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# Revisiting heat treatments for additive manufactured parts: A case study of A20X alloy

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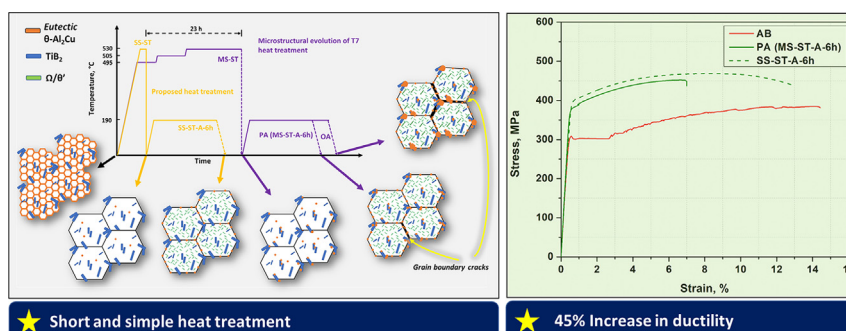
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## HIGHLIGHTS

- Post-processing T7 heat treatment and microstructural evolution of laser powder fused A20X alloy was investigated.
- T7 offers undissolved  $\theta$ -Al<sub>2</sub>Cu phase near TiB<sub>2</sub> particles and promotes early precipitate free zones formation upon aging.
- A novel short and simple heat treatment was proposed for the improved mechanical behaviour.
- Exceptional improvement (~45%) in the tensile ductility was achieved with respect to T7.

## GRAPHICAL ABSTRACT



★ Short and simple heat treatment

★ 45% Increase in ductility

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## ABSTRACT

A20X (Al-Cu-Ag-Mg-TiB<sub>2</sub>) is a precipitation hardening alloy, recently developed for additive manufacturing processing. Printed parts of A20X alloy are usually post-processed with a long T7 heat treatment for improved mechanical properties with respect to its as-built counterparts. However, in the present investigation, it was demonstrated that T7 might not be the best suitable heat treatment available for A20X alloy. A detailed microstructural characterization of A20X samples processed with laser powder bed fusion and post-processed with T7 was carried out. Microstructural features were analysed in terms of grain size, precipitate size, phase quantification, dislocation density and width of the precipitate free zones. After the analysis, a simple and rapid heat treatment was proposed which significantly improved the mechanical properties. The yield strength (YS), ultimate tensile strength (UTS) and elongation to fracture (e) for the T7 heat treatment were  $370 \pm 9$  MPa,  $435 \pm 13$  MPa and  $7.3 \pm 0.3$  % respectively. With the proposed heat treatment, an increment of 7.1 % in YS, 6.3 % in UTS and 45 % in e was witnessed. This exceptional improvement in the mechanical behaviour has been associated with the absence of grain boundary cracking in the proposed heat treatment.

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## 1. Introduction

Additive manufacturing (AM) is a build-up process of 3D parts obtained by joining the material in a layer by layer fashion [1].

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The processing conditions of additive manufacturing technique are quite different than the traditional manufacturing methods [2]. For instance, processing conditions of AM processes such as laser powder bed fusion (LPBF), directed energy deposition (DED) and electron beam melting (EBM) are characterized by the following: (i) high heating (melting) and cooling (solidifying) rates, (ii) melting of a top powder layer and consequent re-melting of an underlying solidified layer, and (iii) repeated thermal cycles of the previously deposited layers. Such a complex processing conditions in combination with materials chemistry causes various microstructural defects [2–4] which severely degrade the mechanical properties of the printed parts which were intended for structural applications. However, now it has become a common practice for the printed parts to undergo a post-processing treatments to mitigate or eliminate such microstructural defects and to modify the microstructure in order to achieve the most suitable properties [3,5].

In particular, post-processing heat treatments becomes essential if the printed parts are of a precipitation-hardening alloy [2,3,6,7]. Recently, certain precipitation-hardening Al alloys have demonstrated good processability by AM. Among those Al alloys, A20X™ has gained quite a success and has been certified by Society of Automotive Engineers (SAE) International as a commercial alloy for aerospace application [8,9]. A20X™ is an Al-Cu-Ag-Mg alloy along with TiB<sub>2</sub> as a reinforcement. This composition has been developed and patented by Aeromet international limited, UK for both casting and AM processes [10,11]. The alloy benefits from the reduced hot cracking susceptibility, which is one of the major problems in the printability of precipitation hardening Al alloys by AM techniques [12] and increased strength due to the precipitation of various phases during aging [13,14].

The company (Aeromet) suggested a modified-T7 post-processing heat treatment for A20X castings [15]. This heat treatment consists of a multi-step solutioning i.e. 495 °C for 5 h + 505 °C for 6 h + 525 °C for 11 h + 538 °C for 24 h followed by artificial aging, such a long solution treatment (about 46 h) has shown a complete dissolution of Cu, which upon aging could effectively take part to the precipitation. Multi-step solution treatments are generally performed to avoid the occurrence of incipient melting in as-cast Al-Cu based alloys [16–19]. This phenomenon can be observed in such alloys due to the presence of phases with lower melting temperatures than the solutioning temperature. The occurrence of incipient melting causes the formation of micro-cracks and pores during solution treatment and degrades the mechanical properties and corrosion resistance of the material.

Due to this, multi-step solution treatments have been widely investigated [16–19]. Zamani et al. [19], utilized two-step solution treatment (490 °C for 5 h + 512 °C for 20 h) for as-cast Al-Cu-(Mg-Ag) alloy. The authors have reported that addition of Mg and Ag favours the formation of S-Al<sub>2</sub>CuMg and Q-Al<sub>7</sub>Cu<sub>3</sub>Mg<sub>6</sub> phases. The S and Q phase showed their peak melting temperature in DSC profile at around 495 °C and 513 °C, respectively, and thus a two-step solution treatment could be able to dissolve these phases into the Al-matrix rather than melting. Similarly, Masuku et al., also suggested a two-step solution treatment (515 °C for 5 h + above 515 °C for 15 h) for an Al-5Cu-Mg-Ag alloy to ensure the complete dissolution of these low temperature melting phases [16]. Daswa et al. performed a comparative study between single-step (525 °C for 16 h) and multi-step solution treatment (controlled heating from 400 °C to 513 °C + 513 °C for 2 h + 525 °C for 16 h). It was reported that, a single-step solution treatment resulted in the formation of incipient melting whereas after a multi-step solution treatment no such defects were detected [18]. Similar heat treatments which were developed for castings have also been utilized for the AM printed parts. Avateffazeli et al. [20] and Shakil et al. [21] recently carried out a post-

processing heat treatment similar to the modified-T7 i.e. 505 °C for 2 h + 530 °C for 4 h followed by aging at 190 °C for 4–6 h, and compared their bulk- and micro-mechanical properties with the as-built A20X counterparts. Both studies have reported an increase in the strength after T7 heat treatment compared to their as-built counterparts.

However, a detailed study on the microstructural evolution and the effects of a T7 heat treatment on additively manufactured A20X alloy has not yet been reported. It is well known that the processing conditions differ significantly between the casting and AM and so does their microstructural features. Thus, heat treatments which are specific to the microstructural features should be redesigned to exploit the full potential of the materials performance for the intended applications such as mechanical behaviour, corrosion resistance etc. In the present study, a detailed investigation of the microstructural evolution during the T7 heat treatment of the A20X alloy processed by LPBF is presented. The outcome of the work has paved the way to the design of a more promising heat treatment.

## 2. Materials and methods

A gas atomized A20X powder provided by ECKART and ALTANA Company was used for printing bulk samples. The composition of the powder, provided by the supplier, is presented in Table 1. The powder particles size distribution (PSD) was evaluated by image analysis of at least 20,000 particles using a Phenom XL table-top Scanning Electron Microscope (SEM).

Parallelepiped samples (12x10x10 mm<sup>3</sup>) were printed by LPBF process using an EOS M270 dual mode system. The optimized parameters employed for printing are presented in Table 2. These parameters were optimized after density measurements. The printing process was performed in an argon atmosphere, platform was pre-heated to 100 °C, and rotated scanning strategy was 67°.

The post-processing multi-step solution treatment (MS-ST) was performed in a horizontal tubular furnace (Nabertherm RHTC 80–710/15) under argon atmosphere, followed by water quenching (WQ) at room temperature. Various interrupted solution treatments were also performed to track the microstructural evolution. The details and the nomenclature of these heat treatments are presented in Table 3. Aging was carried out in an oven at 190 °C for various intervals of time, followed by air cooling. The steps of the heat treatment are presented as a schematic in Fig. 1.

Metallographic sample preparations were performed by polishing according to the standard metallographic procedure and by etching with Keller's reagent for 5 s. Microstructural characterizations were performed using an optical microscope (Leica DMI 5000 M), a Phenom XL table-top Scanning Electron Microscope (SEM) and a TESCAN S900 Field Emission Scanning Electron Microscope (FESEM) equipped with Secondary Electron (SE), Back Scattered Electron (BSE) detectors and an Energy Dispersive Spectroscopy (EDS). Electron backscattered diffraction (EBSD) was performed using 20 kV and 10nA with a step size of 0.2 µm. Transmission electron microscopy (TEM) characterizations were done on a very thin lamella of thickness close to 100 nm prepared from bulk samples using a Helios NanoLab 460 F1 dual beam FIB-SEM equipped with Ga + ion source. A step-by-step procedure for preparing a damage-free lamella can be found elsewhere [22]. High-angle annular dark-field (HAADF) STEM imaging and corresponding EDX chemical mappings were performed using an FEI Titan G2 80–200 ChemiSTEM microscope at 200 kV, equipped with a high-brightness field-emission gun, a probe Cs corrector, and a super-X energy-dispersive X-ray spectroscopy system [23]. The convergence semi-angle for STEM imaging was ~ 25 mrad, and the collection semi-angle was 80–200 mrad. The microstructural

**Table 1**  
Nominal composition of the A20X powder.

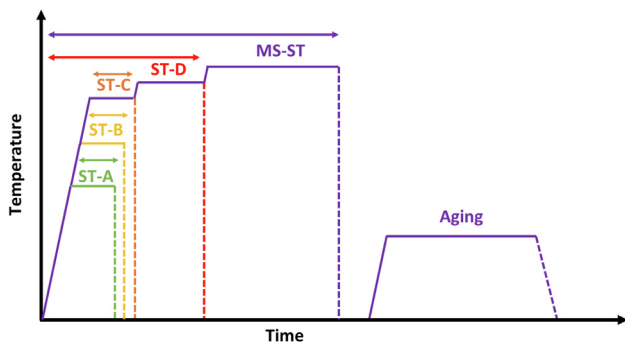
|       | Cu        | Mg        | Si    | Fe     | Ag        | B         | Ti        | Al      |
|-------|-----------|-----------|-------|--------|-----------|-----------|-----------|---------|
| wt. % | 4.20–5.00 | 0.20–0.33 | < 0.1 | < 0.08 | 0.60–0.90 | 1.25–1.55 | 3.00–3.85 | Balance |

**Table 2**  
LPBF optimized process parameters for A20X alloy.

| Parameter                 | Value                  |
|---------------------------|------------------------|
| Laser power               | 195 W                  |
| Scan speed                | 800 mm/s               |
| Hatching distance         | 0.13 mm                |
| Layer thickness           | 30 $\mu$ m             |
| Volumetric energy density | 62.5 J/mm <sup>3</sup> |

**Table 3**  
Various post-processing heat treatments performed on A20X.

| Heat treatment | Temperature and Time                              |
|----------------|---------------------------------------------------|
| ST-A           | 300 °C for 4 h                                    |
| ST-B           | 400 °C for 4 h                                    |
| ST-C           | 495 °C for 4 h                                    |
| ST-D           | 495 °C for 4 h + 505 °C for 6 h                   |
| MS-ST          | 495 °C for 4 h + 505 °C for 6 h + 530 °C for 12 h |
| SS-ST          | 530 °C for 1 h                                    |
| Aging          | 190 °C for 2, 4, 6, 8 and 10 h                    |



**Fig. 1.** A schematic representation of various heat treatments performed on the as-built A20X alloy.

features observed by various characterization techniques were quantified using an ImageJ 1.53 k software.

Phase analysis was carried out by X-ray Diffraction (XRD) in a Panalytical X'Pert PRO PW 3040/60 X-ray diffractometer using a monochromatic Cu-K $\alpha$  radiation of 1.54 Å operated at 40 kV and 40 mA with a step size of 0.013° and 25 s. per step. Individual peak profiles were obtained with a step size of 0.007° and 30 s per step. Phase quantification was performed by Rietveld refinement method using Panalytical X'Pert Highscore Plus software. The refined parameters were phase scale factor, specimen displacement, unit cell, peak broadening factors (U, V and W), peak shape factor and preferred orientation coefficient. Instrumental broadening was obtained from the spectrum of LaB<sub>6</sub> standard sample. Dislocation density was calculated by XRD line profile analysis using the Williamson-Hall (W-H) plot [24,25].

Differential scanning calorimetry (DSC) curves were obtained using NETZSCH DSC 214 Polyma. The DSC samples were weighed around 15 mg and were heated from 27 °C to 550 °C at 10 K/min in N<sub>2</sub> atmosphere. Micro-hardness values were obtained with a micro-Vickers Leica VMHT indenter using 100 g load and 15 s of dwell time (10 measurements/sample). Tensile tests were per-

formed by Zwick-Roell ProLine Z0505 with  $8 \times 10^{-3} \text{ s}^{-1}$  as strain rate, on samples of ASTM E8 standard dimension.

### 3. Results and discussion

#### 3.1. A20X powder characterization

The A20X powder morphology, microstructure and phase analysis were characterized by SEM and XRD as shown in the Fig. 2. The SEM micrograph (in Fig. 2a) shows that the particles were mostly spherical and accompanied by fine satellites. The PSD shows a unimodal distribution with D<sub>10</sub>, D<sub>50</sub> and D<sub>90</sub> as 21  $\mu$ m, 30.6  $\mu$ m and 49.8  $\mu$ m, respectively (see Fig. 2b). The microstructure of the particles from cross-sectional view revealed a cellular structure, where the cell boundaries are enriched with eutectic  $\theta$ -Al<sub>2</sub>Cu whereas the cells are composed of  $\alpha$ -Al matrix, which can be clearly seen in Fig. 2c. The  $\theta$ -Al<sub>2</sub>Cu as cell boundaries was formed as a result of an eutectic reaction ( $\text{Al}_{(\text{liq})} \rightarrow \alpha\text{-Al}_{(\text{s})} + \theta\text{-Al}_2\text{Cu}_{(\text{s})}$ ) [26,27]. The micro-TiB<sub>2</sub> particles were also observed to be present at the cell boundaries, within the cell as well as at the cell triple junctions. These phases were also detected by XRD as shown in Fig. 2d.

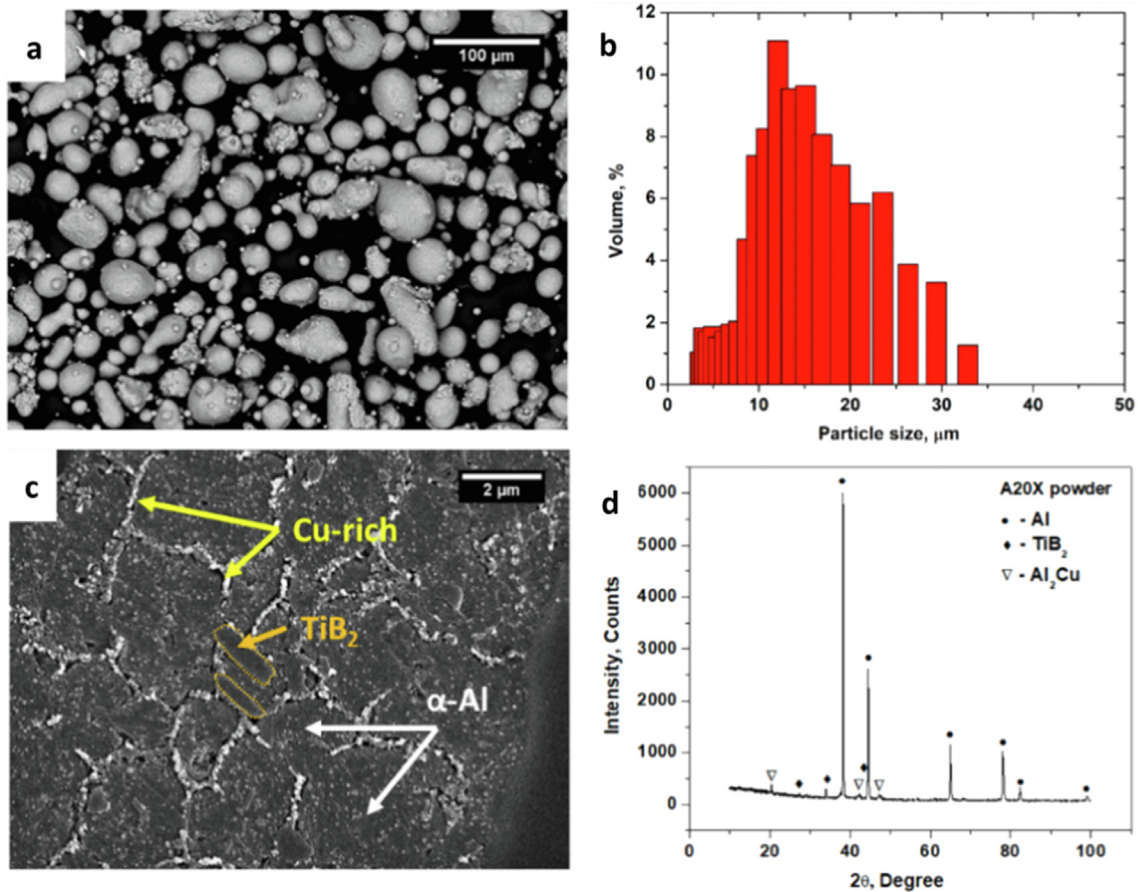
#### 3.2. Microstructural evolution during T7 heat treatment

The micro-hardness values of as-built (AB) and heat-treated samples are shown in a histogram in Fig. 3. The hardness values were reduced from AB ( $\approx 113$  HV) to MS-ST ( $\approx 103$  HV) state. During aging at 190 °C, the micro-hardness values exhibited a typical gaussian curve, where the peak hardness (about 152 HV) was achieved after 6 h, designated as peak aged (PA) state. Continue aging to 10 h reduced the hardness value to 124 HV, designated as over aged (OA) state. To have a better understanding of the salient microstructural features responsible for the peak hardness, the microstructural investigation was carried out using SEM (Fig. 4) and EBSD (Fig. 5) for all 4 different stages: AB, MS-ST, PA and OA.

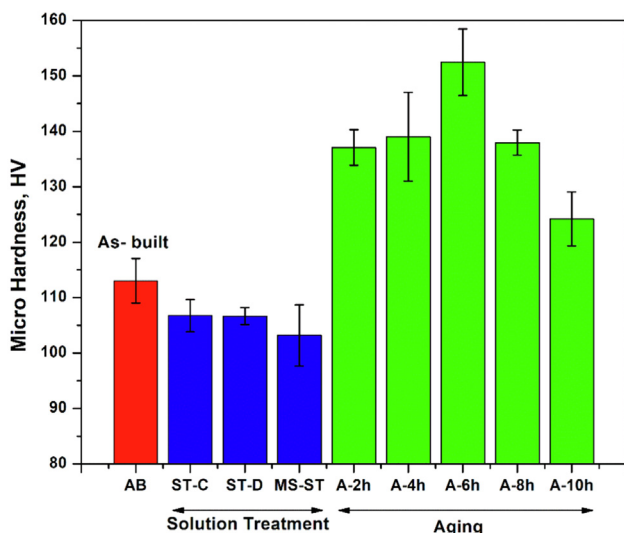
The AB microstructure showed a typical cellular structures of additively manufactured A20X alloy as reported elsewhere [20,27,28]. The cross-sectional SEM-BSE micrograph of the AB sample revealed a well-defined cellular microstructure having cell boundaries enriched with  $\theta$ -Al<sub>2</sub>Cu and  $\alpha$ -Al cells along with elongated micro-TiB<sub>2</sub> particles as shown in Fig. 4a. The AB microstructure was found to possess similar microstructural features to the as-atomized powder (Fig. 2c). However, unlike in the A20X powder, where  $\theta$ -Al<sub>2</sub>Cu formation happens solely by eutectic solidification, in the AB samples, the  $\theta$ -Al<sub>2</sub>Cu phase could be formed in two different ways, majorly as a eutectic product during solidification (cell boundary) and partially as a precipitate by a solid-state process. The precipitation could occur due to multiple thermal cycles experienced by previously solidified layers which are supersaturated with Cu due to the rapid solidification [29].

After MS-ST, the eutectic network of  $\theta$ -Al<sub>2</sub>Cu disappeared (Fig. 4b). The micro-TiB<sub>2</sub> particles were observed majorly at the grain boundary triple junctions or along the grain boundaries. There are also few  $\theta$ -Al<sub>2</sub>Cu precipitates spotted at the micro-TiB<sub>2</sub> particles, see supplementary Fig. S1b. Upon aging, the solutionized Cu precipitated out of the matrix as  $\theta$ -Al<sub>2</sub>Cu phase near micro-TiB<sub>2</sub> particles and at the grain boundaries, Fig. 4c and Fig. S1c. The precipitation of  $\theta$ -Al<sub>2</sub>Cu at the micro-TiB<sub>2</sub> particles can be attributed to





**Fig. 2.** A20X powder characterization showing (a) powder morphology, (b) volume particles size distribution, (c) particle cross-sectional microstructure and (d) XRD phase analysis.



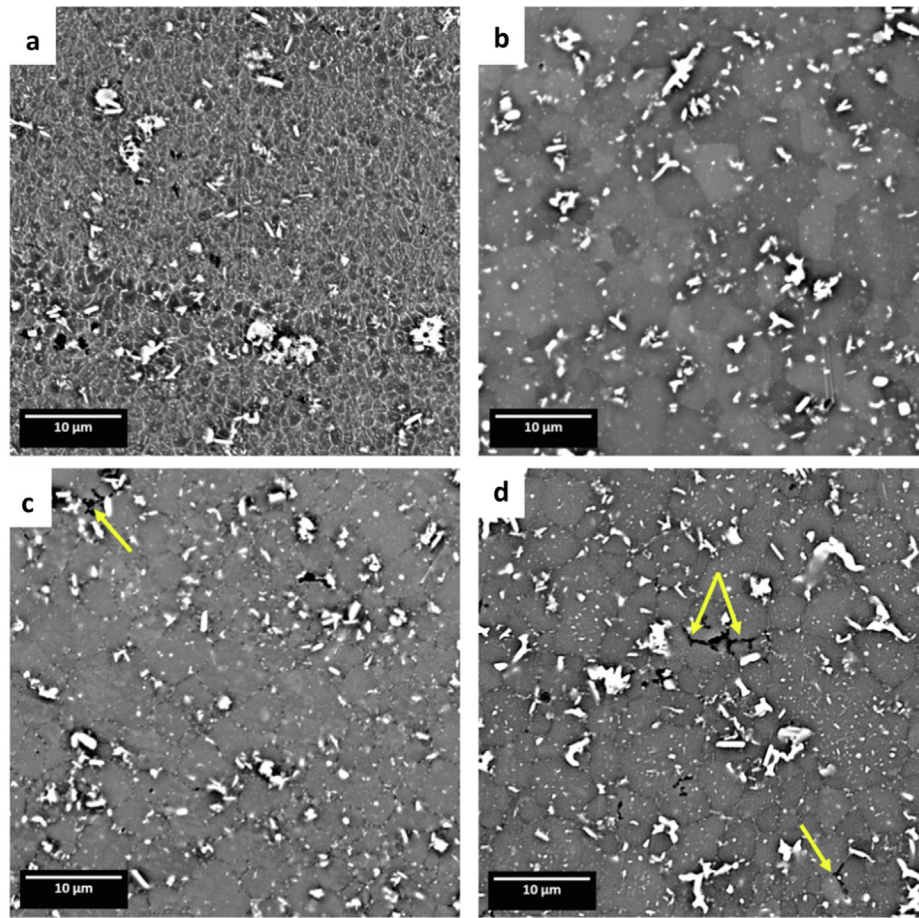
**Fig. 3.** Evolution of micro-hardness of A20X during MS-ST and subsequent aging.

the differences in the thermal expansion coefficient ( $\alpha$ ) of Al ( $26.5 \times 10^{-6} \text{ K}^{-1}$ ) and  $\text{TiB}_2$  ( $7.7 \times 10^{-6} \text{ K}^{-1}$  along c-axis and  $4.1 \times 10^{-6} \text{ K}^{-1}$  along a-axis). Because of this difference, upon quenching, dislocations are trapped at the vicinity of  $\text{TiB}_2$ -Al interface [30–32]. During aging, Cu atoms diffuse through these high dislocation density channels by pipe diffusion mechanism and accelerate the nucleation and growth of  $\theta$ - $\text{Al}_2\text{Cu}$  precipitates at

the  $\text{TiB}_2$  particles [30,31,33,34]. Excessive aging i.e. OA (190 °C for 10 h) significantly enhanced the size of  $\theta$ - $\text{Al}_2\text{Cu}$  precipitates, which nearly engulfed the micro- $\text{TiB}_2$  particles, Fig. 4d and Fig. S1d. In the OA state, it was difficult to clearly distinguish between  $\theta$ - $\text{Al}_2\text{Cu}$  and micro- $\text{TiB}_2$  particles even under low brightness (Fig. S1d). Surprisingly, some grain boundary cracks were also noticed in the PA state (yellow arrow in Fig. 4c). These cracks were more frequent and severe in the OA state. No such cracks were observed in the AB state and thus their origin is likely to be related to the post-processing heat treatment.

Fig. 5 shows the EBSD scans of the 4 states. The grain size in AB state was estimated to be  $1.05 \mu\text{m}$  (Fig. 5a). Moreover, it was observed that, the cell size is equivalent to the grain size. Unindexed black interconnected patches surrounding the grains are likely to be of eutectic  $\theta$ - $\text{Al}_2\text{Cu}$  phase. After MS-ST, the grain size grew twice to that of the AB state ( $\approx 1.97 \mu\text{m}$ ) (Fig. 5b). However, it is important to note that there is no significant increase in the grain size after such a long solution treatment (nearly 24 h). This limited grain growth can be attributed to the Zener pinning effect by the micro- $\text{TiB}_2$  particles at the matrix grain boundaries [35]. The average grain size in the PA ( $2.16 \mu\text{m}$ ) and OA ( $1.93 \mu\text{m}$ ) states are almost similar to that of MS-ST state (Fig. 5c and Fig. 5d).

The phase analysis was carried out by XRD (Fig. 6). The XRD diffractogram of AB, MS-ST, PA and OA revealed primarily-three phases: Al,  $\text{TiB}_2$  and  $\theta$ - $\text{Al}_2\text{Cu}$ , see Fig. 6a. The individual peak profiles of  $\theta$ - $\text{Al}_2\text{Cu}$  (110) and Al (111) phases are shown in Fig. 6b and Fig. 6c, respectively. The increase in the intensity of  $\theta$ - $\text{Al}_2\text{Cu}$  peak during aging is depicting  $\theta$ - $\text{Al}_2\text{Cu}$  precipitation and coarsening. A quantitative estimation of the phase evolution from the AB



**Fig. 4.** SEM-BSE micrographs of A20X samples in the (a) AB, (b) MS-ST, (c) PA and (d) OA states (Yellow arrows depict grain boundary cracks).

to the OA state was performed by Rietveld method and is presented in Fig. 6d. The  $\theta$ -Al<sub>2</sub>Cu phase in the AB state was quantified to be 4.23 wt%. The phase quantification in the MS-ST state was not carried out due to its very low peak-to-background ratio of the  $\theta$ -Al<sub>2</sub>Cu phase, it would avoid the error in quantification during peak fitting. The  $\theta$ -Al<sub>2</sub>Cu phase content in the PA state increased to 1.95 wt% which then significantly increased to 4.20 wt% in the OA state, this is in line with the observation made from SEM micrographs (Fig. 4). The phase quantification of TiB<sub>2</sub> was around 5.5 wt% in all the states.

The individual peak profiles of Al (111) showed a shift towards higher  $2\theta$  values after MS-ST with respect to its reference position (dotted line). This peak shift corresponds to a decrease in its lattice parameters as a result of solid solution by Cu [36]. Interestingly, peak broadening of the Al phase during aging was observed, see Fig. 6c. For instance, the average full-width at half maximum (FWHM) of Al (111) peak in AB, MS-ST, PA and OA are 0.106°, 0.088°, 0.096° and 0.094° ( $2\theta$ ), respectively. The dislocation density in Al matrix associated with the peak broadening was obtained by W-H plot and the values are presented in Fig. 6e. The AB state possessed a high dislocation density of  $6.13 \times 10^{14}/\text{m}^2$ , which is a typical characteristic of additively manufactured Al alloys [37]. The dislocation density decreased after MS-ST ( $2.63 \times 10^{14}/\text{m}^2$ ) but then increased again substantially in the PA ( $5.34 \times 10^{14}/\text{m}^2$ ) and OA ( $4.12 \times 10^{14}/\text{m}^2$ ) states.

In order to understand the origin of such high dislocation densities during aging, DSC profiles were investigated and the sequence of precipitation and phase evolution were analysed (Fig. 7). Majorly 3 peaks were detected in AB and MS-ST specimen

as shown in Fig. 7a. The first peak was an endothermic peak which belongs to the dissolution of GP zones or Mg-Ag co-clustering [38]. The second peak was a large and broad exothermic peak, corresponding to the precipitation and the third peak was also a broad endothermic peak, depicting dissolution of all the precipitates [39].

By comparing the DSC curves, it indicates that the GP zone dissolution in the MS-ST occurred at much lower temperature and possessed larger area under the peak with respect to the AB state. This has been attributed to the dissolution of a larger volume fraction of smaller GP zone in MS-ST state with respect to the AB state [40]. The MS-ST sample displayed the largest area of the exothermic peak i.e. highest amount of precipitation. However, the broadness of the peak indicates that the process of precipitation was not a single event but occurred over a span of temperatures. On deconvolution, 3 sub-peaks were identified as shown in Fig. 7b. Their peak temperatures were about 230 °C, 260 °C and 310 °C which correspond to the precipitation of  $\Omega$  &  $\theta''$ ,  $\theta'$  and  $\theta$  phase, respectively [38,39,41].

Although, DSC hints the formation of multiple precipitates, the resolution of SEM was not sufficient to resolve these precipitates. Therefore, high resolution STEM was used for this. The STEM-EDS mapping (Fig. 8) of the PA state showed fine and discontinuous grain boundary precipitates (Fig. 4). These precipitates were enriched with Cu and accompanied by nano-TiB<sub>2</sub> particles. It was suspected that these precipitates are likely to be of an equilibrium  $\theta$ -Al<sub>2</sub>Cu and might have nucleated due to the difference in CTE of Al and TiB<sub>2</sub> as discussed previously. Additionally, Mg segregation at grain boundaries was also observed. Within the grains, a large number of needle-shaped precipitates, which are of nanosized



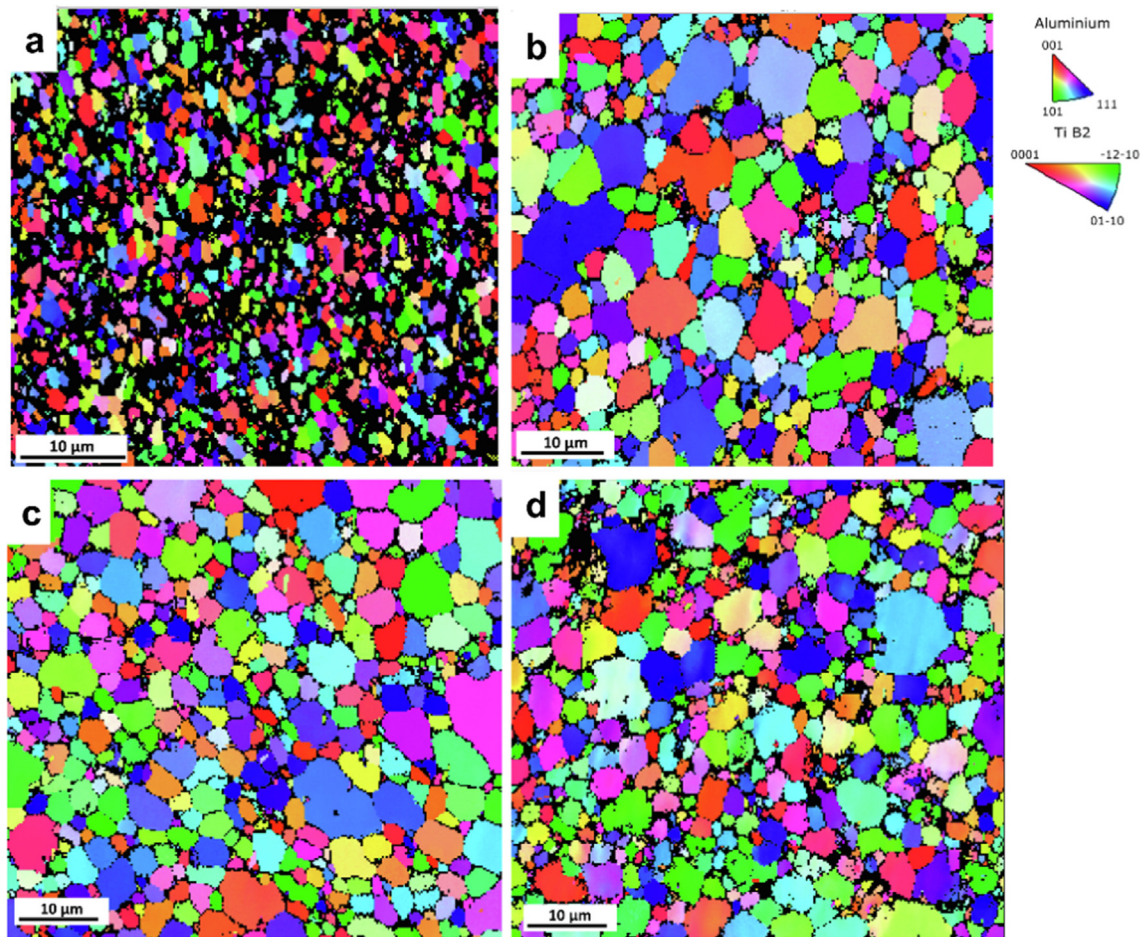


Fig. 5. EBSD micrographs showing evolution of grain structure in various conditions (a) AB, (b) MS-ST (c) PA and (d) OA.

(about 52 nm in length) (Fig. S2a) and enriched with Cu but also partly enriched with Ag and Mg were observed (Fig. 8). These needle-shaped precipitates seem to hold an orientation relationship with the Al-matrix. Three different orientations within a single grain could be detected. According to the earlier researchers, these precipitates are most likely to be of coherent  $\Omega$ -Al<sub>2</sub>Cu and semi-coherent  $\theta'$ -Al<sub>2</sub>Cu [20,21,42].

The  $\Omega$  and  $\theta'$  precipitates prefer to nucleate at the Al {111} and {001} planes, respectively. These phases possess the same chemical composition of an equilibrium  $\theta$ -Al<sub>2</sub>Cu phase [42]. Both  $\Omega$  and  $\theta'$  phase can be generated from a supersaturated solid-solution (SSSS) by two independent precipitation sequences [43]:

SSSS  $\rightarrow$  G.P. Zone (Cu)  $\rightarrow \theta'' \rightarrow \theta'$  on {001} $_{\alpha}$   $\rightarrow \theta$ .

SSSS  $\rightarrow$  G.P. Zone (Mg and Ag)  $\rightarrow \Omega$  on {111} $_{\alpha}$   $\rightarrow \theta$ .

However, precipitation of  $\Omega$  phase is more favourable than  $\theta'$  in Al-Cu-Ag-Mg alloys because of Mg and Ag co-clustering, which reduces the interfacial energy for the precipitating nuclei at coherent {111} $_{\alpha}$  habit planes and moreover facilitates the nucleation of the  $\Omega$  which are coherent with Al matrix [14,44,45]. The  $\Omega$  and  $\theta'$  phases are considered as the predominant hardening phases however,  $\Omega$ -plates offers much higher strength when exposed to higher temperatures or for longer aging times due to higher coarsening resistance than  $\theta'$  [42,45–48]. Thus, few large matrix precipitates (of size  $\approx$  420 nm) which were observed in the OA state might be of  $\theta'$ , (see the inset Fig. S2b). Nonetheless, there is a large lattice misfit associated with these coherent- $\Omega$  and semi-coherent- $\theta'$  precipitates with the Al matrix and this gives rise to micro-strains upon precipitation [49–52]. Thus, a high density of  $\Omega$  and  $\theta'$  precipitates results in large micro-strains which are directly linked to the

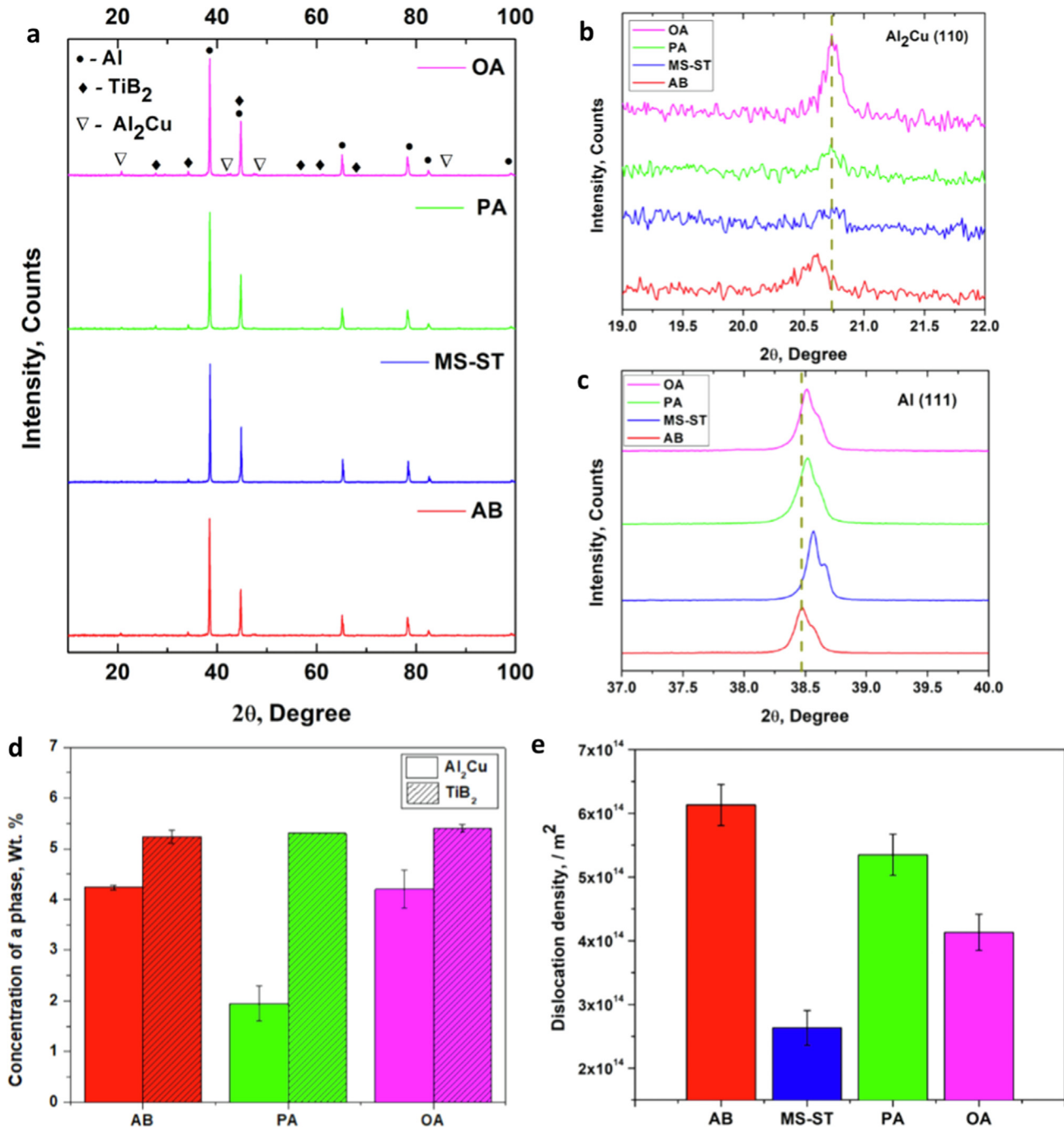
presence of high dislocation densities in the PA and OA state. Hence, these precipitates are responsible for peak broadening which was evident in the XRD diffractograms as shown in Fig. 6c and Fig. 6e.

From the present microstructural evolution analysis, the distinctive features which are responsible for the peak hardness after 6 h of aging are presence of dense coherent- $\Omega$  and semi-coherent- $\theta'$  nano-precipitates within the matrix, optimum precipitation  $\theta$ -Al<sub>2</sub>Cu (OA state significantly coarsened  $\theta$ -Al<sub>2</sub>Cu) and less frequent grain boundary cracks with respect to the OA state.

### 3.3. Grain boundary cracking during aging

During aging, the microstructural characterization revealed grain boundary cracking in both PA and OA states. Grain boundary cracking can be related to the synergetic effect of (i) development of micro-strains (dislocations) (Fig. 6e) during aging due to the presence of coherent ( $\Omega$ -Al<sub>2</sub>Cu) and semi-coherent ( $\theta'$ -Al<sub>2</sub>Cu) nano-precipitates and (ii) simultaneous increase in the width of precipitate free zones (PFZs) at the vicinity of grain boundaries and micro-TiB<sub>2</sub> particles (Fig. 9).

The process of precipitation and growth during aging results in the creation of PFZs in the vicinity of specific microstructural features. From the PA to the OA state, the PFZs width near grain boundaries increased from 30 nm to 85 nm (Fig. 9b and Fig. 9d) whereas PFZs width near micro-TiB<sub>2</sub> particles increased from 78 nm to 153 nm (Fig. 9a and Fig. 9c). It should be noted that in both PA and OA states, the PFZs width near micro-TiB<sub>2</sub> particles was much larger than near grain boundaries. Widening of PFZs



**Fig. 6.** (a) Phase identification by XRD measurements of AB, MS-ST, PA and OA state; individual peaks profiles of (b)  $\text{Al}_2\text{Cu}$  (110) and (c) Al (111); (d) quantification of  $\theta\text{-Al}_2\text{Cu}$  and  $\text{TiB}_2$  phases calculated by Rietveld refinement and (e) an estimated dislocation density by W-H plot. (Reference code for the phases, goodness of fit and weighted R profile values of the refinement are provided in the supplementary file, **Table S1**).

has a direct correlation with the growth of precipitates at these locations. The grain boundary precipitates present at the nano- $\text{TiB}_2$  particles did not show any considerable increase in their size, whereas  $\theta\text{-Al}_2\text{Cu}$  precipitates near micro- $\text{TiB}_2$  particles grew immensely from the PA to OA state and thus consumed higher amounts of solute atoms within its vicinity leading to larger PFZs width.

The precipitation of coherent- $\Omega$  and semi-coherent- $\theta'$  during aging could generate micro-strains, resulting hardening of the grain interiors. The development of micro-strains within the matrix was evident by the presence of high dislocation densities. This results in the localized deformation near the grain boundaries and might induce grain boundary cracking [53], popularly known

as strain-age cracking and is generally observed during the post weld heat treatment of superalloys [53–55]. Moreover, this effect is quite severe if there exists PFZs near grain boundaries and at triple junctions [53]. The localized deformation at the PFZs can cause the nucleation of cracks which can easily propagate along the PFZs leading to grain boundary cracking [56,57].

### 3.4. Microstructural evolution during multi-step solution treatment (MS-ST)

Upon aging, the material experiences cracks formation, which has been correlated with the increase in PFZs width due to significant growth of  $\theta\text{-Al}_2\text{Cu}$  at micro- $\text{TiB}_2$  particles. In order to under-



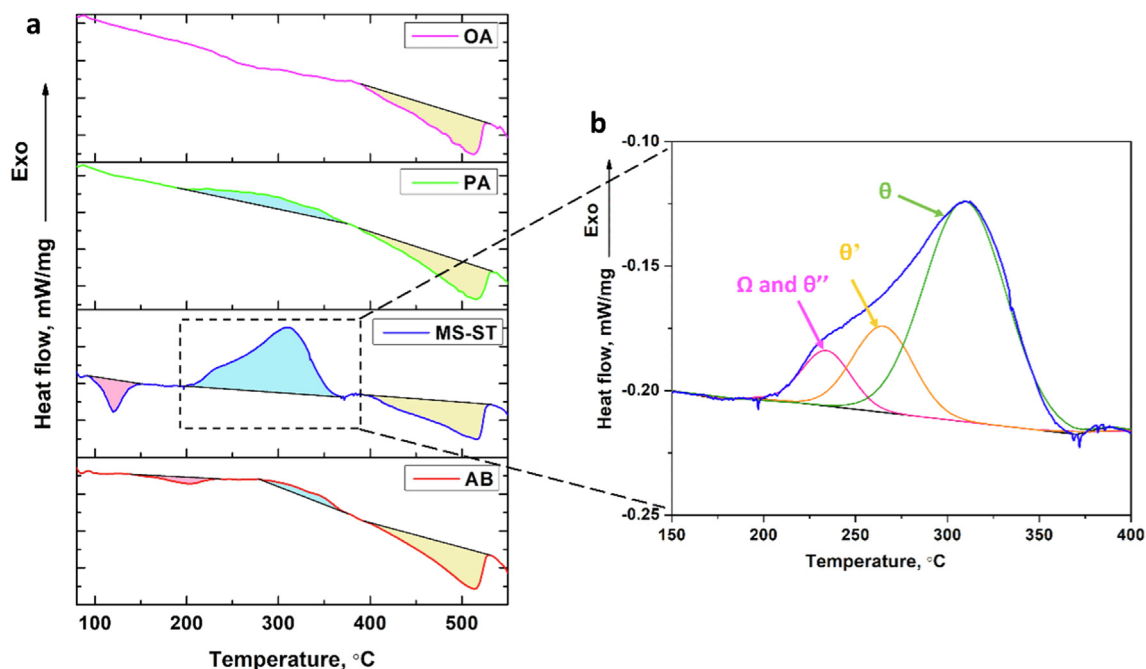


Fig. 7. (a) DSC curves of AB, MS-ST, PA and OA states and (b) deconvolution of the broad exothermic peak of MS-ST state.

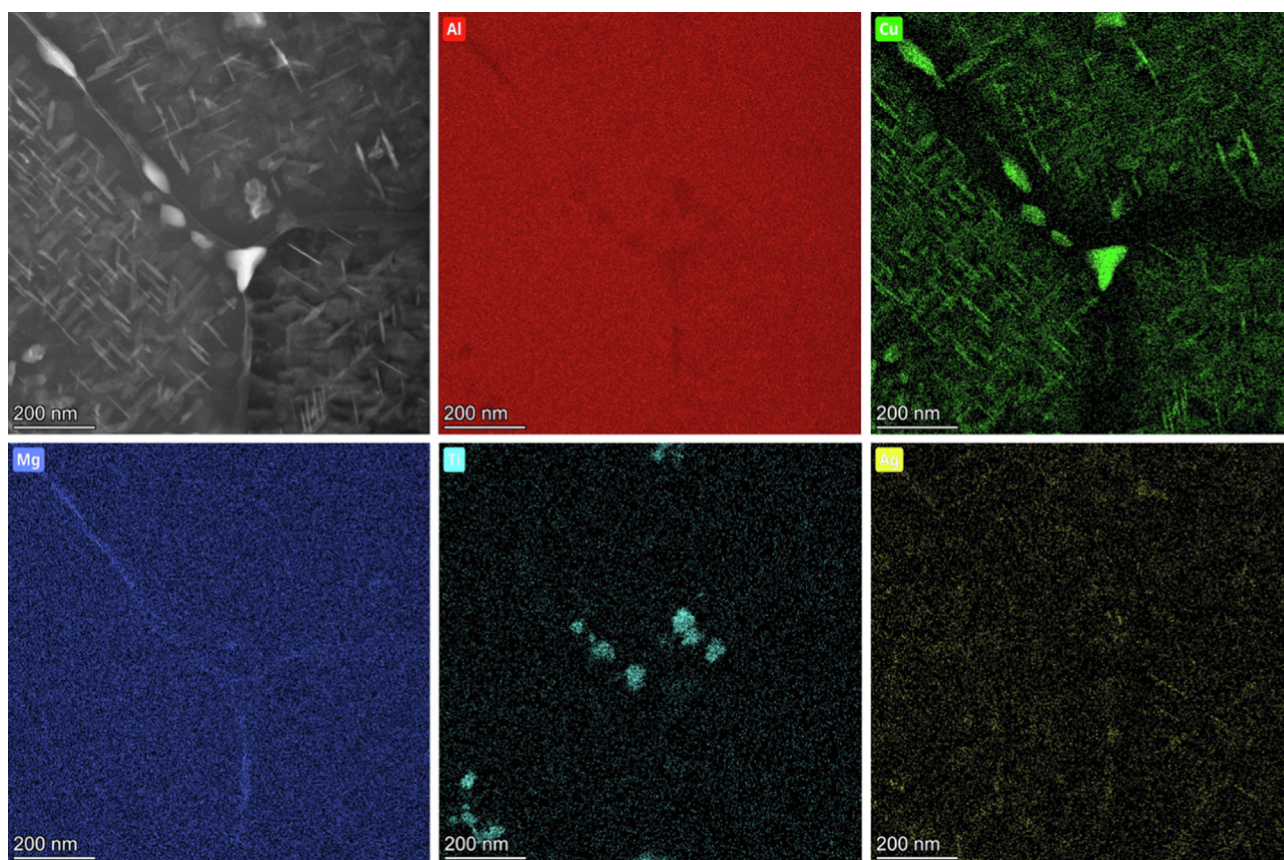
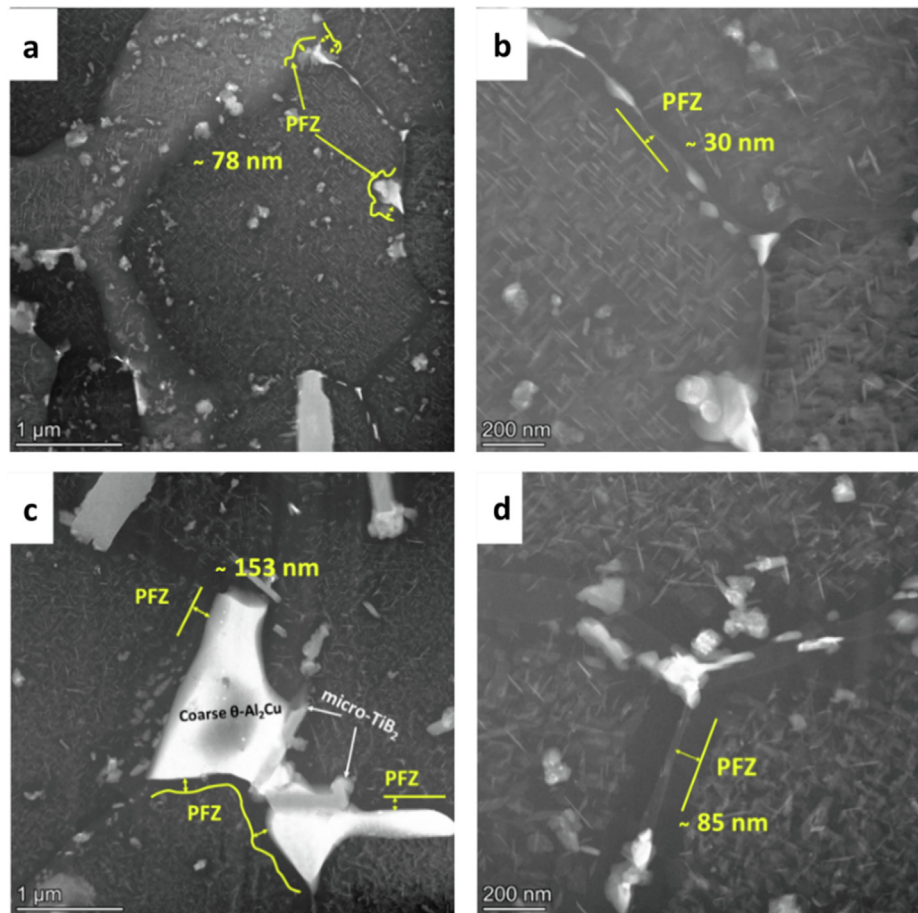


Fig. 8. High resolution HAADF-STEM micrograph and corresponding EDS mapping of a grain boundary triple junction in the PA state.

stand, if the multi-step solution treatment (MS-ST) has any role on this particular aging behaviour, evaluations of the microstructure at various solution steps were performed.

SEM micrographs were taken using backscattered electrons (BSE) mode with both high and low brightness values (Fig. 10 (a-f and a'-f')). These different image settings were used to distin-



**Fig. 9.** High resolution HAADF-STEM micrographs indicating the PFZs in PA (a-b) and OA (c-d) state near micro-TiB<sub>2</sub> particles and grain boundaries.

guish TiB<sub>2</sub> (grey) and  $\theta$ -Al<sub>2</sub>Cu (white) phases by atomic contrast. These micrographs were used to quantify the microstructural features of  $\theta$ -Al<sub>2</sub>Cu in terms of area equivalent average-radius ( $r$ ), circularity ( $C$ ), number density and inter-precipitate spacing ( $\lambda$ ) as shown in Fig. 11.

Heating the AB sample (Fig. 10a and Fig. 10a') to ST-A (300 °C for 4 h) leads to breaking up of the cellular structure as observed in Fig. 10b and Fig. 10b'). The eutectic  $\theta$ -Al<sub>2</sub>Cu network fragments into fine spherical particles ( $r = 73$  nm,  $C = 0.87$ ). The microstructure was also characterized by high number density of  $\theta$ -Al<sub>2</sub>Cu particles ( $0.21/\mu\text{m}^2$ ) and low inter-precipitate spacing ( $\lambda = 0.99$   $\mu\text{m}$ ) (Fig. 11). These fine fragmented particles could then be able to spheroidize by the Rayleigh instability process which occur by the diffusion of atomic species such as Al and Cu from their edges towards the flat surface due to capillary forces [58,59].

After the ST-B heat treatment (400 °C for 4 h), the spheroidized  $\theta$ -Al<sub>2</sub>Cu eutectic particles were happened to be of a larger size ( $r = 251$  nm) than the ST-A ones. They appeared to be migrated towards micro-TiB<sub>2</sub> particles, as seen in Fig. 10c and Fig. 10c'. The migration of  $\theta$ -Al<sub>2</sub>Cu eutectic towards TiB<sub>2</sub> particles has been attributed to the process of Ostwald ripening due to difference in the CTE of Al and TiB<sub>2</sub> and generation of localized dislocation density [60]. The Ostwald ripening was also quantitatively reflected by the significant increase in average size of  $\theta$ -Al<sub>2</sub>Cu from 73 nm to 251 nm (Fig. 11a) and a considerable decrease in the average number density from  $0.21/\mu\text{m}^2$  to  $0.09/\mu\text{m}^2$  (Fig. 11b). The migrated  $\theta$ -Al<sub>2</sub>Cu at micro-TiB<sub>2</sub> particles further coarsened during the ST-C heat treatment (495 °C for 4 h)

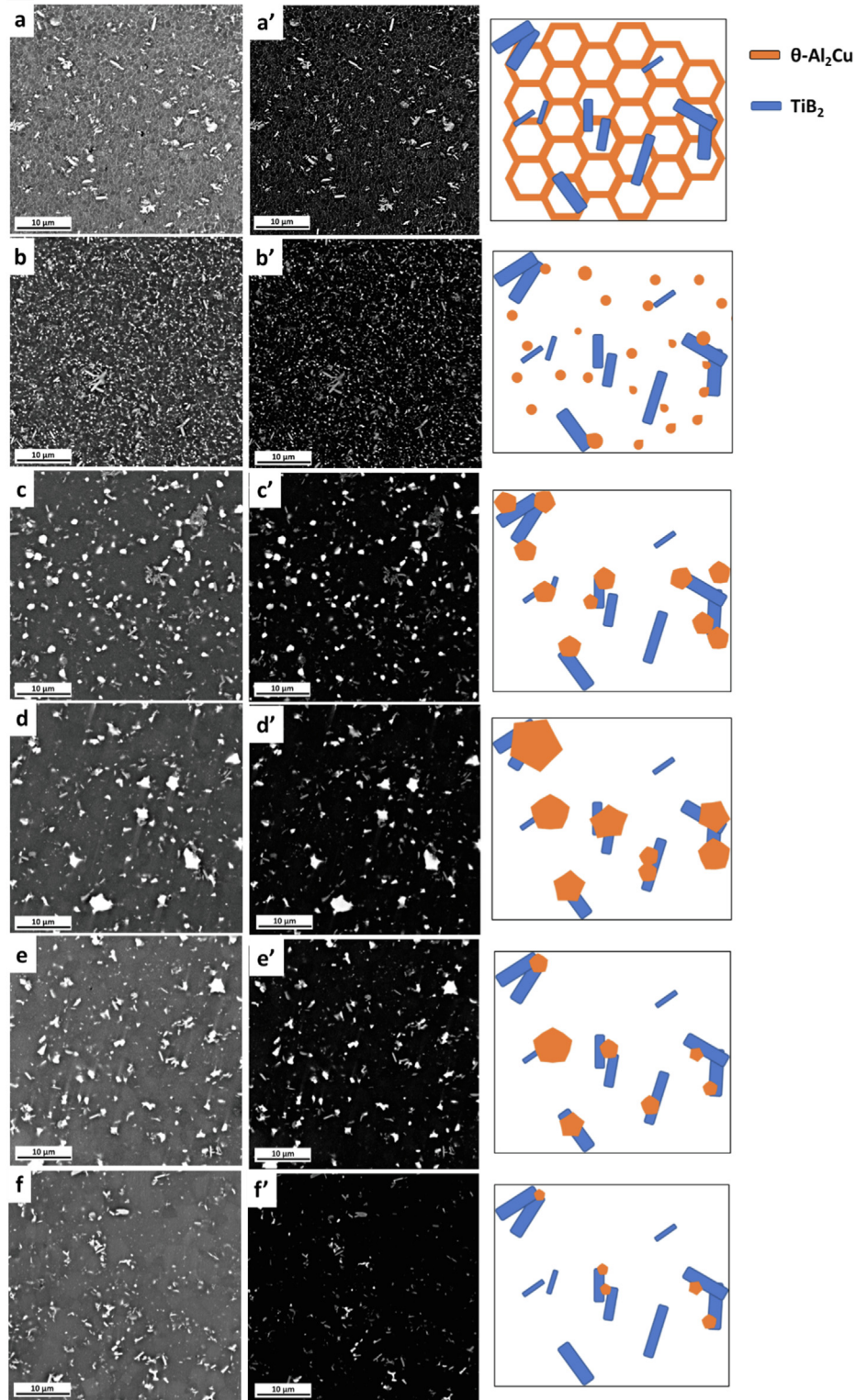
(Fig. 10d and Fig. 10d'), reaching the maximum average size of  $r = 310$  nm (Fig. 11a).

After the ST-D heat treatment (495 °C for 4 h + 505 °C for 6 h), the microstructure showed both fine and coarse  $\theta$ -Al<sub>2</sub>Cu precipitates, which demonstrated that certain  $\theta$ -Al<sub>2</sub>Cu precipitates were dissolved back into the matrix (Fig. 10e and Fig. 10e'). The temperature (505 °C) utilized in ST-D heat treatment was reported to initiate the dissolution of  $\theta$ -Al<sub>2</sub>Cu [19].

Finally, the very long MS-ST (495 °C for 4 h + 505 °C for 6 h + 530 °C for 12 h) which was specifically reported to completely dissolve the large blocky  $\theta$ -Al<sub>2</sub>Cu precipitates in castings of Al-Cu based alloys [61,62], could not be able to completely dissolve the  $\theta$ -Al<sub>2</sub>Cu precipitates in the present case, especially at micro-TiB<sub>2</sub> particles, see Fig. 10f and Fig. 10f'. Overall, upon the MS-ST, the average number density of  $\theta$ -Al<sub>2</sub>Cu significantly reduced and the inter-precipitates spacing increased and reached a value ( $\lambda = 2.5$   $\mu\text{m}$ ) almost equivalent to the average grain size of MS-ST.

The incomplete dissolution of  $\theta$ -Al<sub>2</sub>Cu phase even after long MS-ST could be attributed to the shielding effect of micro-TiB<sub>2</sub> particles which reduce the contact area between  $\theta$ -Al<sub>2</sub>Cu and Al-matrix and greatly limit the dissolution rate [63]. A similar phenomenon was described by Asghar et al. in AlSi12Ni alloy, where 3D interconnectivity of rigid Ni-aluminides with eutectic Si retarded the spheroidization of Si [64]. Recently, Zamani et al. [61], demonstrated that interconnectivity of the TiB<sub>2</sub> and Al<sub>2</sub>Cu retards the diffusivity of Cu atoms along grain boundaries and at Al<sub>2</sub>Cu/ $\alpha$ -Al interface. This phenomenon was modelled by reducing the effective diffusion coefficient of Cu in Al to 2–2.5 times than its typical value.





**Fig. 10.** SEM-BSE micrographs under high (a-f) and low (a'-f') brightness showing microstructural evolution at various interrupted steps during solution treatment. (a and a') AB, (b and b') ST-A, (c and c') ST-B, (d and d') ST-C, (e and e') ST-D and (f and f') MS-ST. ( $\text{TiB}_2$  particles as grey and  $\theta\text{-Al}_2\text{Cu}$  as white). A schematic view of the corresponding microstructure (on the right-hand side) is shown for each interrupted step.

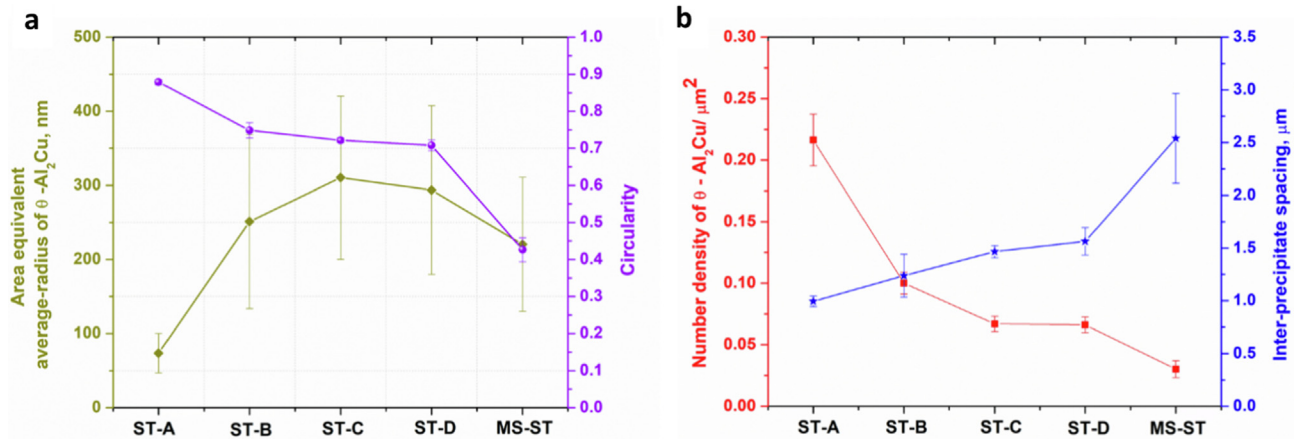
### 3.5. A comparison of multi-step (MS-ST) and single-step (SS-ST) solution treatments

As previously described, since the MS-ST microstructure serves as an initial condition for aging, the presence of undissolved precipitates at micro- $\text{TiB}_2$  particles could induce certain detrimental

effects during the subsequent aging behaviour. To distinguish the effect of a MS-ST on aging behaviour, single-step solution treatment (SS-ST) were performed.

A SS-ST was considered for two key reasons: (i) The presence of low-melting phases such as  $\text{S-Al}_2\text{CuMg}$ ,  $\text{Q-Al}_7\text{Cu}_3\text{Mg}_6$  in the AB state were not observed by any characterization technique. The





**Fig. 11.** Evolution of various features of  $\theta$ -Al<sub>2</sub>Cu precipitate as a function of interrupted solution steps showing (a) area equivalent average-radius and circularity and (b) number density and inter-precipitate spacing.

presence of such phases in the castings are the reason for performing long MS-ST to avoid incipient melting. However, in the LPBF processes, high cooling rates ( $\sim 10^4$ – $10^7$  K/s) could suppress the formation of these phases [19]. This allows applying only a single-step i.e. solutioning at 530 °C to dissolve  $\theta$ -Al<sub>2</sub>Cu eutectic, see Fig. 7a. (ii) It is quite evident now that, MS-ST allows migration and coarsening of  $\theta$ -Al<sub>2</sub>Cu near micro-TiB<sub>2</sub> particles which then become difficult to get completely dissolved even after holding at 530 °C for 12 h, see Fig. 10. Thus, a SS-ST could be able to avoid migration and subsequently coarsening of  $\theta$ -Al<sub>2</sub>Cu precipitates near micro-TiB<sub>2</sub> particles.

Solutioning time was optimized by comparing three-time intervals at 530 °C: 1 h, 6 h and 12 h. All the three conditions could be able to dissolve  $\theta$ -Al<sub>2</sub>Cu effectively, see supplementary Fig S3. However, solution treatment performed for 1 h was considered as it provides higher hardness.

After SS-ST (530 °C for 1 h), the sample was aged at 190 °C for 6 h and compared with MS-ST aged for 6 h i.e. PA state. These samples were named as SS-ST-A-6 h and PA (MS-ST-A-6 h) respectively. The comparative microstructural features were presented in terms of  $\theta$ -Al<sub>2</sub>Cu average size, area fraction and SEM-BSE micrographs (Fig. 12). A summary of these microstructural features is provided in Table 4. SS-ST-A-6 h possessed considerably lower average size and area fraction of  $\theta$ -Al<sub>2</sub>Cu precipitates than PA (MS-ST-A-6 h) as shown in Fig. 12a. In addition, a negligible fraction of  $\theta$ -Al<sub>2</sub>Cu precipitates were located near micro-TiB<sub>2</sub> particles, see Fig. 12b as compared with PA (MS-ST-A-6 h) (Fig. 12c). This indicates that most of the eutectic  $\theta$ -Al<sub>2</sub>Cu could be able to dissolve back into the Al-matrix in a SS-ST. Furthermore, the SS-ST avoids formation of blocky Al<sub>2</sub>Cu phase near micro-TiB<sub>2</sub> particles unlike the case of the MS-ST (Fig. 10). The same was also confirmed from the phase quantification by Rietveld refinement, as shown in Table 4 and Table S1. SS-ST-A-6 h has around 0.9 wt% of Al<sub>2</sub>Cu as compared to 1.9 wt% in PA (MS-ST-A-6 h).

Further characterization by HAADF-STEM of the SS-ST-A-6 h sample displayed grain boundary precipitates and grain interior needle-shaped precipitates, see Fig. 12d and Fig. 12e. These microstructural features are very similar to the PA (MS-ST-A-6 h) state, where the grain boundary precipitates are of equilibrium  $\theta$ -Al<sub>2</sub>Cu and the needle-shaped precipitates are of  $\Omega$  and  $\theta'$  (Fig. 8). The estimated dislocation density due to these matrix precipitates was comparable to the PA (MS-ST-A-6 h) state  $\approx 4.67 \times 10^{14}/\text{m}^2$  and the average length of the precipitates was 58 nm, although the distribution of precipitate length was bimodal, indicating the presence of some fine precipitates in SS-ST-A-6 h state (Fig. S2c). However, the PFZs width near micro-TiB<sub>2</sub> particles ( $\approx 32$  nm)

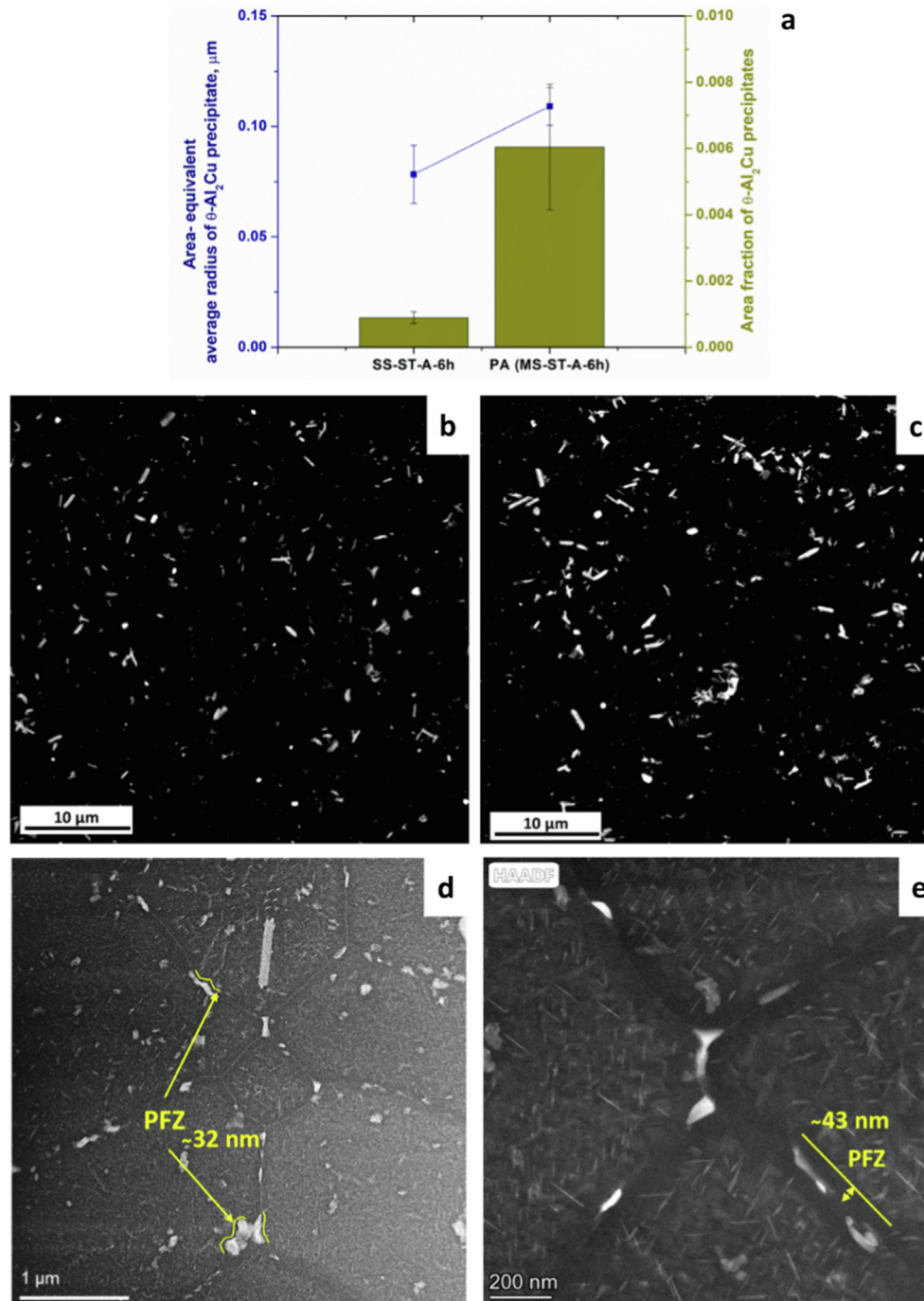
and near grain boundaries ( $\approx 43$  nm) are opposite to that found in PA (MS-ST-A-6 h) state. The PFZs width was much higher near micro-TiB<sub>2</sub> than near grain boundaries in PA (MS-ST-A-6 h), see Fig. 8. This reduced PFZs width near micro-TiB<sub>2</sub> particles has been attributed to more complete dissolution of  $\theta$ -Al<sub>2</sub>Cu in single step solution treatment. As previously described the coarsening of  $\theta$ -Al<sub>2</sub>Cu near micro-TiB<sub>2</sub> particles in PA (MS-ST-A-6 h) can be associated with the presence of pre-existing  $\theta$ -Al<sub>2</sub>Cu precipitates near micro-TiB<sub>2</sub> particles. It is in fact well accepted that the presence of pre-existing precipitates accelerate the precipitation kinetics and showed a pronounced effect on the aging behaviour [65–67]. The  $\theta$ -Al<sub>2</sub>Cu coarsened preferentially at its pre-existing precipitates during secondary aging rather than nucleating as a new precipitate [66]. Rapid coarsening is detrimental to the materials mechanical as well as corrosion properties as it increases the PFZs width [67]. The width of PFZs plays a major role in grain boundary cracking in A20X alloy as a function of aging. Fig. 13 shows multiple grain boundary cracks in PA (MS-ST-A-6 h) whereas no such cracks were observed in SS-ST-A-6 h. Thus, the presence of pre-existing  $\theta$ -Al<sub>2</sub>Cu precipitates (Incomplete dissolution) after solution treatment alters the coarsening kinetics during aging and is undesirable in order to avoid early formation of PFZs.

### 3.6. Tensile behaviour

The mechanical behaviour of A20X alloy was tested for AB and PA (MS-ST-A-6 h) states. The results are shown as a tensile stress-strain curves in Fig. 14. Curve characteristics such as yield strength (YS), ultimate tensile strength (UTS) and total elongation to fracture (e %) are reported in Table 4 along with their microstructural features. The tensile behaviour of SS-ST-A-6 h state have also been shown for the comparison.

The AB sample possessed considerable strength (YS  $\approx 304$  MPa and UTS  $\approx 378$  MPa) and good ductility (e  $\approx 12.5$  %) with respect to an Al-metal matrix composites [68]. The tensile behaviour showed a typical stress-strain curve of an additive manufactured A20X alloy, displaying yield drop phenomena followed by inhomogeneous deformation (Lüders band formation and its spread) and strain hardening with serrated flow behaviour (Portevin-Le Chatelier -PLC effect) [27,28]. The PLC effect is due to repeated locking and unlocking of dislocations by solute atoms such as Cu and Mg. The high percentage of elongation to fracture in AB samples has been attributed to the strong bonding of micro- and nano-TiB<sub>2</sub> particles with the  $\alpha$ -Al matrix [69].

The tensile curve of PA (MS-ST-A-6 h) samples showed a considerable increase in the strength but a reduced ductility with



**Fig. 12.** (a) Area-equivalent average radius and area fraction of  $\theta$ -Al<sub>2</sub>Cu precipitates in SS-ST-A-6 h and PA (MS-ST-A-6 h); and their corresponding SEM-BSE micrographs under low brightness (b) SS-ST-A-6 h and (c) PA (MS-ST-A-6 h) ( $\text{TiB}_2$  particles as grey and  $\theta$ -Al<sub>2</sub>Cu as white). HAADF-STEM micrographs of SS-ST-A-6 h indicating PFZs, (d) near micro-TiB<sub>2</sub> particles and (e) near grain boundaries.

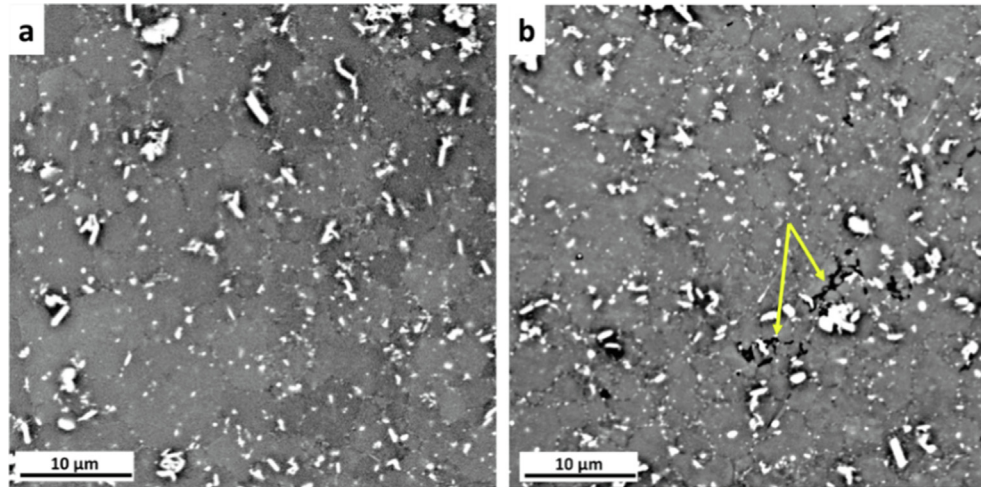
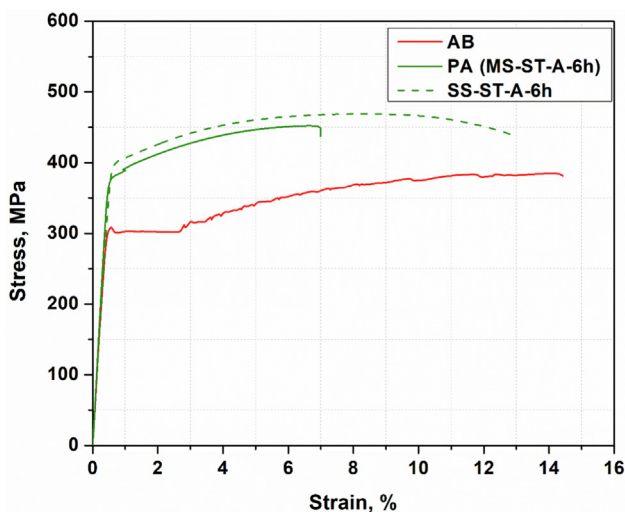
respect to AB samples. Unlike in the AB state, no serrated flow (PLC effect) was observed. It could be related to the consumption of solute atoms (Cu and Mg) in the formation of matrix precipitates such as  $\Omega$  and  $\theta'$  and precipitation of equilibrium  $\theta$ -Al<sub>2</sub>Cu during aging. Furthermore, Mg tends to segregate at the grain boundaries (Fig. 8). The YS and UTS values of PA (MS-ST-A-6 h) samples were 21 % and 15 % higher than the AB ones, respectively (Table 4). However, there was a significant loss in its ductility and the total elongation was drastically reduced to 58 % as that of the AB state. As compared to PA (MS-ST-A-6 h), the tensile curve of SS-ST-A-6 h

demonstrated an improvement in both strength and ductility. The YS and UTS were increased by 7.1 % and 6.3 % respectively. Furthermore, the total elongation significantly increased by 45 % (Table 4). The reduced ductility in PA (MS-ST-A-6 h) state could be attributed to the presence of grain boundary cracks and creation of wide PFZs near micro-TiB<sub>2</sub> particles (Fig. 9). Although, grain boundary cracks have much severe effect on the ductility [56], PFZs possess less strength than grain boundaries which upon tensile loading results in formation of microcracks in these regions and propagate much easier causing a sudden failure [70,71].

**Table 4**

Summary of the mechanical properties and microstructural features at various heat treatment stages of LPBF manufactured A20X alloy.

| Sample           | Microstructural features          |                                       |                                                      |                                                  |                   |              | Mechanical properties |             |              |                |
|------------------|-----------------------------------|---------------------------------------|------------------------------------------------------|--------------------------------------------------|-------------------|--------------|-----------------------|-------------|--------------|----------------|
|                  | Average grain size, $\mu\text{m}$ | $\theta\text{-Al}_2\text{Cu}$ , wt. % | Length of nano-precipitates ( $\Omega/\theta'$ ), nm | Dislocation density, $\times 10^{14}/\text{m}^2$ | Width of PFZs, nm |              | Micro-hardness, HV    | YS, MPa     | UTS, MPa     | e, %           |
| AB               | $1.05 \pm 0.4$                    | $4.2 \pm 0.4$                         | –                                                    | $6.13 \pm 0.32$                                  | –                 | –            | $113 \pm 4$           | $304 \pm 3$ | $378 \pm 6$  | $12.5 \pm 1.6$ |
| MS-ST            | $1.97 \pm 0.5$                    | –                                     | –                                                    | $2.63 \pm 0.27$                                  | –                 | –            | $103 \pm 5$           | –           | –            | –              |
| PA (MS-ST-A-6 h) | $2.17 \pm 0.5$                    | $1.9 \pm 0.3$                         | $54 \pm 17$                                          | $5.34 \pm 0.32$                                  | $30 \pm 6$        | $85 \pm 36$  | $152 \pm 6$           | $370 \pm 9$ | $435 \pm 13$ | $7.3 \pm 0.3$  |
| OA               | $1.93 \pm 0.6$                    | $4.2 \pm 0.3$                         | $82 \pm 20$ & $422 \pm 63$                           | $4.14 \pm 0.28$                                  | $78 \pm 7$        | $153 \pm 58$ | $124 \pm 5$           | –           | –            | –              |
| SS-ST-A-6 h      | –                                 | $0.9 \pm 0.1$                         | $58 \pm 25$                                          | $4.67 \pm 0.99$                                  | $43 \pm 8$        | $32 \pm 13$  | $143 \pm 2$           | $396 \pm 3$ | $462 \pm 4$  | $10.6 \pm 1.8$ |

**Fig. 13.** SEM-BSE micrographs of (a) SS-ST-A-6 h and (b) PA (MS-ST-A-6 h) (Yellow arrows depict grain boundary cracks). (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)**Fig. 14.** Tensile stress–strain curves of A20X alloy in the AB, PA (MS-ST-A-6 h) and SS-ST-A-6 h conditions.

#### 4. Conclusions

In this paper, the microstructural evolution of LPBF A20X samples during a T7 post-processing heat treatment was investigated. Initially, the microstructural evolution was considered for four main states: as-built (AB), multi-step solution treatment (MS-ST), peak aged (PA) and over aged (OA). Then, the microstructural evo-

lution of multi-step solution treatment was done at various interrupted solution steps. Post to the microstructural analysis, a short and simple heat treatment was proposed and compared with T7 in terms of microstructure and mechanical behaviour. The key results of this investigation are as follows:

1. The A20X AB microstructure was similar to the microstructure of the as-atomized powder. It was composed of fine cellular structures with  $\alpha\text{-Al}$  cells and  $\theta\text{-Al}_2\text{Cu}$  eutectic as cell boundaries. The micro- $\text{TiB}_2$  particles were present at triple junctions, within the cells as well as at the cell boundaries. MS-ST dissolved the eutectic  $\theta\text{-Al}_2\text{Cu}$  and solutionized Al matrix with Cu.
2. The PA state was characterized by  $\theta\text{-Al}_2\text{Cu}$  precipitates near the micro- and nano- $\text{TiB}_2$  particles (present specifically at grain boundaries), this is due to the difference in coefficient of thermal expansion between Al (matrix) and  $\text{TiB}_2$  (reinforcement). Excessive aging (OA) resulted in significant growth of  $\theta\text{-Al}_2\text{Cu}$  precipitates and engulfment of micro- $\text{TiB}_2$  particles. PA and OA microstructures displayed grain boundary cracks which were more severe in the case of OA state.
3. The origin of grain boundary cracking has been associated to the synergetic effect of micro-strains developed due to precipitation of  $\Omega$  and  $\theta'$  at grain interiors and simultaneous widening of PFZs near grain boundaries and near micro- $\text{TiB}_2$  particles.
4. The key microstructural features responsible for PA hardness are; (i) presence of coherent- $\Omega$  and semi-coherent- $\theta'$  matrix precipitates of nano-size (about 54 nm in length) and are considered as one of the strengthening phases, (ii) optimum precipitation of  $\theta\text{-Al}_2\text{Cu}$  precipitates and (iii) less frequent grain boundary cracks than OA state.



5. The microstructural evolution of MS-ST revealed that the low temperature solution steps caused break-up of  $\theta$ -Al<sub>2</sub>Cu eutectic network into fine and spheroidal particles that migrated towards micro-TiB<sub>2</sub> particles and coarsened. Additionally, it was pointed out that the very long MS-ST leads to an incomplete dissolution of  $\theta$ -Al<sub>2</sub>Cu. This has been attributed to the shielding effect of TiB<sub>2</sub> particles between  $\theta$ -Al<sub>2</sub>Cu and  $\alpha$ -Al matrix which limits the  $\theta$ -Al<sub>2</sub>Cu dissolution.
6. The proposed single-step solution treatment (SS-ST) allowed a nearly complete dissolution of  $\theta$ -Al<sub>2</sub>Cu. The microstructure of SS-ST aged for 6 h revealed negligible precipitation of  $\theta$ -Al<sub>2</sub>Cu near micro-TiB<sub>2</sub> particles, the width of PFZs near micro-TiB<sub>2</sub> particles ( $\approx$  32 nm) was much less than in the MS-ST aged for 6 h ( $\approx$  85 nm), and thus, no grain boundary cracks were observed in SS-ST aged for 6 h.
7. The mechanical behaviour of AB, PA (MS-ST-A-6 h) and SS-ST-A-6 h were compared. The AB specimen displayed a typical tensile curve with Lüders band formation and PLC effect. PA (MS-ST-A-6 h) showed 21 % and 15 % increase in YS and UTS respectively as compared to the AB state and a reduction of elongation to fracture by 58 %. No PLC effect was observed in PA (MS-ST-A-6 h) state. On the other hand, SS-ST-A-6 h demonstrated a significant increase in elongation to fracture by 45 % with respect to PA (MS-ST-A-6 h). The YS and UTS increased by 7.1 % and 6.3 % respectively. The improved strength and ductility in SS-ST-A-6 h has been attributed to the absence of grain boundary cracks.

Based on the current investigation, the proposed post-processing heat treatment for the AM printed A20X alloy offers better mechanical properties than the conventional T7. It is short and simple, energy efficient and can be easily adaptable for the industrial practices.

## Data availability

Data will be made available on request.

## Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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## Appendix A. Supplementary material

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.matdes.2022.111566>.

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