

Multi-scale modeling to boost fuel cell performance: From pore-scale simulations to better efficiency and durability

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Motivation: High Temp. PEM Fuel Cells





HT PEMFC: main issues









Degradation of electrodes: phosphoric acid loss

Voltage

drop

Phosphoric acid depletion :

Loss of ionic conductivity

- Acid loss may occur through several mechanisms such as :
 - diffusion
 - capillary transport
 - membrane compression
 - evaporation
 - leaching by condensed water during shutdown and cold start
 - T. Sousa, M. Mamlouk, K. Scott*, Int. J. Hydr. En. 35 (2010) 12065
- PA loss as a function of flow rate, temperature, operating conditions

<u>Y. Oono et al. , Journal of Power Sources, 210</u> (2012) 366







Pore-scale modeling of the fluid flow through the electrodes of HT PEMFC



- Development of the reliable algorithm for reconstruction of micro-morphology of electrodes (GDL, CL)
- Challenge for numerical tools and mathematical models for specific degradation phenomenon
- Validate the proposed algorithm and numerical tools computing macroscopic transport coefficients
- Development of mitigation strategy using virtual realization of system

Reconstruction: MEA's real structure SEM images







GDL has mainly ordered structure Example: celtec by BASF® Structure: woven (nonwoven) like Approach: deterministic (stochastic)



00x kV=15 WD=15 P4 H2315T10A Querschnitt BSE

CL has irregular structure Example: celtec by BASF® Vulcan XC - 72R Structure: aggregation of particles Approach: stochastic based on clusterization of particles

Numerical tool: Lattice Boltzmann Method



NUMERICAL TOOL?

- Very complex geometry
- Parallel computation
- Gas rarefaction effect

LBM is the discretization of not only physical space, but also velocity space, which means that particle velocities are restricted to a finite set of orientations.

LBM solves the following Lattice Boltzmann equation:

 $\underbrace{ \underbrace{f_i(\vec{x} + \vec{e_i}\Delta t, t + \Delta t) = f_i(\vec{x}, t)}_{\text{Streaming}} - \underbrace{ \underbrace{ \begin{bmatrix} f_i(\vec{x}, t) - f_i^{eq}(\vec{x}, t) \end{bmatrix}}_{\text{Collision}},$

- It should be noted that in this algebraic equation, "non-locality (streaming term, S) is linear and non-linearity (collision term, C) is local" [Sauro Succi, 2001]
- In LBM fluid models, like BGK or single relaxation time

Computational domain size is ~100 Mcells

lattice

Numerical tool: PALABOS



Palabos (<u>www.palabos.org</u>) is an opensource CFD solver based on the <u>lattice</u> <u>Boltzmann method</u>.



Palabos ingredients:

- Physics: In/weakly compressible, non-thermal Navier-Stokes equations; flows with body-force term, thermal flows with Boussinesq approximation, single-component multi-phase fluids (Shan/Chen model), free surface flows (volume-of-fluid approach), static Smagorinsky model for fluid turbulence.
- Basic fluid models: BGK, MRT, regularized BGK, LW-ACM, entropic model.
- Straight-wall boundary conditions: Zou/He, Inamuro, Skordos, regularized BC, simple equilibrium, bounce-back, periodic.
- Off-lattice boundary conditions: GUO model, generalized off-lattice BC.
- Grid: D2Q9, D3Q13, D3Q15, D3Q19, and D3Q27.
- Parallelism: parallelized with MPI for shared-memory and distributed-memory platforms, including I/O operations that are implemented in terms of MPI's Parallel I/O API.

Validation: Gas diffusion layer



Name	Woven [Celtec BASF [®]]	Non-woven [Freudenberg]
Thickness @0.025MPa	400 µm	256 µm
Computation al domain	534 Mcells	67 Mcells
Fiber diameter	7 μm	7 μm
Porosity	0.78	0.75
Computed Permeability	0.44x10 ⁻¹² m ²	3.2x10 ⁻¹² m ²
Actual permeability	0.50x10 ⁻¹² m ²	2-3x10 ⁻¹² m ²



U. R. SALOMOV, E. CHIAVAZZO, P.ASINARI, Pore-scale modeling of fluid flow through gas diffusion and catalyst layers for high temperature proton exchange membrane (HT-PEM) fuel cells, CAMWA, pp. 19, ISSN: 0898-1221, DOI: 10.1016/j.camwa.2013.08.006

Validation: Catalyst layer



R

Ν

G

Clustering Name	40 nm	400 nm	1500 nm	
Computationa I domain	8 Mcells	8 Mcells	16 Mcells	
Carbon support particles diameter	40 nm	40 nm	40 nm	
Computed	1.1x10 ⁻¹⁶ m ²	4.8x10 ⁻¹⁵ m ²	0.4x10 ⁻¹³ m ²	
Permeability	Increase of permeability			
Actual permeability		1.0x10 ⁻¹³ m ²		

U. R. SALOMOV, E. CHIAVAZZO, P.ASINARI, Pore-scale modeling of fluid flow through gas diffusion and catalyst layers for high temperature proton exchange membrane (HT-PEM) fuel cells, CAMWA, pp. 19, ISSN: 0898-1221, DOI: 10.1016/j.camwa.2013.08.006

Validation: Effect of redistribution of Pt particles on flow field



Name	Non-distrib	Distributed
Computatio nal domain	2.1 MNodes	2.1 MNodes
Average cluster size	1500 nm	1500 nm
Computed flow rate	0.31x10 ⁻⁶	2.3x10 ⁻⁶



Redistribution of catalyst particles inside the microstructure leads to considerable increase of mass flow rate.

This provides an additional tunable parameter of the morphological model.



Mitigation strategy: Redistribution of Pt particles in CL

ECoMaTech, 19-21 September, 2013, Bled (Slovenia)

goal of proposed The morphological model is to design a strategy in order to enhance PEMFC performance and to mitigate degradation phenomena by improving mass transport processes.

Two virtual realizations of morphological model are considered in regards to Pt distribution:

- 1. Homogeneous
- 2. **Exponential decay**





Exponential decay





Mitigation strategy: Results of pore-scale flow simulation



There are quantitative changes (scale factor), but also qualitative changes in the flow field (streamlines). Thus, redistribution of catalyst particles inside the microstructure leads to considerable increase of overall mass flow rate and changing flow field near membrane (flow rate almost zero).



High Temperature PEM FC single MEA performance simulation

Single MEA structure





MEA secondary current distribution





0

Electrolyte: Poisson equation

Electrodes: Poisson equation

$$\nabla \cdot i_l = \mathbf{Q}_l, \, i_l = -\sigma_l \nabla \varphi_l$$

Porous electrodes: mixture of Poisson eqs of electrodes and electrolyte.

Electrode kinetics: Linearized Butler-Volmer for anode CL

$$i_a = i_{a,0} \left(\frac{c_{H_2}}{c_{H_{2,ref}}}\right)^{0.5} \left(\frac{\alpha_a F \eta_a}{RT}\right)$$

Cathodic Tafel equation for cathode CL

$$i_{c} = -i_{c,0} \left(\frac{c_{O_{2}}}{c_{O_{2},ref}}\right) exp\left(\frac{\alpha_{c}F\eta_{c}}{RT}\right)$$

Transport mechanisms

0





Transport of concentrated species:

- Compartments: Channel, GDL and CL
- Anode: Hydrogen and water
- Cathode: Oxygen, water and nitrogen
- Mass transfer : Maxwell-Stefan eqs (for LBM, see <u>P. Asinari, PRE, vol. 80,</u> 056701 (2009) and <u>P. Asinari, PRE, vol.</u> 77, 056706 (2008))
- Momentum transfer: Navier-Stokes and Brinkman equations
- Coupling with Poisson equations as sources and sinks at porous electrode (catalyst layer)

CL with different Pt distribution





Pt particles distribution



C1: Exponential decay C2: Homogeneous (today) C3: Exponential increase

Single MEA performance simulation



Region with higher performance Pc1>Pc2

C1: Exponential decay C2: Homogeneous (today) C3: Exponential increase



Summary



Change flow field (no streamlines) in the region near membrane

Decrease phosphoric acid loss

Improve overall transport of gases through CL porous media

Enhance performance of PEMFC

Almost zero flow rate and its slope near membrane

Decrease possible crossover of gases through membrane

Provide longer path for transport of ions from reaction place

Increase Ohmic resistance



Thank you for your attention !

