CUT tends to maximize the distance from

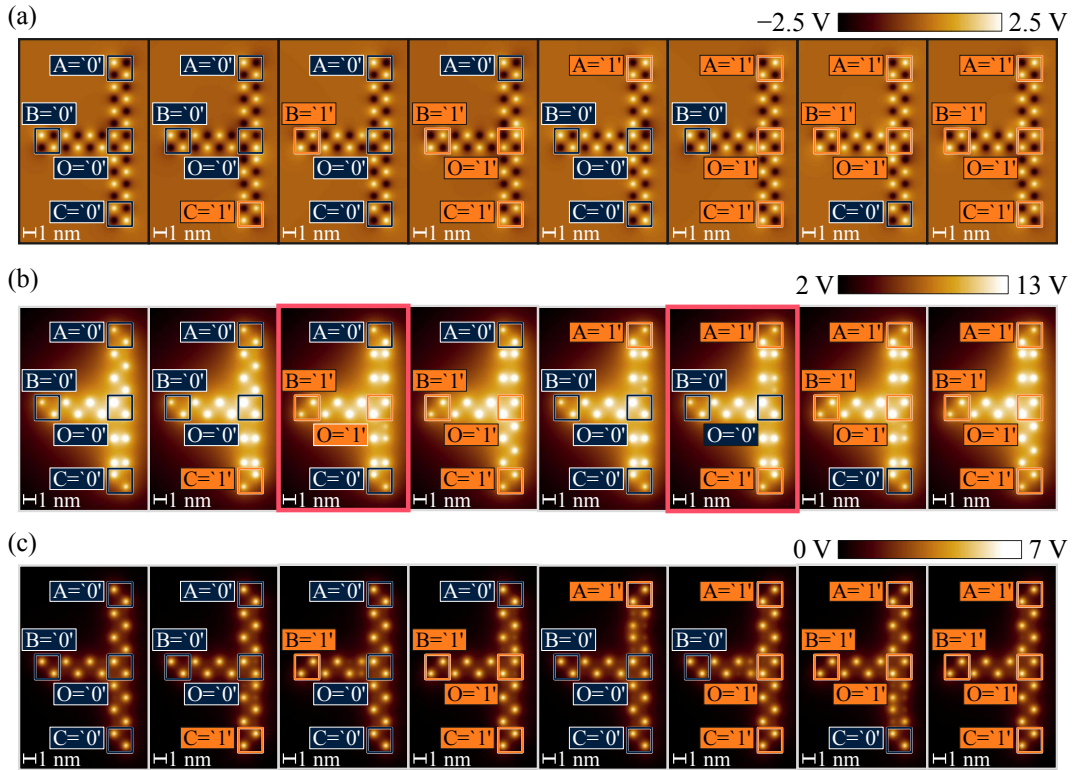


Fig. 3.10 Electrostatic potential obtained by the SCERPA simulation of a MV. The electrostatic potential is evaluated 0.2 nm above the logic dot plane, and the spots indicate the AC. (a) Simulation of the eight input combinations for a MV implemented with neutral molecules. (b) Simulation of the eight input combinations for a MV implemented with oxidized molecules. (c) Simulation of the eight input combinations for a MV implemented with zwitterionic molecules whose counterion position is $x_c = -5 \text{ \AA}$.

the DrC. As seen from the electrostatic potential distribution of the possible input combination, the same effect is found on the branches of *Input A* and *Input C*. In some cases, the incorrect information encoding on input branches does not affect the output correctness, like for the combinations $ABC = '000'$ and $ABC = '001'$. However, the outcome is incorrect in the two cases in which the output should match the information encoded on both the top and bottom inputs. These two cases, corresponding to combinations $ABC = '010'$ and $ABC = '101'$, are bounded in red in Fig. 3.10b. In both situations, the branch of *Input B* dominates the output cell, forcing the incorrect value.

In the case of a MV implemented with zwitterionic molecules, the electrostatic potential of the eight input combinations is shown in Fig. 3.10c. The simulation is performed using zwitterionic molecules with the counterion placed at $x_c = -5 \text{ \AA}$,

that is, 10 Å away from the logic dot plane. In the previous section, we demonstrate that the behavior of the zwitterionic molecules is in between those of neutral and oxidized, depending on the position of the counterion. In the simulated case, the counterion is not very close to the logic dot plane, but still, the behavior is correct for all the input combinations. The only disturbing effect can be noted for the combinations $ABC = '010'$, $ABC = '100'$, $ABC = '101'$, and $ABC = '110'$, where the output cell polarization back-influences the input branch encoding the opposite information. This effect is typically known as *back-propagation*.

3.2.2 Single-branch inverter

The single-branch inverter, whose layout is shown in Fig. 3.11a, has been used in general QCA to invert information. The same layout is often used for molFCN, disregarding the effective functioning of the gate. Since the key interaction in this gate is single-diagonal, we can expect the device to work for neutral molecules and well-engineered zwitterionic molecules.

Fig. 3.11b shows the electrostatic potential evaluated 0.2 nm above the logic dot plane for the simulation of the device implemented with neutral molecules. With both input logic values, information is elaborated correctly, being helped by the fact that one logic dot has a positive charge and a negative charge on the other. The variation of the sign of the charge between logic dots favors, in the same way, the opposite value on the cells near the diagonal. Indeed, focusing on the two nearest dots to the diagonal, we have on the top line a positive (negative) charge aggregation corresponding to a white (black) spot, attracting a negative (positive) charge on the bottom line indicated with a black (white) spot.

Only repulsive forces are involved when the device is implemented with oxidized molecules, so the expected charge configuration is quite different. Fig. 3.11c shows the electrostatic potential evaluated from the SCERPA simulation. The two input combinations produce different situations. When the input is $A = '0'$, the polarization on the output cell is correct. Focusing on the molecules nearest to the diagonal, the molecule on the top branch should theoretically have a positive charge aggregation on *Dot 2*, pushing the cells to invert the charge configuration, producing a correct outcome. Nevertheless, the cells on the bottom line are positively charged, which back-influences the molecules on the top line. This interaction results in the charge

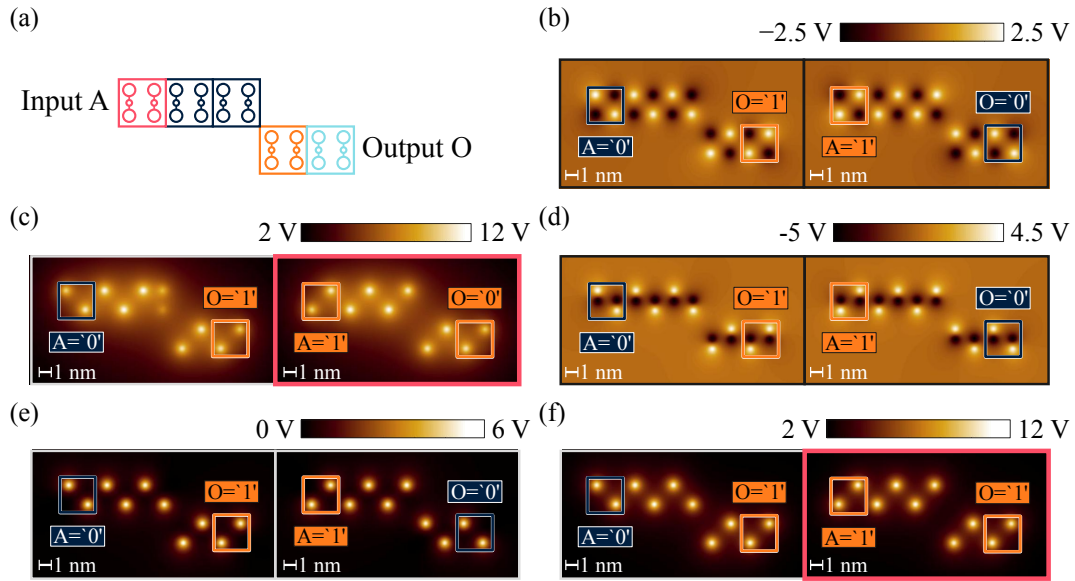


Fig. 3.11 Results obtained by the SCERPA simulation of a single-branch inverter. The electrostatic potential is evaluated 0.2 nm above the logic dot plane, and the spots indicate the AC. (a) Layout of the single-branch inverter used for the electrostatic analysis and input/output definition. The blue cells and input belong to the same clock region, whereas the orange cell and output belong to another. (b) Simulation of the two input combinations for the gate implemented with neutral molecules. (c) Simulation of the two input combinations for the gate implemented with oxidized molecules. (d) Simulation of the two input combinations for the gate implemented with zwitterionic molecules whose counterion position is $x_c = 5 \text{ \AA}$. (e) Simulation of the two input combinations for the gate implemented with zwitterionic molecules whose counterion position is $x_c = 0 \text{ \AA}$. (f) Simulation of the two input combinations for the gate implemented with zwitterionic molecules whose counterion position is $x_c = -5 \text{ \AA}$.

equally distributed on both dots for the last molecule of the top line. Contrarily, the only involved positive charges prevent the device from working when the input is $A = '1'$. In this case, the output still encodes a logic zero because the output configuration globally maximizes the distance of the charges nearest to the diagonal.

Implementation with zwitterionic molecules behaves in the middle of neutral and oxidized versions, and the outcome strictly depends on the position of the counterion that neutralizes the molecules. For this reason, we simulated the single-branch inverter using three different counterion positions, that are $x_c = 5 \text{ \AA}$, $x_c = 0 \text{ \AA}$, and $x_c = -5 \text{ \AA}$. For the reference system that we are using, these positions correspond to a counterion/logic dots distance of 0 \AA , 5 \AA , and 1 nm , respectively. The three values give different kink error values for the single-diagonal interaction and are

highlighted with three different colored lines in Fig. 3.7e. The expectation is that the closer the counterion to logic dots, the stronger the neutralization effect that helps the correct inversion.

Fig. 3.11d shows the electrostatic potential obtained with SCERPA when $x_c = 5 \text{ \AA}$. The presence of the counterion is clearly visible as a black spot between the logic dots. With this implementation, the device works perfectly for both input values, as the negative counterion helps pull the positive charge on *Dot 1* of the first molecule of the bottom line. Fig. 3.11e shows the electrostatic potential obtained with SCERPA when $x_c = 0 \text{ \AA}$. In this situation, the counterion is no longer visible from the plane where the potential is calculated, but its effect is still visible since the information is correctly inverted. Finally, Fig. 3.11f shows the result obtained when the counterion is 1 nm far from logic dots. The increased distance is reflected in the increased potential values measured 0.2 nm above the logic dot plane, which resembles that obtained for oxidized molecules. In this case, the presence of the counterion is insufficient to balance the electrostatic force generated by the positively charged logic dots, so the output is always $O = '1'$, independently from the logic encoded on the input cell.

3.2.3 Double-branch inverter

The single-branch inverter derived by the general QCA paradigm does not always work for molecular implementation and depends on the specific electrostatic of the molecule used. The problem has been addressed in the literature introducing the double-branch inverter layout, reported in Fig. 3.12. The double-branch inverter exploits two branches for the inversion, so the laying interaction is double-diagonal, which the energy analysis proved independent from the molecular species.

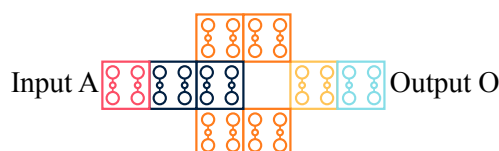


Fig. 3.12 Layout of the double-branch inverter used for the electrostatic analysis and input/output definition. Blue, orange, and yellow cells belong to different clock regions. Input and output cells belong to the same clock region of the nearest cell.

For a system made of neutral molecules, the single-branch inverter was enough to invert the information, so the outcome is correct even in the over-dimensioned double-branch inverter. Fig. 3.13a shows the electrostatic potential obtained with the SCERPA simulation for both input values.

According to the energy analysis, the system should work when the system is implemented with oxidized molecules. However, from the electrostatic potential map shown in Fig. 3.13b, we see that the gate is not inverting for both inputs. The cause is the repulsion that the molecules in the middle line induce on the top and bottom lines since the system is not globally neutral. Anyway, it is possible to simulate only the system of cells involved in the double-diagonal interaction to validate the principle. The system is analyzed in Fig. 3.8. In the simulation, the polarization of the two driver cells is fixed, so only the CUT can vary its polarization. Under this condition, Fig. 3.13c shows the electrostatic potential obtained with SCERPA for the three-cell system. Since the output cell is the only one that can vary, the simulation is coherent with the expectation derived from the energy analysis done in the previous section.

Finally, Fig. 3.13d shows the electrostatic potential for the system implemented with neutral molecules when $x_c = -5 \text{ \AA}$, that is the case making the molecule sufficiently similar to oxidized species. With this counterion position, the single-

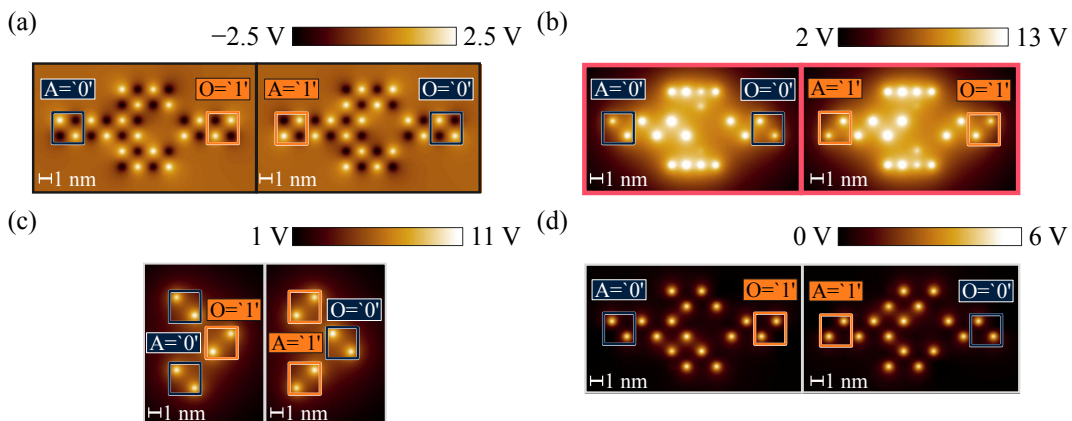


Fig. 3.13 Electrostatic potential obtained by the SCERPA simulation of a double-branch inverter. The electrostatic potential is evaluated 0.2 nm above the logic dot plane, and the spots indicate the AC. (a) Simulation of the two input combinations for the gate implemented with neutral molecules. (b) Simulation of the two input combinations for the gate implemented with oxidized molecules. (c) Simulation of the double-diagonal interaction implemented with oxidized molecules. (d) Simulation of the two input combinations for the gate implemented with zwitterionic molecules whose counterion position is $x_c = -5 \text{ \AA}$.

branch inverter did not work correctly, but in this case, the symmetry given by the layout geometry is helping the system invert the information. As predicted by the energy analysis, the double-branch inverter is insensible to the molecular species used since the electrostatic symmetry is reached through the geometrical symmetry of the layout.

3.2.4 Three-way splitter (fanout circuit)

An interesting device to analyze involving only horizontal and vertical interactions is the three-way splitter, generally called *fanout circuit*, shown in Fig. 3.14. This device has a structure similar to the MV but has one input and three outputs, meaning that the liberty of the electrostatic interaction changes, especially on the top and bottom outputs.

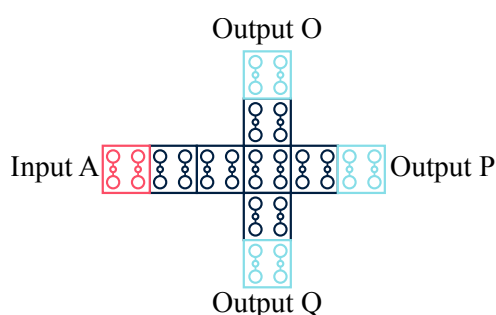


Fig. 3.14 Layout of the three-way splitter (also known as fanout circuit) used for the electrostatic analysis and input/output definition. All cells belong to the same clock region.

Even in this case, we simulated the device with SCERPA considering the various molecular species. Fig. 3.15a shows the electrostatic potential map for the device implemented with neutral molecules. For both inputs, the information is correctly replicated on the three outputs, as could have been expected from the energy analysis. On the other hand, when the device is implemented with oxidized molecules, only the central output *P* propagates. Fig. 3.15b shows the electrostatic potential evaluated 0.2 nm above the logic dots plane, and the behavior is analogous to the one of the MV and the double-branch inverter. Indeed, the charge of the molecules of the line corresponding to outputs *O* and *Q* maximizes the distance from the center, producing an undefined logic value, having the two molecules with concordant dipole moments.

For zwitterionic molecules, we again expect a behavior between the two just presented, depending on the position of the counterion. Fig. 3.15c shows the case

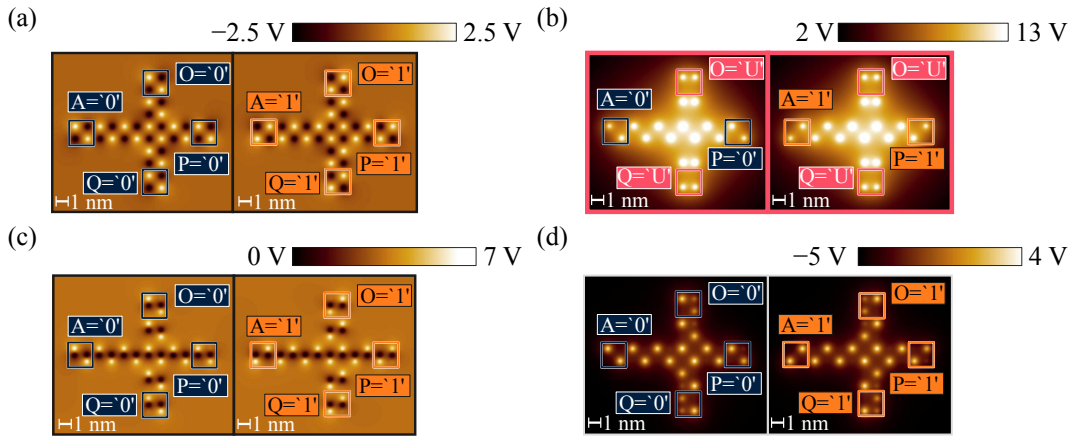


Fig. 3.15 Electrostatic potential obtained by the SCERPA simulation of a three-way splitter. The electrostatic potential is evaluated 0.2 nm above the logic dot plane, and the spots indicate the AC. (a) Simulation of the two input combinations for the gate implemented with neutral molecules. (b) Simulation of the two input combinations for the gate implemented with oxidized molecules. (c) Simulation of the two input combinations for the gate implemented with zwitterionic molecules whose counterion position is $x_c = 5 \text{ \AA}$. (d) Simulation of the two input combinations for the gate implemented with zwitterionic molecules whose counterion position is $x_c = -5 \text{ \AA}$.

more similar to neutral molecules with the counterion in the same plane of logic dots ($x_c = 5 \text{ \AA}$). As for neutral molecules, the polarization of all the cells on the output lines encodes the correct information independently of the logic input. The situation with the fanout circuit implemented with zwitterionic molecules with $x_c = -5 \text{ \AA}$ shown in Fig. 3.15d is different. The behavior in this case resembles more the one of oxidized molecules that, for instance, prevent the correct functioning of the single-branch inverter. In this case, the circuit works, but the disturbance of the central line is still visible. The information encoded on outputs O and Q is far from a perfect diagonal ideal configuration. The analysis reveals that the output cells display a notable charge distribution across three out of the four logic dots within the molecular cell. This observation suggests that one of the molecules may not be fully polarized. Furthermore, these border effects vary depending on the input logic value and directly result from the vertical interactions exerted within the system.

3.3 Crosstalk effect

The situation occurring with oxidized molecules in the devices shown before is the consequence of the so-called *crosstalk field*, which affects the device internally [64]. The crosstalk field can influence undesired parts of the considered circuit or nearby devices. As demonstrated in the previous section, in the context of intra-device crosstalk, the electrostatic properties of individual molecules play a pivotal role.

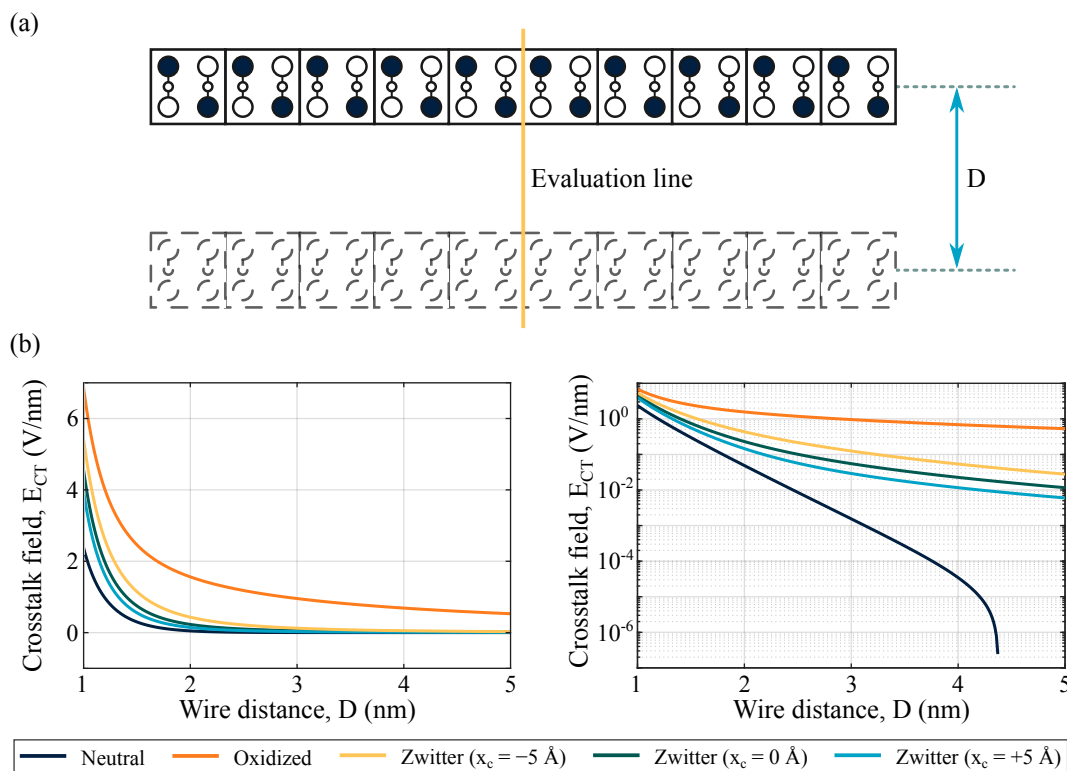


Fig. 3.16 Analysis of crosstalk field E_{CT} . (a) The setup employed for crosstalk analysis highlighting the evaluation line orthogonal to a fully-polarized molecular wire on which E_{CT} is computed. The position of a hypothetical molecular wire is at a distance D measured center-to-center. (b) A comparison of the generated crosstalk field based on the distance D , utilizing various molecular species. The graphs on the right represent the same quantities on a logarithmic scale to appreciate the different behaviors.

This section delves into assessing the impact of the crosstalk field between devices using a fully polarized molecular wire. Fig. 3.16a introduces the setup used to evaluate this field, which occurs at the midpoint of a 20-molecule wire (evaluation line) at the logic dots plane. Subsequently, this calculated field interacts with a hypothetical wire positioned at a distance D , potentially influencing the polarization

of the wire molecules. Fig. 3.16b illustrates the electric field E_{CT} generated by a molecular wire along the orthogonal axis for the various molecular species.

As Fig. 3.16b depicts, oxidized molecules yield a higher crosstalk field E_{CT} , which aligns with expectations considering that an oxidized molecule wire lacks neutrality. Consequently, adhering to Gauss's law, it generates a non-zero field that diminishes with distance. For instance, considering the double-branch inverter described before, the center-to-center distance between the top and bottom lines is $D = 2$ nm. According to Fig. 3.16b, the crosstalk field generated by the polarized wire implemented with oxidized molecules is $E_{CT} = 1.562$ V nm⁻¹, that generates a noise input voltage comparable to those typically found in molFCN devices, thus justifying the incorrect behavior.

On the contrary, when it comes to neutral species, the absence of a positive/negative net charge significantly reduces the electric field generated by the wire, rendering neutral molecules the optimal choice for mitigating crosstalk.

In the case of zwitterionic molecules, the positioning of the counterion assumes paramount importance. While zwitterionic species maintain overall neutrality, the field E_{CT} is notably reduced compared to the oxidized case. Fig. 3.16b further illustrates the crosstalk field generated by zwitterionic species across three possible counterion positions. Notably, when the counterion is aligned with the logic-dot level ($x_c = 5$ Å), the proximity between the positive charge of the logic dots and the counterion's negative charge minimizes the generated net electric field. Conversely, altering the distance between the logical dots and the counterion, as seen in cases $x_c = 0$ Å and $x_c = -5$ Å, amplifies the net electric field at the logic dots level, thereby resembling the behavior observed in oxidized species.

As molecule dimensions define the molecular safe-operating area and intermolecular distance defines the technological safe-operating area, the electrostatic occupation of molecules and devices defines the Device Safe-Operating Area (DSOA). The DSOA measures the electrostatic occupation of a gate, delimiting the region where it is not safe to place another device. The main factor influencing it is the crosstalk field, and we have already seen that not respecting it induces an error in the device and nearby gates.

3.4 Design of a coplanar cross-wire interconnection

Unlike other Field-Coupled Nanocomputing (FCN) implementations, up to now, a prototype or a clear technological direction for fabrication is missing, as discussed in Chapter 1. Considering the few current fabrication concepts, molFCN is considered an in-plane technology. Hence, it is impossible to exploit the vertical direction for signal routing or 3D-stacked systems. This comes with the consequence that signal routing must be performed in the same plane as logic functions. If we imagine a realistic system made of several interconnected gates, it is reasonable to expect that routing would require significant space on the circuit.

The QCA paradigm has proposed two solutions for coplanar signal crossing. The first solution involves one of the two lines having 45° -rotated cells [131]. Fig. 3.17a illustrates the charge distribution for the four possible input combinations. However, this solution has proven to be delicate due to the assumption that horizontal and vertical wires do not interact strongly. From an electrostatic standpoint, this holds true within a minimal range and a slight asymmetry in the system can favor one line over the other [131]. The second issue with this solution is associated with the technological requirement of precisely rotating a few molecules. This necessitates the ability to control the lattice orientation within a small region of a few nanometers so that molecules can self-assemble at different angles.

The second solution proposed for general QCA exploits the diagonal interaction [132]. The system is represented in Fig. 3.17b for the four possible combinations. We must distinguish two situations to understand the behavior. First, if the inputs encode the same logic values, then the central computing cell and the output will all copy the same logic value. If the two inputs encode opposite logic values, the central computing cell is subjected to equal and opposite fields. In an ideal system, the opposite influence leads the central cell not to assume any logic value. Hence, the charge is equally distributed among the four logic dots. If the central computing cell does not assume a logic polarization, the dominating interactions are the two single-diagonals composed of the input and the nearest output cells. Hence, information crossing is performed through a couple of contemporary single-branch inverters in this case.

The second solution is based on two assumptions. First, the system must be perfectly symmetric from an electrostatic point of view rather than geometrically.

Secondly, the single-branch inverter must work correctly. The analysis performed in chapter 3 on the fundamental interactions demonstrated that the two assumptions are guaranteed for neutral molecules and well-engineered zwitterionic molecules.

The following sections discuss various implementations of the coplanar crossover, considering the precise physics of molecules to forecast the functionality of the interconnection [133]. Moreover, we modify the electrostatic characteristics of the involved molecules to assess the resilience of the cross-wire against static electrostatic variations, thereby offering valuable insights for the tailored synthesis of molecules for molFCN applications.

3.4.1 Coplanar crossover simulation

The device behavior depends on the molecule used to implement it, so we analyze it using neutral and zwitterionic molecules, considering the same VACT used in chapter 3 and reported in equation (3.2) and equation (3.4).

We first consider neutral molecules, and the reference structure of the analyzed crosswise is shown in Fig. 3.18. By simulating it with SCERPA, we obtain a correct

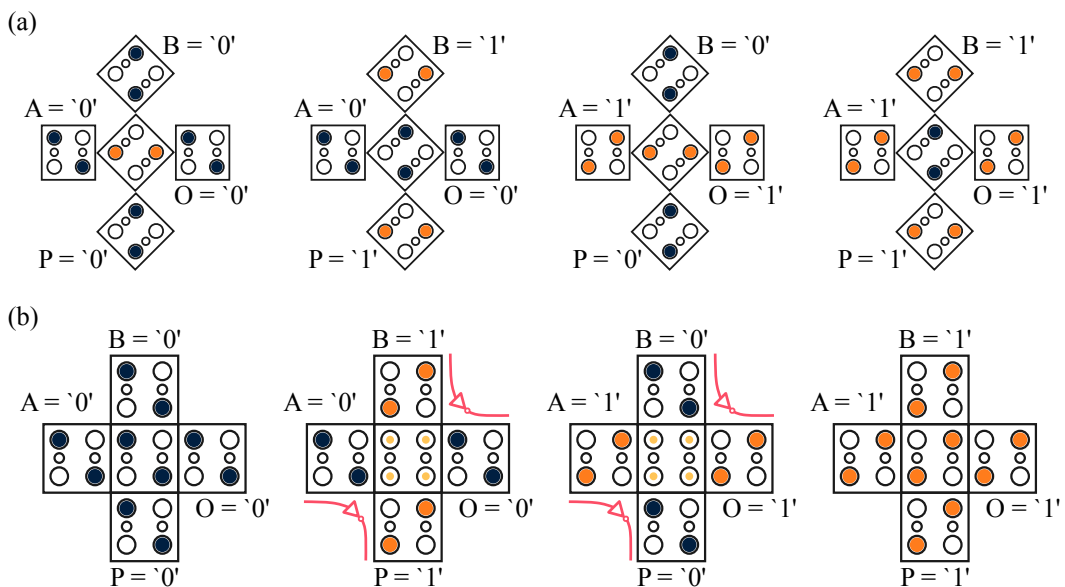


Fig. 3.17 Solutions for coplanar information crossing in the QCA paradigm. (a) Charge configuration for the four possible input combinations in the crossover implemented with rotated cells. (b) Charge configuration for the four possible input combinations in the crossover exploiting diagonal interaction.

propagation of the two logic inputs only if the output wires are in the same clock region of the intersecting computing cell. In Fig. 3.18, the clock zones are identified with different cell colors, so the cells on the input lines are blue-colored. In contrast, the intersecting cell and the cells propagating toward the outputs are orange-colored. For the described layout, Fig. 3.19 shows the electrostatic potential evaluated 0.2 \AA above the logic dot plane in the final step of the four input combinations. From the simulations emerges that information is correctly cross-propagated in all cases.

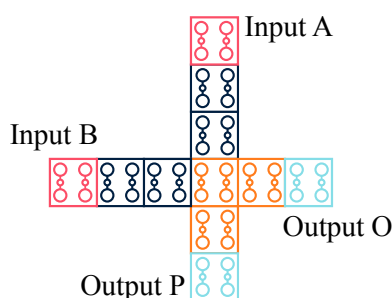


Fig. 3.18 Layout of the coplanar crossover used for the simulation in SCERPA. Blue and orange cells belong to different clock regions. Input and output cells belong to the same clock region of the nearest cell.

From the theoretical behavior, we expect the central cell not to encode any logic value if $A \neq B$. However, from the simulation of those two cases, it is noticeable that the central shared cell retains a particular logic value despite achieving the correct final configurations. This phenomenon remains consistent with the anticipated behavior. Initially, as the clock field initiates the switching process in the output cells, the central cell remains void of encoded information. Consequently, the output branches adhere to the inverting interaction originating from the diagonals. However, towards the end of the propagation, the input voltages on the central molecules cease to be null, resulting in the central cell adopting a logic value as well.

If we want to understand how sensitive the device functioning is to electrostatic variations, we have to modify the VACT. Electrostatic variations can map different molecules themselves or eventual noise affecting the electrostatic behavior of a molecule. Indeed, if some random noise affects the charge distribution on a molecule, we would have a different charge profile for the same input voltage, which is like having a different VACT. Changing the VACT of all the molecules in the circuit is a simplification since noise should be random on the molecules rather than constant. Still, the analysis provides valuable information regarding the device resilience to minor static electrostatic variations.

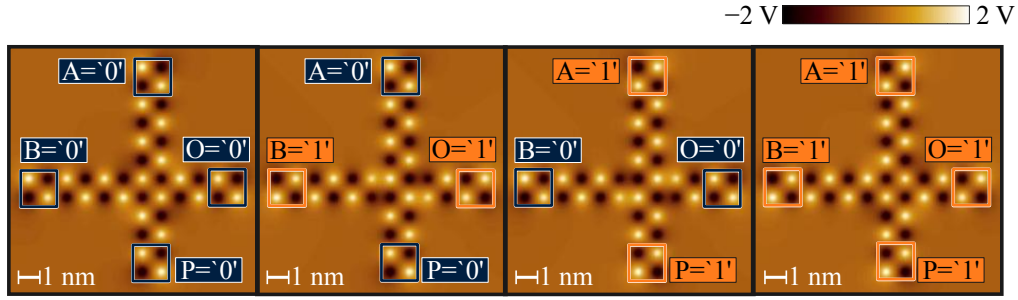


Fig. 3.19 Simulation using SCERPA of a crosswire interconnection utilizing neutral molecules: the four images depict the electrostatic potential for the four input combinations at the final propagation step, evaluated 0.2 nm above the logic dot plane.

The variations we apply are rigid horizontal shifts of the neutral VACT and variations of the steepness. To implement those modifications, we can rewrite the starting equation (3.1) for the charge and write it in a general form, that is:

$$Q_i = \frac{1}{1 + e^{w \cdot (V_{in} + s)}} \quad (3.14)$$

where w acts on the curve steepness and s acts on the rigid shift of the crossing point between AC_{Dot1} and AC_{Dot2} .

Varying the values of w and s , we can obtain the working range for the implementation using neutral molecules. Fixing the value of the weight $w + 10 \text{ V}^{-1}$, the range of allowed shifts to continue having a working interconnection is

$$-30 \text{ mV} \leq s \leq 30 \text{ mV} \quad (3.15)$$

Fig. 3.20a illustrates the shift range on the VACT acceptable to maintain a functional interconnection. If the rigid shift exceeds $|\pm 30 \text{ mV}|$, at least one configuration propagates an incorrect logic value. This result implies that the charge difference at equilibrium ($V_{in} = 0 \text{ V}$) between *Dot 1* and *Dot 2* must be less than $\Delta AC_{MAX} = 0.15 \text{ a.u.}$ when $w = 10 \text{ V}^{-1}$.

Similarly, we can fix the shift $s = 0 \text{ V}$ and vary the steepness of the slope, which is described by w . The working range for the crossover implemented with neutral molecules is $w \in [9; 12] \text{ V}^{-1}$. Fig. 3.20b illustrates the VACT obtained for the two boundary values $w = 9 \text{ V}^{-1}$ and $w = 12 \text{ V}^{-1}$. Besides the neutral molecule being the best-behaving and the device correctly working, small electrostatic variations could generate errors in the propagation.

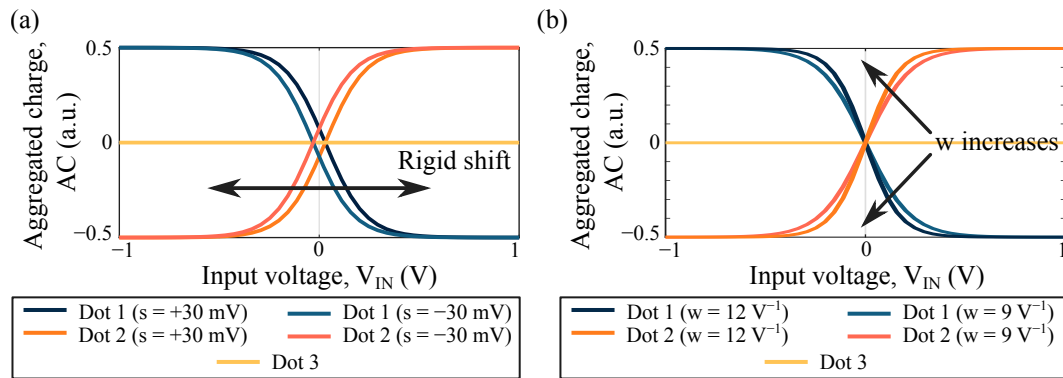


Fig. 3.20 Ranges of VACT variations to maintain functionality in the crosswire interconnection implemented with neutral molecules. (a) Application of rigid shifts to the VACT, with maximum allowable shifts of ± 30 mV. (b) Alteration of the slope of the VACT, where w in equation (3.2) can vary within the interval $[9; 12]$ V⁻¹.

Recently, the literature has demonstrated how single-line devices can be susceptible to weak charge separation in some cases [64]. For instance, the device is susceptible if the molecule used has a bistable factor near one. Hence, information is hardly bistable when there are no other molecules after the output.

Multi-line devices have been proposed to address this problem and strengthen information stability. In the case of a crossover implemented using diagonal interaction, one assumption is null charge separation on the molecule composing the intersecting cell. The need for a weak charge separation for the crosswire could contrast with the need to strengthen the information elsewhere by employing two-line devices. Consequently, we investigate the behavior of a crossover device implemented following the multi-line approach. Fig. 3.21 depicts the proposed layout.

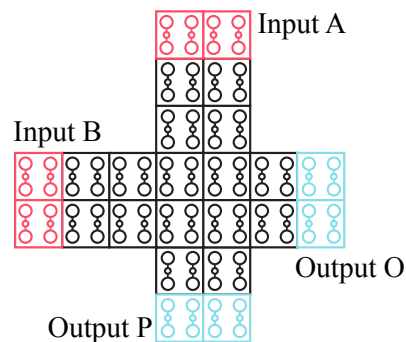


Fig. 3.21 Layout of the multi-line coplanar crossover used for the simulation in SCERPA.

It comes up that the crossing mechanism fails to operate effectively for the multi-line layout implemented with neutral molecules, with any clock distribution. Specifically, in scenarios where A and B differ, the presence of multiple molecules in the central region leads to mutual influence among them. Consequently, these molecules encode distinct logic values, thereby hindering the proper functioning of the interconnection. This underscores the complexity introduced by the collective behavior of molecules in this system, posing challenges to its smooth operation across varied conditions.

In previous sections, zwitterionic molecules showed various behaviors, more or less similar to neutral or oxidized molecules, depending on the counterion position. In this case, the single-line crossover does not work with the considered VACT, independently from the clock field distribution or the counterion position. Fig. 3.22 shows the result of the SCERPA simulation when the crosswire is implemented with zwitterionic molecules whose third dot is on the same plane of logic dots ($x_c = 5 \text{ \AA}$). Even if we demonstrated that this configuration is the most similar to neutral molecules, the information does not correctly cross when inputs encode opposite logic values.

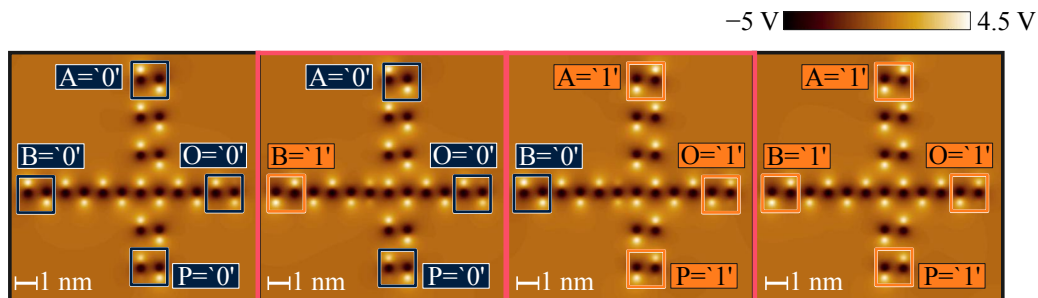


Fig. 3.22 Simulation using SCERPA of a crosswire interconnection utilizing zwitterionic molecules with $x_c = 5 \text{ \AA}$: the four images depict the electrostatic potential for the four input combinations at the final propagation step, evaluated 0.2 nm above the logic dot plane.

While the single-diagonal interaction might manage to invert the logic signal under certain conditions, it's worth noting that the adjacent interaction holds greater energetic favorability, as discussed in previous sections. In this scenario, altering the slope of the VACT or introducing a shift proves ineffective in aiding the propagation. Additionally, simply adjusting the geometrical position of *Dot 3* to favor the diagonal interaction does not affect this setup, unlike the case of the single-branch inverter illustrated in section 3.2.2.

3.4.2 Multi-molecule coplanar crossover

Having valid alternatives for crossing information is crucial for the development of the technology. Without the possibility to perform this operation, it's basically impossible to obtain circuits larger than a few elementary gates.

One possibility is to use tailored molecules behaving in a particular way, enabling the attainment of the target functionality required. The combination of multiple types of molecules has been utilized to achieve the behavior of a three-input neuron using the majority voter layout [124]. In that example, different molecular characteristics serve to implement weights and the activation function in the neuron.

In this case, to determine how molecules should behave, we should start by analyzing the expected functioning of the cells in the crosswire. The crucial behavior is dictated by the interconnecting cell in the middle of the two branches. The central cell must assume two distinct configurations depending on the input signals.

When both inputs are equal, the charge on the logic dots must be well localized and coherent with the logic inputs. Since the two inputs encode the same logic, the global electrostatic influence on the central region favors the charge to stay either on *Dot 1* or *Dot 2*, depending on the logic value. Considering typical VACTs, including those depicted in Fig. 3.1 and used throughout this chapter, when the charge is localized on only one logic dot, we are in the saturation region.

In the case where the two inputs are opposite, theoretically, the electrostatic influence of one input is counterbalanced by the influence of the other input. The two contributions ideally cancel each other out, giving a total null input voltage on the molecules composing the central cell. However, the system composed of the input cells and the central interconnecting cell is not geometrically symmetric, or electrostatically symmetric if we use zwitterionic molecules. This asymmetry leads to small non-null input voltages on the central cell, and is the reason behind the fact that practically the crossover doesn't work for zwitterionic molecules or multi-line devices.

Since the problem lies in the behavior for small voltages, we can anticipate the system to function if the involved molecules exhibit a sort of indecision region. That is, for small voltages, the charge remains constant on logic dots, preventing the molecule from polarizing. This behavior is observed in molecules such as 1,5,9-decatriene cation [62, 51] or in the 1,3,5benzenetrithiol linked bis(hexanedithiol)

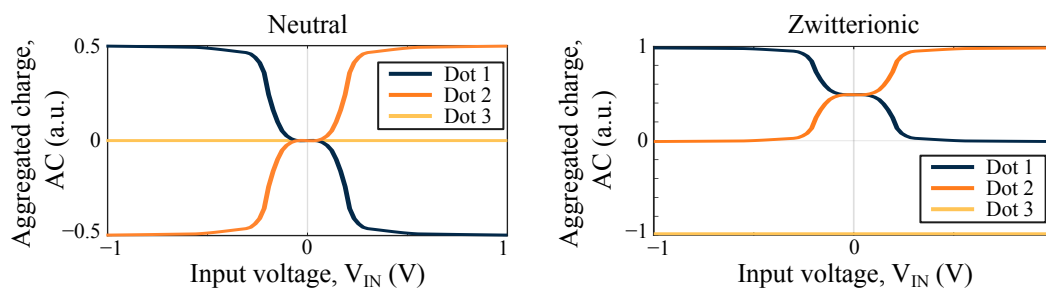


Fig. 3.23 VACT of the neutral and zwitterionic molecules used to favor the crossing mechanism, synthesized inspired by the behavior of the 1,5,9-decatriene cation and the 1,3,5benzenetrithiol linked bis(hexanedithiol) bis(trimethylsilane).

bis(trimethylsilane) presented in section 2.4.4. In both cases, small voltage variations do not affect the equilibrium charge distribution, meaning that the charge is equally distributed among all logic dots, and the molecular cell does not assume a specific logic value.

For the sake of coherence of the work presented here, rather than directly utilizing the VACT of the mentioned molecules, we synthesize the desired behavior on an artificial reference molecule. Fig. 3.23 illustrates the VACT designed for neutral and zwitterionic molecules to reinforce the expected behavior for the crosswire. The geometrical reference system for these molecules is the same as used throughout this chapter and reported in Fig. 3.1a. Since the discussed behavior is expected only from the central cell, the modified molecules should be assigned only to the central cell, creating a device composed of multiple molecular cell types [124].

Single-line multi-molecule coplanar crosswire

The presence of the *ad hoc* designed molecules alters the behavior of the crossover. Fig. 3.24 illustrates the layout used in this case. Unlike the solution that utilized a single molecule type, here the system functions even with the central cell clocked independently, a symptom of the increased robustness of the multi-molecule device. In this layout, the central interconnecting cell is composed of two molecules with a VACT inspired by decatriene, featuring a central plateau. The remaining cells, as well as driver cells, are composed of molecules with the classical VACT used for neutral molecules and reported in equation (3.2).

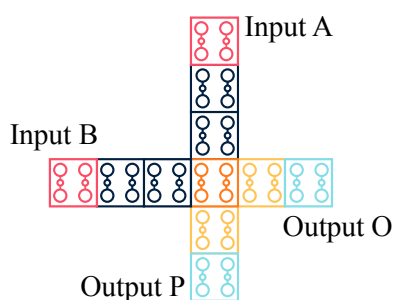


Fig. 3.24 Layout of the coplanar crossover used for the simulation in SCERPA. Blue, orange, and yellow cells belong to different clock regions. Input and output cells belong to the same clock region as the nearest cell.

Fig. 3.25 shows the electrostatic potential calculated with SCERPA for the implementation based on neutral molecules. Since the device is now divided into three clock regions, a single simulation step where both input and output branches are active is missing. While output cells are switching to the *hold* phase, the input cells are transitioning to the *reset* phase. For this reason, to follow information propagation, we need to observe two subsequent clock steps. In Fig. 3.25, the first clock step s is represented on the top line for the four possible input combinations, and the following clock step $s + 1$ is reported in the bottom part of the image.

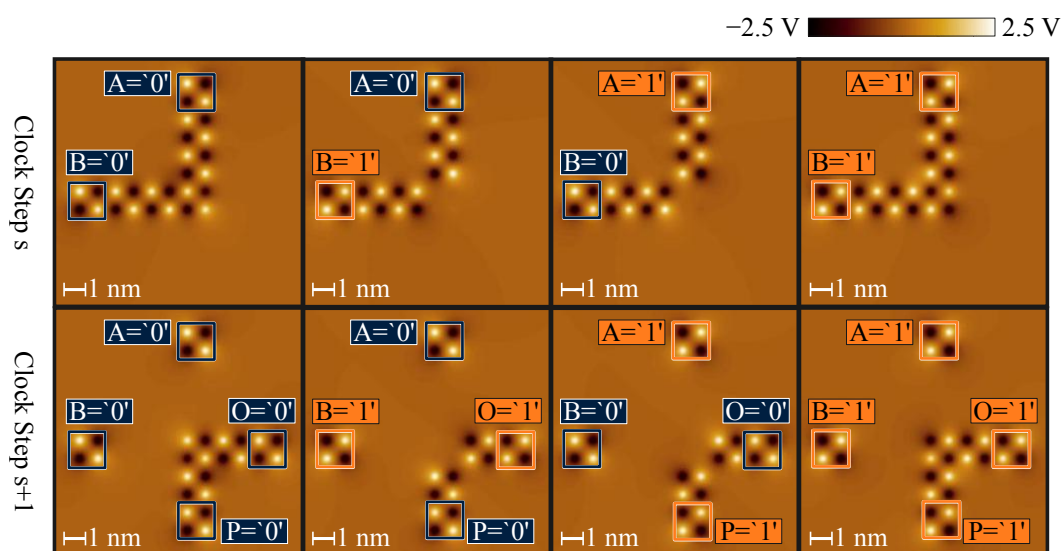


Fig. 3.25 Simulation using SCERPA of a single-line multi-molecule crosswire interconnection based on neutral molecules: the images depict the electrostatic potential for the four input combinations evaluated 0.2 nm above the logic dot plane. The top and bottom lines of the images represent subsequent clock steps.

From the electrostatic potential map, we can observe that when $A = B$, the interconnecting cell encodes the same logic value as them, as for the single-molecule implementation. The most significant difference occurs when $A \neq B$. In that situation, the effects of the two input branches cancel out, resulting in the central cell proposing an equal charge distribution on the four logic dots. Since the charge is almost zero on the logic dots of the interconnecting cell, their presence cannot be discerned from the potential map.

More interestingly, we consider the implementation with zwitterionic molecules. We start from the simplest case from an electrostatic point of view, using the layout shown in Fig. 3.24 implemented with the zwitterionic molecules having *Dot 3* on the same plane as the logic dots on both input and output branches. The interconnecting cell is composed of molecules with the *ad hoc* synthesized VACT presented in Fig. 3.23. For the decatriene-inspired molecules, the position of *Dot 3* is fixed at $x_c = 0 \text{ \AA}$, which is 5 \AA away from the logic dots.

Fig. 3.26a shows the electrostatic potential calculated with SCERPA. The considerations regarding the clock steps are the same as those made previously for the neutral multi-molecule implementation. The electrostatic potential map highlights the correct behavior for all input combinations and the behavior of the interconnecting cell. Again, when inputs are equal, the central cell clearly polarizes according to the logic value encoded by the drivers. Besides the proper behavior, when $AB = '11'$, the central cell presents a slight internal crosstalk phenomenon, evidenced by the distribution on three out of four logic dots, similar to the case of the fanout circuit (Section 3.2.4). When inputs encode opposite logic values, the electrostatic profile clearly highlights the evenly distributed charge among the four logic dots of the interconnecting cell. The evenly distributed charge lets the diagonal interaction dominate on the output branch, ensuring that the correct value is propagated on both outputs O and P .

The situation remains very similar if we move *Dot 3* to $x_c = 0 \text{ \AA}$, even for the molecules on the input and output branches. Fig. 3.26b shows the electrostatic potential calculated with SCERPA. When $A = B$, the considerations are the same as before: the central cell polarizes according to the input values, and a slight presence of crosstalk is evident when $AB = '11'$. The reason for the presence of crosstalk only for that configuration is the asymmetry of the system with branches connected as an ' L' ' or a ' Γ' '. Since with zwitterionic molecules the charge on logic dots is positive, it

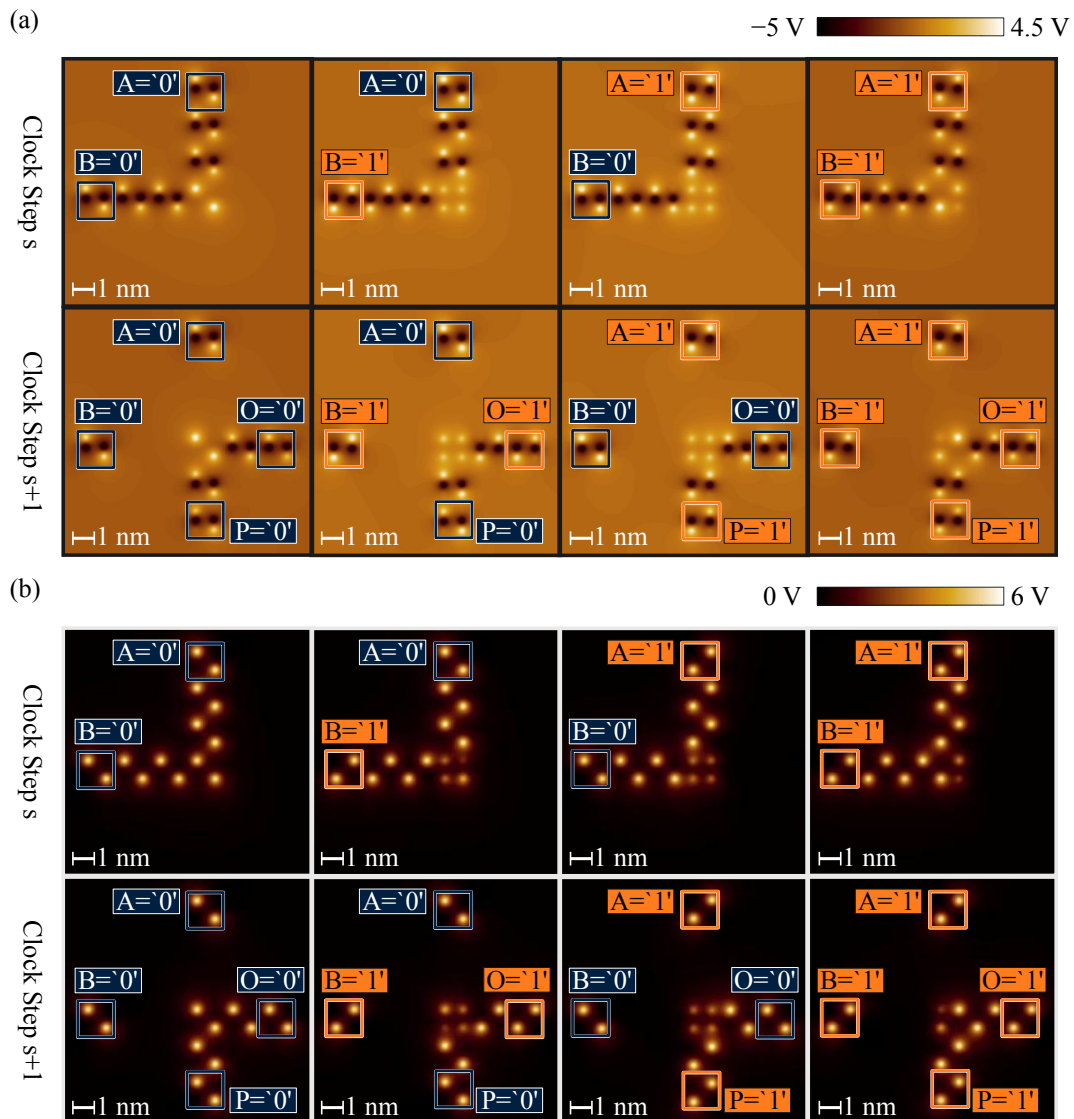


Fig. 3.26 Simulation using SCERPA of a single-line multi-molecule crosswire interconnection based on zwitterionic molecules: the images depict the electrostatic potential for the four input combinations evaluated 0.2 nm above the logic dot plane. The top and bottom lines of the images represent subsequent clock steps. (a) *Dot 3* is on the same plane as logic dots ($x_c = 5 \text{ \AA}$). (b) The position of *Dot 3* is $x_c = 0 \text{ \AA}$.

tends to maximize the distance by occupying *Dot 2* of the right molecule in the first clock step, and *Dot 1* of the left molecule in the second clock step, which are the farthest positions for the charge. For configurations with $A \neq B$, the charge is evenly distributed, favoring diagonal interactions.

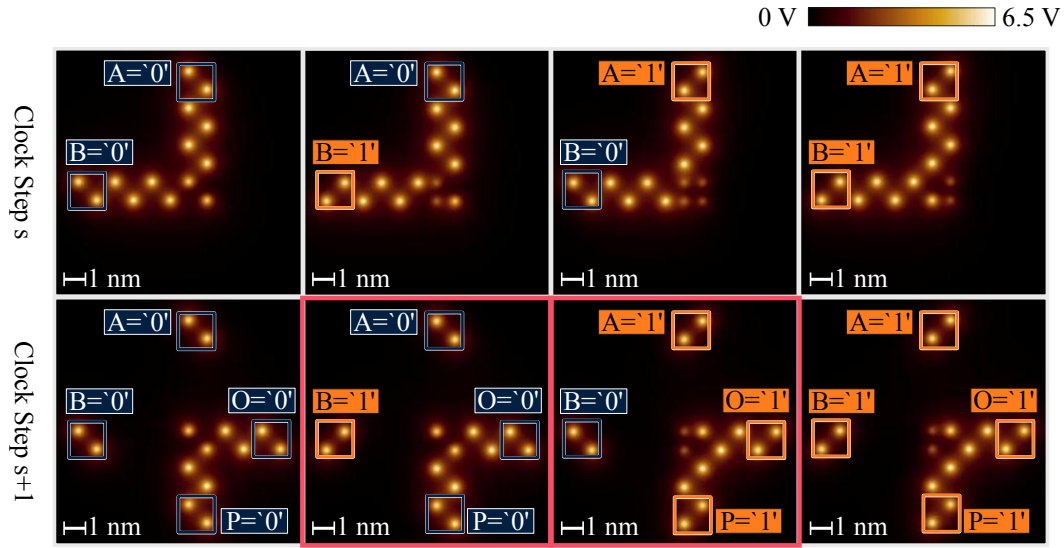


Fig. 3.27 Simulation using SCERPA of a single-line multi-molecule crosswire interconnection based on zwitterionic molecules with $x_c = -5 \text{ \AA}$: the images depict the electrostatic potential for the four input combinations evaluated 0.2 nm above the logic dot plane. The top and bottom lines of images represent subsequent clock steps.

Having moved *Dot 3* far away from the logic dot plane highlights a small crosstalk effect when $AB = '01'$. Even if a small crosstalk effect is present, the solution is resilient against electrostatic variations, demonstrated by the wide working range, i.e., $w \in [6; 13] \text{ V}^{-1}$ referring to equation (3.14).

The multi-molecule crossover implementation increases the set of working devices, including also zwitterionic molecules with various electrostatic footprints. However, we must remember that the working principle is a single-branch inversion, which is dependent on the implementation. Indeed, if we simulate the crossover implemented with zwitterionic molecules having $x_c = -5 \text{ \AA}$, the situation changes drastically, even with the decatriene-inspired molecules in the center. Fig. 3.27 shows the electrostatic potential obtained from the SCERPA simulation. In this implementation, when A and B encode the same logic, information propagates correctly, with just a slight crossover effect. However, when $A \neq B$, both outputs encode the same logic value as input A , demonstrating that a correct single-branch inversion is the essence of the functioning.

Multi-line multi-molecule coplanar crosswire

Multi-line devices are of particular interest for strengthening the information flow in a circuit, as border effects and information decay are strongly reduced. The parallel lines composing the circuits replicate the same information, enhancing the relative influence and allowing the molecular charge to be well localized on one of the two logic dots.

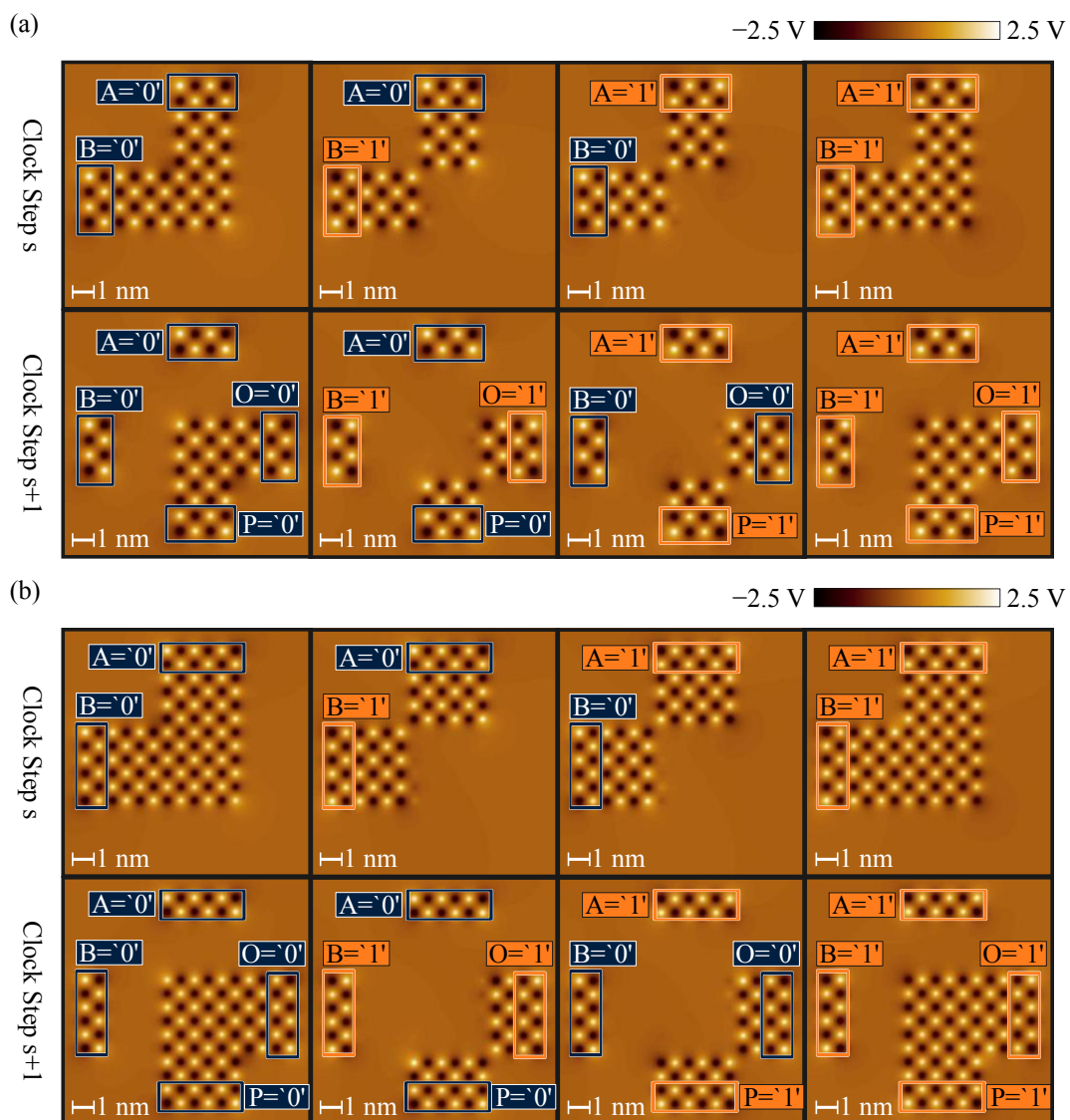


Fig. 3.28 Simulation using SCERPA of a multi-line multi-molecule crosswire interconnection based on neutral molecules: the images depict the electrostatic potential for the four input combinations evaluated 0.2 nm above the logic dot plane. The top and bottom lines of images represent subsequent clock steps. (a) Two-line layout. (b) Three-line layout.

One of the two fundamental assumptions for the crosswire to function properly is that the superposition of the effects of the input branches cancels out in the central region when inputs differ. We have seen that asymmetries, both geometrical and electrostatic, complicate the balancing of the generated electric field in the interconnecting region, leading to unwanted polarization. Indeed, in single-molecule implementations, information crossovers do not work correctly.

Multi-molecule devices that utilize the decatriene-inspired molecules in the center are supposed to mitigate the polarization of the central region. We verify the behavior of multi-line crossover interconnections implemented with neutral molecules using SCERPA. Fig. 3.28a shows the electrostatic potential for the four input combinations in a two-line crosswire. In this case, the central region where the decatriene-inspired molecules are present is composed of four cells. The behavior is exactly as expected based on theoretical principles: when inputs are equal, cells in the central region polarize according to the encoded logic, whereas when inputs differ, the central region is not visible since the charge is nearly zero on logic dots.

To verify the robustness of the implementation with neutral molecules, we also simulate a three-line crosswire, which can be useful in the case of computation on Self-Assembled Monolayers (SAM)[99]. Fig.3.28b shows the electrostatic potential obtained from the SCERPA simulation. Even the increased distance between some cells in the central region from input branches does not affect the behavior of the system, which remains correct in all four input combinations.

Until now, we have observed that implementations with neutral molecules tend to be more straightforward than implementations with zwitterionic molecules. The two-line multi-molecule crossover based on zwitterionic molecules works in a limited set of cases.

We focus on the implementation with *Dot 3* placed 5 \AA away from the logic dots ($x_c = 0 \text{ \AA}$), since it is borderline for functioning. In this condition, the crossover provides correct propagation only when $w = 6 \text{ V}^{-1}$ or $w = 7 \text{ V}^{-1}$, referring to equation (3.14). Fig. 3.29 shows the electrostatic potential obtained with SCERPA in the case with $w = 7 \text{ V}^{-1}$. For higher values of w , the electrostatics are sufficiently strong to polarize the central cells when $A \neq B$. On the contrary, for smaller values of w , the electrostatics are not strong enough to induce inversion when $A \neq B$.

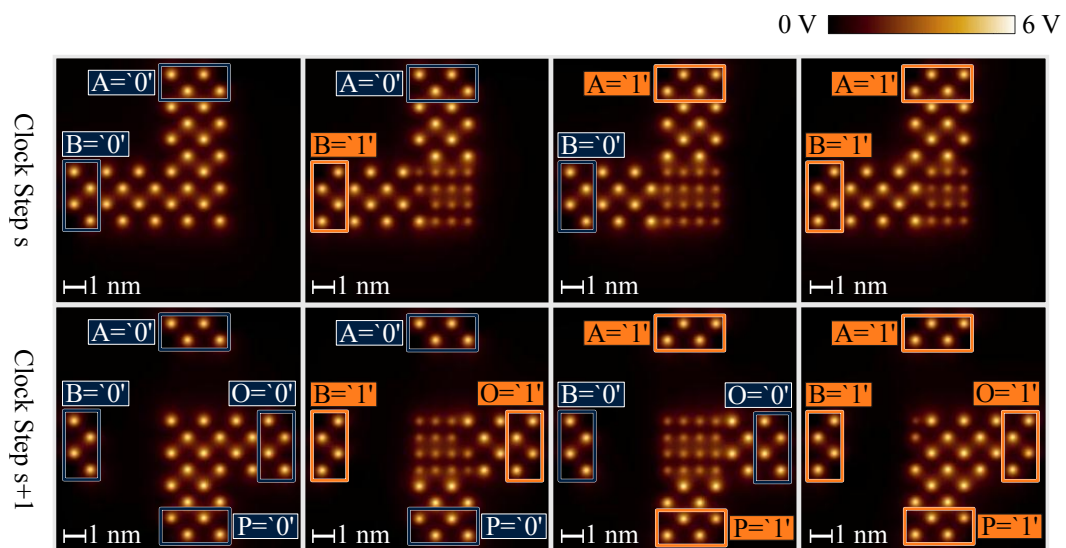


Fig. 3.29 Simulation using SCERPA of a two-line multi-molecule crosswire interconnection based on zwitterionic molecules with $x_c = 0 \text{ \AA}$: the images depict the electrostatic potential for the four input combinations evaluated 0.2 nm above the logic dot plane. The top and bottom lines of images represent subsequent clock steps.

Chapter 4

Device modeling

Performing experiments in molecular-based technologies such as molecular Field-Coupled Nanocomputing (molFCN) poses significant challenges and expenses. Consequently, adopting a prudent approach by anticipating potential outcomes beforehand is advisable. In contemporary times, simulation tools exhibit increasing accuracy, and the availability of computational power facilitates the exploration of larger systems.

In the case of molFCN, as discussed in Chapter 2, *ab initio* simulation and Density Functional Theory (DFT) enable the precise study of molecules with dozens of atoms. However, comprehensive optimization and calculation for each V_{in} -AC transcharacteristic (VACT) can take up to a week or more, depending on the molecule. Moreover, the time required does not increase linearly with the complexity of the analyzed system but exponentially.

This necessity led to the development of an electrostatic model specific to molFCN, simplifying the treatment while remaining linked to the physical behavior of the molecule. The Molecular Simulator Quantum-dot cellular automata Torino (MoSQuiTo) methodology and the Self-Consistent Electrostatic Potential Algorithm (SCERPA) tool allow the simulation of small devices, as those presented in Chapter 3, in short times, often less than a minute. Additionally, SCERPA enables simulations with different types of molecules, corroborating the predictions of energy calculations, which make it physically accurate.

However, SCERPA also has its limits. It can simulate devices with a few hundred molecules in a few hours. Such times are prohibitive if one aims to accurately simulate even a simple arithmetic unit composed of thousands of molecules.

This chapter delves into enhancing the level of abstraction to develop physically accurate models of entire devices. This approach allows us to treat them as black boxes whose behavior is delineated by a specific number of variables and tables.

The methodology we employ consists of several key steps. First, we establish clear rules to describe the behavior of gates in terms of their input and output interactions. This step helps us understand how each gate functions within a circuit. Next, we automate the characterization process for elementary gates, which essentially means we analyze their behavior systematically. This analysis helps us build a comprehensive library of devices, where each device is characterized by its unique input/output behavior.

Once we have this library in place, we can move on to evaluating the behavior of more complex circuits. By utilizing the predefined characteristics of the elementary gates, we can predict how these gates will interact with each other in larger circuit configurations.

To ensure the reliability of our methodology, we rigorously validate the accuracy of the gate models. We do this by simulating designs that incorporate these models and comparing the results with SCERPA simulations that we demonstrate to be accurate. This validation step is crucial as it ensures that our models accurately represent the actual behavior of the gates.

4.1 Device figures of merit

Imagine a molFCN circuit, such as the XOR gate, comprising various elementary gates like ANDs, ORs, and inverters [64]. Our methodology suggests simulating each of these elementary gates individually once and then storing the simulation results in a Look-Up Table (LUT). This LUT serves as a reference for obtaining information about any circuit or architecture containing these gates without the need to simulate the elementary blocks again. Essentially, each circuit can be viewed as a collection of black boxes, where the behavior of each box is cataloged in library files that link input and output signals.

This approach remains effective even for more complex systems. For example, we can describe a ripple-carry adder as an interconnection of XORs and ANDs, using the simulative data we already have to predict the final result. In practice, we replace each elementary gate with a black box featuring input and output ports. The behavior of each block is described by a LUT that relates input to output, possibly including other parameters.

In molFCN, information travels through electric field coupling, a long-range field. This implies that breaking a circuit into smaller blocks and simulating these blocks in isolation may not accurately represent the global behavior of the original circuit. Additionally, changes in the molecule or the extension of the clock region can produce different behavioral results, highlighting the significance of physics at this scale, as we demonstrated in Chapter 3. Therefore, simulating the elementary gates used to construct the library files must consider the physics of the interaction within and between devices to accurately describe the expected outcome. This section proposes a framework for a comprehensive and precise description of a generic device, incorporating the underlying physics to model its behavior accurately.

4.1.1 Input and output signals

To begin, it's essential to establish how signals represent the inputs and outputs of the device. This ensures that they accurately capture the charge state of molecules. By storing the input values, we can uniquely identify a particular physical state, enabling us to predict the correct output under similar conditions.

In previous chapters, two main approaches have been used to describe the charge state of a molecule or a molecular cell. The first method involves considering the molecule input voltage V_{IN} and its VACT, describing the charge configuration on dots for each value of V_{IN} . The second approach involves cell polarization, a measure of the charge distribution within the molecular cell as used for the energy analysis in Chapter 3. Between the two options, we opt for input voltages, as the SCERPA simulator operates with the transcharacteristics of molecules.

Then, we know that information propagates through the sequential activation of clock regions. This kind of propagation is similar to having information packages moving along the circuit, so information at the output is meaningful only if sampled in the correct time step. We are interested in sampling the information on the output

when the clock signal is in the hold state. The intermediate states are not of any influence in the input/output model of a device, so there is no need to store them. This means that for each information package sent into the device input, we will store in the LUT only the output values during the hold phase. If we know the molecule used and the clock field value, then the molecule input voltage is sufficient to describe the molecular charge state thoroughly and uniquely.

After deciding that the molecular input voltage is the signal of interest for inputs and outputs, we must decide on which cells we have to read that information on order to obtain a meaningful input/output description of the Device Under Test (DUT). Specifically, we must ensure that the model aligns with the actual circuit implementation. This means that even spatially we have to remember that inputs and outputs correspond to actual molecular cells.

For instance, consider the situation shown in Fig. 4.1a with two single-line wires, *Wire A* and *Wire B*, each composed of 6 molecular cells, receiving an input

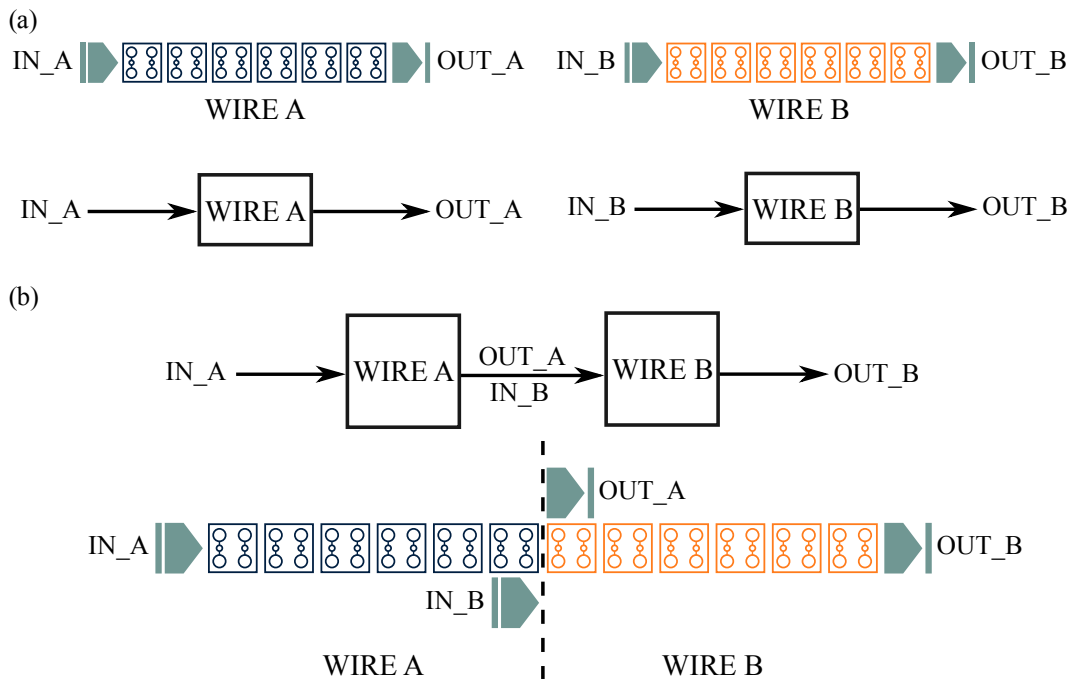


Fig. 4.1 Choice of the input and output signals of a device on an example system. (a) Scheme of the two wires *A* and *B* used for the example, with input and output pins highlighted. At the bottom is shown an input/output schematic representation. (b) Connection between *Wire A* and *Wire B*. The schematic view of the signals is on top, whereas at the bottom, the corresponding physical circuit shows the overlapping of the cells and pins at the joint, giving two possible choices for signal compliance.

signal from the driver pin and generating an output state on the output pin. In an input/output model, the 6-cell wires can both be substituted by a generic box, which receives an input signal and produces an output signal.

When we connect *Wire B* to *Wire A*, we expect a 12-cell single-line wire, with the output of *Wire A* (OUT_A) corresponding to the input signal of *Wire B* (IN_B). Fig. 4.1b shows that we cannot use both the output pin of *Wire A* and the driver pin of *Wire B*, otherwise the real implementation would not correspond to a 12-cell wire.

Two equivalent cases show up for signal compliance: either we use the first cell of *Wire B* and the output pin of *Wire A*, or we use the driver pin of *Wire B* and the last cell of *Wire A*. We select the latter option, ensuring that the input refers to the driver pin and the output to the last cell of the device under test.

Taking into account the decisions made thus far, we assign distinct names to the molecules of interest to designate the device's input and output signals and their positions. In Fig. 4.2a, we illustrate the example of a Majority Voter (MV) with multiple inputs (IN_1 , IN_2 , IN_3) and a single output (OUT). Input signals must be taken on driver pins, which are thus substituted with controllable molecular cells. On these cells, we are interested in the molecule voltage (V), and because it is for identifying the device input, we attach the label IN . The branch related to input IN_1 is associated with the identifier UP , since it comes from upward. Similarly, the branch related to input IN_3 is associated with the label DW , as the signal comes from downward. The signal associated with input IN_3 does not have any additional identifying label. The voltages related to the device output signal are identified with the label OUT . Regarding molecule labeling, the first and second molecules in a cell are denoted by the letters A and B , respectively.

The title entry of the MV LUT will look like Table 4.1 for the single-line MV shown in Fig. 4.2a.

Table 4.1 Title entry of the LUT describing the input/output behavior of a single-line MV.

$V_{IN_UP_A}$	$V_{IN_UP_B}$	V_{IN_A}	V_{IN_B}	$V_{IN_DW_A}$	$V_{IN_DW_B}$	V_{OUT_A}	V_{OUT_B}
...

Fig. 4.2b shows the example of a fanout circuit with a single input (IN) and three outputs (OUT_1 , OUT_2 , OUT_3). As described before, V_{IN} is associated with device input signals, and V_{OUT} describes the device's output. The branch related to

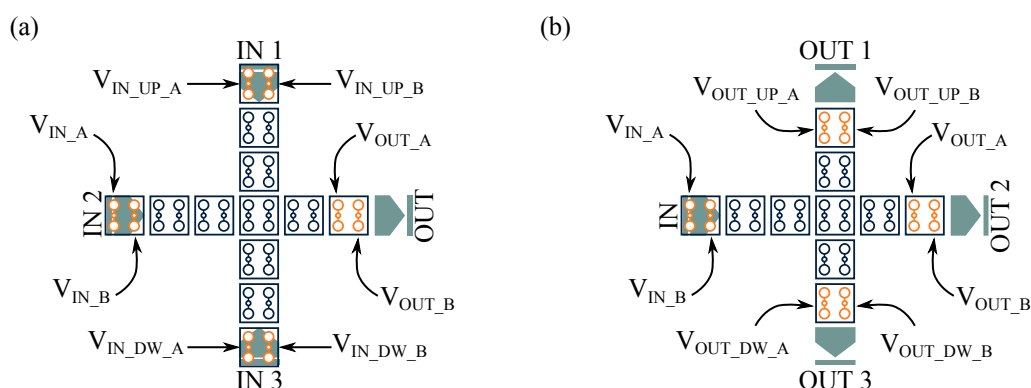


Fig. 4.2 Rules for naming device's input and output signals before being stored in the LUT. The molecules of interest are orange-colored. (a) Example on the MV layout with three inputs and one output. (b) Example on the fanout circuit layout with one input and three outputs.

output OUT 1 is associated with the identifier *UP*, as it goes upward. Similarly, the branch related to output OUT 3 is associated with the label *DW*, as the signal goes downward. The signal associated with input OUT 3 does not have any additional identifying label. Additionally, letters *A* and *B* identify the molecules on the left and right of the cell, respectively. Table 4.2 reports the title entry for the input/output signals of a single-line three-way splitter.

Table 4.2 Title entry of the LUT describing the input/output behavior of a single-line fanout circuit.

V_{IN_A}	V_{IN_B}	$V_{OUT_UP_A}$	$V_{OUT_UP_B}$	V_{OUT_A}	V_{OUT_B}	$V_{OUT_DW_A}$	$V_{OUT_DW_B}$
...

For multi-line devices, which present duplicated transmission lines, we must store the voltages of both the cells constituting the input and output signals. To distinguish them, we add the labels *H* and *L*. On horizontal lines, *H* is associated with the higher line, and *L* with the one on the bottom. In vertical lines, *H* is associated with the line on the left, whereas *L* identifies the line on the right.

4.1.2 Information stability

After selecting the input and output signals, we must also consider the simulation conditions, particularly regarding the output. Devices may exhibit vanishing information

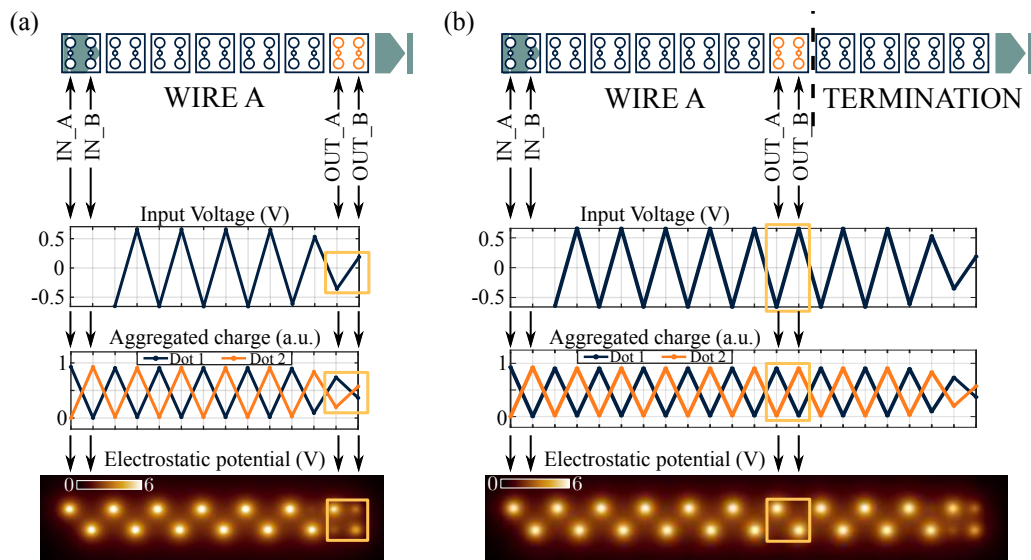


Fig. 4.3 Border effect on information propagation. (a) Propagation in a single-line wire simulated with SCERPA, showing the vanishing data at the end of the wire, corresponding to output molecules A and B. The three graphs represent the input voltage, the aggregated charge value, and the electrostatic potential on each molecule of *Wire A*. (b) Propagation in a single-line wire connected to a termination block composed of 4 cells simulated with SCERPA, showing the restored data on the output of *Wire A*. The three graphs represent the input voltage, the aggregated charge value, and the electrostatic potential on each molecule of the system composed with the wire and the termination.

at their terminus, even in cases of bistable propagation [64]. In an isolated device, no molecules reinforce the information beyond the output, leading to less pronounced charge separation between the dots of the final molecules. Fig. 4.3a showcases the SCERPA results of the same *Wire A* mentioned previously and implemented with bisferrocene molecules.

Fig. 4.3a illustrates that if *Wire A* is simulated in isolation, the wire output voltages (V_{OUT_A} and V_{OUT_B}) are lower than the voltage on other molecules because of the information decay. However, when we connect a second device at the end of *Wire A*, the diminishing information on *Wire A* output is restored by the presence of the attached device. Fig. 4.3b presents the results obtained from SCERPA of *Wire A* with a 4-cell termination wire connected at its end. The simulation demonstrates that information on the original output cell of *Wire A*, boxed in yellow in Fig. 4.3, is restored. The termination wire acts as a stub at the end of the DUT helping to restore information.

When conducting simulations to assess the input/output behavior of a device, with the intention of incorporating it as a connected module within a larger design, it's crucial to account for the phenomenon just described. Thus, as we isolate and simulate each gate for characterization, we need to anticipate its connection to other circuits during actual operation. Even if we simulate our DUT isolated, we must remember that in the steady-state functioning, the DUT is supposed to be connected to other gates that will act as the termination wire in the example of Fig. 4.3b. To consider this, we enhance the device model by simulating it in conjunction with a termination circuit, mimicking the presence of a cascaded device. This ensures that the stored output voltages align with the behavior expected in the context of a cascade setup, maintaining precision to the underlying physical system.

Termination circuits can take various forms, and in this chapter, we utilize a short wire to reinforce information retention at the output cell, without introducing excessive complexity that may slow down simulations. Specifically, our termination wire consists of four cells, and in the case of multi-line setups, it is replicated across both lines.

4.1.3 Additional information for the model

When explaining the functionality of a device, it's crucial to account for various factors, including the distribution of the clock signal, the voltages at input and output nodes, and the incorporation of a termination circuit. The division of the device into phases significantly impacts signal synchronization and is intimately tied to latency considerations, which are pivotal in circuit design as they dictate path balancing. By ensuring that a device terminates with the maximum available clock phase, we can characterize latency and synchronization using two key parameters: the maximum number of phases (*maxPhases*) and the number of repetitions of those phases (*phaseRepetition*). For instance, in the XOR gate illustrated in Fig. 4.4a [64], with a maximum of four phases (*maxPhases* = 4) repeated three times (*phaseRepetition* = 3), the latency amounts to 12.

Additional critical parameters relate to the physical and electrostatic footprint of the device. Physical occupation can be quantified in nanometers (absolute area) or in terms of unit cells within a grid system (relative area). On the other hand, electrostatic occupancy primarily considers crosstalk effects within the device, as

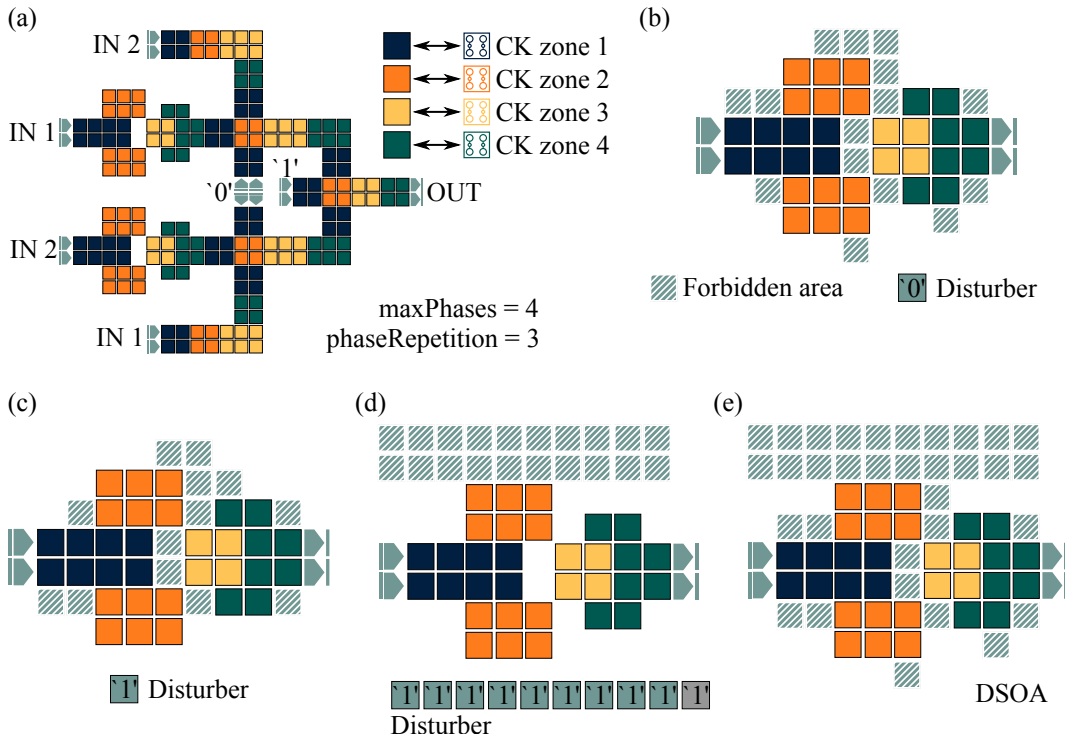


Fig. 4.4 Additional information needed for a complete description of a device. (a) Layout of a XOR gate highlighting clock zones for synchronization and their repetition for latency. (b) Forbidden area for a multi-line double-branch NOT gate when the disturber is a single driver cell encoding a logic '0'. (c) Forbidden area for a multi-line double-branch NOT gate when the disturber is a single driver cell encoding a logic '1'. (d) Forbidden area for a multi-line double-branch NOT gate when the disturber is a whole polarized wire encoding a logic '1'. (e) Device safe operating area of the multi-line double-branch NOT gate.

described in section 3.3. To determine the Device Safe-Operating Area (DSOA) and prevent errors, we must conduct multiple simulations of the DUT under various conditions, incorporating different types of disturbers in its vicinity. These disturbers are modeled by fixed-logic driver cells or wires. For instance, in the case of the multi-line double-branched NOT gate, Fig. 4.4b illustrates the forbidden area obtained using a single-cell disturber encoding a logic '0', while Fig. 4.4c depicts the forbidden area derived using a single-cell disturber encoding a logic '1'. Disturber can be whole polarized wires, as in the case of Fig. 4.4d, showing the forbidden area using a disturber wire encoding a logic '1'. By aggregating all the forbidden regions, we delineate the DSOA of the gate, as shown in Fig. 4.4e.

The library files must encapsulate all the parameters discussed in this section to adequately describe the DUT behavior. To streamline this process, we've developed

a program that automatically evaluates the requisite parameters and generates the library files for each device. However, manual intervention is still required for DSOA analysis, as automating the design process for composite circuits remains a work in progress. Standardizing this process may entail numerous simulations tailored to the specific device. Nonetheless, the designer expertise is sufficient to build small circuits that adhere to the gate's DSOA.

4.2 Automated algorithm for device characterization

The previous section highlighted the key elements that must be accounted for when simulating a device. The general idea is to obtain a set of files for each gate, containing all the relevant information, starting from the input/output behavior, the latency, and the area occupation. Then, once we have all these files for each gate, we should be able to simulate larger designs that are a collection of these gates interconnected in some way.

Aiming to standardize the process and simplify the treatment for a general user, we developed an automatic tool that handles the simulation of the DUT and the creation of the relative files in the library. The Block-Based characterization (BBchar) tool automatically simulates devices and extracts the needed properties.¹

The following sections describe the principal elements of BBchar to create a complete set of library files and the written code. The program applies to molFCN circuits designed in MagCAD [134, 135], which produces a file describing the layout of a circuit, including input and output pins. The BBchar tool is able to:

- automatically generate input files for the simulation of the circuit in SCERPA;
- automatically characterize the circuit to provide a LUT with the input/output description plus some additional information regarding timing and area;
- provide the output of a circuit in the library given a generic combination of the inputs.

The expected workflow for each DUT is to simulate it multiple times with SCERPA until we obtain a correct behavior. Then we have to attach the termination

¹The tool and all the files presented in this chapter are available online at <https://github.com/vlsi-nanocomputing/BBchar>.

before launching the complete simulation that we use to extract the data for the library files. Finally, we can extract the information we need from the simulation data and store them in the library files related to the DUT. All the steps are handled automatically by the BBchar tool.

4.2.1 Drivers and clocks

The focal point of the entire process lies in conducting a comprehensive simulation, allowing us to gather a sufficiently extensive dataset to extract and generate valuable library files. Indeed, the richness of data in these files directly impacts the precision of output predictions for any given input. Consequently, meticulous attention must be given to configuring the clock signal for each clock region and determining driver voltages.

Initially, it's necessary to establish the structure of a single clock cycle. Common clock signals typically exhibit trapezoidal waveforms as described in chapter 1, exemplified in Fig. 4.5a. The clock signal is divided into four stages (*num_stages*): switch, hold, release, and reset. Each of these stages is divided into a specific number of steps that can be set by the user with the variable *clock_step*. The variables *clock_low* and *clock_high* define the lower and upper voltage levels for the clock signal. Once we know these three variables, we can obtain the clock voltage for each simulation step. For instance, if *clock_step* = 5, *clock_low* = -2 V, and *clock_high* = 2 V, we obtain the following clock signal sequence for the various phases:

$$\begin{aligned}
 pSwitch &\rightarrow -2 \quad -1 \quad 0 \quad +1 \quad +2 \\
 pHold &\rightarrow +2 \quad +2 \quad +2 \quad +2 \quad +2 \\
 pRelease &\rightarrow +2 \quad +1 \quad 0 \quad -1 \quad -2 \\
 pReset &\rightarrow -2 \quad -2 \quad -2 \quad -2 \quad -2
 \end{aligned}$$

The complete unit cycle (*pCycle*) is defined as the union of the four stages just described and is replicated as needed.

Circuits are divided into clock regions, typically four regions are sufficient to correctly propagate the information by sequentially activating each region. Each

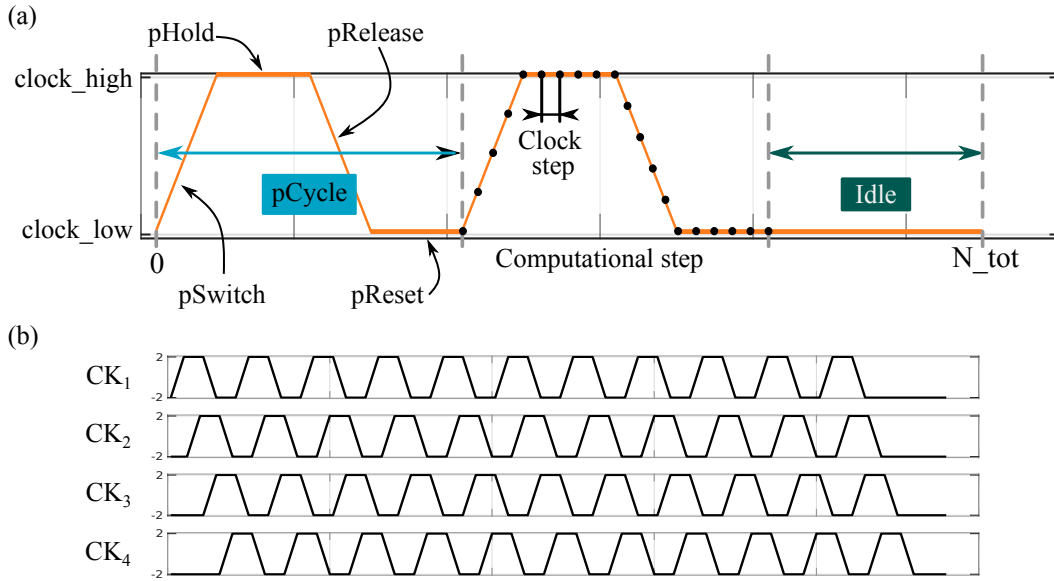


Fig. 4.5 Clock signal for DUT simulations. (a) Definition of the terminology for a trapezoidal clock signal. (b) Example of a complete set of clock signal for a device having four clock regions, with 11 periods plus the idle phase.

clock region receives a waveform as the one described before, just out of phase of one stage. Fig. 4.5b shows the clock waveform for a circuit having four regions. For the steps added to comply with the waveform shift, we can associate the clock value of the reset, i.e., *clock_low*. The length of this *idle* period is the same as the period *pCycle* for each region and divided between the beginning and end of the simulation according to the specific region. So, for instance, for region1, the idle phase is fully at the end, for region3 it is divided between the beginning and end, and for region 4 it is all at the beginning.

In addition to the voltage values and the shape of the period for the clock cycle, we need to define how many periods we need to propagate the set of inputs we are interested in simulating. The total number of periods is linked to how many times the four clock regions are repeated between the input and the output, that is the latency of the DUT. In the case of a single input value to be propagated from the input to the output, the total number of computational steps is

$$N_{tot} = clock_step * num_stages * latency + idle \quad (4.1)$$

Together with the clock signal, the signals associated with the inputs are fundamental for performing a complete simulation of the DUT under different conditions.

First, we need to determine how many different inputs are present in the design and how they are identified in the MagCAD layout file, so that we can locate their positions. Even if inputs are replicated, as in the example of the XOR gate shown in Fig. 4.4a, they are considered as a single input, and thus, they receive the same signal.

As discussed earlier, input signals are represented by voltages on molecules. Therefore, we need to determine the maximum voltage to be applied (V_{inMAX}). This value depends on the specific molecule used for the DUT implementation. For instance, for bisferrocene molecules, an input voltage range of ± 1 V is sufficient to saturate the charge on one logic dot rather than the other. Testing the DUT only for the maximum voltages could lead to a limited set of simulation data from which to extrapolate entries for the LUT. To address this, in addition to permitting the simulation of constant ‘0’ and ‘1’ (associated with $\pm V_{inMAX}$), we allow for more signals: a positive and a negative ramp between the maximum allowable voltages. The positive ramp corresponds to a sweep from the voltage associated with ‘0’ to the voltage associated with ‘1’, while the negative ramp sweeps from the voltage associated with ‘1’ to the voltage associated with ‘0’. These ramps can be linearly or logarithmically divided into N_{driver_step} , depending on the user’s requirements. The ability to simulate the DUT for voltages that do not saturate the molecular charge helps in predicting all simulation cases, including non-ideal scenarios, thereby improving confidence in the data stored in the LUT. A higher value of N_{driver_step} results in more entries in the input/output LUT that describe the circuit behavior, enhancing the predictive capability when the library file is used.

The preceding section highlighted that for each physical input, we require two signals, one for each molecule composing the cell. Typically, since charge in a cell localizes on antipodal dots, the two molecules composing the cell exhibit opposite voltages. Moreover, in multi-line devices, inputs are composed of multiple cells. For instance, considering the multi-line NOT gate shown in Fig. 4.4b, we will have four distinct input signals. Fig. 4.6 illustrates a set of possible inputs for the multi-line NOT gate, where we request a positive ramp divided into $N_{driver_step} = 10$. From the waveforms, we observe that the two molecules in the cell (A and B) exhibit opposite ramps, while the upper line (H) and the lower line (L) show the same behavior. The ramp is executed in small steps, and each value is held constant for a period. This choice is rooted in the physical system. Information propagates in

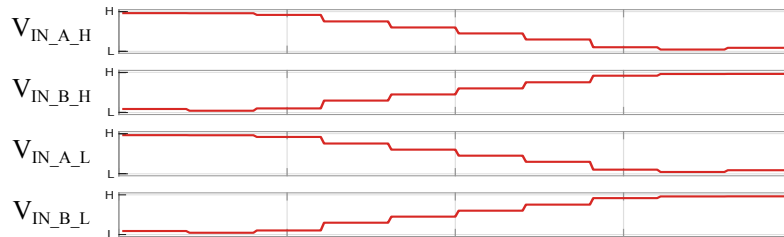


Fig. 4.6 Example of input signals for the simulation of a two-line NOT gate.

the form of packages, and inputs must remain stable for an entire *pCycle* to avoid generating conflicts.

Finally, to formulate a comprehensive algorithm, we must also account for specifying multiple contiguous input combinations (N_{comb}) for a single simulation.

Considering all constraints, the number of clock cycles necessary to propagate all the input values is given by:

$$N_{cycles} = N_{driver_step} * N_{comb} + (phasesRepetition - 1) + idle_phase \quad (4.2)$$

The first row of equation (4.2) is related to the number of provided inputs. The second row determines the cycles required to propagate the last input to the output, effectively clearing the pipeline.

4.2.2 Termination

Section 4.1.2 emphasized the importance of adding a termination to the DUT to adequately characterize it in case a second device is attached. We have already mentioned that the termination circuit can be of any type. In BBchar, the default termination is a 4-cells-long wire, single-line, or multi-line depending on the DUT, and is added automatically.

A termination is added to each physical output in the following clock phase, ensuring that a new clock region is not added. Therefore, for instance, in a circuit with four clock regions, if the output is in clock region 4, the termination will be added in region 1. Similarly, if the output is in region 3, the termination will be added in clock region 4. Once the termination is added, the driver and clock matrices must

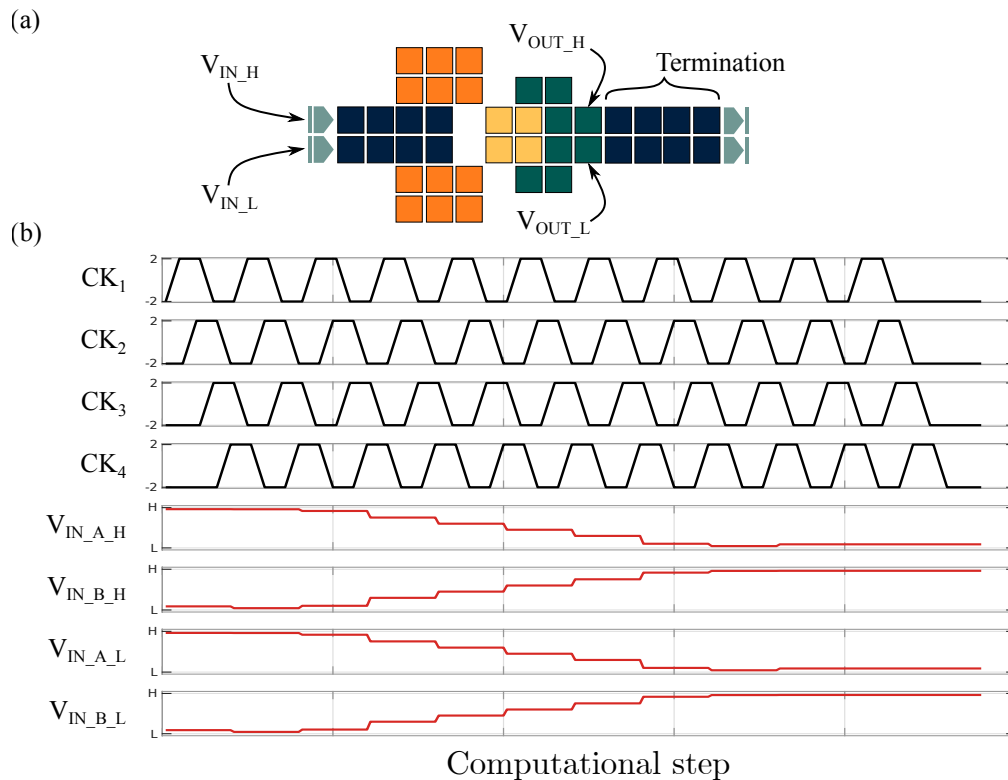


Fig. 4.7 Example of the simulation of a two-line NOT gate with a termination attached to the end. (a) Layout of the gate highlighting the termination and the cells on which input and output signals are considered. (b) Clock and input signals for the simulation of a two-line NOT gate with a termination attached to the end.

be updated to include the new region: a clock cycle is added to the phase matrix, and the last driver value is replicated for its length.

Consider again the example of a two-line NOT gate divided into four clock regions. The termination circuit will be in clock region 1, as shown in Fig. 4.7a. Input signal refers to the driver pins. Output signals will be read on the two cells right before the termination, and because of it, information is not vanishing. For the simulation, we fix $N_{driver_step} = 10$ and $N_{comb} = 1$ since we want to simulate a single positive sweep. The original circuit has four clock regions repeated a single time $phasesRepetition = 1$. According to equation (4.2), we obtain $N_{cycles} = 10 + idle_phase$. However, because of the presence of the termination, we need an additional cycle, as correctly represented in Fig. 4.7b, where 11 cycles are visible.

4.2.3 Library files creation

The generation of library files serves the crucial function of ultimately predicting the logic value at the output of a circuit. Therefore, their generation is fundamental to obtaining high accuracy in output prediction.

The primary focus lies in the voltage values of both the input and output molecular cells. We must ensure meticulous sampling of signals during the appropriate hold clock phase, considering the delay between input and output due to latency. Moreover, we should obtain a sufficiently large set of input/output pairs to increase the number of entries in the Look-Up Table (LUT). The more entries we have, the more input/output associations we have when using the LUT.

Understanding the waveforms utilized for simulation alongside the circuit's clock phase repetitions simplifies the identification of simulation steps aligning with the correct sampling intervals. Each input voltage employed for simulation is logged in the LUT, recording the corresponding voltage values on the output molecules during the hold state of the clock region to which the output belongs and after the correct number of repetitions.

In molFCN technology, the functionality of a device is directly related to its geometrical occupancy. In addition, interconnections between gates are not time-independent and are subject to all the behavioral rules of a gate. This means that the library not only should encompass several gates, with multiple versions, but also interconnections. Fig. 4.8 illustrates the layout of the interconnections used in the following section for the validation of the device modeling methodology with their identification names. We simulate them using SCERPA, using $N_{clock_step} = 5$, $N_{driver_step} = 10$, and $V_{inMAX} = 1$ V. Table 4.3 lists the SCERPA simulation time for the various layouts and the number of cells composing each interconnection. Once we have performed the simulations, we have all the data to extract and calculate the parameters for the library files that we previously described. The last column of Table 4.3 reports the overhead time needed to generate and store the library files. As can be seen from the table, the time for the generation of library files is negligible for interconnections comprising 20 or 30 molecular cells.

The two fundamental gates used to build other functions are the MV and the NOT gate. Even in this case, we need more than one version of this gate, which can be connected in different ways in a layout that uses them. Fig. 4.9 depicts

the layout that we use in the next section for the validation of the methodology, together with their identification names. The NOT gate is identified with ‘inv’. The gate called ‘invDr’ is an inverter gate that also provides the output equal to the input, serving as a generic driver gate. For the ‘MVlong’ gate, we simulate three different output configurations. The SCERPA simulations are performed using the same parameters as before, namely $N_{clock_step} = 5$, $N_{driver_step} = 10$, and $V_{inMAX} = 1$ V. Table 4.4 reports the simulation time obtained with those parameters

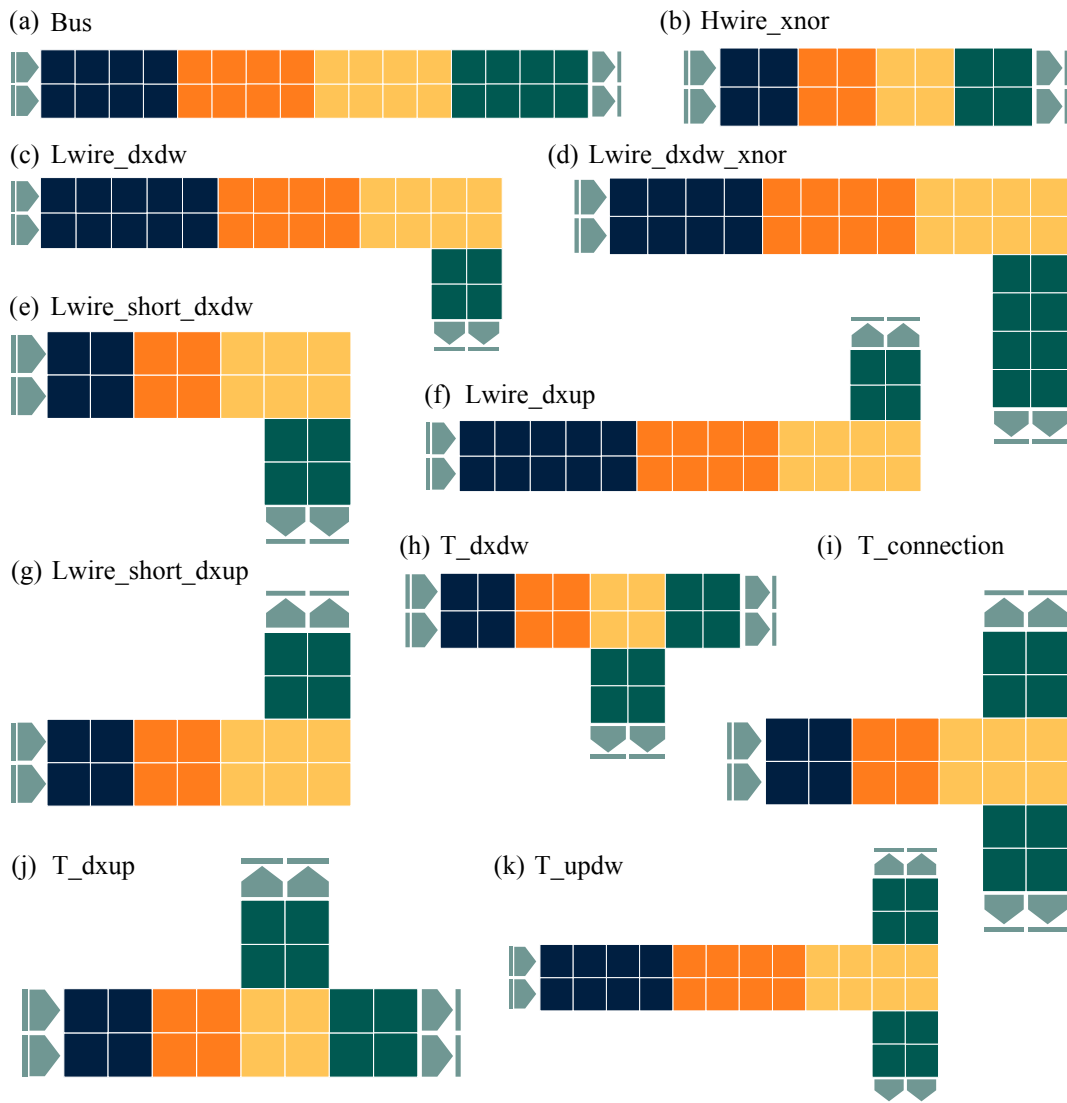


Fig. 4.8 Layout of the interconnections used in this chapter and characterized for the library. (a) Bus. (b) Hwire_xnor. (c) Lwire_dxdw. (d) Lwire_dxdw_xnor. (e) Lwire_short_dxdw. (f) Lwire_dxup. (g) Lwire_short_dxup. (h) T_dxdw. (i) T_connection. (j) T_dxup. (k) T_updw.

Table 4.3 Simulation time for a complete characterization performed with $N_{clock_step} = 5$, $N_{driver_step} = 10$, and $V_{inMAX} = 1$ V, number of cells, and overhead time for generating the library files of some interconnections.

Name	Sim. time (min:sec)	# cells (-)	Lib. gen. time (s)
bus	05:14	32	0.1242
Hwire_xnor	02:24	16	0.0762
Lwire_dxdw	06:28	30	0.0904
Lwire_dxdw_xnor	07:31	32	0.0958
Lwire_dxup	07:34	30	0.0341
Lwire_short_dxdw	03:52	18	0.1068
Lwire_short_dxup	04:52	18	0.0928
T_connection	09:20	22	0.1234
T_dxdw	07:18	20	0.0786
T_dxup	09:26	20	0.1380
T_updw	12:30	32	0.1372

and the cell count. Intuitively, the time needed for the SCERPA simulation depends on the cell count: the more cells, the longer the simulation time. However, simulation time is also proportional to the number of sequential input combinations, as in the case of the MV, which, besides the limited number of cells, requires the simulation of 8 input combinations. The last factor influencing the simulation time is the number of interacting molecules, depending on clock regions distribution and number of cells per region. The last column of Table 4.4 lists the time needed for the extraction of the interesting data and the creation of the library files. Conversely to simulation time, the library files generation time demonstrates minimal variance and is negligible when compared to the simulation time.

4.2.4 Input file analysis and algorithm description

The BBchar tool is a MATLAB program for characterizing and simulating the behavior of a molFCN circuit layout designed in MagCAD. Summing up all the focal points previously described, Fig. 4.10 illustrates a flowchart of the characterization of a device.

At the beginning, we must provide several settings and parameters. The launching script begins by defining various file paths for data storage and retrieval. The main fields are:

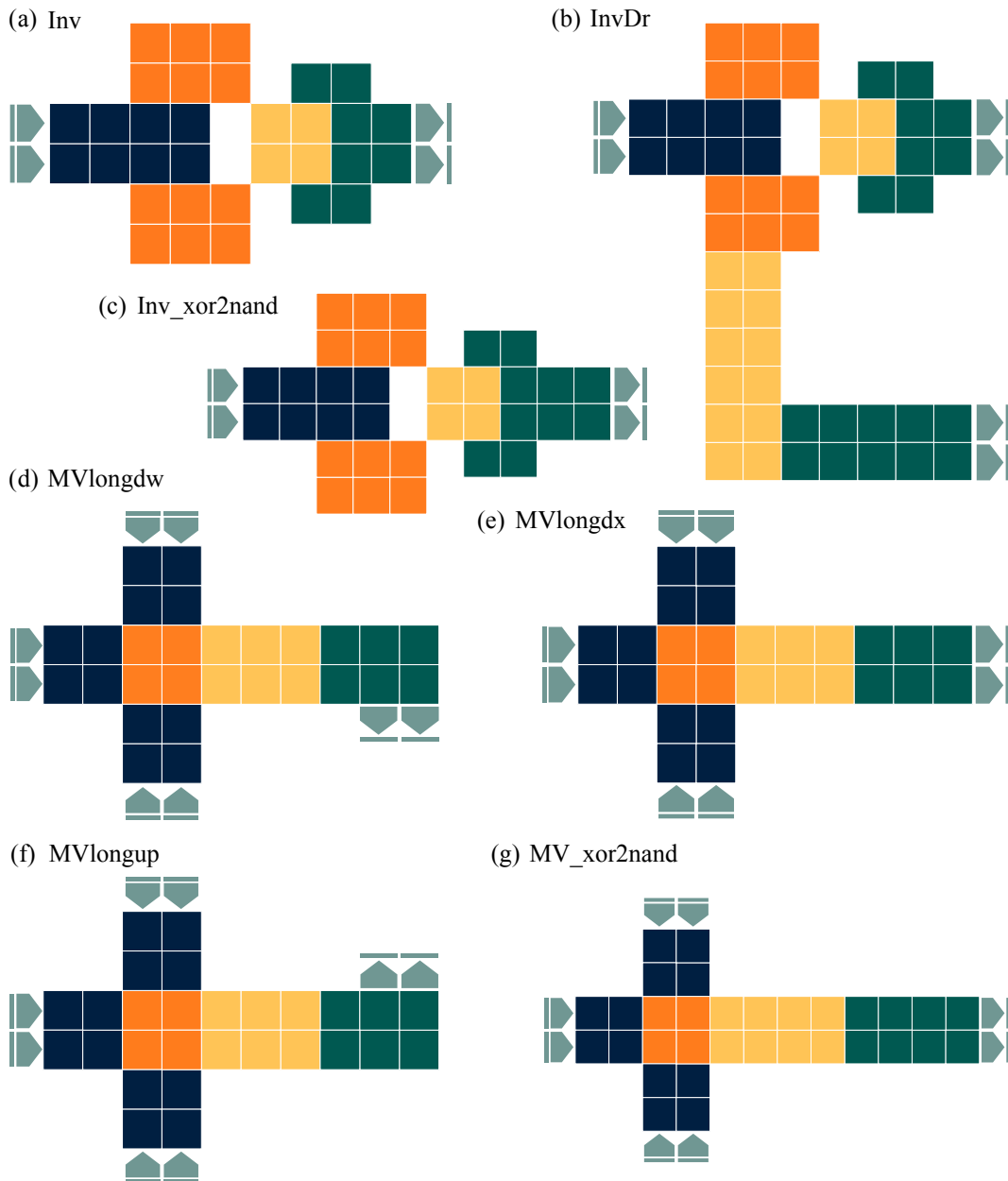


Fig. 4.9 Layout of the gates used in this chapter and characterized for the library. (a) Inv. (b) InvDr. (c) Inv_xor2nand. (d) MVlongdw. (e) MVlongdx. (f) MVlongup. (g) MV_xor2nand.

- **myDataPath**: The root path to data storage.
- **BBcharPath**: Path to BBchar program in the file systems
- **BBcharCodePath**: Path to BBchar code files.

Table 4.4 Simulation time for a complete characterization performed with $N_{clock_step} = 5$, $N_{driver_step} = 10$, and $V_{in_{MAX}} = 1$ V, cell counts, and overhead time for generating the library files of the interconnections used in this work.

Name	Sim. time (min:sec)	# cells (-)	Lib. gen. time (s)
inv	09:46	32	0.1116
invDr	29:58	54	0.1126
inv_xor2nand	09:52	34	0.1480
MV_xor2nand	33:20	32	0.1867
MVlongdw	31:33	28	0.3592
MVlongdx	29:22	28	0.2868
MVlongup	36:00	28	0.1699

- **thisPath**: Current working directory.
- **scerpaPath**: Path to SCERPA program.
- **libraryPath**: Path to the device library.
- **outputPath**: Path for storing SCERPA simulation output.
- **file**: Name of the MagCAD layout file of the DUT.

After that, we must define the parameters for the design of the input signals for drivers and clock waveforms. The main fields are:

- **clock_low** and **clock_high**: Low and high voltage levels for the clock signal.
- **clock_step**: Number of steps for each clock phase.
- **pSwitch**, **pHold**, **pRelease**, and **pReset**: Clock signal profiles for different phases.
- **pCycle**: Combined clock signal for a full simulation cycle.
- **Ninputs**: Number of physical inputs. If a driver is repeated several times in the circuit (repeated input, constant input, bus layout, etc.), it still counts as 1.
- **driverNames**: Names of the drivers as they are in the .qll layout file.
- **maxVoltage**: Maximum voltage value in volts for the drivers.

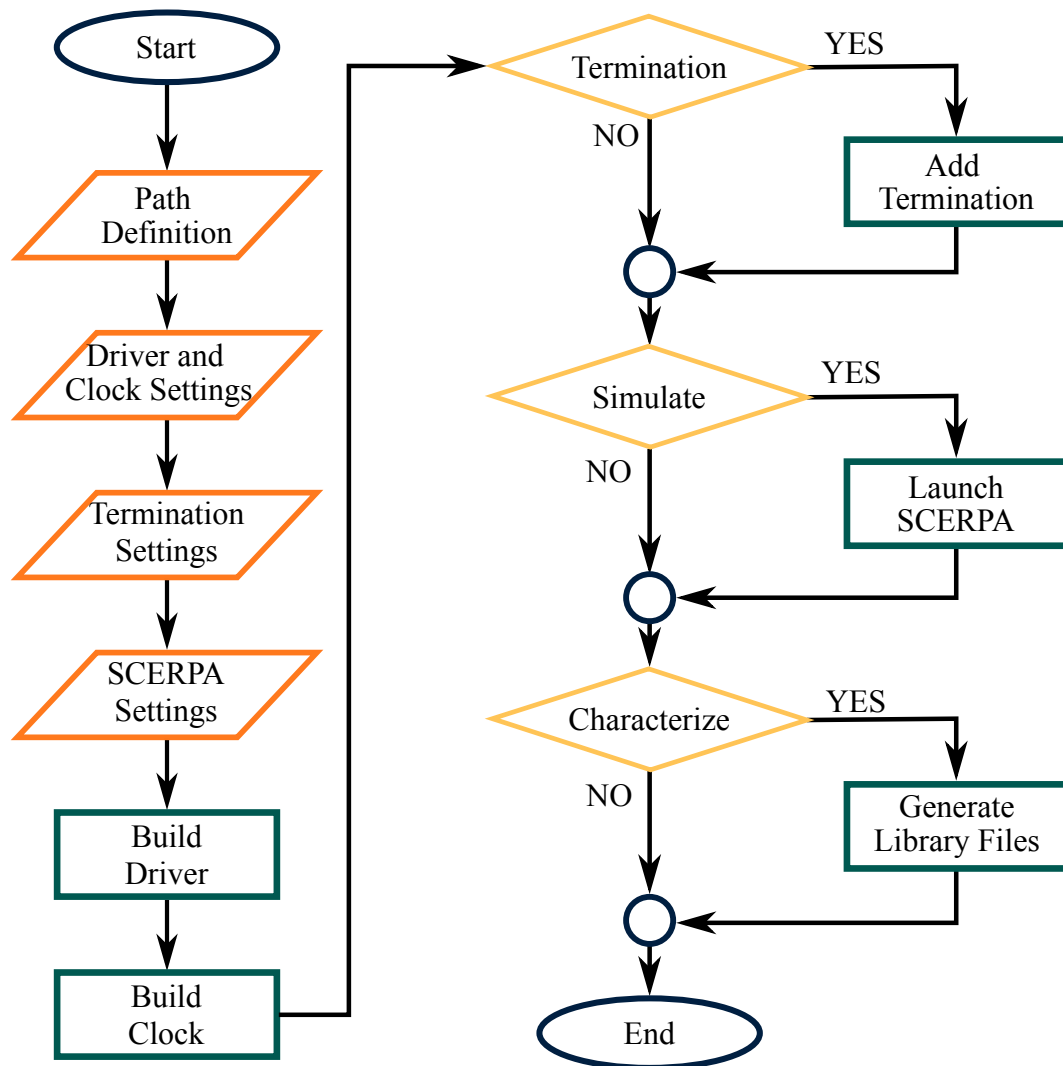


Fig. 4.10 Flowchart of the BBchar program to simulate and characterize the DUT.

- **driverModes:** Values assumed by the drivers in the simulations, can be more than 1 value at a time. The possible values are:
 - *sweep*: variation vector between '0' and '1' (voltage variation between $-maxVoltage$ and $maxVoltage$)
 - *not_sweep*: variation vector between '1' and '0' (voltage variation between $maxVoltage$ and $-maxVoltage$)
 - *0*: constant value equal to logic '0'
 - *1*: constant value equal to logic '1'

- **sweepType**: Type of sweep for driver simulation. It can be *lin* for linear variations or *log* for logarithmic variations.
- **NsweepSteps**: Number of steps in the sweep, if the driver modes specify a constant value, than it is repeated.
- **cycleLength**: Length of a complete clock cycle.
- **NclockRegions**: Number of clock regions.
- **phasesRepetition**: Number of times clock regions repeat.

Finally, we have to provide information regarding the termination circuit. The main fields are:

- **enableTermination**: Flag for enabling termination addition.
- **customLength**: Custom number of cells for termination.
- **busLayout**: Flag for bus or single line layouts.

Once we have defined the parameters for the simulations, we call the function that builds the driver and clock signal matrix for the SCERPA simulation, for which we already specified the parameters. If the user wants to simulate the device with the termination attached to its outputs, then we call the routine to generate the new layout file including the terminations and to adjust the driver and clock matrices.

Finally, if we want to simulate the DUT, we call SCERPA; otherwise, we call the characterization function. The characterization function analyzes the output files produced by SCERPA and extracts the significant data. It creates a new folder in the device library where it copies the layout used for the modeling, the generated input/output table, and a text file containing the additional information discussed previously.

The MATLAB code of the main functions is reported in Appendix B. There's an example of a launching script, the functions to build drivers and clocks, the routine to attach the termination, and the characterization function.

4.3 Validation of the block-based simulation

Once we have a populated device library, the idea is that we do not need to simulate those devices again, and we can instead use the stored values themselves. For example, if we have a NAND gate and we want to test an input combination, instead of simulating it, we interrogate the library files of the MV for the input combination we're interested in and read the output values. Those outputs become the inputs of an inverter, so we interrogate the library again. The output of the inverter coincides with the output of the NAND gate we had at the beginning.

Therefore, the main operation to perform on the library is to read from the LUT the proper value for a generic input set. If the set of input voltages is present as an entry of the LUT, then we just read the corresponding output voltages from the same entry. If the input voltages are not present in the table, then we extract the row that minimizes the distance, quantified using the Euclidean norm. Appendix B reports the code of the evaluation function, providing the output of a specified device for the required input combination.

Also, by scrutinizing the LUTs of various devices, it emerges a remarkable similarity for input combinations yielding the same logic. As an example, Table 4.5 reports the LUT of the *Inv* gate. The uniformity of the output values is a direct consequence of the termination attachment making information bistable. For this reason, the number of entries of the LUT can be reduced without losing precision on the predicted output.

Table 4.5 LUT describing the input/output behavior of a two-line inverter (*Inv*). All the values are expressed in volts.

$V_{IN_1A_H}$	$V_{IN_1B_H}$	$V_{IN_1A_L}$	$V_{IN_1B_L}$	$V_{OUT_1A_H}$	$V_{OUT_1B_H}$	$V_{OUT_1A_L}$	$V_{OUT_1B_L}$
-1.0000	+1.0000	-1.0000	+1.0000	+0.3453	-0.7601	+0.8578	-0.3568
-0.7777	+0.7777	-0.7777	+0.7777	+0.3453	-0.7601	+0.8578	-0.3568
-0.5555	+0.5555	-0.5555	+0.5555	+0.3453	-0.7601	+0.8578	-0.3568
-0.3333	+0.3333	-0.3333	+0.3333	+0.3453	-0.7601	+0.8577	-0.3568
-0.1111	+0.1111	-0.1111	+0.1111	+0.3454	-0.7600	+0.8581	-0.3566
+0.1111	-0.1111	+0.1111	-0.1111	-0.8212	+0.3628	-0.3297	+0.8222
+0.3333	-0.3333	+0.3333	-0.3333	-0.8212	+0.3628	-0.3297	+0.8222
+0.5555	-0.5555	+0.5555	-0.5555	-0.8211	+0.3628	-0.3297	+0.8222
+0.7777	-0.7777	+0.7777	-0.7777	-0.8212	+0.3628	-0.3297	+0.8222
+1.0000	-1.0000	+1.0000	-1.0000	-0.8212	+0.3628	-0.3297	+0.8222

In order to validate the characterization procedure, we tested it on some simple devices built with interconnections of MV and inverters that we characterized in the previous section. Specifically, we used the NAND/NOR, the XOR, the XNOR, and the MUX21 gates.

Fig. 4.11 shows the layout of a NAND/NOR gate, which is composed of an MV followed by an inverter. As for the MV, by fixing one of the majority voter inputs we obtain either a NAND gate (fixing a ‘0’) or a NOR gate (fixing a ‘1’). The global truth table is reported in Table 4.6.

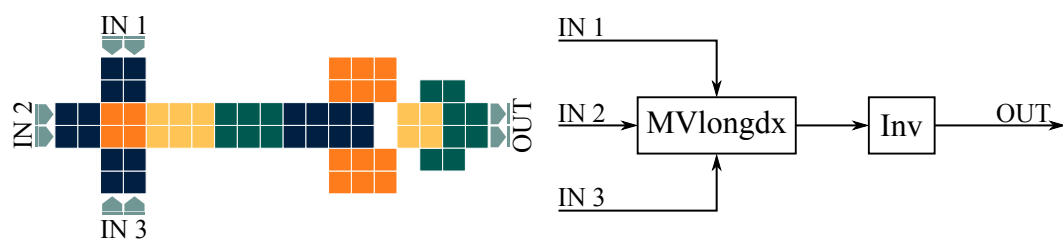


Fig. 4.11 Layout of the NAND/NOR gate with a schematic representation of the fundamental devices composing it.

First, we simulated the device in SCERPA for all the input combinations to have a reference. The simulation is performed with $N_clock_step = 5$ and $V_{in_MAX} = 1$ V as for fundamental gates and interconnections. However, in this case, we are interested solely in the functional behavior, not in the complete characterization, so we simulate only one voltage value for each combination, setting $N_driver_step = 1$. This situation corresponds to testing just one voltage value for each combination to obtain a basic logic correctness check of the truth table. The SCERPA simulation for the

Table 4.6 Truth table of the NAND/NOR gate, if IN 1 is fixed to ‘0’ than we have the NAND behavior (top half), if IN 1 is fixed to ‘1’ than we obtain a NOR gate (bottom half).

IN 1	IN 2	IN 3	OUT
0	0	0	1
0	0	1	1
0	1	0	1
0	1	1	0
1	0	0	1
1	0	1	0
1	1	0	0
1	1	1	0

eight combinations required 24 min and 8 s. For completeness, we generate the library files based on this simulation, which will present a large granularity between inputs. The time required for the library files generation is comparable to the one seen for the fundamental gates and interconnections, being 0.0838 s. The characterization time is still comparable, demonstrating that the overhead needed to create the library files is negligible.

We then compare the results provided by SCERPA for the whole device with the result obtained using the library. Starting from the input values we want to test, we access the library for the first time to read the corresponding output of the *MVlongdx* gate. Then, we use the output provided by the first access to the library to interrogate it again for the *Inv* gate, which gives us the final output voltage set. Since the device is multi-line, the output is composed of four voltages. As an example, equation (4.3) reports the output voltages of the four involved molecules obtained from SCERPA, and equation (4.4) reports the output from the library, for the input combination with all zeros considered with maximum voltage ($V_{inMAX} = \pm 1$ V).

$$OUT_{SCERPA} = \begin{pmatrix} 0.3453408 & -0.7601388 \\ 0.8577583 & -0.3568273 \end{pmatrix} \text{V} \quad (4.3)$$

$$OUT_{Lib} = \begin{pmatrix} 0.3453358 & 0.7601387 \\ 0.8577604 & 0.3568268 \end{pmatrix} \text{V} \quad (4.4)$$

In order to validate the accuracy of the BBchar procedure, we evaluate the minimum, maximum, and average values of the difference between OUT_{SCERPA} and OUT_{Lib} for each input combination. Table 4.7 lists the obtained values for the errors, and Fig. 4.12 provides a graphical representation of the same values.

The maximum error observed across all input combinations is on the order of $100 \mu\text{V}$. Considering that typical voltage values on molecules align closely with those presented in equations (4.3) and (4.4), mean errors fall within the range of 0.001 %. This remarkably low value underscores the extreme accuracy of the methodology proposed for device modeling and the reliable functioning of the BBchar tool. Moreover, achieving such precision comes with minimal time overhead. The time required for the library evaluation of all input combinations amounts to a few tens of milliseconds, rendering it negligible compared to the characterization time and the simulation of the entire device.

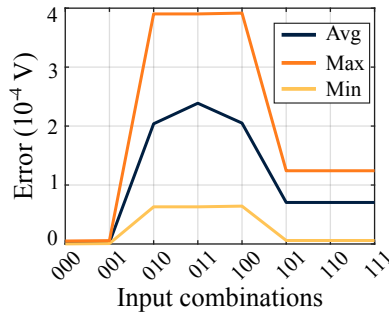


Fig. 4.12 Average, minimum, and maximum errors for the NAND/NOR gate between the output voltages calculated directly with SCERPA and using the library files.

The same procedure has been applied to the other tested devices, namely the XOR, XNOR, and MUX21 devices. Fig. 4.13 shows their layout with the subdivision into smaller blocks taken from the library.

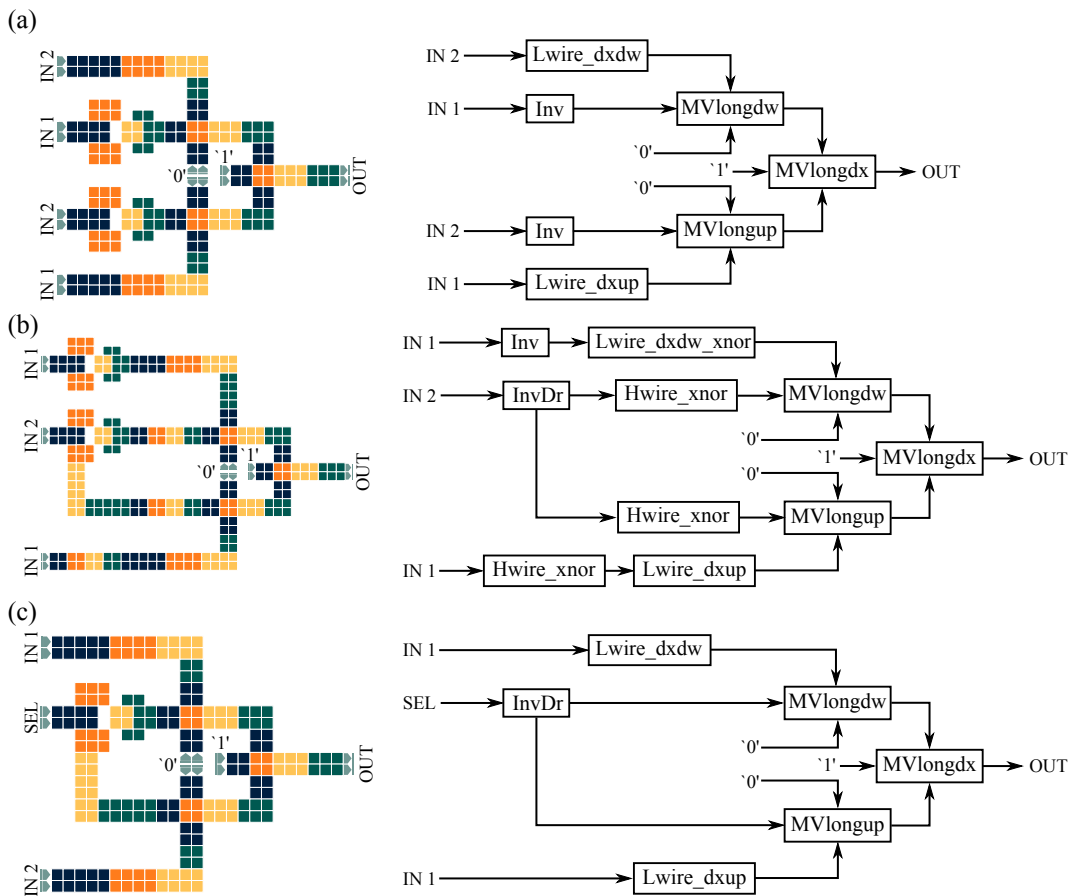


Fig. 4.13 Layout of the tested gates with a schematic representation of the fundamental device composing them. (a) XOR gate. (b) XNOR gate. (c) MUX21 gate.

Table 4.7 Average, minimum, and maximum errors for each input combination, evaluated on the molecules composing the output of the tested devices.

Gate	error	Input combinations			
		000	001	010	011
NAND	avg	1.9×10^{-6}	2.4×10^{-6}	2.0×10^{-4}	2.4×10^{-4}
	min	1.0×10^{-7}	7.0×10^{-7}	6.3×10^{-5}	6.3×10^{-5}
	max	5.0×10^{-6}	5.7×10^{-6}	3.9×10^{-4}	3.9×10^{-4}
NOR		100	101	110	111
	avg	2.0×10^{-4}	7.0×10^{-5}	7.0×10^{-5}	7.0×10^{-5}
	min	6.4×10^{-5}	5.9×10^{-6}	5.9×10^{-6}	5.9×10^{-6}
	max	3.9×10^{-4}	1.2×10^{-4}	1.2×10^{-4}	1.2×10^{-4}
MUX21 ($s = 0$)		000	001	010	011
	avg	4.3×10^{-5}	4.6×10^{-5}	1.2×10^{-4}	1.2×10^{-4}
	min	1.4×10^{-6}	2.6×10^{-5}	2.7×10^{-5}	3×10^{-6}
	max	1.5×10^{-4}	1.6×10^{-4}	2.2×10^{-4}	4.0×10^{-4}
MUX21 ($s = 0$)		100	101	110	111
	avg	6.9×10^{-5}	9.5×10^{-5}	4.5×10^{-5}	9.4×10^{-5}
	min	1.7×10^{-5}	1.2×10^{-5}	1.2×10^{-5}	1.3×10^{-5}
	max	1.1×10^{-4}	3.1×10^{-4}	7.6×10^{-5}	2.4×10^{-4}
XOR		00	01	10	11
	avg	1.3×10^{-4}	1.2×10^{-4}	6.3×10^{-5}	4.5×10^{-5}
	min	2.9×10^{-5}	4.1×10^{-6}	1.6×10^{-5}	8.0×10^{-7}
	max	2.4×10^{-4}	3.9×10^{-4}	1.0×10^{-4}	1.6×10^{-4}
XNOR		00	01	10	11
	avg	6.6×10^{-5}	4.2×10^{-5}	1.1×10^{-4}	9.4×10^{-5}
	min	1.7×10^{-5}	1.8×10^{-6}	2.5×10^{-5}	1.3×10^{-5}
	max	1.1×10^{-4}	1.5×10^{-4}	2.0×10^{-4}	2.4×10^{-4}
XOR_2		00	01	10	11
	avg	2.2×10^{-2}	1.3×10^{-2}	1.3×10^{-2}	2.2×10^{-2}
	min	6.2×10^{-3}	4.0×10^{-3}	3.9×10^{-3}	6.0×10^{-3}
	max	5.6×10^{-2}	2.6×10^{-2}	2.6×10^{-2}	5.6×10^{-2}

We compared the output voltages obtained from the SCERPA simulation of the entire device and the values obtained by the interconnection of smaller gates using the library files. Even in this case, the simulation of the whole device in SCERPA has been performed setting $N_clock_step = 5$, $V_{in_MAX} = 1$ V, and $N_driver_step = 1$, meaning that we only test the logic functionality of these devices.

Table 4.8 Simulation time to validate the truth table of the gate used for testing the BBchar methodology, their cell counts, and the respective overhead time for creating the library files. Simulation are performed with $N_clock_step = 5$, $N_driver_step = 1$ and $V_{inMAX} = 1$ V.

Gate	Sim. time	# cells (-)	Lib. gen. time (s)
NAND/NOR	24 min 8 s	60	0.0838
MUX21	6 h 40 min 8 s	198	0.1286
XOR	4 h 19 min 59 s	208	0.1375
XNOR	9 h 5 min 9 s	280	0.1716

Starting with the same combination tested in SCERPA for the whole device, we follow the signal through the intermediate gates and use the library files to predict the values of the cascade layouts. The simulations for testing each combination using the library files last just a few milliseconds for all the devices, confirming the substantial enhancement of the simulation time using the proposed methodology. The improvement is substantial, as the simulation in SCERPA only for logic functionality last hours for each of the three devices. Table 4.8 lists the SCERPA simulation time and the cell counts, which also substantially increases with respect to the cell count of the NAND/NOR gate. For the sake of completeness, we perform the characterization of these devices to have their library files, and the last column of Table 4.8 reports the negligible time required.

Even in this case, we calculate the voltage error on output molecules to verify the precision of the simulation using the proposed device model. The obtained values for each combination are listed in Table 4.7 and graphically shown in Fig. 4.14. The errors are of the same order of magnitude as the NAND/NOR gate, confirming the accuracy of 0.001 %.

As a final validation, we implemented an XOR gate made with NAND gates, named XOR_2 hereafter and shown in Fig. 4.15 with the schematic representation of the fundamental device composing it. The NAND gate composing it is a bit longer than the one presented before, due to signal synchronization, and is here called NAND_XOR2, built up with the interconnection of an MV_xor2nand and an Inv_xor2nand. The time required for a complete characterization of the NAND_XOR2 with $N_driver_step = 10$, as we did for elementary gates and interconnections, becomes prohibitive. We want to test the accuracy of the procedure when one of the internal gates has a less granular LUT. The LUT of the

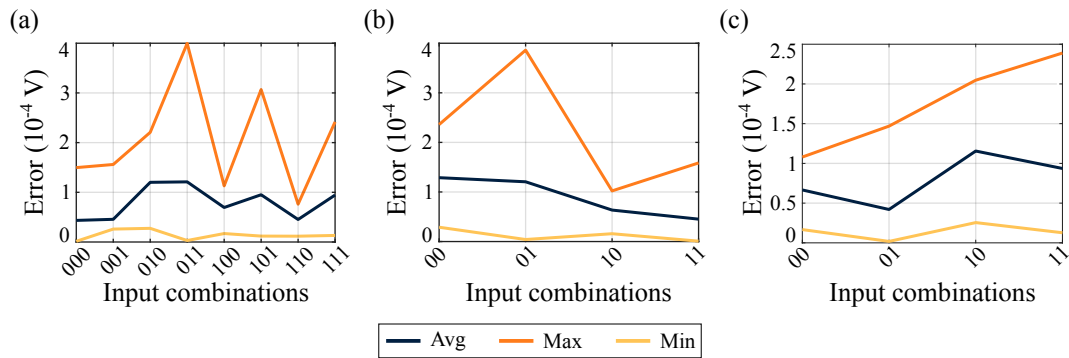


Fig. 4.14 Average, minimum, and maximum errors for the tested devices between the output voltages calculated directly with SCERPA and using the library files. (a) MUX21. (b) XOR. (c) XNOR.

NAND_XOR2 only comprises the eight entries for the input combinations obtained with the maximum voltage ($V_{in_{MAX}} = 1$ V).

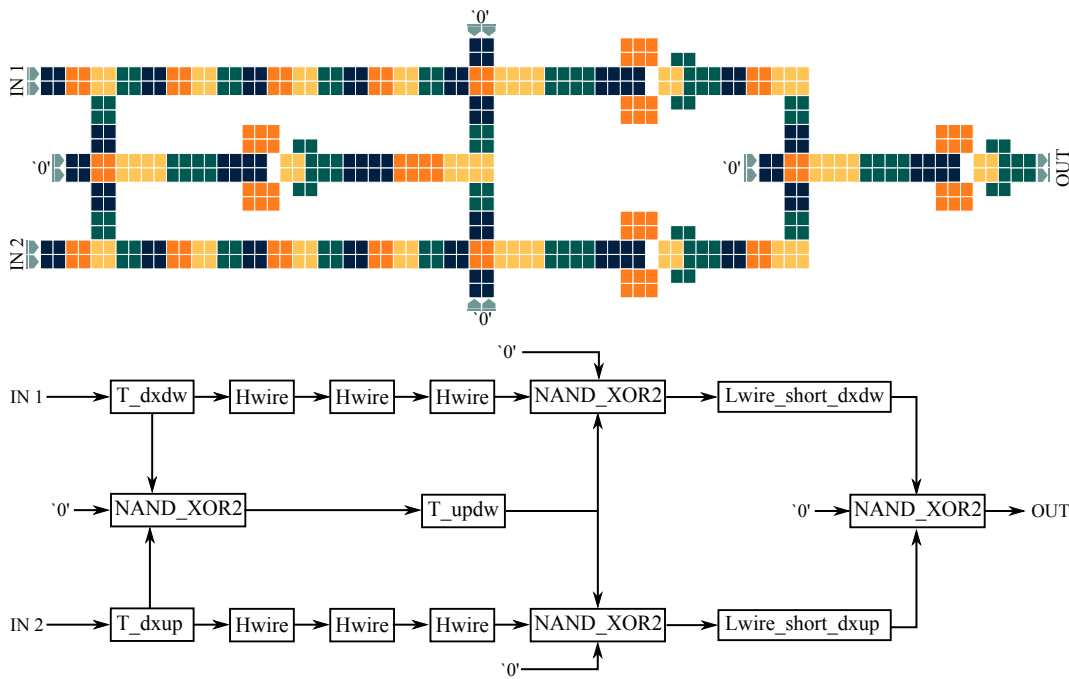


Fig. 4.15 Layout of the XOR₂ gate with a schematic representation of the fundamental device composing it.

The XOR₂ gate is made with 468 cells, and the time required for the SCERPA simulation of the four input combinations calculated with the maximum voltage required 1 d16 h43 min34 s. Even with this large design, the time required to generate the library files is comparable to others (0.2724 s). The time needed to calculate the

output using the library files is a few tens of milliseconds, demonstrating once again the huge improvement in terms of simulation time.

We test the accuracy of the library-based simulation by calculating the error with respect to the SCERPA simulation. Fig. 4.16 depicts the obtained result and the exact values are reported in Table 4.7.

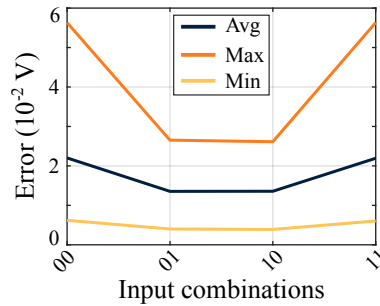


Fig. 4.16 Average, minimum, and maximum errors for the XOR_2 gate between the output voltages calculated directly with SCERPA and using the library files.

The device shows errors two orders of magnitude higher than other devices. The origin of that has to be found in the poor characterization of the NAND_XOR2 device since we only stored its output voltages for the maximum input voltages. Nevertheless, the mean error in the prediction of the XOR_2 gate is around 1 %, which is still acceptable, demonstrating the robustness of the proposed methodology to poor characterizations. The case of the XOR_2 gate also demonstrates that it is unnecessary to divide larger designs into the smallest possible gates with a fine characterization. Indeed, errors are negligible even if we do not use densely populated library files, simplifying the future design process and simulation of architectures.

Chapter 5

Interface with the external world

Previous chapters deeply investigate with a simulative approach what we can expect from a molecular Field-Coupled Nanocomputing (molFCN) device, for example depending on the molecule used, and describe the effective model at different abstraction levels that let us make such predictions. The first level is related to the single molecule study and modeling to validate and simplify the molecular behavior (Chapter 2). The second abstraction level deals with the electrostatic interaction of a bunch of molecules, and uses Self-Consistent Electrostatic Potential Algorithm (SCERPA) to validate the expectation from the energetic analysis (Chapter 3). SCERPA proves physically accurate and provides a set of various functionalities that address possible process variations [136]. The highest abstraction level describes how to effectively model entire devices and reuse simulation data in larger designs including them (Chapter 4). The simulations performed with Block-Based characterization (BBchar) demonstrate the combination of high accuracy with the prediction made by SCERPA, while strongly cutting simulation time.

All this work helps delineate the safe working space, excluding non-interacting molecules, or non-propagating layouts, simplifying the experimental investigation on the molFCN technology. It is established that experimental validations of the technology are hard to obtain, as introduced in Chapter 1 because of the combination of several challenging aspects. One major hurdle is the difficulty in accurately measuring molecule charges for validation purposes, which is the core procedure of experimental validation.

This chapter proposes a novel modified implementation of the molFCN paradigm, that builds upon the strengths of molFCN while addressing the challenges of reading the stored information onto a molecule [137]. The approach is named bend-boosted molFCN and exploits the spatial arrangement of fluctuating molecules to enhance information encoding capabilities. Through Density Functional Theory (DFT) simulations using 6-(ferrocenyl)hexanethiol cations, this chapter demonstrates the feasibility of this paradigm and its potential for practical implementation.

This chapter also discusses the compatibility with the existing molFCN paradigm, that here we call charge-based to differentiate it. By exploiting a molecular junction built with a 6-(ferrocenyl)hexanethiol cation, and thanks to the successful integration with charge-based molFCN, the proposed system establishes a pathway for the realization of a read-out system. This read-out mechanism not only enables experimental validation of molFCN principles but also contributes to the integration of molFCN with CMOS technology.

5.1 A new paradigm for molecular FCN

Despite its potential advantages, the practical realization of a functional prototype of molFCN faces several challenges. The obstacles include aspects from different areas of expertise. First of all, the synthesis of tailored molecules is difficult, and up to today is still missing a set of working molecules presenting the minimum requirements. Another strong challenge is related to the difficulty of precisely fabricating nanopatterned electrodes or trenches to host the molecules, and depositing molecules onto these nanopatterned structures increases even more the task difficulty even more. Finally, it's challenging to read the charge-encoded logic information, because of the non-idealities of the possible sample and because we want the measure to not back influence the sample.

All these aspects are of great interest to researchers, and some innovative approaches have been proposed to address the challenges. Novel molecules are frequently proposed, as discussed in 2. To avoid the need for nanopatterned structures, we previously proposed to directly use uniform self-assembled monolayers for computing purposes [99]. Additionally, advancements in scanning-probe techniques are promising for in-lab measurement of the information stored in the molecule charge distribution [81, 80]. Another solution could be the use of Raman scattering to detect

the state of the molecule at the laboratory level [138]. Regarding in-circuit reading, authors in [139] proposed to use asymmetric mixed-valence molecules to detect information. Besides the promising and interesting results over different aspects, remains challenging to put things together and have a solution that inglobates several of these aspects.

The following sections discuss a novel molFCN implementation, named bend-boosted molFCN, aimed at simplifying in-circuit and in-lab read-out operations. The laying idea is to associate the information with different molecule orientations. Different orientations mean different spatial occupations depending on the information. This association could simplify in-lab read-out, as it is established that is possible to perform atomic-precise topographic measurements [140–142]. Additionally, the proposed principle can be integrated with a molecular junction, serving as a single-molecule sensor [143].

5.1.1 Conceptual vision of the new paradigm

Starting again from the information encoding in molFCN, we associate a logic value to the charge localization in a molecule. By applying an electric field, the charge moves to different regions of the molecule, but the molecule stays ideally fixed in the same position, as reported in Fig. 5.1a with bisferrocene molecule. In the bend-boosted paradigm, the molecule is no more fixed, but able to move as a sort of inverted pendulum, with the pivot placed on a substrate.

As an example, we here use the 6-(ferrocenyl)hexanethiol cation ($\text{Fc}^+\text{C}_6\text{S}$) as a reference molecule, shown in Fig. 5.1b. $\text{Fc}^+\text{C}_6\text{S}$ is a molecule known for its unique ability to form uniform Self-Assembled Monolayers (SAM)s on various substrates [144]. This compound consists of a ferrocene unit attached to a hexanethiol chain, enabling it to form robust and well-ordered monolayers through chemisorption on metal surfaces such as gold and silver. The ferrocene moiety provides stability and control over the monolayer structure, while the hexanethiol chain facilitates the attachment to the substrate, ensuring uniform coverage.

Fig. 5.1c depicts how the application of an external electric field affects the molecule if we imagine it anchored to a substrate. Being the molecule positively charged, the presence of a uniform field aligns the molecule with the field lines, but since the molecule is anchored, the resulting movement is bending. The sensibility

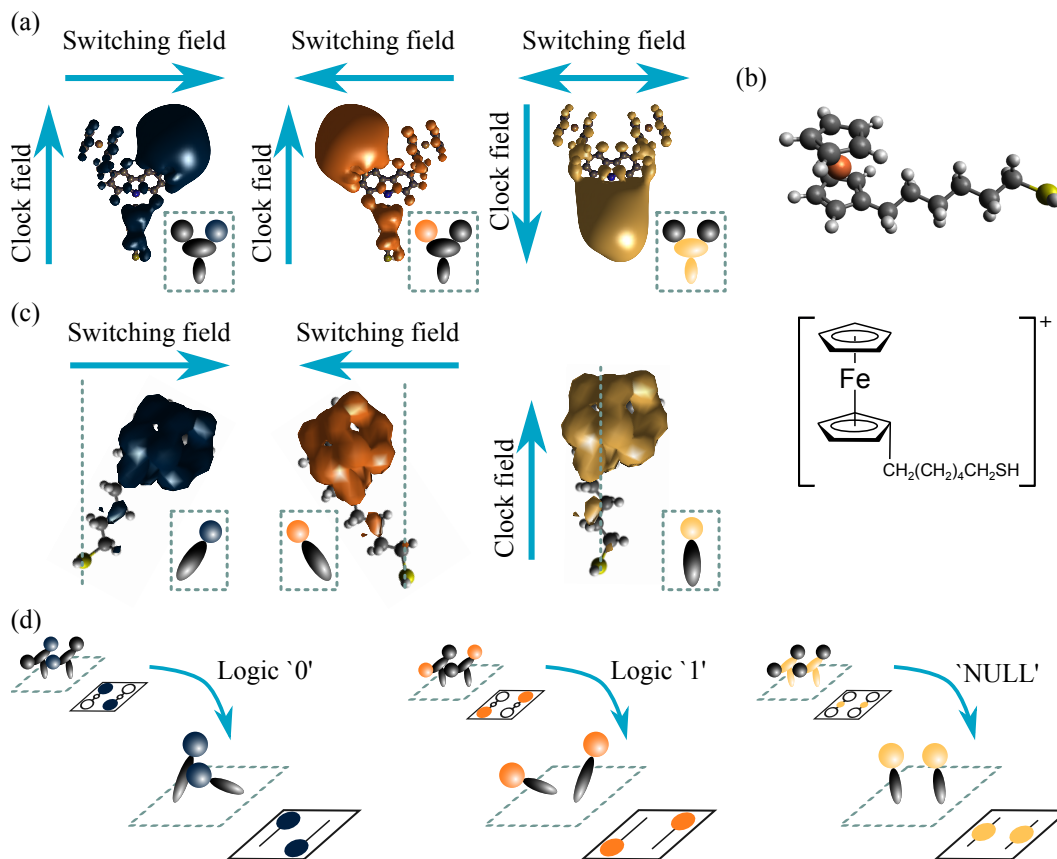


Fig. 5.1 Representation of the conceptual behavior of the bend-boosted molFCN. (a) Charge localization on the bisferrocene molecule varying the applied fields. (b) 3D geometry shown in a ball-and-stick diagram and chemical structure of the $\text{Fc}^+\text{C}_6\text{S}$ cation. (c) Charge localization and bending of the $\text{Fc}^+\text{C}_6\text{S}$ cation varying the applied fields. (d) Unit cell in the three logic states, compared to the charge-based unit cell to show the equivalence from a conceptual point of view. Atoms in figure are colored as follows: H = white, C = charcoal, N = blue, S = yellow, Fe = orange.

of the molecule to an external field guarantees the write-in operation, fundamental for driving circuits.

When we place two of these molecules next to each other, the presence of the oxidation charge will make them repel, so they'll bend in opposite directions, as shown in Fig. 5.1d. Assuming that the oxidation charge localizes on the ferrocene groups of the two $\text{Fc}^+\text{C}_6\text{S}$, we obtain a charge configuration equivalent to a Quantum-dot Cellular Automata (QCA) cell. The similarity of the charge behavior leads to the same logic encoding as the charge-based molFCN, so from a logical point of view, we can assume a similar gate design.

To validate the conceptual working, the next two sections analyze the isolated $\text{Fc}^+\text{C}_6\text{S}$ as a case of study. The results aim to validate the working principle and the encoding capabilities, proving the ground base for the development of the bend-boosted molFCN paradigm.

5.1.2 Theoretical background

The bend-boosted molFCN paradigm is based on molecular movements, meaning that we need to extend the simulation of a single molecule to the time domain. We exploit two computational techniques: Ab Initio Molecular Dynamics (AIMD) and classical Molecular Dynamics (MD).

AIMD is a computational technique used to simulate the dynamic behavior of atoms and molecules from first principles, without the need for empirical force fields [145].

In AIMD simulations, the positions and velocities of atoms are propagated forward in time using Newton's equations of motion, while the electronic structure is updated at each time step based on the instantaneous atomic configurations. These simulations are also known as Born-Oppenheimer molecular dynamics simulations. This is because they approximately solve the time-independent Schrödinger equation to calculate gradients, which are then used to guide the movement of the atoms [146].

However, AIMD also comes with computational challenges. The computational cost scales with the system size, limiting its applicability to small systems.

Unlike AIMD, classical MD simulations rely on the principle of Newtonian mechanics, where the forces acting on each atom are determined based on their interactions with neighboring atoms [147]. These interactions are typically described using interatomic potentials, which can be derived from empirical observations or calculated from quantum mechanical principles. Common types of interatomic potentials include pairwise potentials, such as Lennard-Jones or Coulombic potentials, as well as more sophisticated potentials like the reactive force field [148].

MD simulations are more computationally favorable with respect to AIMD methods, with the accuracy dependent on the accuracy of the interatomic potentials used.

In this context, we first simulate the behavior of the $\text{Fc}^+\text{C}_6\text{S}$ with AIMD to validate the conceptual behavior. Subsequently, we compare the obtained result with a more simplistic MD simulation to verify the coherence of the results. The possibility of using MD for technological validation would release some of the computational complexity typically needed for molecular simulations.

5.1.3 Single-molecule simulation with *ab-initio* techniques

As already mentioned, we use the $\text{Fc}^+\text{C}_6\text{S}$ cation as a reference molecule to validate the working principle. The molecule is shown in Fig. 5.2a in a ball-and-stick diagram, highlighting the two functional groups *Dot 1* and *Dot 2* associated with the ferrocene and the hexanethiol, respectively. The coordinates have been obtained from the geometry optimization of the isolated molecule with the Unrestricted Kohn-Shann (UKS) method using B3LYP functional with def2-TZVP basis set [119]. Grimme D3 correction is applied [120, 121].

The charge distribution is analyzed by exploiting the ESP-derived CHarges from ELeCtrostatic Potentials using a Grid (CHELPG) paradigm. Following the same procedure as in Chapter 2, we use the ORCA tool to compute the electrostatic potential produced by the molecule and shown in Fig. 5.2b, which is radial because we are analyzing the oxidized molecule. The maximum is located on the ferrocene group, suggesting where the oxidation charge localizes. Fig. 5.2c shows the ESP charges on each atom, presenting maximum values on the carbon atom of the

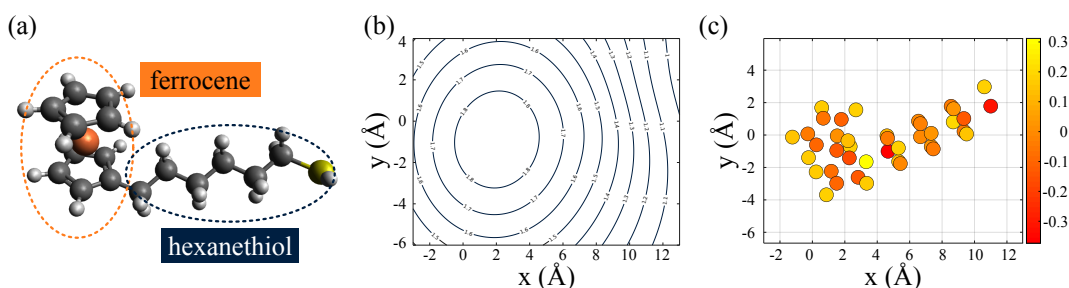


Fig. 5.2 Analysis of the isolated $\text{Fc}^+\text{C}_6\text{S}$ cation at equilibrium through DFT simulation. (a) 3D geometry shown in a ball-and-stick diagram of the relaxed molecule highlighting the atom groups for the calculation of the aggregated charges. (b) Molecular electrostatic potential evaluated with DFT calculation, the solid blue lines are isopotential curves. (c) ElectroStatic Potential (ESP) charges on each atom of the molecule. Atoms in figure are colored as follows: H = white, C = charcoal, S = yellow, Fe = orange.

pentacene connecting to the hexane. On the contrary, the most negative charge appears on the sulfur atom of the thiol group. The confirmation of the positive charge aggregation is given also by the charge distribution shown in Fig. 5.1c, where we show the conceptual working principle of the bend-boosted molFCN paradigm.

By calculating the Aggregated Charge (AC) value on the ferrocene and the hexanethiol we obtain the following values

$$\begin{cases} Q_{Fc} = 0.9595 \text{ a.u.} \\ Q_{C6S} = 0.0405 \text{ a.u.} \end{cases} \quad (5.1)$$

confirming the positive charge aggregation on the ferrocene and the electrostatic neutrality of the hexanethiol, as already demonstrate in Chapter 2.

Because of the charge distribution, when the molecule is subjected to an electric field and anchored to a substrate through the thiol, we can expect that the movement is dictated by the ferrocene. The hexane, being neutral, would behave as a bar holding the ferrocene.

To study the dynamic behavior of the isolated Fc^+C_6S cation we exploit AIMD and we apply an electric field in time to the molecule. The simulation of the molecule in the presence of constant electric fields is crucial to validate the working principle. As already mentioned in Chapter 2, electric fields are applied in ORCA by positioning point charges in space. Fig. 5.3 shows a 3D representation of the system consisting of the Fc^+C_6S cation and the point charges used for initiating the AIMD simulations.

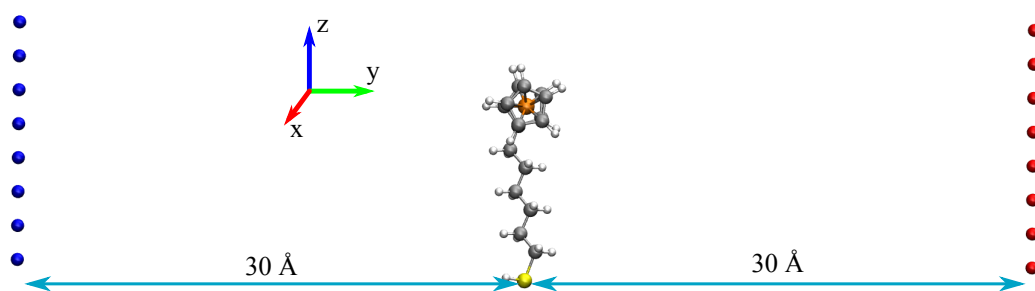


Fig. 5.3 The Fc^+C_6S cation alongside the point charges employed for generating the electric field in AIMD simulations. Negative point charges are denoted by blue dots, while positive ones are indicated by red dots. This configuration yields a positive uniform electric field aligned along the y-axis. The molecule has the sulfur atom placed in the origin of the axis. Atoms in figure are colored as follows: H = white, C = charcoal, S = yellow, Fe = orange.

Table 5.1 Values and coordinates of the point charges used to generate a uniform electric field along the positive y-axis equal to 4.9451 V nm^{-1} .

Value (a.u.)	x (Å)	y (Å)	z (Å)
+2	0.0	+30.0	1.0
+2	0.0	+30.0	3.0
+2	0.0	+30.0	5.0
+2	0.0	+30.0	7.0
+2	0.0	+30.0	9.0
+2	0.0	+30.0	11.0
+2	0.0	+30.0	13.0
+2	0.0	+30.0	15.0
-2	0.0	-30.0	1.0
-2	0.0	-30.0	3.0
-2	0.0	-30.0	5.0
-2	0.0	-30.0	7.0
-2	0.0	-30.0	9.0
-2	0.0	-30.0	11.0
-2	0.0	-30.0	13.0
-2	0.0	-30.0	15.0

Referring to the initial position of the molecule as identified in Fig. 5.3 with the S atom placed in the origin, Table 5.1 lists the values and positions of the point charges used to generate a uniform electric field along the positive y-axis.

The value of the electric field generated is calculated at the geometrical center ($x = 0.3145 \text{ Å}$, $y = 0.0549 \text{ Å}$, $z = 7.3003 \text{ Å}$) of the molecule. It is computed as the sum of contributions from all point charges, as expressed by equation (5.2), where N_{pc} is the total number of point charges, ϵ_0 is the vacuum permittivity, q is the electron charge, pc_i is the value of the i -th point charge, and r_i is the distance between the i -th point charge and the molecule geometrical center.

$$E(x, y, z) = \sum_{i=1}^{N_{pc}} \frac{1}{4\pi\epsilon_0} \frac{q \cdot pc_i}{r_i^2} \quad (5.2)$$

Using the charge configuration specified in Table 5.1, we obtain $E_x = 0.0003 \text{ V nm}^{-1}$, $E_y = -4.9451 \text{ V nm}^{-1}$, and $E_z = -0.0006 \text{ V nm}^{-1}$. Consequently, the dominant field is along the y-axis, and the other components can be neglected. By

following the same procedure but changing the sign of the point charges, a positive uniform electric field equal to $E_y = 4.9451 \text{ V nm}^{-1}$ can be obtained.

We thus perform two AIMD simulations, applying an electric field equal to $\pm 4.9451 \text{ V nm}^{-1}$, respectively. The value of the electric field is high with respect to typical fields we used in DFT simulation in Chapter 2. The choice helps limiting the computational time required by the simulation of the complete molecule bending. The next section will verify the paradigm under reduced electric fields and extended durations using classical molecular dynamics simulations.

In Fig. 5.4 is represented the molecule behavior obtained from the AIMD simulations for both the positive and negative field. The blue dots are the atoms in the initial position, corresponding with the 'NULL' state. The simulation reproduces the case in which the clock field is instantaneously released and the molecule can thus bend on one side or another depending on the surrounding electric field. During the simulation, the thiol group is fixed to the same coordinates, emulating the anchoring to a substrate placed on the plane $z = 0$. After 3 ps, the molecule reaches the position represented with the orange dot configuration, and the iron atom of the ferrocene follows the dotted light blue line.

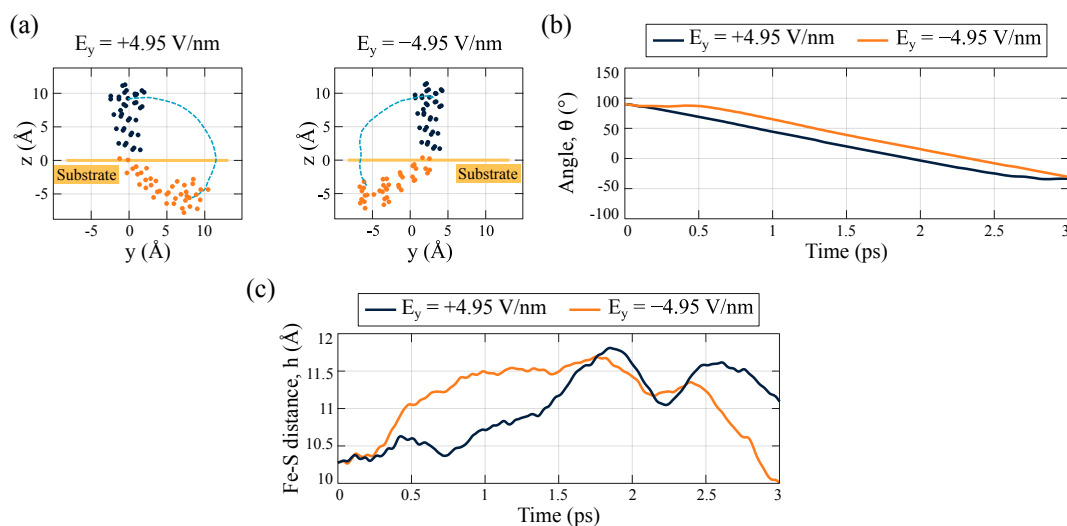


Fig. 5.4 Results of the AIMD simulation of the $\text{Fc}^+\text{C}_6\text{S}$ cation. (a) Simulation with a constant electric field ($E_y = \pm 4.95 \text{ V nm}^{-1}$). The light blue dotted line represents the trajectory followed by the Fe atom from the starting position (blue dots) and after 3 ps (orange dots). The molecule rotates around the thiol group $-\text{SH}$, fixed on the fictitious substrate line. (b) Temporal evolution of the angle (θ) between the Fe-S segment and the plane $z = 0$ where the molecule is anchored. (c) Temporal evolution of the Fe-S distance (h) during the AIMD calculation.

When the applied field is positive, the molecule bends clockwise, whereas the negative field bends the molecule counterclockwise around the anchoring point. Since the simulation are performed with the isolated molecule, the molecule can overcome the plane $z = 0$, as the absence of the substrate is reflected in a missing repulsive force that blocks the molecule. The molecule behaves similarly for both fields and the angular rotation has the same amplitude, demonstrating encoding capability and equivalent movements.

Fig. 5.4b illustrates the progression of the angle (θ) between the Fe–S segment and the substrate line ($z = 0$) over the course of the simulation. The angle measurements are referenced to the positive y-axis for clockwise rotation and the negative y-axis for counterclockwise rotation to ensure comparability. At its maximum value ($\theta = 90^\circ$), the molecule is in the ‘NULL’ state, perpendicular to the substrate. The graph in Fig. 5.4b showcases a linear bending dynamic, where the molecule undergoes controlled rotation. The rotational speed is almost identical between different electric field directions, measuring approximately 0.0528 deg/fs for the negative field value and 0.0522 deg/fs for the positive one. The curve for the angle also show a small plateau at the beginning or at the end of the simulations, modeling a sort of inertia of the molecule to variation from the equilibrium position. The difference in the plateau can be a consequence of the asymmetry of the anchoring point, as the molecule show resistance to bend on the side where there is the hydrogen atom of the thiol, i.e., toward negative y-axis.

Fig. 5.4c provides insights into the Fe–S distance (h) during the simulation. The average strains recorded for the two electric fields are 1.1131 Å and 1.1026 Å, respectively. Comparatively, the equilibrium value is $h = 1.0283$ nm, indicating relative strains of 10.8% and 10.7%, respectively. These results suggest that the molecule experiences minimal strain during the rotational movement, underscoring its rigid-bar behavior. This observation is significant as it emphasizes the structural stability of the hexanethiol under the influence of external electric fields, which is crucial for its functional integrity. The results also confirms the expectations of the hexanethiol behaving as a rigid-bar connecting the ferrocene to the substrate.

For the paradigm definition, the encoding dynamic range is delineated by the angles $\theta = 90^\circ$ and $\theta = 0^\circ$. We adapt to the traditional definition of rising (falling) time as defined for CMOS technology, i.e., the time required for a transition between 10% and 90% (90% and 10%) of the full dynamic. Thus, we define the encoding time

(τ_e) as the time required to transition from $\theta = 81^\circ$ to $\theta = 9^\circ$, representing 90% and 10% of the total encoding dynamic, respectively. For the performed simulations, the encoding time is $\tau_{e,cw} = 1381.4180$ fs for the positive field (clockwise rotation) and $\tau_{e,ccw} = 1421.6907$ fs for the negative field (counterclockwise rotation). On average, the molecule switches in $\bar{\tau}_e = 1401.5544$ fs.

To estimate the maximum switching frequency, we consider driving the molecule from the 'NULL' state ($\theta = 90^\circ$) to encode a specific logic value and then resetting it to replicate an entire switching operation. For instance, we imagine the molecule tilting to $\theta = 0^\circ$ with a clockwise rotation and subsequently returning to the standing configuration with a counterclockwise rotation.

To complete the switching dynamic, we should simulate also the reset condition, where the molecule is initially laying on the substrate ($\theta = 0^\circ$) and we apply the clock field to make the molecule standing up again. For consistency, from that simulation we could extract the reset time (τ_r) as the time required to transition the molecule from 9° to 81° . However, from the simulation described previously, we obtained very similar encoding times for both rotations, where differences are due to constraints on the thiol group. Given the similarity between the encoding time, we can approximate the reset time to be equivalent to the switching time corresponding to the same rotation. In simpler terms, if the encoding time is $\tau_{e,cw}$, the reset time is approximately $\tau_{e,ccw}$, and vice versa. This approximation allows us to have a first-order approximation of the expected frequency at which the molecule can switch states, crucial for assessing the operational limits and performance of molecular-scale information processing systems.

Considering the previous assumptions, the complete switching operation is performed in a time equal to $\tau_{e,cw} + \tau_{e,ccw}$, giving a switching frequency (f_{sw}) for $E_y = \pm 4.9451$ V nm⁻¹ of:

$$f_{sw} \approx \frac{1}{\tau_{e,cw} + \tau_{e,ccw}} = 356.75 \text{ GHz} \quad (5.3)$$

The obtained value provides an initial estimation of the achievable switching frequency. In addition to approximating the reset time as the encoding time with the same rotation, it's essential to note that the simulations are conducted without a substrate, which could potentially attract or repulse the molecule, thus impacting the effective dynamics by either accelerating or slowing down the process. Furthermore,

the encoding time calculation does not incorporate potential delays in the first and last femtoseconds, such as the time required for the molecule to move out of equilibrium commented before. Conversely, it's conceivable to achieve encoding with lower angles, allowing for a reduction in the switching time. This acknowledgment underscores the need for further refinement and consideration of various factors to provide a more accurate assessment of the switching frequency achievable in practical applications.

5.1.4 Single-molecule simulation using molecular mechanics

The results obtained with the AIMD calculations suggest the feasibility of utilizing electric fields to encode information through the bending of $\text{Fc}^+\text{C}_6\text{S}$. However, the simulations are done with quite high electric field values to reduce the computational time as already stated. Indeed, we can expect that the intensity of the applied electric field is directly related to the velocity of the molecule to cross the substrate line. Reducing the electric field would thus reduce the maximum obtainable switching frequency, although offering the potential for reducing associated power dissipation.

To explore this further while mitigating computational costs, we employ classical MD simulations, initially validating the approach with the same field values used in the AIMD simulations. Thus, starting from the same initial configuration, we conduct dynamic simulations over a period of 3 ps, applying electric fields of $E_y = 4.9451 \text{ V nm}^{-1}$ and $E_y = -4.9451 \text{ V nm}^{-1}$. Considering the results from the AIMD simulations, where the molecule exhibited rigidity during bending, we treat the molecule as a rigid system in the MD simulations. In these simulations, the only permitted movements are stretches of the S-molecule bond and rotations around the sulfur anchoring point, simulating its interaction with an imaginary substrate. This approach allows for a more detailed exploration of the dynamics of the system under lower electric field intensities, yet is an approximation of the actual complexity of the system.

Fig. 5.5a displays the overlay of the final configurations of the molecule obtained through both AIMD and MD simulations after 3 ps, along with the trajectory traced by the iron atom. Notably, the last configurations exhibit remarkable similarities between the two approaches, with minor variations attributed to the differences in molecular rigidity observed in MD simulations. Particularly, significant differences

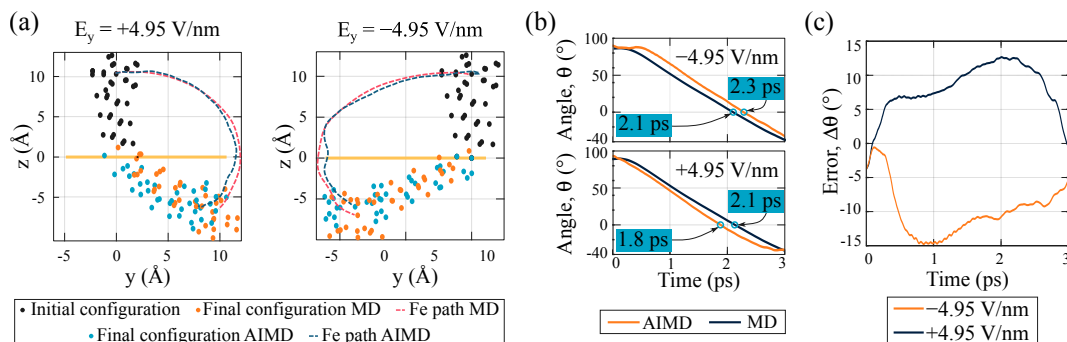


Fig. 5.5 Results of the MD simulation of the $\text{Fc}^+\text{C}_6\text{S}$ cation compared to the results obtained with AIMD. (a) Comparison of AIMD and MD trajectories of the $\text{Fc}^+\text{C}_6\text{S}$ cation when subjected to positive and negative electric fields. The blue/red dotted line represents the movement of the Fe atom in the AIMD/MD simulation from the starting position (black circles) and after 3 ps (light blue/orange circles). (b) Comparison of the angle created by the $\text{Fc}^+\text{C}_6\text{S}$ cation along the evolution of AIMD and MD simulations, with both a positive and a negative electric field. The points highlighted identify the time value where the angle θ equals 0° . (c) Absolute error between MD and AIMD in the angle created by the $\text{Fc}^+\text{C}_6\text{S}$ with the substrate for both positive and negative driving electric fields.

emerge when the molecule aligns with the y-axis, experiencing the maximum impact of the applied electric field, leading to slight deformations. In MD simulations, these deformations manifest primarily as elongations in the bond between S and the hexane, owing to the rigid nature of the rest of the system. Despite the simplified model utilized in MD, the trajectory closely mirrors the more detailed AIMD calculation, demonstrating the reliability of the results obtained from both simulation methods.

Fig. 5.5b provides further insights into the comparison by depicting the angle between the molecule and the $z = 0$ plane during both AIMD and MD simulations, with emphasis on the points where $\theta = 0^\circ$. Remarkably, the MD curves exhibit almost identical behavior for both positive and negative fields due to the rigid nature of the system. Additionally, the steepness of the curves appears comparable in AIMD and MD for both field orientations. In the case of the two MD simulations, the encoding times are determined to be $\tau_{e,cw} = 1495.4204$ fs for the positive field (clockwise rotation) and $\tau_{e,ccw} = 1488.5061$ fs for the negative field (counterclockwise rotation), resulting in a lower switching frequency compared to AIMD. The switching frequency equals $f_{sw} \approx 335.129$ GHz, evaluated using equation (5.3).

The primary difference between AIMD and MD emerges in the initial femtoseconds, affecting the calculated switching time values. Specifically, in AIMD simulations, the initial resistance of the molecule to movement varies depending

on the direction of the applied field. Conversely, this initial resistance is absent in MD simulations due to the simplification of the model. This variation in the initial instants leads to discrepancies in the angle measurements. Fig. 5.5c illustrates the differences in the angle between AIMD and MD simulations for both positive and negative electric fields. Notably, during the transient phase, the error may be significant (approximately $\sim 15^\circ$), primarily attributed to the assumption of a rigid molecule model in MD. However, the trajectory followed by both AIMD and MD converges toward the end of the simulation, indicating the feasibility of employing MD to estimate the behavior of the paradigm with minimal loss in precision.

Since both AIMD and MD simulations yield consistent results, yet with MD simulations offering faster computation times, we leverage the latter to validate the behavior across various applied electric fields and assess the switching time. Fig. 5.6a collect the temporal variation of the angles formed by the molecule with the $z = 0$ plane for the tested field values. It is observed that reducing the field intensity decelerates the bending of the molecule and amplifies the discrepancies between opposite signs of the same field, while still preserving the fundamental principle of the paradigm. Notably, for smaller fields, the angle requires more time to traverse the $z = 0$ plane, where the hypothetical substrate is located.

The switching time, depicted in Fig. 5.6b, exhibits variation with the driving electric field as a consequence of the variation in the bending velocity. The dashed line represents the average encoding time $\tau_{e,\mu} = 2.1$ ps. It is evident that reducing the applied field intensity prolongs the switching time. For instance, with a field of $E_y = -0.4945 \text{ V nm}^{-1}$, the switching frequency is approximately $f_{sw} \approx 186.440 \text{ GHz}$, as calculated using equation (5.3), indicating that halving the applied field decreases the frequency by half. Furthermore, beyond a certain threshold, increasing the driving field results in a disproportionate increase in power consumption without yielding significant improvements in frequency performance. For instance, with $E_y = -7.4177 \text{ V nm}^{-1}$, the frequency is approximately $f_{sw} \approx 349.075 \text{ GHz}$, signifying that a 33% increase in the field only yields a roughly 4% increase in frequency.

To further explore the behavior of the molecule over an extended duration, we exploit MD to perform a simulation that surpasses the typical timeframes utilized in previous analyses. By applying an electric field along the y-axis, we expect the $\text{Fc}^+\text{C}_6\text{S}$ molecule to align with the substrate plane $z = 0$. Fig. 5.6c depicts the angle oscillations around the y-axis when the driving electric field is set to 2.48 V nm^{-1} .

The dashed lines delineate the region where oscillations settle below 9° , representing 10% of the encoding dynamic. This simulation corroborates the alignment of the molecule with the applied field. Furthermore, the angle exhibits characteristics indicative of an underdamped second-order system, with two complex conjugate roots, as it approaches the regime value.

Introducing a substrate is expected to impact the molecule's behavior significantly. Specifically, the attraction and repulsion exerted by the substrate may either speed up or decelerate the bending of the molecule, thereby altering the settling angle. Subsequent research endeavors should explore the electrostatic influence exerted by the substrate and the potential for charge exchange between the two entities. Additionally, exploring the effects of different substrate materials must be investigated in future research.

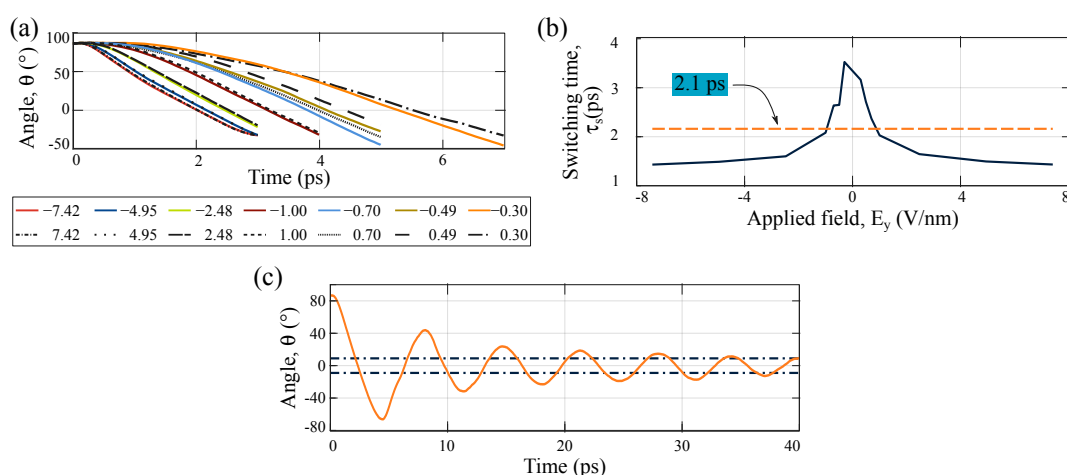


Fig. 5.6 Results of the MD simulation of the $\text{Fc}^+\text{C}_6\text{S}$ cation for various applied electric field. (a) Angle θ created by the $\text{Fc}^+\text{C}_6\text{S}$ with the substrate plane ($z = 0$) calculated with MD for different electric fields. The smaller the applied field, the slower the molecule to reach the substrate. (b) Switching time evaluated as the difference between the time needed to move from an angle of 81° to 9° (i.e., 90% and 10% of the total dynamic, respectively) varying the intensity of the electric field. (c) Angle θ created by the $\text{Fc}^+\text{C}_6\text{S}$ with the plane $z = 0$ calculated with a long MD simulation when the driving field equals 2.48 V nm^{-1} . The angle oscillated around 0° , which is the expected regime value. The black dotted line identifies the range between -9° and 9° (i.e., $\pm 10\%$ of the switching dynamic).

5.2 Read-out architecture

The comprehensive analyses presented in the previous section demonstrate the potential of utilizing an electric field to encode binary information into the bending of the molecule. It is noteworthy that various cations possess the capacity to generate an electric field, which could drive neighboring molecules, thereby facilitating the propagation of information. However, successful information propagation depends on overcoming steric hindrance during the molecular rearrangement process. While a detailed examination of logic propagation and elaboration is deferred to subsequent studies, it is imperative to highlight the promising readout capabilities offered by the bend-boosted molFCN mechanism. This mechanism enables the detection of encoded information at the laboratory level by measuring the orientation of the molecule through scanning probe microscopy [142]. Such advancements hold significant implications for the future development and practical application of molFCN.

The bend-boosted molFCN concept offers the possibility of integration with the charge-based paradigm, presenting exciting prospects for enhanced functionality and experimental validation. Notably, prior research has highlighted that the bis-ferrocene cation can generate switching fields within the range of $[-0.4; 0.4] \text{ V nm}^{-1}$ [62], which align with the requirements for driving the bending mechanism as elucidated in the previous section. Consequently, the electric field produced by the bis-ferrocene molecule, serving as a representative example, holds the potential to serve as a driving signal for the $\text{Fc}^+\text{C}_6\text{S}$ molecule.

By integrating bend-boosted molFCN with the charge-based paradigm, it becomes feasible to conduct in-lab measurements of charge-based molFCN logic using scanning probe microscopy. This integration offers significant advantages, facilitating the assessment and validation of charge-based molFCN technology under controlled laboratory conditions. Such synergistic integration holds promise for advancing our understanding of molecular field-coupled nanocomputing and unlocking its full potential for future technological applications.

In the following sections, we illustrate the modulation of current in a molecular junction, which is facilitated by the encoded information stored within the molFCN system. This capability enables real-time readout directly within the circuit, providing a means for evaluating the performance and feasibility of molFCN technology.

Furthermore, this in-circuit readout capability opens possibilities for exploring the integration of molFCN with CMOS technology. By integrating molFCN functionality into existing CMOS architectures, we can exploit the benefits of both technologies, paving the way for novel computing paradigms and enhanced device functionalities.

5.2.1 Theory of electronic transport

The molecular junction used for the readout system is analyzed by exploiting the Non-Equilibrium Green's Functions (NEGF) formalism. The NEGF formalism is a theoretical framework used to study quantum transport phenomena in nanoscale electronic devices [149]. It provides a rigorous mathematical description of electron transport in systems where quantum effects dominate. NEGF is particularly well-suited for modeling devices operating far from thermal equilibrium, where traditional transport theories based on equilibrium assumptions fall short.

The NEGF formalism provides a rigorous treatment of electronic conduction through molecular systems by accounting for contact self-energies and broadening functions, which reflect the influence of molecular contacts on conduction. These factors allow for a comprehensive description of the system's response to external stimuli.

Moreover, the transmission coefficient derived from NEGF facilitates the calculation of current flow through the system, enabling researchers to evaluate electronic transport properties with greater precision. Landauer's equation, integrated within the NEGF framework, offers insights into ballistic transport phenomena, while the flexibility of NEGF allows for the modeling of incoherent scattering effects, further enhancing its applicability to real-world scenarios.

The simulations described in the following section are performed using QuantumATK [150, 151]. The geometry relaxation process is carried out utilizing employing DFT calculations at the generalized gradient approximation (GGA) level within the Perdew, Burke, and Ernzerhof (PBE) framework. A double- ζ plus polarization (DZP) basis set is employed for all elements except gold, for which a single- ζ plus polarization (SZP) basis set is utilized.

To simulate the transport process, Dirichlet Boundary Conditions (DBC) are applied in the transport direction, while Periodic Boundary Conditions (PBC) are implemented in the transverse directions. The limited memory quasi-Newton method

for large-scale optimization (LBFGS) optimization algorithm is employed, with a force tolerance set to $0.05 \text{ eV}/\text{\AA}$ and a maximum step size of 0.02 \AA . These computational techniques ensure efficient and accurate relaxation of the molecular configurations, enabling reliable predictions of transport properties in the $\text{Fc}^+\text{C}_6\text{S}$ -gold junction system.

The electronic structure properties of the Au– FcC_6S junction are thoroughly investigated using QuantumATK, employing DFT calculations with GGA-PBE exchange-correlation functional. To accurately capture the electronic interactions, Hubbard (on-site - shell) corrections are incorporated for exchange-correlation terms.

For the atomic basis sets, a SZP basis set is employed for gold (Au), while a DZP basis set is utilized for the other elements involved. The Brillouin zone integration is carried out using a Monkhorst-Pack grid with (3,1,135) k-points.

To accurately describe the electronic structure in both equilibrium and non-equilibrium situations, a sparse recursion method for self-energy calculation is employed. Additionally, an additional continuum spatial region is introduced within the simulation boundary cell. This region allows for the consideration of the vanishing electrostatic potential, mimicking the behavior in a dielectric substrate. By incorporating this continuum spatial region, abrupt potential decay to zero at the simulation cell boundary is avoided, ensuring more accurate and physically realistic simulations of the Au– FcC_6S junction.

5.2.2 Charge sensing

This section illustrates how bend-boosted molFCN facilitates the integration of molFCN with standard current-based technologies, and ultimately with CMOS technology. As depicted in Fig. 5.7a, the setup for information readout involves a single $\text{Fc}^+\text{C}_6\text{S}$ cation anchored to a nanowire (NW) positioned within a metallic nanogap, adjacent to a charge-based molFCN wire. The metallic nanogap comprises electrodes labeled as E0 and E1, allowing for a tunneling current between either E1 or E0 and the NW via the molecule.

The encoded information alters the orientation of the $\text{Fc}^+\text{C}_6\text{S}$ molecule, consequently affecting its distance from the electrodes. For instance, when the molecule encodes a logic ‘1’, the ferrocene group moves closer to E1, enhancing the tunneling current through the E1 electrode while significantly reducing the current through

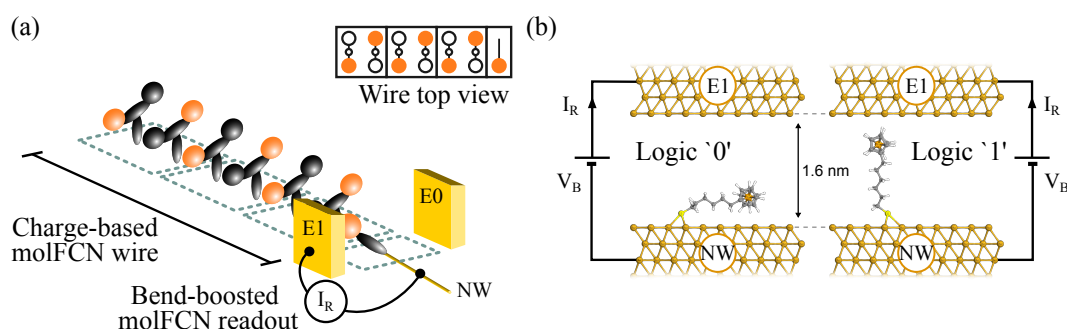


Fig. 5.7 Architecture of the readout system based on bend-boosted molFCN. (a) Charge-based molFCN wire propagating a logic '1' terminated with a $\text{Fc}^+\text{C}_6\text{S}$ cation placed between two electrodes (E0, E1) for the conductive measurement. Molecules are supposed anchored to an ideal nanowire (NW). The schematic amperometer measures the readout current (I_R). (b) Geometries used to study the conduction between the electrode E1 and the nanowire NW with DFT. In the left figure, the cation encodes a logic '0' (far from the E1 electrode), whereas in the right figure the cation encodes a logic '1' (close to the E1 electrode). A bias voltage (V_B) is applied between E1 and NW.

E0. Conversely, when encoding a logic '0', the ferrocene group shifts towards E0, enabling the conversion of the information into a current.

This operational principle is demonstrated through modeling a $\text{Fc}^+\text{C}_6\text{S}$ single-molecule junction, showcasing two distinct configurations as depicted in Fig. 5.7b. These configurations arise from a geometric relaxation of the pre-optimized $\text{Fc}^+\text{C}_6\text{S}$ structure, representing logic '0' (Fc positioned away from E1) and logic '1' (Fc situated near E1), respectively. In this model, the cation is anchored onto a face-centered-cubic (111) gold substrate, specifically a 10×4 atoms slab. The distance between E1 and the NW is set to 1.6 nm, with both E1 and NW contacts fabricated from gold.

The configurations utilized for the transport measurements are generated by initially specifying a tilting angle as the initial guess for the $\text{Fc}^+\text{C}_6\text{S}$ molecule anchored to the gold substrate, with a Au–S distance set to 2.39 \AA according to literature values [152]. Subsequently, these configurations undergo relaxation using DFT simulations, wherein the $\text{Fc}^+\text{C}_6\text{S}$ molecule is relaxed together with the first layer of gold atoms while keeping the remaining gold atoms fixed.

To verify the functionality of the readout structure, we assess the electronic transport properties of the junction using both DFT and NEGF methodologies. The transport equations are solved self-consistently with the system electrostatics, calculated by solving Poisson equation with the parallel conjugate gradient method

with DBC in the transport direction and PBC in the transverse ones. The readout current I_R is computed through a Landauer-like formula for current [153]:

$$I_R = \frac{2q}{h} \int_{-\infty}^{+\infty} T(E, V_B) [f_{NW}(E) - f_{E1}(E)] dE \quad (5.4)$$

where h is the Planck constant, E is the electron energy, V_B is the bias voltage, and f_{NW} and f_{E1} are the electrode Fermi-Dirac distributions. $T(E, V_B)$ is the transmission function representing the NW-to-E1 electron transmission probability of the device. Six bias points are considered from $V_B = 0$ V to $V_B = 1$ V. In each of them, the $T(E)$ function and its integral (5.4) in the following are calculated considering an energy array from -1 eV to 1 eV with 201 values; transverse k-point (Monkhorst grid) (7,3); Krylov method for self-energy calculation.

In Fig. 5.8a, we present the resulting readout current I_R as a function of the applied bias voltage (V_B) between electrodes E1 and NW for both configurations. In the logic ‘1’ configuration, a notable current I_R is observed. For instance, at a bias voltage of $V_B = 0.4$ V, the readout current is measured at $I_R = 32.53$ nA. Conversely, in the logic ‘0’ configuration, the current is zero, indicating a clear distinction between the two logic states. This confirms the efficacy of the readout structure in accurately detecting and distinguishing between different logic states based on the molecular orientation.

By analyzing the transmission spectrum $T(E)$, we aim to elucidate the underlying cause of the current discrepancy. In Fig. 5.8b, $T(E)$ is depicted as a function of the energy shift (E) with $V_B = 0.4$ V held constant. Here, $E = 0$ eV represents the equilibrium Fermi level of the system. The inset in Fig. X provides a closer look at $T(E)$ within the range $[-0.4, 0]$ eV. Notably, a discernible peak in electron transmission probability emerges in the logic ‘1’ configuration, with a peak value of $T(E) = 0.682$ observed at $E = -0.15$ eV. These peaks in transmittivity predominantly occur at negative energy levels, corresponding to occupied $\text{Fc}^+\text{C}_6\text{S}$ molecular orbitals. Specifically, the conduction is primarily governed by the Highest Occupied Molecular Orbital (HOMO). Conversely, in the logic ‘0’ configuration, electron transmission probability remains negligible across all considered energy values, resulting in negligible I_R .

Moreover, by examining the NW-to-E1 Transmission Pathways (TP) at a bias voltage of $V_B = 0.4$ V and an energy level of $E = -0.15$ eV, we discern the underlying

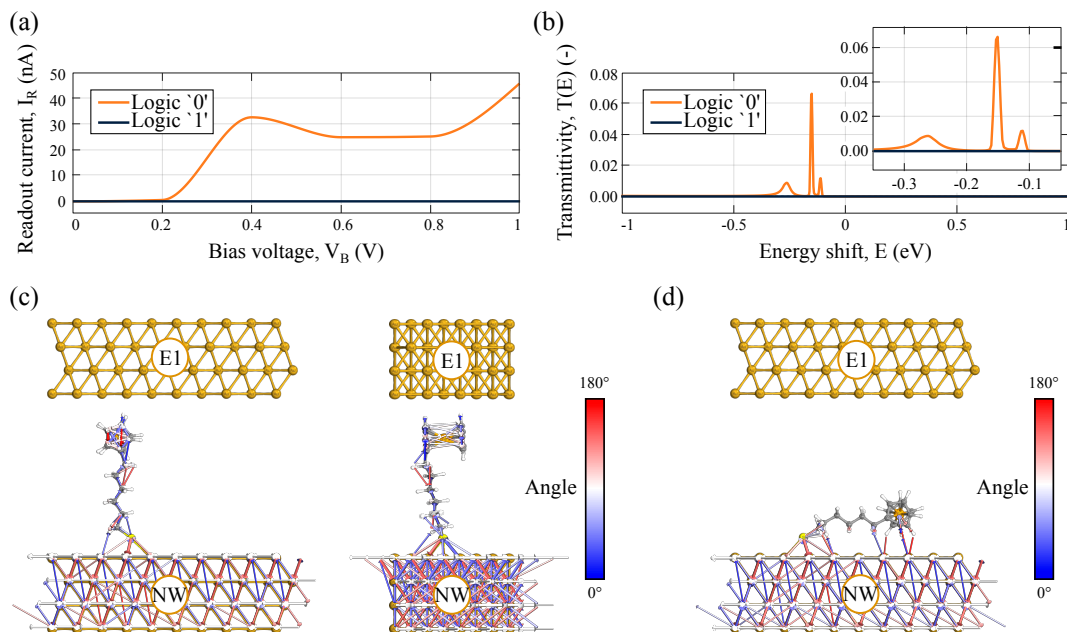


Fig. 5.8 Results of the DFT analysis of the molecular junction representing the charge-sensing readout system. (a) Readout current (I_R) calculated with DFT between E1 and NW when the $\text{Fc}^+\text{C}_6\text{S}$ encodes the two logic values. (b) Transmission spectrum ($T(E)$) calculated with DFT when the $\text{Fc}^+\text{C}_6\text{S}$ encodes the two logic values. The inset shows an enlargement of the transmittivity where is non-zero. (c) Transmission pathways for the electrons transmitted from the NW to E1 electrode when the cation encodes a logic '1'. (d) Transmission pathways for the electrons transmitted from the NW to E1 electrode when the cation encodes a logic '0'.

factors contributing to the disparities in $T(E)$ between the logic '0' and logic '1' scenarios. The TP analysis, as detailed in [154], elucidates the local influences on $T(E)$, with arrow direction indicating the direction of propagation and arrow size denoting transmission probability.

Fig. 5.8c illustrates the TPs for the logic '1' configuration, providing both frontal and lateral perspectives. Notably, transmission is highly probable from the NW to the sulfur atom, as indicated by the presence of large arrows, whereas transmission through the saturated chain is less likely, depicted by smaller arrows. No arrows connect the ferrocene to E1, indicating a significant tunneling barrier. Arrows become discernible when a lower transmission probability is portrayed, suggesting the presence of a finite $T(E)$.

These findings align with expectations: the anchoring sulfur atom contributes well-delocalized states due to its covalent bond with gold, while the saturated hex-

anethiol exhibits localized orbitals with minimal transmission probability along the carbon atoms. Conversely, the ferrocene displays good delocalization, attributed to the presence of Fe and cyclopentadiene π -type orbitals, resulting in numerous and relatively large arrows. The primary bottleneck to transmission is the tunneling barrier between $\text{Fc}^+\text{C}_6\text{S}$ and E1. In the logic '1' configuration, the $\text{Fc}^+\text{C}_6\text{S}$ molecule acts as a molecular bridge between NW and E1, yet with a small tunneling barrier.

In contrast, in the logic '0' configuration, depicted in Fig. 5.8d, the role of $\text{Fc}^+\text{C}_6\text{S}$ as a molecular bridge is prevented. Despite notable electron transmission between NW and sulfur, the rotational freedom of the molecule results in minimal transmission along the hexane chain. The TP analysis reveals sparse and diminutive arrows, primarily directed toward the NW, thereby forming limited connections between NW and Fc. Consequently, electrons originating from NW encounter a tunneling barrier approximately equivalent to the NW-E1 distance, rendering $T(E)$ negligible.

The distinct separation observed in I_R between the two logic states underlines the efficacy of the readout system. However, the reliability of the readout process could be susceptible to external influences. For instance, an external electric field generated by nearby electrodes might perturb the positioning of the molecule. Hence, future investigations should consider such external factors to validate the robustness of the readout mechanism.

The discussion and the results presented so far represent a conceptual idea for one possible alternative way of solving a specific issue typically related to charge-based molFCN: the conversion of charge into readable information. To bring this theoretical investigation to reality, there are several feasibility aspects to consider.

One of the key challenges lies in fabricating electrodes at the nanoscale with the requisite precision and alignment. Current techniques such as Electron Beam Lithography (EBL) and Focused Ion-Beam (FIB) offer the high resolutions needed for such work. Nevertheless, these methods also come with challenges related to complexity, time consumption, and cost. Achieving consistent reproducibility at this scale is critical for ensuring reliable device performance, making this an area ripe for innovation. Advances in these nanofabrication techniques could significantly enhance the feasibility of the device, paving the way for more consistent and cost-effective production methods. Further study should address the impact of process variations and include them in the model.

Material selection is another vital aspect to consider. The materials used for electrodes and nanowires must combine excellent electrical conductivity with mechanical stability and compatibility with nanofabrication processes. At such small scales, even slight changes in physical and chemical properties can have a pronounced impact on device performance. Research into new materials or improved processing methods could help overcome these challenges, leading to more robust and reliable devices.

The presence of counterions, necessary for stabilizing the ferrocenium state, introduces additional complexity but also opportunities for innovation. These counterions can shield the charge, potentially reducing the switching efficiency. However, understanding and controlling the Debye screening effect at this scale can lead to new ways to mitigate this issue. By carefully managing the distribution and concentration of counterions, it may be possible to minimize their impact on the intended electronic response. Addressing the impacts of unwanted scattering sites on transport involves developing strategies to control or limit the mobility of counterions, thereby preserving the electrical conductivity and mobility essential for devices. Innovations in materials and device design could help mitigate these effects, leading to improved performance and stability.

Moreover, the current choice of using a cation has been made to address the problem gradually, but in future works, it can be substituted with polar or engineered zwitterionic molecules, preventing the need for counterions in the substrate for charge balance.

Environmental factors such as temperature play a significant role in device performance. Nanoscale devices are highly sensitive to these conditions, which can influence the behavior of both nanostructures and counterions. Developing devices that are less sensitive to environmental variations or incorporating protective measures could enhance their robustness and reliability in real-world applications.

Chapter 6

Conclusion and future perspective

The research presented in this thesis explores the use of molecular Field-Coupled Nanocomputing (molFCN) as a promising paradigm for future electronics. The ultimate goal of this technology is to facilitate the realization of a prototype to validate the paradigm and its working principles. Achieving this goal necessitates vast knowledge in various research areas, such as the synthesis of molecules and their deposition, as well as the fabrication of nanoelectrodes for the clock system and nanostructures patterned following the desired logic. The driving force behind investments in these directions lies in the potential for obtaining significant performance improvements in specific target areas, such as compute-in-memory, neural computing, or arithmetic acceleration. With the increased precision and computational power of simulation frameworks, performance validation can be conducted before the development of a working prototype.

In this context, the present work focuses on analyzing the relationship between variations at the physical level and their impact on circuit behavior, thereby expanding the current simulation framework for molFCN technology. The exploration is structured across multiple abstraction levels, each starting from a modelistic simplification of the previous layers. This approach enables the decomposition of the overall goal into smaller, addressable tasks.

The first level of analysis regards molecules, the elemental units of the technology. This work collects and discusses a set of properties that molecules should possess to behave suitably for the molFCN application. Other than constraints on dimensions or symmetries, the key features are related to the electrostatic behavior,

such as polarizability. Subsequently, the work presents a case study: starting from bis(trimethylsilyl) sulfide, we evaluate, through Density Functional Theory (DFT), the impact of modifications to subparts of the molecules on their electrostatic characteristics (mainly the charge distribution and the V_{in} -AC transcharacteristic (VACT)). This process aims to identify suitable chemical composites or specific molecular compositions for the ultimate definition of a set of candidate molecules for different purposes within a circuit.

Since electrostatic interactions among a set of molecules are at the core of the technology, we proceed with their analysis at the following level. Having defined a suitable electric model for molecules, we can study their interactions without the need for heavy DFT calculations. Focusing on three molecular species, namely neutral, oxidized, and zwitterionic, to target a complete set of behaviors, this work analyzes the system energy of fundamental interactions. The analysis relates molecular behavior to the logical behavior expected from various cell layouts found in fundamental gates, defining more molecular properties for successful information processing. The energy analysis is then compared with the results of SCERPA simulations of fundamental gates to demonstrate the coherence of the predictions. The final step advances this process by defining a set of feasible VACTs for the correct realization of in-plane information crossing.

The confident simulations of elementary gates using SCERPA are the starting point for the design of more elaborate circuits. Due to its tight integration with single-molecule behavior, the complexity of the simulation and thus its execution time increases exponentially with the number of cells. In order to advance the analysis of potential use-cases for the molFCN technology, there is a need to add an abstraction layer. This work proposes a precise and comprehensive method to model molFCN gates, making it possible to consider them as whole entities rather than collections of molecules. We develop a tool to automatically handle the characterization and treatment of the library of created devices, enabling the speeding up of circuit simulations from days to milliseconds while maintaining an accuracy higher than 99%.

Finally, the last step of this work aims to simplify the possibility of experimentally measuring the outcome of a molFCN circuit. This work proposes a modification of the current charge-based molFCN to include molecule bending in the information encoding. The addition of physical movement simplifies the measurement in a

laboratory and is compatible with the charge-based paradigm. Moreover, by integrating the new bend-boosted paradigm with a molecular junction, the conversion of the information stored in the charge into a current provides a way for in-circuit information readout and integration with standard current-based technologies.

With regard to the ultimate goal of developing molFCN technology, this work has taken a small yet significant step forward in providing the tools and knowledge for future technology assessment. The research has yielded several significant outcomes aimed at establishing clear parameters and insightful predictions of expected behaviors. These efforts are crucial for facilitating future experimental validations and the development of functional prototypes.

One key achievement is the establishment of precise characteristics required for molecules to exhibit suitable behaviors for molFCN. Additionally, validating the impact of their electrostatic footprint on circuit-level simulations has resulted in the formulation of specific guidelines for identifying suitable molecules.

Moreover, the development and publication of tools capable of generating valuable predictions in a short period have been instrumental. These tools not only expedite research progress but also encourage contributions from other researchers, fostering collaborative advancements in the field.

Furthermore, a noteworthy contribution involves proposing a modified version of the paradigm. This modified paradigm retains the inherent advantages of molFCN while simplifying information measurement in laboratory settings and facilitating integration with conventional current-based devices.

The research activity pursued during the development of the present work has been characterized by a miscellaneous and interdisciplinary context. This necessitated addressing various topics, including single-molecule simulations, understanding technological limitations and possibilities, abstracting physical behavior to develop manageable models, and creating algorithms and tools to predict the digital behavior of circuits effectively. Single-molecule simulations have been particularly demanding computationally, often taking weeks to complete or sometimes not being feasible due to limited computational resources. Additionally, the ability to develop robust models has been hindered by a lack of experimental data in the literature on the molecules or structures of interest. These two factors have constrained the scope of exploration within a three-year program.

Collectively, the results obtained can converge into a comprehensive roadmap outlining fundamental next steps essential for the experimental validation of the technology. This roadmap serves as a guiding framework for future research endeavors in the field. The first key element to be analyzed is the molecular design process, delineating the requisite criteria that molecules must meet to ensure their efficacy and reliability in operation. Although this task has been partially addressed in this work, the sky is the limit for its completion. With some of the constraints delineated, chemists are empowered to embark on the synthesis of molecules that conform to the prescribed requirements, thus ensuring optimal performance within the designated operational parameters. This iterative process of theoretical analysis, simulation, and synthesis constitutes a crucial step toward the advancement and refinement of molFCN technology, facilitating the development of robust and reliable molecular circuits for diverse applications.

The second aspect that must be part of the roadmap for the future development of molFCN is the exploration of the intricate physical and structural prerequisites necessary for the fabrication of sub-nanometric molecular patterns, crucial for achieving desired functionalities. The meticulous design of sub-nanometric molecular patterns stands as a pivotal determinant in ensuring the operation of molFCN circuits. As such, the achievement of a functional molFCN prototype depends on the precise control of atomic deposition processes. Future works should explore contemporary nanofabrication methodologies. The outcomes of these analyses furnish crucial insights that can be incorporated within the SCERPA framework to assess propagation characteristics and discern feasible fabrication pathways.

Lastly, future research should further investigate the intricate terrain of hybrid integration between molFCN and CMOS, paving the way for seamless convergence between molecular and semiconductor technologies. The distinguishing feature of molFCN technology lies in its unique characteristic of information propagation and processing without the flow of current. While this attribute affords minimal power consumption, it poses challenges in interfacing with established technologies reliant on current conduction. Consequently, the transduction of information encoded within the charge distribution assumes crucial importance in validating a prototype and facilitating its hybrid integration with other technologies. This work offers one possible solution to be further investigated. Future research should perform a comprehensive exploration of the mechanisms underlying the transduction of charge-encoded logic information into an electrical quantity. The potential of single-

molecule junction sensors in facilitating the integration of molFCN technology with existing electronic architectures must follow rigorous theoretical analyses and experimental validation.

In conclusion, this thesis has presented a comprehensive exploration of molFCN technology. Through rigorous investigation and analysis, significant contributions have been made to understanding the relationship between underlying physics and the logic behavior of devices. The findings of this research not only advance the theoretical understanding of molFCN technology but also hold practical implications for the future realization of a working prototype. By addressing key research questions and objectives, this study has laid the groundwork for future research endeavors in the technological realization of a prototype. Moreover, the methodologies and approaches developed herein provide valuable insights and tools for researchers and the engineering community. As we look to the future, the findings of this thesis serve as a springboard for further exploration and innovation, with the ultimate goal of driving positive change and progress in future electronics.

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Appendix A

Molecules geometry

Appendix A reports the coordinates of all the molecules shown in this work in order of appearance.

- A.1 Thiolated-carbazole linked bisferrocenes
- A.2 Creutz-Taube ion
- A.3 Double-cage fluorinated fullerenes
- A.4 Diferrocenyl carborane
- A.5 Zwitterionic three-dot molecule with di-anionic pentaborate counterion
- A.6 Three-dot 'Y'-shaped functionalized zwitterionic neutral closo-carborane
- A.7 Bis(trimethylsilyl) sulfide
- A.8 Hexanedithiol bis(trimethylsilane)
- A.9 Dodecanedithiol bis(trimethylsilane)
- A.10 Tridecanedithiol bis(trimethylsilane)
- A.11 1,3,5benzenetrithiol linked bis(hexanedithiol)bis(trimethylsilane)
- A.12 Trithiocyanuric acid linked bis(hexanedithiol)bis(trimethylsilane)
- A.13 Trithiocyanuric acid linked tris(hexanedithiol)bis(trimethylsilane)
- A.14 6-(ferrocenyl)hexanethiol cation

A.1 Thiolated-carbazole linked bisferrocenes

Table A.1 reports the coordinates of the optimized geometry of the isolated thiolated-carbazole linked bisferrocenes with the Unrestricted Kohn-Shann (UKS) method using Becke3 Lee–Yang–Parr (B3LYP) functional with LANL2DZ basis set¹. The geometry optimization is performed using the ORCA package [117, 118].

Table A.1 Coordinates of the optimized geometry of the thiolated-carbazole linked bisferrocenes.

Atom	x (Å)	y (Å)	z (Å)
C	-4.9188270000	-6.5031650000	-0.8587530000
C	-3.5793170000	-6.9858170000	-0.8968480000
C	-3.0811900000	-7.0062560000	0.4368760000
C	-4.1126780000	-6.5356030000	1.2994850000
C	-5.2484660000	-6.2257030000	0.4986850000
Fe	-3.6217760000	-5.0615550000	-0.0939200000
C	-3.0508360000	-3.3213890000	0.8995480000
C	-1.9625760000	-3.7964630000	0.1050180000
C	-2.4166390000	-3.8223460000	-1.2495530000
C	-3.7634760000	-3.3657300000	-1.2892450000
C	-4.1585670000	-3.0586390000	0.0428770000
C	-0.5582570000	-4.1310580000	0.5536160000
C	-0.4652460000	-4.4392800000	2.0581700000
C	0.4441240000	-3.0445410000	0.1639090000
C	0.1395530000	-1.6916540000	0.2833520000
C	1.0960900000	-0.7296370000	-0.0413320000
C	2.3738640000	-1.1401250000	-0.4918530000
C	2.6965410000	-2.4868420000	-0.6196260000
C	1.7215770000	-3.4189030000	-0.2889600000
C	1.1008250000	0.7194320000	-0.0412840000
C	2.3812740000	1.1215920000	-0.4917430000
N	3.1335150000	-0.0117310000	-0.7554380000

¹[H - Ne]: T. H. Dunning Jr. and P. J. Hay, *Methods El. Str. Th.* Vol. 2, PLENUM PRESS (1977).
[Na - Hg]: P. J. Hay and W. R. Wadt, *J. Chem. Phys.* 82, 270 (1985) // P. J. Hay and W. R. Wadt, *J. Chem. Phys.* 82, 284 (1985) // P. J. Hay and W. R. Wadt, *J. Chem. Phys.* 82, 299 (1985)

C	0.1506340000	1.6876640000	0.2835270000
C	0.4640790000	3.0385460000	0.1642530000
C	1.7439360000	3.4045770000	-0.2886090000
C	2.7127700000	2.4661720000	-0.6193800000
C	-0.5311710000	4.1315020000	0.5541900000
C	-0.4363400000	4.4386040000	2.0588550000
C	-1.9376210000	3.8062770000	0.1053370000
C	-2.3913780000	3.8354980000	-1.2492670000
C	-3.7412770000	3.3880170000	-1.2891560000
C	-4.1385450000	3.0833300000	0.0428750000
C	-3.0291350000	3.3384160000	0.8996890000
Fe	-3.5882010000	5.0825740000	-0.0935380000
C	-5.2072030000	6.2577310000	0.4984930000
C	-4.8753910000	6.5326720000	-0.8589280000
C	-3.5326750000	7.0063390000	-0.8968080000
C	-3.0347280000	7.0237190000	0.4370340000
C	-4.0695530000	6.5601770000	1.2994980000
H	-0.2590570000	-5.0363310000	0.0166720000
H	-1.1460800000	-5.2457240000	2.3369930000
H	0.5509870000	-4.7382870000	2.3199200000
H	-0.7105020000	-3.5592170000	2.6561410000
H	-1.1118790000	5.2494350000	2.3378330000
H	-0.6874840000	3.5599870000	2.6565090000
H	0.5818070000	4.7308230000	2.3208380000
H	-5.5209280000	6.3894550000	-1.7106570000
H	-2.9833760000	7.2839840000	-1.7823070000
H	-2.0437840000	7.3235600000	0.7394440000
H	-3.9993490000	6.4438930000	2.3692920000
H	-6.1479530000	5.8692480000	0.8547970000
H	-1.8033050000	4.1508180000	-2.0971830000
H	-3.0272300000	3.2211060000	1.9713650000
H	-5.1126950000	2.7403010000	0.3534950000
H	-4.3615770000	3.3192620000	-2.1686130000
H	-1.8307520000	-4.1414420000	-2.0975700000
H	-4.3833560000	-3.2926400000	-2.1686480000

H	-5.1303610000	-2.7091200000	0.3536370000
H	-3.0480810000	-3.2043100000	1.9712480000
H	-5.5636110000	-6.3558540000	-1.7103550000
H	-6.1865260000	-5.8309030000	0.8551380000
H	-4.0414210000	-6.4195020000	2.3692290000
H	-2.0921750000	-7.3125600000	0.7391150000
H	-3.0321230000	-7.2673380000	-1.7824260000
H	-0.8459960000	-1.3865790000	0.6138220000
H	1.9530010000	-4.4739140000	-0.3875970000
H	3.6706100000	-2.8049730000	-0.9719930000
H	3.6889010000	2.7779530000	-0.9717200000
H	1.9822740000	4.4580580000	-0.3871300000
H	-0.8368880000	1.3890330000	0.6140000000
H	-0.2259770000	5.0349770000	0.0175930000
C	5.0775300000	-0.0180640000	-1.4665380000
C	5.9729190000	-0.0238510000	-0.3644280000
H	5.2476050000	0.8561080000	-2.0596640000
H	5.2406300000	-0.8911760000	-2.0631730000
H	5.8098200000	0.8492620000	0.2322060000
H	5.8028450000	-0.8980230000	0.2286980000
C	7.3104610000	-0.0282330000	-0.8412520000
H	7.4735580000	-0.9013430000	-1.4378910000
H	7.4805370000	0.8459430000	-1.4343720000
C	8.2058500000	-0.0340200000	0.2608580000
C	9.5433800000	-0.0410050000	-0.2159690000
H	8.0344630000	-0.9072070000	0.8550560000
H	8.0440740000	0.8400680000	0.8564230000
H	9.7147690000	0.8321860000	-0.8101620000
H	9.7051540000	-0.9150900000	-0.8115400000
C	10.4387680000	-0.0467920000	0.8861400000
H	10.2769940000	0.8272980000	1.4817040000
H	10.2673800000	-0.9199780000	1.4803390000
S	11.7762980000	-0.0537770000	0.4093130000
H	11.7125470000	0.9439170000	0.0279430000

A.2 Creutz-Taube ion

Table A.2 reports the coordinates of the optimized geometry of the isolated Creutz-Taube ion with the UKS method using CAM-B3LYP functional with def2-TZVP basis set [119]. Grimme D3 correction is applied [120, 121]. The geometry optimization is performed using the ORCA package [117, 118].

Table A.2 Coordinates of the optimized geometry of the Creutz-Taube ion.

Atom	x (Å)	y (Å)	z (Å)
N	-1.47988946796756	-0.05285205457158	1.27498448071720
C	-0.67077241974449	1.09292716963293	1.35149917597568
C	0.67160970221453	1.09249718589177	1.35097621146075
H	-1.18595971395240	2.03672658501147	1.51016666325216
N	1.47976804961815	-0.05391181022150	1.27452063224758
H	1.18744248777038	2.03607278141805	1.50871394894585
Ru	3.09643885330220	-0.00967568741406	-0.08532731192698
C	0.67126167251913	-1.20061422690920	1.27746499279927
C	-0.67227305606569	-1.20013257057835	1.27731432270635
Ru	-3.09629329637808	-0.00608626616805	-0.08511290511556
H	1.18399076803482	-2.15423063898605	1.38091806757177
H	-1.18558092467358	-2.15349957384486	1.37989236542018
H	1.28689912937046	1.76566808346104	-0.87630653443178
N	2.20081636645352	1.53178165198214	-1.26791162733666
H	5.60207210769173	-0.54457653546232	-1.04413963817974
N	2.18093457967615	-1.49199405646246	-1.33867583571480
H	1.27064714763617	-1.72793775877195	-0.93977510940189
N	-4.84246119740889	0.06015263861448	-1.38440552787630
N	-2.20691001983119	1.54720832918151	-1.25718200973636
N	-3.92584592963764	-1.51967255033729	1.18435013577577
N	-3.95970781905486	1.42946564996908	1.23833841081172
N	-2.17346408860936	-1.47619379765541	-1.34751320389788
H	4.65016844873175	-0.22818017349962	-2.35124617982749
N	3.92916385647976	-1.51100322750842	1.19658245330711
H	5.22083715488400	1.00740210184025	-1.42788480511639
H	2.07669224715364	1.26933392626952	-2.24493928639198

H	2.77097738418381	2.37818101294587	-1.27939573667679
N	4.84270375417842	0.05429515309954	-1.38458128288286
H	2.73083238939539	-2.34635596762969	-1.40862848185735
H	2.04075089265852	-1.17043878902753	-2.29513762286367
N	3.95513590412887	1.43881798139578	1.22735637763319
H	4.90723323952343	-1.33922467627333	1.42057794772940
H	3.87622826022995	-2.46976314842196	0.86082726447609
H	3.37691922544517	-1.43691106825587	2.05096437785074
H	4.93501898475252	1.24025226701994	1.42230285530781
H	3.42921118006169	1.39492630956840	2.09990096443082
H	3.92752877790984	2.39634274198853	0.87928465815331
H	-3.37756653938518	-1.45033325325653	2.04152908205282
H	-3.86816916226101	-2.47572228922758	0.84161694581887
H	-4.90538706002103	-1.35325035558085	1.40598002683170
H	-1.26460531802086	-1.71410157994343	-0.94633105158835
H	-2.02936364196797	-1.14672924013530	-2.30071436376009
H	-2.72201201188931	-2.33057350677314	-1.42680237620439
H	-5.59957029347015	-0.54378950771531	-1.04781392216471
H	-4.64886310530859	-0.21557320094577	-2.35278254118530
H	-5.22429001881935	1.01203069402461	-1.42152005804121
H	-1.29235846019742	1.78020414938377	-0.86634759754982
H	-2.77907529563520	2.39227889287884	-1.26019594968467
H	-2.08507932786773	1.29287229227844	-2.23665453496341
H	-3.43072147014224	1.38161092075632	2.10886190963294
H	-4.93802532047200	1.22469431933907	1.43471011049103
H	-3.93781160522220	2.38954367362604	0.89701311297634

A.3 Double-cage fluorinated fullerenes

Table A.3 reports the coordinates of the optimized geometry of the isolated double-cage fluorinated fullerenes with the UKS method using B3LYP functional with TZVP basis set and Grimme D3 correction [120, 121]. The geometry optimization is performed using the ORCA package [117, 118].

Table A.3 Coordinates of the optimized geometry of the double-cage fluorinated fullerenes.

Atom	x (Å)	y (Å)	z (Å)
C	1.81539158216088	-0.74594889026984	-1.39606926306569
C	2.04126291165057	0.71316181663375	-0.89550066793158
C	0.58964565572261	-0.74263980663103	-2.36226660300715
C	0.05666230477152	0.72052421932033	-2.45727259522941
C	0.95379207928282	1.61896719864214	-1.54933992032502
C	-0.52978775232601	-1.62169795431806	-1.72260546450494
C	-1.75435468067518	-0.70443743476046	-1.42408434248723
C	-1.39135992379999	0.74317982343612	-1.87593667590415
C	0.00378577053614	-2.16091585572254	-0.36066484347232
C	-0.89145466296637	-1.58442299175537	0.77389533772276
C	-1.97366498341561	-0.68420108883679	0.12253338493782
C	1.44859501579587	-1.62115222565608	-0.15951545483734
C	1.45459952822716	-0.70816765482609	1.10081307424631
C	-0.00034769840341	-0.68990278529434	1.73746077487520
C	1.81679317052332	0.73146757203071	0.64980814267185
C	0.59771382373814	1.64279306604825	0.94861356327690
C	-0.53182202174231	0.76890331191782	1.64297478991630
C	-1.74801880239871	0.76683226190623	0.62161225063527
C	-1.38439187644502	1.64929188032809	-0.60726687455903
C	0.06044231009516	2.18866813347419	-0.40605702301986
F	2.91944008500152	-1.22255009513440	-2.01719514429658
F	3.28458694642525	1.14560251661433	-1.20973997457316
F	0.93533397016145	-1.21147527718804	-3.58379828916108
F	0.07110092248249	1.15861776268652	-3.73783628282942
F	1.52062798404762	2.61767447487444	-2.26571803865727
F	-0.87474603959810	-2.63990139980677	-2.54523768721507
F	-2.86108090245779	-1.14909172420606	-2.06365606585681
F	-2.27457900267978	1.20114486556982	-2.79375189071492
F	-0.00581695195562	-3.51689836428172	-0.34285319788836
F	-1.47383335948789	-2.60372350456759	1.45507397899091
F	-3.21896203459351	-1.11714307051947	0.43647556952485
F	2.32541199784574	-2.64596198816361	-0.01856942571290

F	2.36945704299609	-1.16933238101248	1.98986618605916
N	0.00169890428261	-1.22559139725723	3.04935814858203
F	2.92483779484543	1.17705527141180	1.29097064106155
F	0.96549124050988	2.68044537035698	1.74088377876521
N	-0.92713675787288	1.32543437753883	2.88499490517672
F	-2.87570322483590	1.24626909527837	1.20592653419205
F	-2.26246643566861	2.67369237190854	-0.74345952883154
F	0.06851682673349	3.54474347257972	-0.41889466359684
C	0.89249354156603	-0.64159187102952	5.29560295983004
C	1.12701474554284	0.81277610689620	5.78158320300001
C	-0.34310052873227	-0.65062449483338	4.29760412732807
C	-0.87619239761260	0.80787606470728	4.20294207460420
C	0.03334115486010	1.70880264226395	5.14376202235827
C	-1.45887197349415	-1.51904014542860	5.01950685465332
C	-2.67255960398907	-0.60607832621234	5.33354534050435
C	-2.31851640404902	0.83025233740845	4.86680810983997
C	-0.89632011084181	-2.05601471451330	6.36733647767087
C	-1.76870495231265	-1.47844232741107	7.52283085433154
C	-2.86814273867407	-0.57708539544688	6.88308582320062
C	0.55187605402187	-1.51534281000912	6.53752540049872
C	0.58228080496890	-0.60083461343408	7.79985566767638
C	-0.85495272572742	-0.57409187052758	8.40768821442661
C	0.93672256189543	0.84359097115355	7.33172557211209
C	-0.28237150833643	1.76304575036022	7.64652509145300
C	-1.38969926220863	0.88843724445968	8.31261133430682
C	-2.63300771908201	0.88520003825609	7.36927447677068
C	-2.28914771765967	1.75209477157462	6.12045916221683
C	-0.84072259557726	2.29289297912105	6.29091074014041
F	2.00779326986174	-1.12493530165951	4.69398844092096
F	2.36634490016301	1.24342666855175	5.44137112463020
F	0.60167696320481	2.72234497477800	4.44494481975384
F	-3.79220662116143	-1.05640047086833	4.71673618899468
F	-0.90494628571587	-3.41208844957915	6.38892790829061
F	-2.32223829961080	-2.47243020975218	8.25605458069338
F	-4.10552039002880	-1.00744206529758	7.22275067658503

F	1.43176876058948	-2.53934897692465	6.66463930612187
F	1.48179030420437	-1.05310098904944	8.70456727364989
F	-0.84612678916298	-1.00350129957839	9.69126729553365
F	2.05481206696742	1.29295251140102	7.94785438418088
F	0.07714467158049	2.78674167348535	8.45606244902639
F	-1.71327651481305	1.36525152209803	9.53710070751009
F	-3.72565121353595	1.36599020194070	8.00738754798605
F	-3.16884105344898	2.77558903698442	5.98809339907591
F	-0.83240121824824	3.64880168038538	6.26384081350284
F	-1.84220709685123	-2.56336461283919	4.24108713957359
F	-3.25129349584466	1.28556089608532	3.99148476426679
H	0.28447713941442	-2.19090672175873	3.10153266699895
H	-1.33022570263486	2.24598772189295	2.81733586282389

A.4 Diferrocenyl carborane

Table A.4 reports the coordinates of the mean geometry of the diferrocenyl carborane obtained from the optimization of the isolated molecule with the UKS method using B3LYP functional with TZVP basis set and Grimme D3 correction [120, 121]. The geometry optimization is performed using the ORCA package [117, 118].

Table A.4 Coordinates of the mean geometry of the diferrocenyl carborane.

Atom	x (Å)	y (Å)	z (Å)
C	1.415708	0.262184	-0.819146
C	1.182495	-0.980907	-0.745066
C	2.039432	-1.764391	-1.452743
C	2.830782	-1.022917	-1.956793
C	2.458709	0.212152	-1.554627
C	4.793123	-1.597985	0.238682
C	4.066247	-2.388661	0.700172
C	3.190470	-1.683018	1.396297
C	3.376987	-0.465480	1.370291
C	4.362661	-0.407320	0.654408

Fe	2.977211	-0.978202	-0.228484
H	5.523789	-1.855540	-0.330375
H	4.146251	-3.347343	0.533744
H	2.482883	-1.999708	1.833422
H	2.826928	0.297343	1.776322
H	4.698499	0.403369	0.443879
H	2.878737	0.993040	-1.758348
H	0.490735	-1.269883	-0.228468
H	2.098794	-2.746543	-1.569916
H	3.598181	-1.338821	-2.521591
C	-4.362661	-0.407320	0.654408
C	-3.376987	-0.465480	1.370291
C	-3.190470	-1.683018	1.396297
C	-4.066247	-2.388661	0.700172
C	-4.793123	-1.597985	0.238682
C	-2.830782	-1.022917	-1.956793
C	-2.039432	-1.764391	-1.452743
C	-1.182495	-0.980907	-0.745066
C	-1.415708	0.262184	-0.819146
C	-2.458709	0.212152	-1.554627
Fe	-2.977211	-0.978202	-0.228484
H	-3.598181	-1.338821	-2.521591
H	-2.098794	-2.746543	-1.569916
H	-0.490735	-1.269883	-0.228468
H	-2.878737	0.993040	-1.758348
H	-5.523789	-1.855540	-0.330375
H	-4.698499	0.403369	0.443879
H	-2.826928	0.297343	1.776322
H	-2.482883	-1.999708	1.833422
H	-4.146251	-3.347343	0.533744
C	0.734285	1.351622	-0.187265
C	-0.734285	1.351622	-0.187265
B	1.311997	2.047716	1.149081
B	0.000000	2.629473	2.132789
B	-1.311997	2.047716	1.149081

B	1.321962	2.894172	-0.270800
B	0.844219	3.682381	1.184686
B	-0.844219	3.682381	1.184686
B	-1.321962	2.894172	-0.270800
B	0.000000	2.382840	-1.163567
B	0.000000	3.898688	-0.266638
H	0.000000	2.153295	-2.292258
H	0.000000	2.804657	3.260792
H	2.256123	1.688517	1.557704
H	1.413045	4.545933	1.698549
H	0.000000	4.893835	-0.784879
H	-2.269949	3.039389	-0.840151
H	2.269949	3.039389	-0.840151
H	-1.413045	4.545933	1.698549
H	0.000000	1.585608	1.728480
H	-2.256123	1.688517	1.557704

A.5 Zwitterionic three-dot molecule with di-anionic pentaborate counterion

Table A.5 reports the coordinates of the optimized geometry of the isolated molecule with the UKS method using CAM-B3LYP functional with def2-TZVP basis set [119]. Grimme D3 correction is applied [120, 121]. The geometry optimization is performed using the ORCA package [117, 118].

Table A.5 Coordinates of the optimized geometry of the zwitterionic three-dot molecule with di-anionic pentaborate counterion.

Atom	x (Å)	y (Å)	z (Å)
B	-4.0508705411	2.2269955274	0.0000000000
B	-2.7630687737	-0.1309997415	-0.0000000000
B	-1.2307474684	2.2924954286	0.0000000000
B	-2.0897841250	1.3873041204	-2.1486281880
B	-2.6323755121	1.7132962755	1.5431915407

C	-5.3520195017	3.1883118428	0.0000000000
C	-6.4531917126	3.9203221867	0.0000000000
C	-7.5219760826	4.6197985839	0.0000000000
C	-7.0037778180	5.6934138471	0.0000000000
C	-8.8984413462	4.1968592127	0.0000000000
C	0.1700339006	3.2371123481	0.0000000000
C	1.1254625385	3.9691226920	0.0000000000
C	2.3561840391	4.7011330358	0.0000000000
C	1.6598548618	5.8886164211	0.0000000000
C	3.7636249159	4.3619494998	0.0000000000
C	-2.7448331523	-1.5778889481	0.0000000000
C	-2.6962519440	-3.0256426625	0.0000000000
C	-2.6800583001	-4.4083288983	-0.0000000000
C	-3.8621987306	-4.8963356099	0.0000000000
C	-1.4493366613	-5.1566060772	-0.0000000000
H	-9.2061216868	5.2704744758	0.0000000000
H	-9.7243199514	3.4485820337	0.0000000000
H	-7.8944315519	6.5230257544	0.0000000000
H	-5.7892495467	5.8723495860	0.0000000000
H	2.7610267962	6.5718262597	0.0000000000
H	0.5262954322	6.3440894625	0.0000000000
H	4.2346539234	5.3518092042	-0.0000000000
H	4.1860727151	3.6112509376	0.0000000000
H	-0.2995835532	-4.6685993656	-0.0000000000
H	-1.6598548618	-6.2627552871	-0.0000000000
H	-3.5383247461	-5.9862179846	-0.0000000000
H	-4.8500148810	-4.2781936646	0.0000000000
C	-1.9127327478	1.3253392168	-3.9186138548
C	-2.8051913624	1.7862900835	3.2561574684
N	-2.9546815133	1.9095771713	4.6673595751
N	-1.7610660536	1.1525658817	-5.2944156550

A.6 Three-dot 'Y'-shaped functionalized zwitterionic closo-carborane

Table A.6 reports the coordinates of the optimized geometry of the isolated molecule with the UKS method using CAM-B3LYP functional with def2-TZVP basis set [119]. Grimme D3 correction is applied [120, 121]. The geometry optimization is performed using the ORCA package [117, 118].

Table A.6 Coordinates of the optimized geometry of the zwitterionic three-dot 'Y'-shaped functionalized zwitterionic closo-carborane.

Atom	x (Å)	y (Å)	z (Å)
B	-3.507300000	-0.788500000	1.428900000
H	-2.968800000	-0.828500000	2.484700000
B	-3.385600000	-1.955100000	-0.013300000
H	-2.752000000	-2.950700000	-0.134000000
B	-3.596200000	-0.527700000	-1.193900000
H	-3.130300000	-0.362400000	-2.272100000
B	-3.698600000	0.643100000	0.247600000
H	-3.327300000	1.765900000	0.346500000
B	-4.888100000	-1.859700000	0.981500000
H	-5.178300000	-2.762700000	1.702200000
B	-4.953200000	-1.676800000	-0.867700000
H	-5.298400000	-2.423000000	-1.730200000
B	-5.169100000	0.166100000	-0.674900000
H	-5.714600000	0.975200000	-1.370500000
B	-5.111200000	-0.013500000	1.163000000
H	-5.595300000	0.644500000	2.036400000
B	-6.118100000	-0.991000000	0.176200000
C	-2.595800000	-0.538000000	0.096900000
C	-1.188300000	-0.367500000	0.066900000
C	-7.624800000	-1.031800000	0.180500000
C	-8.863300000	-0.877300000	0.145300000
Fe	-10.556700000	-0.175200000	-0.028100000
P	-9.593200000	1.113900000	-1.652600000

P	-9.741 200 0000	1.408 500 0000	1.400 400 0000
C	-8.635 300 0000	2.567 700 0000	-0.890 800 0000
H	-10.380 000 0000	1.772 800 0000	-2.657 000 0000
H	-8.638 800 0000	0.431 500 0000	-2.459 900 0000
C	-8.288 300 0000	2.279 800 0000	0.565 600 0000
H	-10.568 100 0000	2.497 200 0000	1.850 000 0000
H	-9.212 800 0000	0.942 500 0000	2.638 700 0000
H	-7.725 500 0000	2.737 300 0000	-1.473 700 0000
H	-9.275 700 0000	3.451 500 0000	-0.984 700 0000
H	-7.435 500 0000	1.599 100 0000	0.642 600 0000
H	-8.047 300 0000	3.200 500 0000	1.104 900 0000
C	-11.732 400 0000	-1.668 600 0000	0.951 900 0000
C	-12.444 200 0000	-0.429 900 0000	0.977 100 0000
H	-11.388 500 0000	-2.224 700 0000	1.810 800 0000
C	-11.518 500 0000	-2.024 900 0000	-0.413 000 0000
C	-12.660 200 0000	-0.003 500 0000	-0.366 700 0000
H	-12.757 600 0000	0.098 200 0000	1.866 200 0000
H	-10.977 000 0000	-2.890 900 0000	-0.761 300 0000
C	-12.081 400 0000	-0.997 500 0000	-1.229 100 0000
H	-13.193 600 0000	0.882 100 0000	-0.678 300 0000
H	-12.093 100 0000	-0.985 600 0000	-2.309 100 0000
C	0.022 000 0000	-0.225 900 0000	0.042 400 0000
C	1.442 600 0000	-0.082 500 0000	0.015 800 0000
C	2.033 000 0000	1.192 100 0000	0.008 400 0000
C	2.260 000 0000	-1.225 100 0000	-0.001 300 0000
C	3.433 400 0000	1.335 600 0000	-0.015 400 0000
H	1.405 000 0000	2.075 700 0000	0.022 700 0000
C	3.662 500 0000	-1.111 900 0000	-0.024 800 0000
H	1.799 700 0000	-2.207 300 0000	0.000 600 0000
C	4.230 700 0000	0.175 400 0000	-0.032 200 0000
H	5.311 200 0000	0.272 200 0000	-0.053 600 0000
C	4.032 900 0000	2.631 600 0000	-0.020 900 0000
C	4.514 200 0000	3.773 100 0000	-0.024 600 0000
Fe	5.316 200 0000	5.490 400 0000	-0.020 300 0000
P	6.824 100 0000	4.621 000 0000	-1.438 500 0000

P	6.581 100 0000	4.629 700 0000	1.625 500 0000
C	8.212 500 0000	3.736 800 0000	-0.468 800 0000
H	7.607 700 0000	5.410 900 0000	-2.351 200 0000
H	6.392 900 0000	3.589 600 0000	-2.337 600 0000
C	7.666 400 0000	3.261 200 0000	0.874 100 0000
H	7.564 500 0000	5.394 100 0000	2.356 500 0000
H	5.969 700 0000	3.938 500 0000	2.721 100 0000
H	8.613 800 0000	2.904 900 0000	-1.055 400 0000
H	9.014 600 0000	4.470 600 0000	-0.327 800 0000
H	7.000 100 0000	2.403 000 0000	0.744 300 0000
H	8.470 900 0000	2.984 800 0000	1.562 300 0000
C	3.776 400 0000	6.633 500 0000	0.837 700 0000
C	5.006 100 0000	7.368 200 0000	0.886 900 0000
H	3.175 200 0000	6.327 300 0000	1.680 000 0000
C	3.485 500 0000	6.381 800 0000	-0.541 600 0000
C	5.478 700 0000	7.562 700 0000	-0.442 000 0000
H	5.499 100 0000	7.712 100 0000	1.785 600 0000
H	2.647 900 0000	5.812 500 0000	-0.914 000 0000
C	4.528 000 0000	6.941 800 0000	-1.330 100 0000
H	6.369 200 0000	8.101 400 0000	-0.731 200 0000
H	4.585 600 0000	6.917 800 0000	-2.408 600 0000
C	4.482 400 0000	-2.280 900 0000	-0.040 000 0000
C	5.194 900 0000	-3.294 300 0000	-0.053 800 0000
Fe	6.204 500 0000	-4.898 500 0000	-0.071 200 0000
P	4.621 200 0000	-5.691 700 0000	-1.456 400 0000
P	4.829 000 0000	-5.500 600 0000	1.603 500 0000
C	3.147 400 0000	-6.380 400 0000	-0.455 900 0000
H	4.842 500 0000	-6.776 000 0000	-2.375 800 0000
H	3.971 900 0000	-4.767 200 0000	-2.339 900 0000
C	3.074 600 0000	-5.669 900 0000	0.892 100 0000
H	4.960 000 0000	-6.744 100 0000	2.325 300 0000
H	4.605 400 0000	-4.617 600 0000	2.709 000 0000
H	2.217 800 0000	-6.262 000 0000	-1.020 600 0000
H	3.330 100 0000	-7.453 200 0000	-0.324 000 0000
H	2.708 300 0000	-4.645 500 0000	0.776 100 0000

H	2.422900000	-6.199800000	1.593200000
C	8.019300000	-4.216900000	0.737600000
C	7.969000000	-5.647100000	0.807800000
H	8.105800000	-3.535000000	1.569500000
C	7.934700000	-3.855300000	-0.645900000
C	7.846600000	-6.168700000	-0.511700000
H	8.009400000	-6.235000000	1.714200000
H	7.902200000	-2.848500000	-1.032800000
C	7.821800000	-5.046000000	-1.415000000
H	7.809500000	-7.212900000	-0.785400000
H	7.745200000	-5.097000000	-2.491400000

A.7 Bis(trimethylsilyl) sulfide

Table A.7 reports the coordinates of the optimized geometry of the isolated molecule with the UKS method using B3LYP functional with def2-TZVP basis set [119]. Grimme D3 correction is applied [120, 121]. The geometry optimization is performed using the ORCA package [117, 118].

Table A.7 Coordinates of the optimized geometry of the bis(trimethylsilyl) sulfide.

Atom	x (Å)	y (Å)	z (Å)
S	-0.95428891730269	-1.23955758881714	-0.65422183709048
Si	0.71788101489163	-0.07791862998613	0.06533307580146
Si	-2.73052351358668	-0.06213213009761	-0.30330461975319
C	-2.91334687918615	1.26016438578659	-1.62003360000493
C	-2.74164052147983	0.72668482086830	1.39952829840927
C	-4.14077240406354	-1.28780842821987	-0.44033807157811
C	0.61024194056148	1.71927354146955	-0.46235533963979
C	0.84338648383641	-0.18701357125193	1.93277964083188
C	2.21933903720987	-0.88312106726522	-0.71450267626607
H	-2.90583137960204	0.81207375902380	-2.61552864085001
H	-2.10448005896503	1.99090480809122	-1.57585594165865
H	-3.85911592059259	1.79597797385716	-1.49510050838090

H	-1.95378603670091	1.47162609839399	1.51897489826883
H	-2.62209612595437	-0.02459884981018	2.18158548675627
H	-3.69926532286390	1.23165383673496	1.55753058338410
H	-4.13409316193386	-1.78573782210617	-1.41215347262263
H	-5.10237212221819	-0.77896806274947	-0.32902003404887
H	-4.07013140118295	-2.05681608389303	0.33140264032402
H	1.52184929390463	2.24471922655522	-0.16281960306773
H	-0.23294439332726	2.23327009863857	0.00044404114527
H	0.50925717544397	1.80622727464723	-1.54550015884279
H	-0.02255050997634	0.25508885199681	2.42677287730842
H	1.73648259240332	0.33743233117615	2.28547158183736
H	0.91389925530236	-1.22880965127333	2.25080783471470
H	3.13471648044431	-0.39141234904786	-0.37401560021193
H	2.18001979227684	-0.81756873546974	-1.80357839294147
H	2.28229638536161	-1.93876470445185	-0.44230246192397

A.8 Hexanedithiol bis(trimethylsilane)

Table A.8 reports the coordinates of the optimized geometry of the isolated molecule with the UKS method using CAM-B3LYP functional with def2-TZVP basis set [119]. Grimme D3 correction is applied [120, 121]. The geometry optimization is performed using the ORCA package [117, 118].

Table A.8 Coordinates of the optimized hexanedithiol bis(trimethylsilane).

Atom	x (Å)	y (Å)	z (Å)
S	-0.83585053254184	-1.58207501005048	-0.32739638942515
Si	9.99195868703724	-0.47383403737463	-0.10255607835101
Si	-2.62274411872835	-0.29843134635611	-0.52274506115594
C	-2.46187687749834	0.60389538035764	-2.14034866838474
C	-2.62074477340903	0.85859999900031	0.93150127132285
C	-4.02343520176948	-1.50932234172472	-0.50069287595469
C	11.39260018534678	0.73710478792717	-0.12451643384845
C	9.83113441628682	-1.37643766417207	1.51488711967175

C	9.98974556594540	-1.63052113669947	-1.55706149769350
H	-2.40332974955817	-0.08704507412834	-2.98066276689756
H	-1.58867926722657	1.25587489576756	-2.16700719479696
H	-3.34322799866315	1.23309804715449	-2.28496841991277
H	-1.76270568622300	1.53094487712623	0.92239350302476
H	-2.62737750327083	0.31768989891517	1.87749019979080
H	-3.52052999454416	1.47707278185984	0.89540487766327
H	-3.95120623989942	-2.22491338692948	-1.32034041037698
H	-4.97070832205732	-0.97744793951997	-0.60935783319476
H	-4.06163858789384	-2.06521909988418	0.43627757954051
H	12.33989637590161	0.20522242177477	-0.01609640779498
H	11.32047634443243	1.45252261448207	0.69528704686979
H	11.43066253013015	1.29318882637304	-1.06137431672301
H	9.77275722885230	-0.68567064737749	2.35535065346154
H	10.71243297450204	-2.00576799661366	1.65927117290981
H	8.95786855004762	-2.02832150433346	1.54149893265568
H	10.88956054797038	-2.24896462845333	-1.52123682521240
H	9.99624330314863	-1.08939073848228	-2.50292184806347
H	9.13173543710622	-2.30289725926287	-1.54799647147244
C	0.50992520284617	-0.39205222317457	-0.14039763987405
C	1.85863680164284	-1.05086781396677	-0.37584258447542
C	3.00459491479102	-0.06559105821017	-0.19768324304947
C	4.36447616557796	-0.70651198338665	-0.42697292167789
C	5.51044261356931	0.27876119740002	-0.24883043589070
C	6.85913873008341	-0.38010952620947	-0.48425647996161
S	8.20497380104159	0.80985924144899	-0.29745592337145
H	0.44938988762511	0.01605127462467	0.87167152766081
H	0.34880059845532	0.43505852509107	-0.83367690773073
H	1.88411127286569	-1.47164301289341	-1.38324934977342
H	1.97828704512787	-1.88791989130207	0.31571984442565
H	2.96749291889679	0.35831556058453	0.81032569581830
H	2.87768920718936	0.77297839330844	-0.88878738126393
H	4.40157156604088	-1.13042502137437	-1.43497773197558
H	4.49138461487604	-1.54507769950319	0.26413597807654
H	5.48496864872242	0.69956831468188	0.75855946894997

H	5.39081581795642	1.11578910980542	-0.94042207481880
H	6.91965198668198	-0.78823971433365	-1.49631596038119
H	7.02025821398516	-1.20720595906629	0.20902950206090

A.9 Dodecanedithiol bis(trimethylsilane)

Table A.9 reports the coordinates of the optimized geometry of the isolated molecule with the UKS method using CAM-B3LYP functional with def2-TZVP basis set [119]. Grimme D3 correction is applied [120, 121]. The geometry optimization is performed using the ORCA package [117, 118].

Table A.9 Coordinates of the optimized dodecanedithiol bis(trimethylsilane).

Atom	x (Å)	y (Å)	z (Å)
S	-8.64697994303781	-3.18196317606984	-0.91161159035013
Si	9.53261606944893	-0.00421820537965	0.22377115488636
Si	-10.45539199413120	-2.03541420802959	-1.01705966419252
C	-10.44781563924897	-0.93350641417057	-2.52642114831194
C	-10.65391251539637	-0.98226442367879	0.51373562316797
C	-11.81969414027215	-3.30105081302062	-1.13305113932966
C	10.89436737209092	1.26326114653629	0.34999851955232
C	9.53161842630917	-1.12088517770414	1.72211019145693
C	9.72956403589288	-1.04108524141851	-1.31824330305211
H	-11.37953681969218	-0.36708439204776	-2.59117166704196
H	-10.34438384797749	-1.52271655824595	-3.43782991397573
H	-9.62856188397092	-0.21350175743868	-2.49675684225747
H	-11.59577587164076	-0.43049182589409	0.47630989546870
H	-10.65561209123743	-1.59680426257896	1.41423608219595
H	-9.84981082880683	-0.25089164581156	0.60650336889230
H	-12.79208688012799	-2.80961538177094	-1.20195470654637
H	-11.69920871652048	-3.93139345563950	-2.01471738669593
H	-11.83225360076495	-3.94890707192275	-0.25580898562295
H	11.86765747151157	0.77270840711530	0.41279763722174
H	10.77402618164697	1.88360602915099	1.23889786736986

H	10.90486030679183	1.92013885641326	-0.52049561925744
H	9.42723198986330	-0.54139322755921	2.63965580922134
H	10.46538052555757	-1.68460517749305	1.78005426363328
H	8.71516919526533	-1.84366578464269	1.68693036388023
H	10.67358111156498	-1.58970291899021	-1.29039198619420
H	9.72541745626030	-0.41715847385450	-2.21227315431855
H	8.92779026450387	-1.77438396439724	-1.41582388984999
C	0.08469891392364	-0.50289237545118	-0.30745921328041
C	1.48912623217659	-1.08413245023235	-0.35058075685395
C	2.58057079530140	-0.03994147921891	-0.17772809481499
C	3.98473625652000	-0.62301025440002	-0.20998375760509
C	5.07090836490062	0.42866619776020	-0.04104101872888
C	6.46515765969721	-0.17197519179980	-0.05521752060931
S	7.72116407426038	1.13885195210732	0.13441824173081
H	1.63528819407253	-1.60652682116101	-1.30167255559805
H	1.58937546309946	-1.84514320863831	0.43044908624020
H	2.43088398505136	0.48715006938169	0.76996580062246
H	2.48523852833541	0.71719606610222	-0.96281483136240
H	4.13484532823677	-1.15304903121752	-1.15605880979322
H	4.08120617164352	-1.37604586487781	0.57911033011307
H	4.92179913758084	0.96641593437145	0.89871246018088
H	4.99110107098232	1.17270137432609	-0.83797969080283
H	6.63593726305619	-0.69832701005855	-0.99431101223039
H	6.56750275950535	-0.89033479802697	0.75846745257453
H	-2.55725976891221	-0.44283259149203	0.51402498800406
H	-5.05603288969586	-0.89666881298832	0.37284007975170
C	-2.41190844671142	-0.96462483926810	-0.43752911310790
C	-4.90766963886130	-1.42462192872894	-0.57467192123981
C	-7.38850751717960	-1.87306696469156	-0.72484449993462
C	-1.00759772883897	-1.54603209516163	-0.48188410823004
H	-3.40744792922325	-2.76679956888325	0.17327091317058
C	-3.50384936343874	-2.00830680978116	-0.61036449454476
H	-2.51261121466590	-0.20300651121628	-1.21790724746020
H	-5.91378522888313	-3.22160801272490	0.05012801254816
C	-5.99497124558624	-2.47504928095808	-0.74432272468595

H	-5.00490337378496	-0.66969978781054	-1.36187220144679
H	-7.49266422791870	-1.15343368737616	-1.53717831277693
H	-3.35592230057982	-2.53380458275983	-1.55921772637706
H	-5.84855866377603	-3.00999355461827	-1.68607940275040
H	-0.01532421606612	0.25947690459621	-1.08713232027283
H	-0.90711174700926	-2.30877172439946	0.29736891779758
H	-0.06140548768655	0.01793965134101	0.64447874690037
H	-0.86217056189228	-2.06655529318009	-1.43409935287087
H	-7.55585776851528	-1.34774977502203	0.21545359569336

A.10 Tridecanedithiol bis(trimethylsilane)

Table A.10 reports the coordinates of the optimized geometry of the isolated molecule with the UKS method using CAM-B3LYP functional with def2-TZVP basis set [119]. Grimme D3 correction is applied [120, 121]. The geometry optimization is performed using the ORCA package [117, 118].

Table A.10 Coordinates of the optimized tridecanedithiol bis(trimethylsilane).

Atom	x (Å)	y (Å)	z (Å)
S	-8.68288935391500	-3.03800955887075	-0.83762130549265
Si	10.66254358643075	1.38174944668350	-0.15974699084829
Si	-10.50245365358928	-1.80378023692650	-0.98986931175209
C	-10.37828546823184	-0.84047993749771	-2.57594124231798
C	-10.54247901297963	-0.69147485083810	0.49931105551024
C	-11.87257004136661	-3.05112412202916	-1.00770696853688
C	12.33384600984349	0.58310677288970	-0.10805532935641
C	10.36105227293856	2.33794362871720	-1.72608728491784
C	10.28202285661878	2.38707741389310	1.35723123233646
H	-11.28170985176335	-0.23845746864104	-2.69806286320525
H	-10.29559268000767	-1.49874229799844	-3.44013745878296
H	-9.52817900754408	-0.15810106612674	-2.57917500952226
H	-11.46820725076710	-0.11177921721769	0.48594990403724
H	-10.51904675273386	-1.25980328948367	1.42883437220832

H	-9.71381892186110	0.01669791370287	0.50539126311851
H	-12.83373616523754	-2.54144072104079	-1.09903104983198
H	-11.78147715571472	-3.73766579003668	-1.85000898378383
H	-11.89488429458254	-3.63860714252510	-0.08959290578030
H	13.11012987366725	1.35062704862440	-0.11138071907086
H	12.49923818986201	-0.05912934804113	-0.97356274537559
H	12.46460667713840	-0.01772417423055	0.79223147102990
H	10.51514807797597	1.71952631585333	-2.60978604973683
H	11.06292967773311	3.17346040795885	-1.77472492978586
H	9.35454035219491	2.75352402871462	-1.76679359337141
H	10.99365647409099	3.21271188652162	1.42635251610834
H	10.37086879080439	1.79318419383882	2.26661021119036
H	9.28223398690175	2.81934823096206	1.32248750870506
C	0.13643819147257	-0.66436248547339	-0.33441056108129
C	1.51873846326207	-1.29246162295371	-0.41770480879179
C	2.64914170562391	-0.28917436574321	-0.25128921810876
C	4.02870508448457	-0.92302668268202	-0.33635772868826
C	5.16068925393786	0.07924675987320	-0.17281476681029
C	6.53535346796290	-0.56822783213928	-0.25732779036376
S	9.27621111364764	-0.33279261380938	-0.17397878607838
H	1.62672579806200	-1.80203545631237	-1.38053249988775
H	1.60983112669204	-2.06988600035263	0.34778112867393
H	2.54290079239331	0.22005803036967	0.71191432795775
H	2.55906645259618	0.48810738160527	-1.01710380744531
H	4.13438679896233	-1.43352938275192	-1.29862213202203
H	4.11955859496572	-1.69876765636202	0.43045016735802
H	5.05720746565573	0.59053146938592	0.78941351338913
H	5.07387941595985	0.85331685259935	-0.94203453943261
H	-2.48972734571243	-0.53808639253805	0.53690585501564
H	-5.00472164335195	-0.92535162380789	0.43291977652225
C	-2.37501945514252	-1.04616477488504	-0.42592815965020
C	-4.88315203152303	-1.43385119524551	-0.52864430761279
C	-7.36926498316596	-1.81101993548011	-0.64267950356000
C	-0.99075501957189	-1.67064806342174	-0.50416397630447
H	-3.40981681916261	-2.83176983225254	0.16889114632719

C	-3.49756958438932	-2.05700113576975	-0.59925486824523
H	-2.46578566441277	-0.26909783570312	-1.19188486359989
H	-5.92057303304191	-3.22437019731207	0.06707085606701
C	-5.99630056190832	-2.45628193959113	-0.70623411257252
H	-4.97252775591174	-0.66005967802254	-1.29785242801624
H	-7.48506648184374	-1.05575387360773	-1.42272410900199
H	-3.38011929214978	-2.56756491330519	-1.56013033218768
H	-5.88047107531333	-2.96643315679157	-1.66512830073500
H	0.04525522363079	0.11375331373866	-1.09932606813683
H	-0.90030918146656	-2.44760721531063	0.26186537089622
H	0.02669678640400	-0.15630571663152	0.62908734087760
H	-0.87835273496234	-2.18026905358536	-1.46646531313086
H	-7.52322825808903	-1.31249318688841	0.31728307726765
C	7.64892200663661	0.45020697292529	-0.09066724733578
H	6.64805509396198	-1.07381173403580	-1.21904886812033
H	6.63046779923964	-1.33730764949572	0.51295591466439
H	7.57481709379605	0.95508222303095	0.87522851861666
H	7.59779250626543	1.21956795827708	-0.86395047358854

A.11 **1,3,5benzenetrithiol** **linked** **bis(hexanedithiol)bis(trimethylsilane)**

Table A.11 reports the coordinates of the optimized geometry of the isolated molecule with the UKS method using CAM-B3LYP functional with def2-TZVP basis set [119]. Grimme D3 correction is applied [120, 121]. The geometry optimization is performed using the ORCA package [117, 118].

Table A.11 Coordinates of the optimized geometry of the 1,3,5benzenetrithiol linked bis(hexanedithiol)bis(trimethylsilane).

Atom	x (Å)	y (Å)	z (Å)
S	-8.58129054339771	-3.35206521211445	-0.95002193976839
Si	16.22331636920907	1.13106637947843	-0.23569158232155
Si	-10.44545700723592	-2.27858541567044	-1.02842812429166

C	-10.45604580984135	-1.11747642960202	-2.49033796068127
C	-10.69387599966172	-1.30637409790943	0.54608161719978
C	-11.73887278124683	-3.60487315895150	-1.21208723758811
C	17.81760717502618	0.16915282952959	-0.23988941411193
C	16.04159836496769	2.14033618472074	-1.79613573743481
C	16.11571782557443	2.25246778994362	1.25315568664634
H	-11.41351248788633	-0.59620797813437	-2.55457467323285
H	-10.30392407881440	-1.66074921393319	-3.42319226611904
H	-9.67718333775028	-0.35758747236537	-2.41068380681333
H	-11.65816921281088	-0.79419625856219	0.52661600601834
H	-10.67646962077550	-1.96143679519738	1.41732049598936
H	-9.92392745270742	-0.54516862615087	0.67982552966189
H	-12.73405200417296	-3.15940597811696	-1.26479925284755
H	-11.58429888900933	-4.18625967931391	-2.12160161856473
H	-11.72465541694391	-4.29207454094440	-0.36536860263434
H	18.67018682856321	0.84931259614834	-0.28428675359183
H	17.87637604744820	-0.49750304630601	-1.10087804116929
H	17.91803291382272	-0.43497511239257	0.66234526719293
H	16.08750282470196	1.50409031503529	-2.68014356383565
H	16.84374874763234	2.87814242801347	-1.86756575269865
H	15.09637732141539	2.68426250371927	-1.82012943909005
H	16.91937752780029	2.99180696264856	1.23083948785970
H	16.20523835067759	1.68357167881275	2.17879581489598
H	15.17185249781540	2.79858553836595	1.28253428897704
C	6.92106607937861	-0.57341588460690	-0.12048407101151
C	8.11307385200183	0.37088202267036	-0.08814040955488
C	9.42604622705997	-0.39860610259495	-0.11268301392958
C	10.64506232832225	0.51032363628302	-0.10413126213364
C	11.95118104534657	-0.26911514217087	-0.12278659751769
S	14.68795285654786	-0.37414198818138	-0.14363115596836
H	8.06606908909326	0.99162914297098	0.80982330259574
H	8.06674213143767	1.04823682784577	-0.94447596866348
H	9.45755461273324	-1.03582310153064	-1.00195070989212
H	9.46950142859000	-1.07077658017553	0.74999111704459
H	10.61394900800286	1.15226791982796	0.78176247769127

H	10.60503032274302	1.17860396427934	-0.97008812117543
H	-2.59117981939022	-0.42350637112508	0.53872623467469
H	-5.09158736563642	-0.91919997563575	0.35718211209275
C	-2.42423565757831	-1.00127718611171	-0.37325112123382
C	-4.90501275068088	-1.49349355726500	-0.55522646149732
C	-7.37527842100844	-1.99929192632313	-0.76076583827545
C	-1.03441088563338	-1.61936271872844	-0.34005881066623
H	-3.42256291708840	-2.77358161442659	0.32568326414672
C	-3.49849761801765	-2.07096664677433	-0.50993504102103
H	-2.49411573252194	-0.30109943724721	-1.20914649702350
H	-5.90117250473420	-3.27689663966659	0.12276922040968
C	-5.96909309049525	-2.56972447479436	-0.70807588423002
H	-4.97901420448975	-0.78449867634336	-1.38565532476382
H	-7.46619266840205	-1.31002369106902	-1.60075317431535
H	-3.31802261403540	-2.65227884554030	-1.41919881271262
H	-5.78318453825400	-3.14478754587880	-1.61882646968323
H	-7.58865769679936	-1.44645203909788	0.15434177351500
C	13.17077139394311	0.63477385071454	-0.12941191204347
H	11.98180283162209	-0.91640430442612	-1.00325736486704
H	11.99903665185509	-0.92870897038427	0.74782010198229
H	13.16733025239829	1.27330376981586	0.75439603434550
H	13.15240476148665	1.27766355280141	-1.00997480206330
S	0.20126302274699	-0.31456383552780	-0.21638093039620
H	-0.84000180175256	-2.18717616160809	-1.25134359148303
H	-0.92333680493238	-2.28268926858746	0.51929390988614
S	5.39544907739287	0.38569358457692	-0.08819695040587
H	6.93058105857968	-1.23842455533401	0.74468525476503
H	6.93227834582546	-1.17633101133197	-1.02980042841656
C	1.70460501377602	-1.14504191441708	-0.21990852805345
C	4.15106216566244	-0.80791187935951	-0.15911574835852
C	2.83487012767540	-0.28766170000752	-0.15613326590187
C	1.89325130788457	-2.50778182879394	-0.28031382990971
C	3.19717657516818	-3.01613302938741	-0.27780056194665
C	4.32133807862342	-2.17047128772341	-0.21882224997428
S	3.53113796536130	-4.70432118722300	-0.34655023599143

H	2.26246947099242	-5.13440379089044	-0.38949285688845
H	1.05470700174315	-3.18350033236638	-0.32828186355633
H	2.68554267835695	0.78427776151407	-0.10682663727703
H	5.30681184049864	-2.61049620599474	-0.22082114379333

A.12 Trithiocyanuric acid linked bis(hexanedithiol)bis(trimethylsilane)

Table A.12 reports the coordinates of the optimized geometry of the isolated molecule with the UKS method using CAM-B3LYP functional with def2-TZVP basis set [119]. Grimme D3 correction is applied [120, 121]. The geometry optimization is performed using the ORCA package [117, 118].

Table A.12 Coordinates of the optimized geometry of the trithiocyanuric acid linked bis(hexanedithiol)bis(trimethylsilane).

Atom	x (Å)	y (Å)	z (Å)
S	-8.34648208703203	-3.54473554127529	-0.87571781757833
Si	16.17375725399849	0.83751384026756	-0.03521281156306
Si	-10.31784884078585	-2.56956606337411	-0.81549557675702
C	-10.39749185272021	-1.37770153957980	-2.24087935591048
C	-10.45540648580594	-1.69952284676253	0.82306064810186
C	-11.50633080192691	-3.98202583807463	-0.98284816773805
C	17.67677924784672	-0.24455564096257	-0.07351144897494
C	16.04855753577710	1.97476149113014	-1.49971581547270
C	15.95991321977316	1.73808620034465	1.57671957494881
H	-11.38909400046756	-0.92068455455054	-2.26918055100478
H	-10.23415197546379	-1.87881380108539	-3.19476447410591
H	-9.66984524905852	-0.57181715924015	-2.14378306939767
H	-11.44265119716072	-1.23841976389162	0.89971680131962
H	-10.34422413923322	-2.39296818693664	1.65607065749849
H	-9.71532599598873	-0.90680223159490	0.93257730407003
H	-12.53148136467276	-3.60767300137400	-0.95006050187308
H	-11.37555055462689	-4.50800014212075	-1.92886812922940

H	-11.39130022989846	-4.70307952941723	-0.17289451053486
H	18.57475862204111	0.37188555850924	0.00054140436832
H	17.74115997106963	-0.81286747170280	-1.00163315104020
H	17.68925222218725	-0.94793622868823	0.75958527146041
H	16.08170959152135	1.42678759783716	-2.44100311434536
H	16.89537688597951	2.66494816700627	-1.48569103258753
H	15.13859629745732	2.57434637307824	-1.47939093294725
H	16.80058669610344	2.42224569492462	1.71371957567518
H	15.94699643209687	1.05258334463789	2.42349921131763
H	15.04756135864461	2.33386817901709	1.59875905581649
C	6.71966784177200	-0.22880999566073	-0.32145139234056
C	7.99768986781866	0.59237781331663	-0.31648747787372
C	9.23625032953018	-0.28982060648291	-0.27687409183445
C	10.52859274290502	0.51159328214071	-0.26994799052627
C	11.76621761151076	-0.37286444296374	-0.23570725140801
S	14.50809053222383	-0.60619180110361	-0.19255231363575
H	7.99988511000194	1.26488082983883	0.54589778104980
H	8.03035912371122	1.22735788614764	-1.20614178515249
H	9.23109688310599	-0.96300633056897	-1.13955919427592
H	9.20033253129094	-0.92852492013981	0.61081885347644
H	10.53469781488225	1.18234641052124	0.59517405667456
H	10.56259721112433	1.15293201701054	-1.15633356425599
H	-2.57973215492926	-0.07342289376798	0.23460995898452
H	-5.01960336060806	-0.78417966277852	0.11618440837918
C	-2.35535050129654	-0.73397374476547	-0.60707119021716
C	-4.78369157840976	-1.44971560814711	-0.71970281796511
C	-7.19906836628112	-2.14631730431200	-0.84046413871690
C	-0.91475193741499	-1.20902798674659	-0.51732469557555
H	-3.20254655557925	-2.49095281353599	0.29581981140415
C	-3.33435991866763	-1.89888396723085	-0.61458719922880
H	-2.48963124978929	-0.13852860171000	-1.51398341419241
H	-5.64050126533187	-3.20851989299651	0.18036170987523
C	-5.75974722843639	-2.61746533099803	-0.73044056209150
H	-4.91436222458064	-0.85706600503812	-1.63026068686720
H	-7.35033979876665	-1.56839583420430	-1.75481730673643

H	-3.09972589426856	-2.56475590473876	-1.45017963539716
H	-5.53143638006086	-3.28233819040417	-1.56664520081843
H	-7.46610099512396	-1.50263850180212	-0.00018894256700
C	13.04240855170968	0.44942234855826	-0.22459932279051
H	11.77070917300437	-1.03765411609379	-1.10277018334238
H	11.73989489968746	-1.01269287857683	0.64940275512909
H	13.07965984321157	1.10733514803619	0.64659584791297
H	13.10925749482514	1.08317043505474	-1.11231437607006
S	0.18102746907807	0.22823602824267	-0.51524004800592
H	-0.65766219139211	-1.84402401663492	-1.36289870854875
H	-0.74513395202927	-1.78246371053617	0.39213759535495
S	5.30097228193905	0.88958124214911	-0.38254210851484
H	6.64332224438541	-0.84532239228232	0.57228658101553
H	6.68063605605071	-0.89176205715541	-1.18391539225278
C	1.75166509050955	-0.51523661676719	-0.44598986286701
C	3.96810559657348	-0.22730323505068	-0.38932279742910
N	1.85913588811798	-1.84122090861491	-0.40647408213622
C	3.10685571565712	-2.28633867001450	-0.35782917935760
N	4.19882389650461	-1.53567830481177	-0.34736111712994
S	3.39362317459677	-4.00871689488662	-0.30159490148057
H	2.09558759406645	-4.33600684565736	-0.32814610498660
N	2.76846056671646	0.33562023224060	-0.43876413738357

A.13 Trithiocyanuric acid linked tris(hexanedithiol)bis(trimethylsilane)

Table A.13 reports the coordinates of the optimized geometry of the isolated molecule with the UKS method using CAM-B3LYP functional with def2-TZVP basis set [119]. Grimme D3 correction is applied [120, 121]. The geometry optimization is performed using the ORCA package [117, 118].

Table A.13 Coordinates of the optimized geometry of the trithiocyanuric acid linked tris(hexanedithiol)bis(trimethylsilane).

Atom	x (Å)	y (Å)	z (Å)
S	-7.39752355419722	-4.72985791977839	-0.53616545041344
Si	16.72687845580355	1.27834651104394	-0.16512486416505
Si	-9.43267240173060	-3.95049744211450	-0.83946824568485
C	-9.47918443577476	-3.16152786688394	-2.52348525429100
C	-9.76781682919887	-2.71879883941863	0.51322220060403
C	-10.49627373962013	-5.46399806664262	-0.72182590315962
C	18.28619627272992	0.29263110451421	0.00501622585912
C	16.51818422685716	2.03426243644176	-1.85068605649017
C	16.48823860667525	2.52066024167975	1.19652244055019
H	-10.49295875149348	-2.80761519973413	-2.72375574431198
H	-9.21213953293970	-3.87113534996946	-3.30607617014627
H	-8.81304395915215	-2.30195939279183	-2.59406878959305
H	-10.79209001888537	-2.35201564280626	0.41675003421172
H	-9.66616615561241	-3.16933597293899	1.50029635000443
H	-9.10554343117614	-1.85486978653701	0.45728598633275
H	-11.54330279532440	-5.19186614997284	-0.87011503511366
H	-10.23400729182401	-6.19979072306699	-1.48262483445339
H	-10.41232730328934	-5.93917683095642	0.25588760524335
H	19.15050940202168	0.95173989505106	-0.09671905454633
H	18.36397209042998	-0.47594030174054	-0.76447208842550
H	18.35164225427732	-0.19155520813834	0.97948758151882
H	16.57764049315861	1.28467253789675	-2.63919524862569
H	17.31947353967383	2.75819241087583	-2.01600906209286
H	15.57204858160003	2.56614383003481	-1.94902021763676
H	17.29812984041273	3.25281888337572	1.15540350108227
H	16.51303925935465	2.05217973454637	2.18012500645452
H	15.54997713304530	3.06595478681395	1.09553713873596
C	7.34449037614793	-0.28719804610915	-0.08175361779329
C	8.57311625718491	0.60566383281925	-0.06357981200676
C	9.85881742621740	-0.20678178111944	-0.03281769971261
C	11.10602149336042	0.66281387108174	-0.02158408043537
C	12.38908872991639	-0.15474753187502	0.00535790070557
S	15.13970377054532	-0.24177908153593	0.05766453873361
H	8.53664958476702	1.26626538165122	0.80712880532638

H	8.57113494298601	1.25257534744894	-0.94529747052356
H	9.88839665094885	-0.87207622909903	-0.90108057546239
H	9.85930131964807	-0.85417048437806	0.84933186423053
H	11.07820862292243	1.32742274433931	0.84774749404518
H	11.10430101854438	1.31067742424571	-0.90395229933919
H	-1.93523742327675	-0.77249311283800	0.44317948177901
H	-4.30802477377550	-1.68230359240546	0.34917211515112
C	-1.66885859523398	-1.40784216350314	-0.40566050465839
C	-4.02979145989728	-2.32978949183458	-0.48791345180100
C	-6.37690927036842	-3.23670732091933	-0.56950263677788
C	-0.19339384502672	-1.76292705631453	-0.33749192573160
H	-2.35063508139702	-3.23320725221057	0.50026449417144
C	-2.54562259400072	-2.65159391768067	-0.40559664882454
H	-1.86493477874553	-0.82098560080799	-1.30691632165530
H	-4.71793216862950	-4.14746467995665	0.43858417932331
C	-4.90033911215258	-3.57811429370525	-0.47579788986470
H	-4.22617176117907	-1.75538230565904	-1.39837053923662
H	-6.59510844718684	-2.69389570354387	-1.49139501718074
H	-2.26708284358713	-3.29318655182468	-1.24689379661019
H	-4.62626089669598	-4.22948832650732	-1.30884603390049
H	-6.68445811857829	-2.60122215981081	0.26367853956771
C	13.61982231960524	0.73390926282590	0.00759228290494
H	12.42315326976786	-0.81877592543019	-0.86154052388850
H	12.40206235547240	-0.79467523637226	0.89081858106233
H	13.62412643086676	1.39616156305770	0.87691093135617
H	13.65050447864366	1.36709627076321	-0.88182266584354
S	0.78862732964179	-0.24562578045066	-0.32187389810406
H	0.10444167111780	-2.36457905988529	-1.19407907252822
H	0.03088847328604	-2.33443269999158	0.56127311971855
S	5.86115332473717	0.74386984342125	-0.13022191448545
H	7.30369722663525	-0.92035702654609	0.80268738034555
H	7.34500328603196	-0.93966780164304	-0.95294107200285
C	2.40989360478355	-0.87838947448266	-0.25392139633429
C	4.60137493002337	-0.45697904411005	-0.17014283557485
N	2.60009001343310	-2.19530766814242	-0.23799033630941

C	3.87057705012469	-2.56922302857256	-0.18541887251512
N	4.91390011767163	-1.74465031082347	-0.14941382948024
S	4.27545163181807	-4.26201814722085	-0.15901826224769
H	-0.38707978105048	-9.06176079948164	-1.26955662877004
N	3.36724383325481	0.03485056198955	-0.22190245934497
C	2.63980182741951	-5.02796832515006	-0.23288378463530
S	-1.47906359076070	-11.59260322909230	-0.44761386326044
H	-1.14115580507700	-12.88791264858177	-0.43715433314020
C	2.77598216444795	-6.54065568586504	-0.21982179737716
H	2.05611021952988	-4.68045972502369	0.61750036230790
H	2.14058607105618	-4.68836039994048	-1.13858488561551
C	1.41572991162869	-7.21838269245512	-0.29520940130167
H	3.39433638051308	-6.86421738370862	-1.06136450345823
H	3.29495258940554	-6.85789285742549	0.68858525470467
C	1.49615081040439	-8.73660675698151	-0.28434575088042
H	0.79820234723161	-6.88401333989485	0.54437646487203
H	0.89952214742428	-6.89110462576745	-1.20328764616044
C	0.12627853615634	-9.39357111989009	-0.36258497698783
H	2.11188594712423	-9.07386394450170	-1.12366552425964
H	2.00879310230835	-9.06708460023862	0.62418741038890
C	0.20365190965437	-10.90978984157476	-0.35834292433810
H	-0.48702250895303	-9.06219017998229	0.48021073834653
H	0.68808649999688	-11.25996093600127	0.55233041514199
H	0.78120989534540	-11.26011580356057	-1.21295292527209

A.14 6-(ferrocenyl)hexanethiol cation

Table A.13 reports the coordinates of the optimized geometry of the isolated molecule with the UKS method using B3LYP functional with def2-TZVP basis set [119]. Grimme D3 correction is applied [120, 121]. The geometry optimization is performed using the ORCA package [117, 118].

Table A.14 Coordinates of the optimized geometry of the 6-(ferrocenyl)hexanethiol cation.

Atom	x (Å)	y (Å)	z (Å)
Fe	0.00000000	0.00000000	10.27544414
C	1.75741181	1.19931111	10.20897015
C	1.71406847	-0.97754321	9.51374548
C	1.73030159	0.37966955	11.36995726
C	1.70313906	-0.97742935	10.93794539
C	1.76800553	0.36878248	9.05407579
C	-1.81805619	0.52682872	9.24946272
C	-1.78035235	1.19297611	10.49807779
C	-1.75108029	-0.87044459	9.49021903
C	-1.69260774	0.21267599	11.52107193
C	-1.67593609	-1.06924566	10.89998045
H	1.75271051	2.27835367	10.19685004
H	1.67382297	-1.85221549	8.88543586
H	1.72291843	0.72432396	12.39139080
H	1.67621685	-1.84871587	11.57212586
H	-1.84691330	1.00217965	8.28195304
H	-1.77910736	2.26155943	10.64463971
H	-1.74808767	-1.64307959	8.73853645
H	-1.63721399	0.40692972	12.57996902
H	-1.62162310	-2.01972459	11.40513971
C	1.81721890	0.84228397	7.63759988
C	1.10802695	-0.04733842	6.61502470
H	1.42163781	1.85973562	7.58828306
H	2.87652122	0.92330267	7.36503746
H	0.05823055	-0.17053626	6.90232651
H	1.54964274	-1.04833471	6.62321803
C	1.18544820	0.52590366	5.20125409
H	2.23629667	0.65196415	4.92082039
H	0.74531611	1.52905982	5.19354071
C	0.48444659	-0.34244069	4.15933679
C	0.56498369	0.24709108	2.75001673
H	0.93088532	-1.34290832	4.16124239
H	-0.56688054	-0.47432003	4.44085693
H	0.11623486	1.24553697	2.74574776

H	1.61099389	0.37881021	2.46127191
C	-0.13298535	-0.62243683	1.71174449
H	-1.18586833	-0.76469807	1.96086864
H	0.32439258	-1.61333006	1.67428074
S	0.00000000	0.00000000	0.00000000
H	-0.71343658	1.12433965	0.18369314

Appendix B

BBchar code

Appendix B reports the the MATLAB code of the main scripts and functions of the Block-Based characterization (BBchar) tool.

The following code reports, as an example, the launching script for the simulation of all the input combinations and subsequent characterization of a XOR gate.

```
clear variables
close all

%% Paths definition
5 myDataPath = '~';
  BBcharPath = fullfile(myDataPath, 'BBchar');
  BBcharCodePath = fullfile(BBcharPath, 'Code');
  thisPath = pwd;
  scerpaPath = fullfile(myDataPath, 'scerpa');
10 libraryPath = fullfile(BBcharPath, 'Lib');

  outputPath = fullfile(BBcharPath, 'Layouts', 'xor');
  file = 'xor.qll';

15

%% Clock signal parameters

%definitions
clock_low = -2;
```

```
20 clock_high = +2;
   clock_step = 5;

   %Step simulation implementation
   pSwitch = linspace(clock_low, clock_high, clock_step);
25 pHold = linspace(clock_high, clock_high, clock_step);
   pRelease = linspace(clock_high, clock_low, clock_step);
   driverPara.pReset = linspace(clock_low, clock_low, ...
       clock_step);

30 %Cycle to simulate
   driverPara.pCycle = [pSwitch pHold pRelease ...
       driverPara.pReset];

   %% Driver parameters
35 driverPara.doubleMolDriver = 1;
   driverPara.Ninputs = 4; %Number of layout physical inputs
   driverPara.driverNames = [{'Dr1'} {'Dr2'} {'DrC0'} ...
       {'DrC1'}]; %list of the drivers name as they are in
       % the .qll file
40 driverPara.driverModes = [{'0'} {'0'} {'1'} {'1'}];
       % Dr1 combinations
       {'0'} {'1'} {'0'} {'1'}];
       % Dr2 combinations
       {'0'} {'0'} {'0'} {'0'}];
45 % DrC0 combinations
       {'1'} {'1'} {'1'} {'1'}];
       % DrC1 combinations

   driverPara.sweepType = 'lin'; %sweep creation following
50 % a linspace ('lin')
       % or a logspace ('log')

   driverPara.NsweepSteps = 1;
   driverPara.cycleLength = length(driverPara.pCycle);
   driverPara.clockStep = clock_step;
55 driverPara.NclockRegions = 4; % number of clock regions in
       % the layout
   driverPara.phasesRepetition = 3; % How many time regions
```



```
                                % repeat in the layout
driverPara.maxVoltage = 1; % maximum voltage (absolute
60                                % value) the driver will assume

%% Termination settings
%set to '1' if you want to add a termination to the layout
terminationSettings.enableTermination = 1;
65 %specify a custom number of cells to realize each the
% termination. Default is 8 molecules
terminationSettings.customLength = 0;
%set to 1 for bus layouts, set to 0 for single line layouts
terminationSettings.busLayout = 1;
70

%% SCERPA settings
%layout (MagCAD)
circuit.qllFile = fullfile(pwd,file);
circuit.magcadImporter = 1;
75 circuit.doubleMolDriverMode = driverPara.doubleMolDriver;
circuit.outIsPin = 0;

%algorithm settings
settings.out_path = outputPath;
80 settings.damping = 0.6;
settings.verbosity = 0;
settings.dumpDriver = 1;
settings.dumpOutput = 1;
settings.dumpClock = 1;
85 settings.dumpVout = 1;

%viewer settings
plotSettings.plot_waveform = 1;
plotSettings.plot_waveform_index = 1;
90 plotSettings.plot_3dfig = 0;
plotSettings.plot_1DCharge = 0;
plotSettings.plot_logic = 1;
plotSettings.plot_potential = 1;
plotSettings.plotSpan = clock_step;
95 plotSettings.fig_saver = 0;
```

```
plotSettings.HQimage = 0;

if isfield(settings, 'out_path')
    plotSettings.out_path = settings.out_path;
100 end

%% Characterization settings
charSettings.LibPath = libraryPath;
charSettings.LibDeviceName = 'xor';
105 charSettings.out_path = outputPath;

%% Launch the BBchar software
simulate = 1;
characterize = 1;
110

cd(BBcharCodePath)
circuit.Values_Dr = buildDriver(driverPara);
circuit.stack_phase = buildClock(driverPara);
% if the user want the termination
115 if terminationSettings.enableTermination
    [circuit, terminationCircuit] = add_termination(...
        circuit, terminationSettings, driverPara.pCycle, ...
        length(driverPara.pReset));
else %termination not enabled
120     termination.num = 0;
end
cd(thisPath)

% launcher
125 if simulate
    if terminationSettings.enableTermination
        circuit.qllFile = terminationCircuit.filepath;
        settings.out_path = outputPath;
        plotSettings.out_path = settings.out_path;
130     end
    cd(scerpaPath)
    diary on
    SCERPA('generateLaunchView', circuit, settings, ...
```

```

        plotSettings);
135  diary off
      if isfield(settings,'out_path')
          movefile('diary',fullfile(settings.out_path,...
              'logfile.log'))
      end
140  cd(thisPath)
end
if characterize
    cd(BBcharCodePath)
    tic
145  characterization(charSettings,terminationSettings,...
        terminationCircuit,driverPara,circuit.Values_Dr);
    charTime = toc;
    cd(thisPath)
end

```

The following function generated the driver matrix to perform a Self-Consistent Electrostatic Potential Algorithm (SCERPA) simulation considering the user inputs.

```

function [valuesDr] = buildDriver(driverSettings)

if ~isfield(driverSettings,'sweepType')
    warning(['Missing sweep type for drivers, default' ...
5         ' will be used, default value is ''lin'' '])
    driverSettings.sweepType = 'lin';
end

switch(driverSettings.sweepType)
10  case 'lin'
        % from smaller to bigger value
        variation = linspace(driverSettings.maxVoltage,...
            -driverSettings.maxVoltage, ...
            driverSettings.NsweepSteps);
15  %from bigger to smaller value
        not_variation = linspace(...
            -driverSettings.maxVoltage,...
            driverSettings.maxVoltage, ...

```

```
        driverSettings.NsweepSteps);
20 case 'log'
    % generates NsweepSteps/2 points between decades
    % 10^-2 and 10^0. 0.01 1
    half_pos_variation = driverSettings.maxVoltage*...
        logspace(-2, 0, driverSettings.NsweepSteps/2);
25 overturned_half_pos_variation = ...
        driverSettings.maxVoltage*logspace(...
        0, -2, driverSettings.NsweepSteps/2);
%1 0.01

    % Add a central '0' for compliance with odd
    % NsweepSteps
30 if mod(driverSettings.NsweepSteps,2)
        variation = [ overturned_half_pos_variation...
            0 -half_pos_variation];
        not_variation = ...
35         [-overturned_half_pos_variation 0 ...
            half_pos_variation];
    else
        variation = [ overturned_half_pos_variation ...
            -half_pos_variation];
40 not_variation = ...
            [-overturned_half_pos_variation ...
            half_pos_variation];
    end
otherwise
45 warning(['Sweep type for drivers not ' ...
        'recognized, default will be used, ' ...
        'default value is ''lin'' '])
    variation = linspace(driverSettings.maxVoltage, ...
        -driverSettings.maxVoltage, ...
50 driverSettings.NsweepSteps);
    not_variation = linspace(...
        -driverSettings.maxVoltage, ...
        driverSettings.maxVoltage, ...
        driverSettings.NsweepSteps);
55 end
```

```

Ndrivers = driverSettings.Ninputs;

% each element of variation is repeated cycleLength times
60 D_sweep = repelem(variation, driverSettings.cycleLength);
% each element of not_variation is repeated cycleLength
% times
D_not_sweep = repelem(not_variation, ...
    driverSettings.cycleLength);
65
D_one = -driverSettings.maxVoltage * ones(1, ...
    length(D_sweep));
D_zero = driverSettings.maxVoltage * ones(1, ...
    length(D_sweep));
70
% the number of columns of driverModes tells the number of
% driver values' combinations
[~, Ncomb] = size(driverSettings.driverModes);

75 %length of the values to insert to empty the pipe and make
% the last input to propagate till the end of the circuit
empty_pipe = (driverSettings.phasesRepetition)* ...
    driverSettings.cycleLength;
Dr_tot = zeros(1,length(D_sweep)*Ncomb + empty_pipe);
80 % + 1 for the input name, times 2 if doubleMolDriver
valuesDr = cell(Ndrivers+Ndrivers* ...
    driverSettings.doubleMolDriver, length(Dr_tot)+1);

for nDr = 1:Ndrivers %for each input branch
85 for jj = 0:Ncomb-1 % for each combination value
    currentArrayElements = ...
        (jj*length(D_sweep) + 1):(length(D_sweep)+ ...
        jj*length(D_sweep));
    switch(driverSettings.driverModes{nDr, jj+1})
90 case '0'
        Dr_tot(currentArrayElements) = D_zero;
    case '1'
        Dr_tot(currentArrayElements) = D_one;

```

```

    case 'sweep'
        Dr_tot(currentArrayElements) = D_sweep;
    case 'not_sweep'
        Dr_tot(currentArrayElements) = D_not_sweep;
    end
end
100
% filling elements for the empty pipe phase looking the
% last value of the combinations
if strcmp(driverSettings.driverModes{nDr,end},...
    'sweep') || strcmp(...
105
    driverSettings.driverModes{nDr,end},'1')
    Dr_tot(currentArrayElements(end)+1:end) = ...
        -driverSettings.maxVoltage * ...
        ones(1,empty_pipe);
else % not_sweep or '0'
110
    Dr_tot(currentArrayElements(end)+1:end) = ...
        driverSettings.maxVoltage * ones(1,empty_pipe);
end

%write dr name in row ii and column 1
115
valuesDr{nDr,1} = driverSettings.driverNames{nDr};
valuesDr(nDr,2:end) = [num2cell(Dr_tot)];
if driverSettings.doubleMolDriver
    %write dr_c name in row ii+Ndrives and column 1,
    % so we obtain Dr1,Dr2,...,DrN,Dr1_c,Dr2_c,...,DrN_c
120
    valuesDr{nDr+Ndrivers,1} = ...
        [driverSettings.driverNames{nDr} '_c'];
    valuesDr(nDr+Ndrivers,2:end) = ...
        [num2cell(-Dr_tot)];
end
125
end
end

```

The following code report the function to create the clock matrix to perform a SCERPA simulation considering the user inputs.

```
function [stack_phase] = buildClock(settings)
```

```

    [~,Ncomb] = size(settings.driverModes);
    filler = repmat(settings.pReset,1,...
5         settings.NclockRegions);
    completeCycle = [repmat(settings.pCycle,1,...
        settings.NsweepSteps*Ncomb+(...
        settings.phasesRepetition-1))...
        filler];
10    stack_phase = zeros(settings.NclockRegions,length(...
        completeCycle));

    for ii = 1:settings.NclockRegions
        stack_phase(ii,:) = circshift(completeCycle,...
15         length(settings.pReset)*(ii-1));
    end

end

```

The following code describes the function used to attach a termination to each Device Under Test (DUT) output, creating a new MagCAD input file. It also adjust the driver and clock matrices to include the newly added termination circuit.

```

function [QCAcircuit, termination] = add_termination(...
    QCAcircuit, settings, pCycle, pResetLength)

% Read the circuit qll file and extract the name and the
% number of the independent output
5    CUTstruct = xmlRead(QCAcircuit.qllFile);
    [outNames,indexName] = unique( CUTstruct.output.name );
    Noutputs = length(outNames);

10    if settings.busLayout && Noutputs == length(...
        CUTstruct.output.name)
        error("Dependent output in bus layout MUST have" + ...
            " the same name. Please correct the layout!");
    end

15    if isfield(settings,'customLength') && ...

```

```
        settings.customLength ~= 0
        termination.Length = settings.customLength;
else
20     %default termination has 8 molecules (4 cells)
        termination.Length = 4;
end

termination.numOutput = Noutputs;
25 termination.filepath = insertBefore(QCAcircuit.qllFile,...
    '.qll','_termination');
copyfile(QCAcircuit.qllFile, termination.filepath);

%add for cycle for each output
30 for nOut = 1:Noutputs
    % get output info
    OUT_x = CUTstruct.output.x(indexName(nOut));
    OUT_y = CUTstruct.output.y(indexName(nOut));
    OUT_angle = CUTstruct.output.angle(indexName(nOut));
35
    if OUT_angle == 0
        test_x = OUT_x - 1;
    elseif OUT_angle == 90
        test_y = OUT_y - 1;
40    elseif OUT_angle == 270
        test_y = OUT_y + 1;
    else
        error(['Angle not compatible with the actual' ...
            'program. Please correct the .qll and try' ...
45            'again']);
    end

    %find the molecule near the output and read the phase
    if exist('test_x','var') && ...
50        any([CUTstruct.molecules.x] == test_x)
        index = find([CUTstruct.molecules(:).x] == test_x);
        nearMolPhase = CUTstruct.molecules(index(1)).phase;
    elseif exist('test_y','var') && ...
        any([CUTstruct.molecules.y] == test_y)
```



```
55     index = find([CUTstruct.molecules(:).y] == test_y);
        nearMolPhase = CUTstruct.molecules(index(1)).phase;
    else
        error(['No molecule near the output: check and' ...
              'correct the layout'])
    end
60     Nphases = max([CUTstruct.molecules(:).phase]) + 1;
    termination.stack(nOut).phase = ...
        mod(nearMolPhase+1,Nphases);

65     % Compute every id used in the original qll to extract
    % the next one to use
    mol_id_MAX = ...
        max(str2double([CUTstruct.molecules(:).id]));
    out_id_MAX = max(str2double([CUTstruct.output(:).id]));
70     drv_id_MAX = max(str2double([CUTstruct.driver(:).id]));
    id_MAX = max([mol_id_MAX out_id_MAX drv_id_MAX]);
    termination.stack(nOut).StartID = id_MAX + 1;

    if settings.busLayout
75         termination.stack(nOut).StartX = ...
            [CUTstruct.output.x(indexName(nOut)) ...
            CUTstruct.output.x(indexName(nOut) + 1)];
        termination.stack(nOut).StartY = ...
            [CUTstruct.output.y(indexName(nOut)) ...
80             CUTstruct.output.y(indexName(nOut) + 1)];
    else
        termination.stack(nOut).StartX = ...
            CUTstruct.output.x(indexName(nOut));
        termination.stack(nOut).StartY = ...
85         CUTstruct.output.y(indexName(nOut));
    end
    termination.stack(nOut).angle = OUT_angle;

    terminationFileContent = createQLLfile(...
90         termination,nOut,settings.busLayout);
    fileID = fopen(termination.filepath,'w');
    fprintf(fileID,terminationFileContent);
```

```

        fclose(fileID);
95 end

    % Update Values_Dr and stack_phase

    %Evaluate the last driver value to copy it n-times
100 % at the end
    driver_last_value = QCAcircuit.Values_Dr(:,end);
    %Create the matrix of repeated values to attach at the end
    % of Values_Dr
    flap_matrix = repmat(driver_last_value,1,length(pCycle));
105 %Attach flap matrix
    QCAcircuit.Values_Dr = [QCAcircuit.Values_Dr flap_matrix];

    %Insert a pCycle in stack_phase in the correct position for
    % each row, considering the presence of the pReset filler
110 newPhaseMatrix = zeros(size(QCAcircuit.stack_phase,1),...
        size(QCAcircuit.stack_phase,2) + length(pCycle));
    for ii = 1:size(QCAcircuit.stack_phase,1)
        pos = [1:length(pCycle)] + pResetLength*(ii-1);
        idx = ones(1,length(newPhaseMatrix));
115         idx(pos) = 0;
        newPhaseMatrix(ii,pos) = pCycle;
        newPhaseMatrix(ii,logical(idx)) = ...
            QCAcircuit.stack_phase(ii,:);
    end
120 QCAcircuit.stack_phase = newPhaseMatrix;

end

%% Function used to generate the new qll file with the
125 % termination
function [terminationFile] = createQLLfile(...
    termination,nOut,busLayout)

    terminationFile = fileread(termination.filepath);
130    molLineTemplate = fileread('molLineTemplate.qll');

```

```
newID = termination.stack(nOut).StartID;
newX = termination.stack(nOut).StartX;
newY = termination.stack(nOut).StartY;
135
for ii = 1:termination.Length
    strToAdd = molLineTemplate;
    strToAdd = strrep(strToAdd,"$ID$",num2str(newID));
    strToAdd = strrep(strToAdd,"$PHASE$",num2str(...
140         termination.stack(nOut).phase));
    strToAdd = strrep(strToAdd,"$X$",num2str(newX(1)));
    strToAdd = strrep(strToAdd,"$Y$",num2str(newY(1)));
    terminationFile = insertBefore(terminationFile,...
        '    </layout>',strToAdd);
145
if busLayout
    newID = newID + 1;
    strToAdd = molLineTemplate;
    strToAdd = ...
150         strrep(strToAdd,"$ID$",num2str(newID));
    strToAdd = ...
        strrep(strToAdd,"$PHASE$",num2str(...
            termination.stack(nOut).phase));
    strToAdd = ...
155         strrep(strToAdd,"$X$",num2str(newX(2)));
    strToAdd = ...
        strrep(strToAdd,"$Y$",num2str(newY(2)));
    terminationFile = ...
        insertBefore(terminationFile,...
160         '    </layout>',strToAdd);
end

newID = newID + 1;
if termination.stack(nOut).angle == 0
165     newX = newX + 1;
elseif termination.stack(nOut).angle == 90
    newY = newY + 1;
elseif termination.stack(nOut).angle == 270
```

```

    newY = newY - 1;
170     end

    end

end

```

The following function is responsible for the analysis of the output produced by SCERPA to extract the characterization data. It creates a new folder in the library where it places the characterization files.

```

%% VIN/VOUT - CHARACTERISTIC

function characterization(charSettings,...
    terminationSettings,terminationCircuit,...
5     driverPara,drivers)

% Paths definition
if ~isfolder(charSettings.LibPath)
    % creating main Library directory if it doesn't exist
10     mkdir(charSettings.LibPath);
end
% path of the directory related to the specific circuit
currentDeviceLibPath = fullfile(charSettings.LibPath,...
    charSettings.LibDeviceName);
15 if ~isfolder(currentDeviceLibPath)
    mkdir(currentDeviceLibPath);
end
outFolderPath = fullfile(charSettings.out_path,...
    'SCERPA_OUTPUT_FILES');
20

% Read simulation data
table_AI = readtable(fullfile(outFolderPath,...
    'Additional_Information.txt'),'VariableNamesLine',1);
%rename time column
25 table_AI.Properties.VariableNames{1} = 'Time';
availableRows = table_AI.Properties.VariableNames;
table_AI_array = table2array(table_AI);

```

```
%flag that tells whether it is a bus layout or not
30 bus_flag = terminationSettings.busLayout;

%% Extract output values from the table
% Evaluating the position as expressed in the qll file of
35 % the molecules right before the terminations and associate
% the new names for the table
for oo = 1:terminationCircuit.numOutput
    switch terminationCircuit.stack(oo).angle
        case 0
40             if bus_flag
                    % order output as from lowest y to the
                    % highest, for bus structure H before L
                    out_pos_qll((2*oo-1):(2*oo),:) = ...
                        sortrows([zeros(2,1) ...
45                             terminationCircuit.stack(oo).StartY'...
                                (terminationCircuit.stack(oo).StartX...
                                    -1)'] ,2);
                    new_outNames{4*oo-3} = ...
                        sprintf('Vou_%dA_H',oo);
50                    new_outNames{4*oo-2} = ...
                        sprintf('Vou_%dB_H',oo);
                    new_outNames{4*oo-1} = ...
                        sprintf('Vou_%dA_L',oo);
                    new_outNames{4*oo} = sprintf('Vou_%dB_L',oo);
55                else
                    out_pos_qll(oo,:) = ...
                        sortrows([0 ...
                                terminationCircuit.stack(oo).StartY'...
                                    (terminationCircuit.stack(oo).StartX...
                                        -1)'] ,2);
60                    new_outNames{2*oo-1} = ...
                        sprintf('Vou_%dA',oo);
                    new_outNames{2*oo} = ...
                        sprintf('Vou_%dB',oo);
65                end
            end
        end
    end
end
```

```

70     case 90
       if bus_flag
           out_pos_qll((2*oo-1):(2*oo),:) = ...
               sortrows([zeros(2,1) ...
               (terminationCircuit.stack(oo).StartY...
               -1)' ...
               terminationCircuit.stack(oo).StartX'...
               ],3);
       new_outNames{4*oo-3} = ...
75           sprintf('Vou_%dA_dH',oo);
       new_outNames{4*oo-2} = ...
           sprintf('Vou_%dB_dH',oo);
       new_outNames{4*oo-1} = ...
           sprintf('Vou_%dA_dL',oo);
80       new_outNames{4*oo} = ...
           sprintf('Vou_%dB_dL',oo);
       else
           out_pos_qll(oo,:) = sortrows([0 ...
           (terminationCircuit.stack(oo).StartY...
           -1)' ...
           terminationCircuit.stack(oo).StartX'],...
           3);
           new_outNames{2*oo-1} = ...
90             sprintf('Vou_%dA_d',oo);
           new_outNames{2*oo} = ...
               sprintf('Vou_%dB_d',oo);
       end
       case 270
       if bus_flag
95           out_pos_qll((2*oo-1):(2*oo),:) = ...
               sortrows([zeros(2,1) ...
               (terminationCircuit.stack(oo).StartY+...
               1)' ...
               terminationCircuit.stack(oo).StartX'],...
               3);
           new_outNames{4*oo-3} = ...
100             sprintf('Vou_%dA_uH',oo);
           new_outNames{4*oo-2} = ...

```

```

        sprintf('Vou_%dB_uH',oo);
105     new_outNames{4*oo-1} = ...
        sprintf('Vou_%dA_uL',oo);
        new_outNames{4*oo} = ...
        sprintf('Vou_%dB_uL',oo);
    else
110     out_pos_qll(oo,:) = sortrows([0 ...
        (terminationCircuit.stack(oo).StartY...
        +1)' ...
        terminationCircuit.stack(oo).StartX']...
        ,3);
115     new_outNames{2*oo-1} = ...
        sprintf('Vou_%dA_u',oo);
        new_outNames{2*oo} = ...
        sprintf('Vou_%dB_u',oo);
    end
120 end
end

%read the qll to extract the qll_identifier and the phase
qllFile_term = fileread(fullfile(charSettings.out_path,...
125     strcat(charSettings.LibDeviceName,...
        '_termination.qll')));
abq = '([~"]+)'; % anything but quotation mark
for oo = 1:size(out_pos_qll,1)
    xpr = ['<item comp="',abq,'" id="',abq,'"( angle="',...
130     abq,'"? x="',num2str(out_pos_qll(oo,3)),...
        '" y="',num2str(out_pos_qll(oo,2)),...
        '" layer="0">(\s*<property' ...
        ' name="',abq,'" value="',abq,'"/>)*'];
    tmp = regexp(qllFile_term, xpr, 'tokens');
135     outID{2*oo-1} = ['Vout_' sprintf('%.4da',...
        str2double(tmp{1,1}{1,2}))];
    outID{2*oo} = ['Vout_' sprintf('%.4db',...
        str2double(tmp{1,1}{1,2}))];
    % get the phase of each output molecules
140     phasecell = regexp(tmp{1,1}{1,4}, ['\s*<property ' ...
        'name="phase" value="',abq,'"/>'], 'tokens');

```

```

        outPhase(2*oo-1) = str2double(phasecell{1,1})+1;
        outPhase(2*oo) = str2double(phasecell{1,1})+1;
    end
145 % extract the column indexes of the table where output
    % voltages are stores in the correct order
    [tf,loc] = ismember(availableRows,outID);
    [~,p] = sort(loc(tf));
    out_cols = find(tf);
150 out_cols = out_cols(p);

    %% Get the driver values list from the driver
    % values are the first occurrence
    [driverValues,index] = unique(cell2mat(...
155     drivers(:,2:end)).','rows','stable');
    in_firsthalf_col = 1:(size(driverValues,2)/2);
    in_secondhalf_col = (size(driverValues,2)/2+1):size(...
        driverValues,2);
    %drivers are reversed in the origianl matrix
160 interleaved_idx = [in_secondhalf_col;in_firsthalf_col];
    interleaved_idx = repelem(interleaved_idx,1,2);
    interleaved_idx = interleaved_idx(:)';
    driverValues = driverValues(:,interleaved_idx).';
    Ndrivers = length(driverPara.driverNames);
165 for dd = 1:Ndrivers
        if bus_flag
            new_driverNames{4*dd-3} = sprintf('Vin_%dA_H',dd);
            new_driverNames{4*dd-2} = sprintf('Vin_%dB_H',dd);
            new_driverNames{4*dd-1} = sprintf('Vin_%dA_L',dd);
170            new_driverNames{4*dd} = sprintf('Vin_%dB_L',dd);
        else
            new_driverNames{2*dd-1} = sprintf('Vin_%dA',dd);
            new_driverNames{2*dd} = sprintf('Vin_%dB',dd);
        end
175 end

    %% Extract the time instants (table rows) corresponding
    %% with the output in the hold phase for the selected
    %% input combination

```



```
180 %the first time depend on the output phase and the phase
% repetition in the circuit, the others are separated by a
% cycle lenght
time_rows = index + (outPhase+1).*driverPara.clockStep ...
    + (driverPara.cycleLength * ...
185     (driverPara.phasesRepetition -1));

%% Generate the lookup table and save it to the library
table_header = [new_driverNames new_outNames];
table_values = zeros(size(driverValues,2),...
190     size(driverValues,1)+length(out_cols));
for comb = 1:size(driverValues,2)
    table_values(comb,:) = [driverValues(:,comb)' ...
        diag((table_AI_array(time_rows(comb,:),...
            out_cols)),0)'];
195 end

%creating the table containing output and drivers
T = array2table(table_values, 'VariableNames', table_header);
path = fullfile(currentDeviceLibPath, 'table.csv');
writetable(T, path); %creation of the .csv file
200

%% Generate the additional information file and save it to
%% the library
% saving the min and max coordinates in y-axis and z-axis
% (to evaluate the number of cells)
205 qllFile = fullfile(charSettings.out_path,...
    strcat(charSettings.LibDeviceName, '.qll'));
copyfile(qllFile, currentDeviceLibPath)
circuit = xmlRead(qllFile);

210 ymin = min([circuit.molecules.y]);
ymax = max([circuit.molecules.y]);
zmin = min([circuit.molecules.x]);
zmax = max([circuit.molecules.x]);
y_cell_num = (ymax - ymin + 1);
215 z_cell_num = (zmax - zmin + 1);
area_cells = y_cell_num*z_cell_num;
```

```

%latency of the circuit
latency = driverPara.phasesRepetition*...
220     driverPara.NclockRegions;

% writing the .txt file with the additional informations
fileid = fopen(fullfile(currentDeviceLibPath,...
    'info.txt'),'w');
225 fprintf(fileid,'Number of outputs\t%d\n',length(outID));
for w = 1:length(outID)
    fprintf(fileid,'%s\t',new_outNames{w});
end
fprintf(fileid,'\nLatency\t%d',latency);
230 fprintf(fileid,'\nNumber of clock phases\t%d',...
    driverPara.NclockRegions);
fprintf(fileid,'\nMaximum driver voltage\t%.2f V',...
    driverPara.maxVoltage);
fprintf(fileid,'\nTotal area\t%dx%d=%d',y_cell_num,...
235     z_cell_num,area_cells);
fclose(fileid);

end

```

The following function is used for reading the Look-Up Table (LUT) of a device and providing the output voltages corresponding to the specified input.

```

function [Vout] = InOut_eval(LibPath,device,input)

    %open the file that contains the names of
    % the output files
5     infoFile = fileread(fullfile(LibPath,device,...
        'info.txt'));
    tmp = regexp(infoFile,...
        'Number of outputs\s([0-9]+\)\n','tokens');
    num_out = str2double(cell2mat(tmp{1}));
10
    LUTtable = readtable(fullfile(LibPath,device,...
        'table.csv'),'VariableNamesLine',1);
    availableRows = LUTtable.Properties.VariableNames;

```

```
15     if isempty(regexp(availableRows{1,end},...
        'Vou_[0-9]*[A|B]_[a-z]?[H|L]', 'once'))
        busFlag = 0;
    else
        busFlag = 1;
    end
20     LUT = table2array(LUTtable);
    num_in = size(LUT,2) - num_out;
    LUT_in = LUT(:,1:num_in);

    if num_in ~= size(input,1)*size(input,2)
25         error("The size of the provided input " + ...
            "is not consistent with the device.")
    end

    in_total = input';
30     in_total = in_total(:)';

    % calculate the norm of each row of LUT-input
    dist = vecnorm(LUT_in - in_total, 2, 2);
    %the nearest vector has the minimum norm
35     [~, match] = min(dist);
    Vout = reshape(LUT(match,num_in+1:end),2,num_out/2)';

end
```