Polyurethane-based thiomers: A new multifunctional copolymer platform for biomedical applications

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Attenuated Total Reflectance Fourier Transform Infrared (ATR-FTIR) spectroscopy

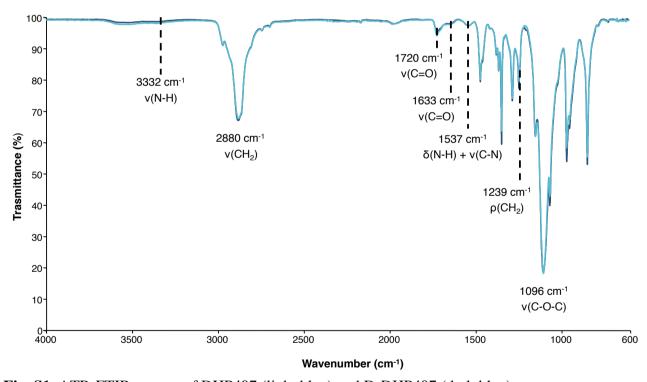


Fig. S1. ATR-FTIR spectra of DHP407 (light blue) and D-DHP407 (dark blue).

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Size Exclusion chromatography

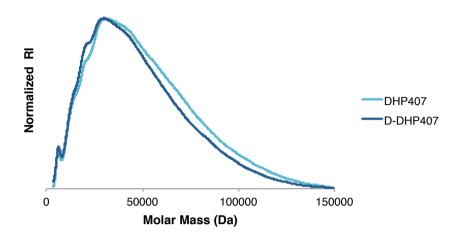


Fig. S2. Molecular weight distribution profiles (normalized Refractive Index (RI) *vs* molar mass) of the synthesised polyurethane before (DHP407 – light blue) and after (D-DHP407 – dark blue) acid treatment to remove Boc protecting groups.

Orange II Sodium Salt colorimetric assay

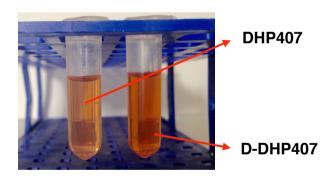


Fig. S3. Orange II Sodium Salt colorimetric assay performed on DHP407 (light orange – Orange molecules absorbed) and on D-DHP407 (dark orange – Orange molecules absorbed and grafted to – NH groups) samples.

Dynamic Light Scattering measurements

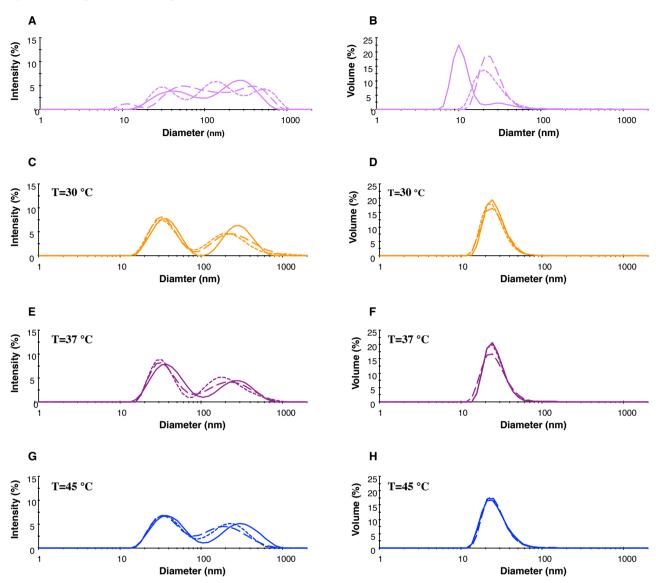


Fig. S4. Distribution by intensity (A, C, E, G) and by volume (B, D, F) and (B, D, F)

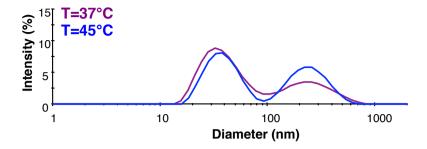


Fig. S5. Hydrodynamic average diameter profiles (intensity patterns) of D-DHP407 at 0.5% w/V concentration measured at 37 °C (purple) and 45 °C (blue).

Attenuated Total Reflectance Fourier Transform Infrared (ATR_FTIR) spectroscopy

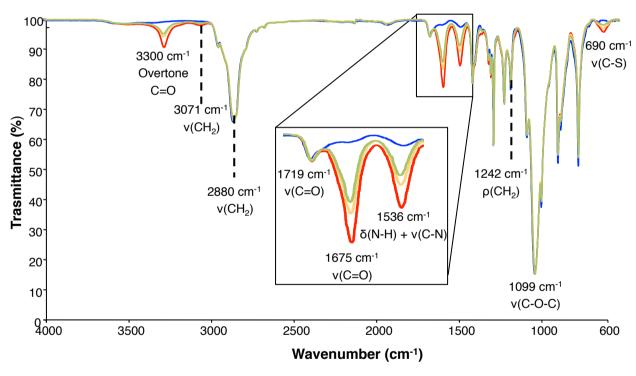
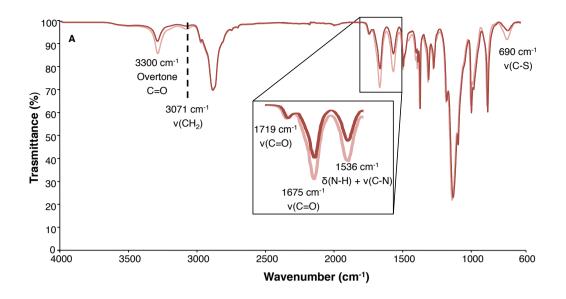
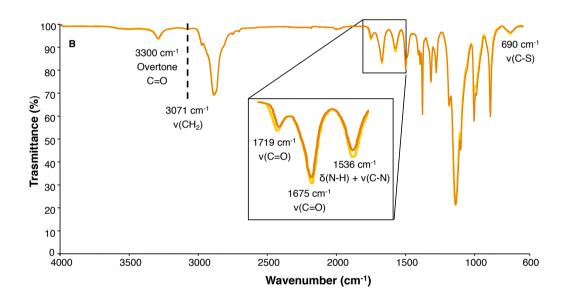


Fig. S6. Averaged ATR-FTIR spectra of D-DHP407 (blue) and S-DHP407 after TGA grafting at pH 4 (red), 5 (yellow) and 7 (green) and 24 h reaction time. Differences between spectra are reported as magnifications.





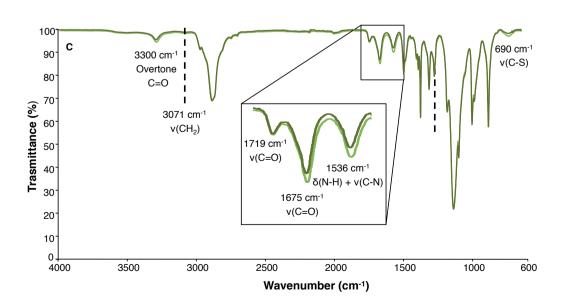


Fig. S7. Averaged ATR-FTIR spectra of S-DHP407 samples obtained by adjusting the grafting reaction pH at 4 (A), 5 (B) and 7 (C) for 6 h (light line) and 24 h (dark line). Main differences between spectra are reported as magnifications.

Dynamic Light Scattering measurements on S-DHP407_pH4_6h

To investigate whether TGA grafting could influence polyurethane thermo-sensitivity, DLS measurements were also performed on thiomer-based solution at 0.5% w/V concentration at 25 °C, 37 °C and 45 °C, according to the protocol adopted for D-DHP407 characterisation (Fig. S8). Specifically, the polymer exposing the highest amount of thiol groups/g of polymer (i.e., S-DHP407_pH4_6h) was considered for this analysis.

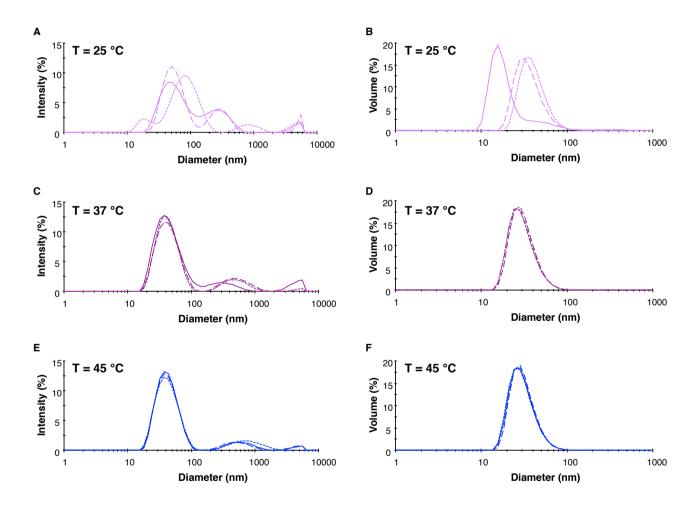


Fig. S8. Distribution by intensity (A, C, E) and by volume (B, D, F) of the hydrodynamic diameter of the polymeric micelles and aggregates in S-DHP407_pH4_6h solutions with 0.5% *w/V* concentration measured at different temperatures (25 °C –37 °C and 45 °C). Continuous, dash and dot lines represent measurements conducted on three different samples prepared in the same conditions.

According to data reported for D-DHP407 samples at 25 °C (Fig. 4), no clear chain organisation was detected also for thiomer-based solutions at the lowest tested temperature (Fig. S8_A and B), due to the high system instability. Upon temperature increase up to 37 °C (Fig. S8_C) a well-defined double peak appeared, ascribed to single micelles (46.24±1.9 nm) and aggregates (451.38±25.08 nm), respectively. This bimodal chain organisation was further confirmed at 45 °C (Fig. S8_E) with micelles and aggregates showing a mean hydrodynamic diameter of 46.01±0.9 nm and 622.91±54.89 nm, respectively, thus proving that micelle nucleation proceeded upon temperature increase. Hence, TGA grafting did not affect polymer thermo-responsiveness. On the contrary, both S-DHP407-based micelles and clusters showed higher hydrodynamic diameters compared to D-DHP407-based ones, suggesting that the exposure of thiol groups induced the formation of a thicker hydrated shell around them which accounts for the observed increased in their dimension. However, only one peak at about 30 nm was identified from the volume patterns (Fig. S8_D and F), according to D-DHP407 results. Thus, also for thiomer-based solutions the amount of aggregates present in the sample was negligible (< 1% at each analysed temperature).