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#### Novel infiltration method of membranes over activated carbons electrodes

Original

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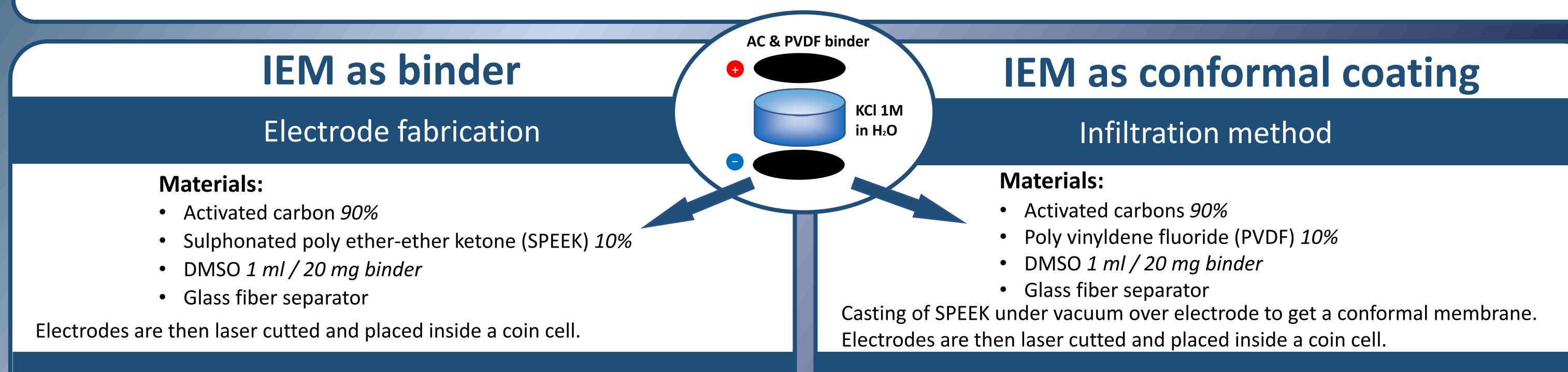


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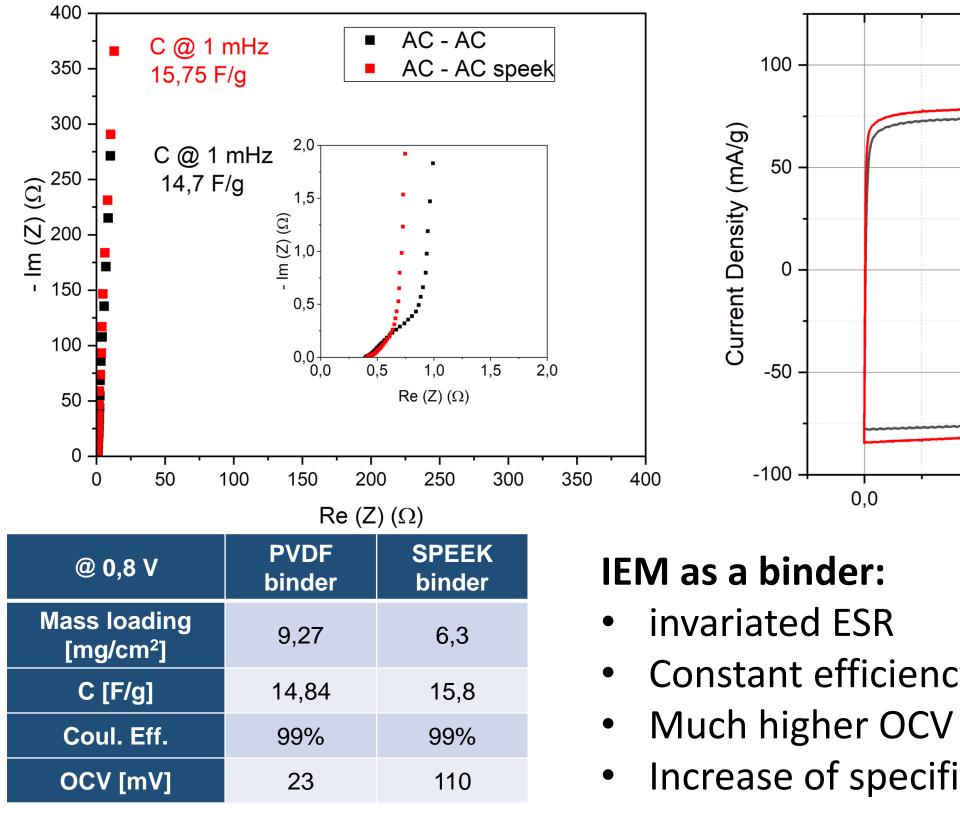
# Background

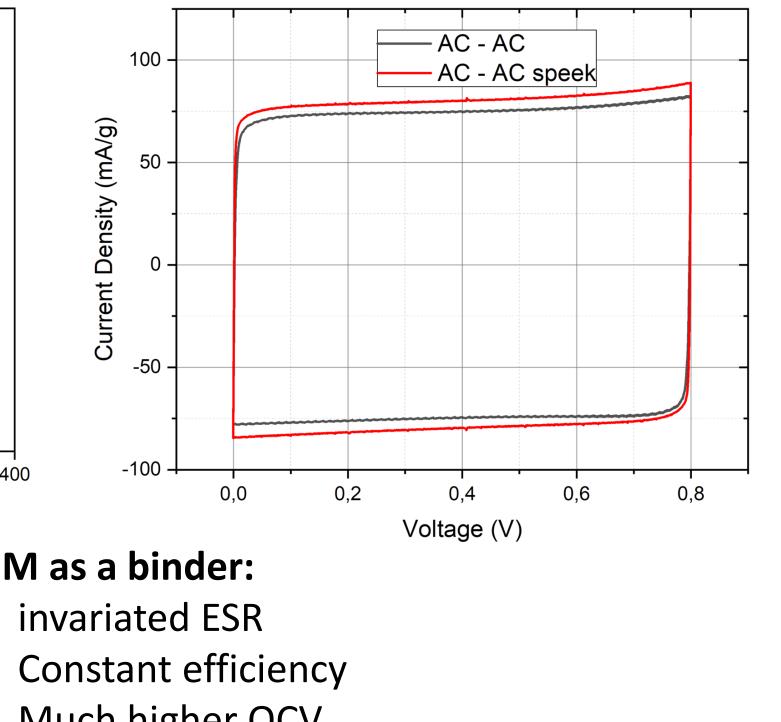
A novel method to modify activated carbon electrodes through addiction of selective ion exchange polymers (IEM) is presented. Nowadays the preferred method to self-polarize an electrode in an EDLC is to place a stand-alone membrane in proximity of the electrode itself [1, 2]. This technique produces some drawbacks brought about the dimension of the device: an increase of the internal series resistance and reduction of overall capacitance. With the propoesed methods, it is possible to use the IEM directly in the production of the slurry (instead of traditional binders) or in a conformal contact with the electrode.



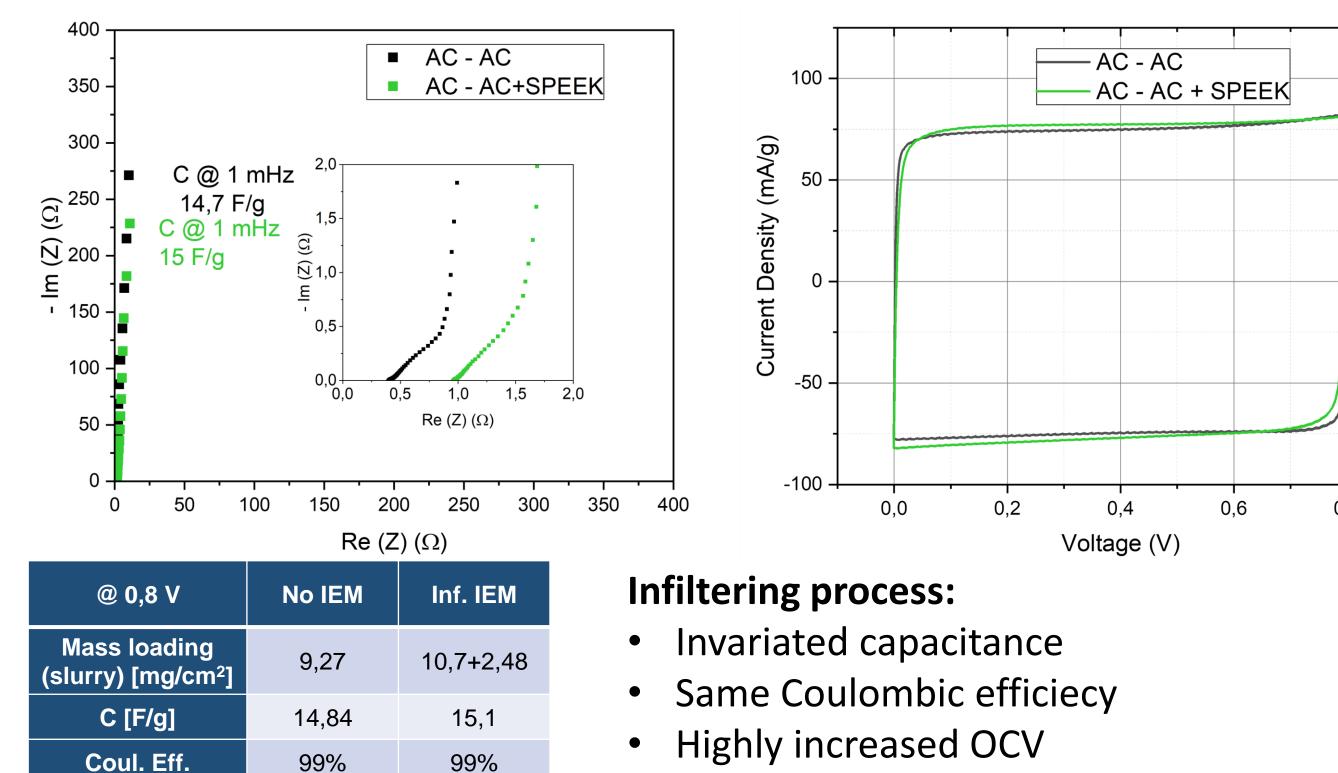


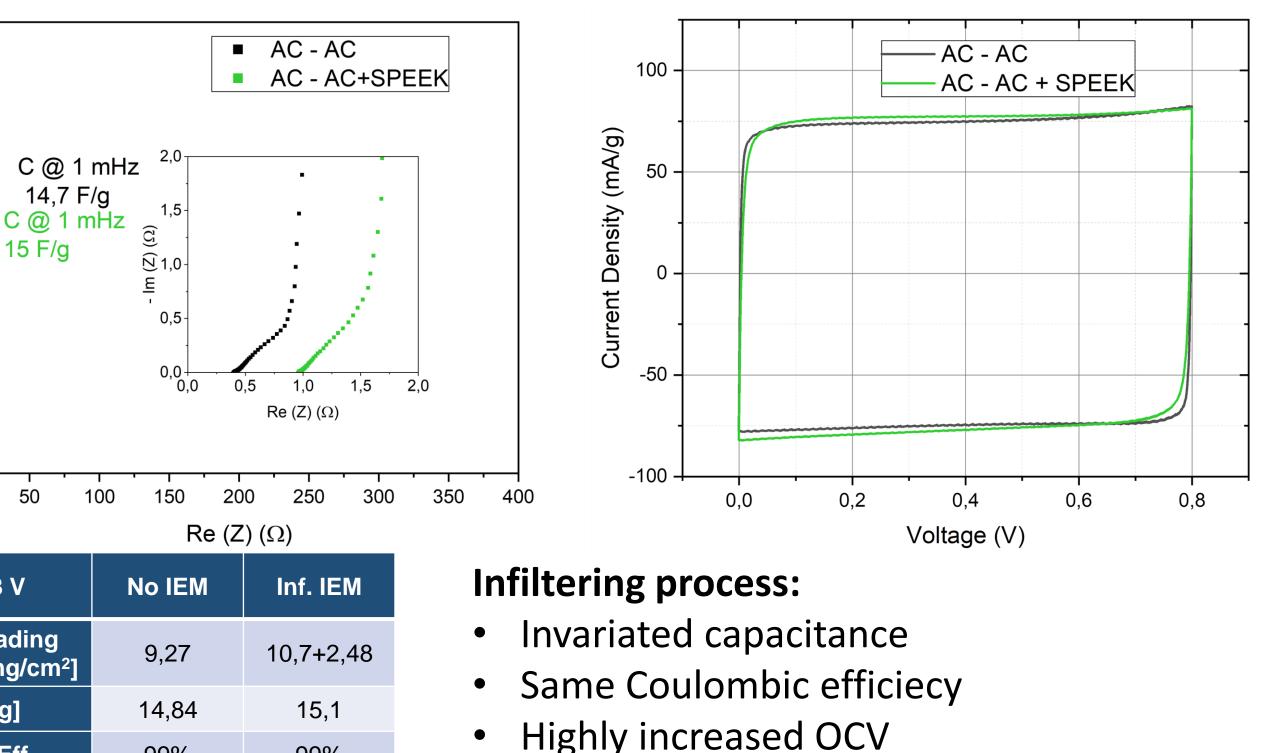
### Electrochemical characterization





#### **Electrochemical characterization**





Increase of specific capacitance

### Adhesion test

Direct contact electrolyte - membrane

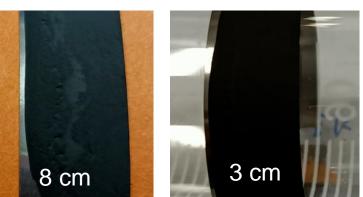
### SEM comparison



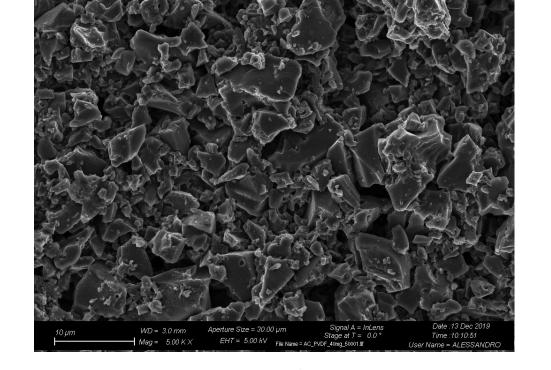
1.5 cm

#### **PVDF** binder **SPEEK binder**

No delamination No delamination due due to deformation to deformation of the of the substrate. substrate. Remotion Remotion of all the only of all the upper deposited material layer of carbon during peeling test. during peeling test.



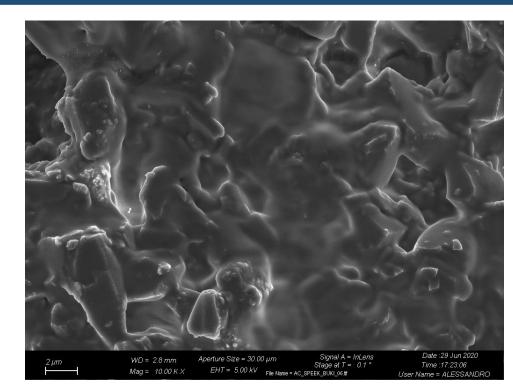




23

458

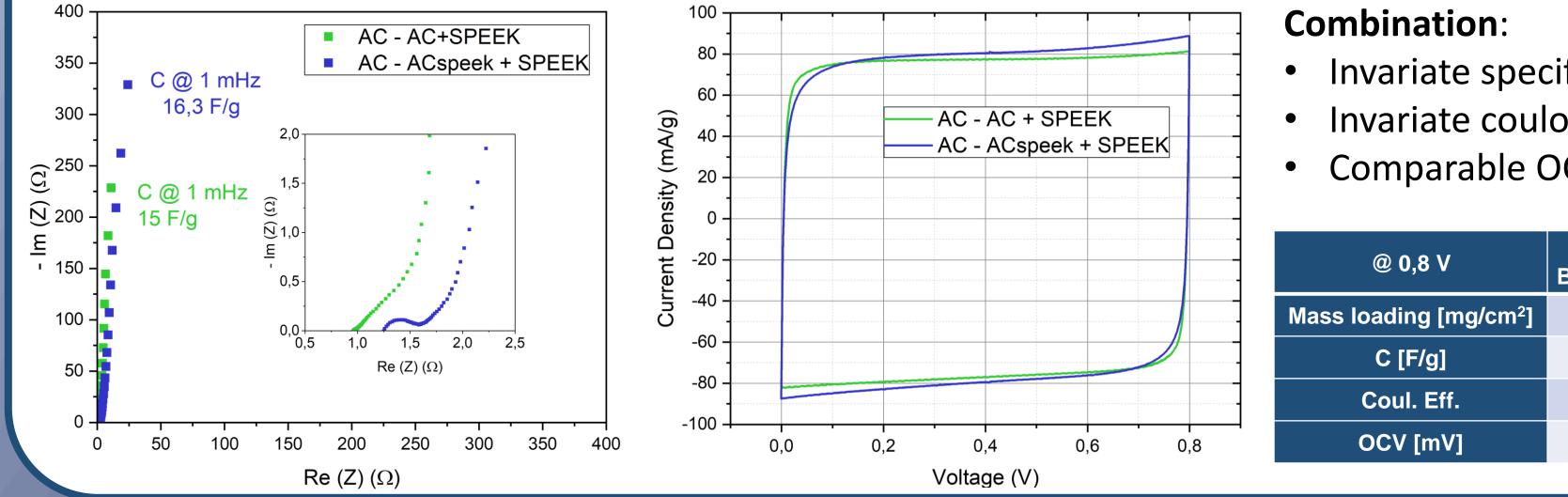
No membrane



Vacuum infiltred membrane

## **Combination of binder + infiltred IEM**

# Conclusions



Invariate specific capacitance

OCV [mV]

- Invariate coulombic efficiency
- Comparable OCV

@ 0,8 V	PVDF Binder + IEM inf	IEM binder + IEM inf
Mass loading [mg/cm <sup>2</sup> ]	10,7+2,48	6,7+3,98
C [F/g]	15,1	15,2
Coul. Eff.	99%	99%
OCV [mV]	458	442

Speek binder provides a better adhesion to Ti

- Infiltering process allows to maintain constant the ESR of the device, but highly increases its OCV
- It is possible to combine both approaches presented

## **Future outlooks**

• Repeat same tests with a anion exchange membrane and combine with SPEEK

• Test the obtained electrodes in a device built for capmix or RED

## **Bibliography:**

[1] Sales, Bruno B., et al. << Electrochemical characterization of a supercapacitor flow cell for power production from salinity gradients>> *Electrochemica acta*, 86 (2012) 298-304 [2] Wang, Xingfeng, et al. <<A 1.8 V aqueous supercapacitor with a bipolar assembly of ion-exchange membranes as separator>> Journal of The Electrochemical Society, 163.9 (2016): A1853

