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Energy Harvesting from Carbon Dioxide Capture through an Ionic Liquid Based Supercapacitor

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(Article begins on next page)



# Energy harvesting from carbon dioxide capture through an ionic liquid based supercapacitor

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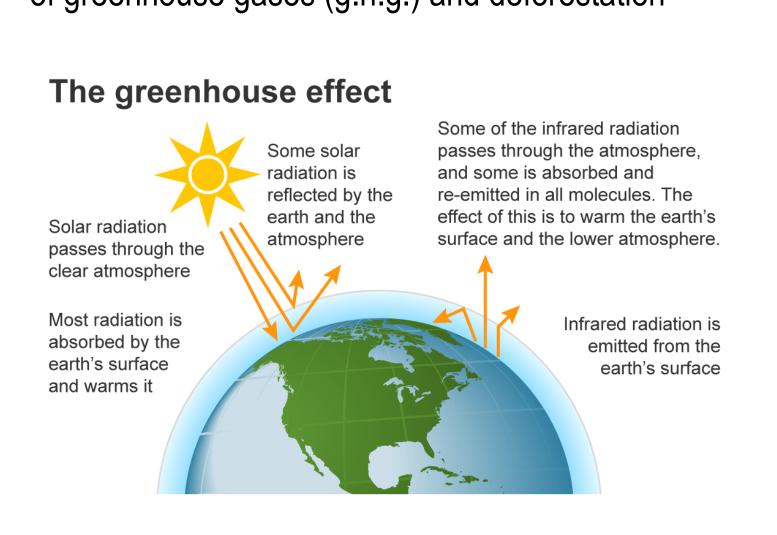
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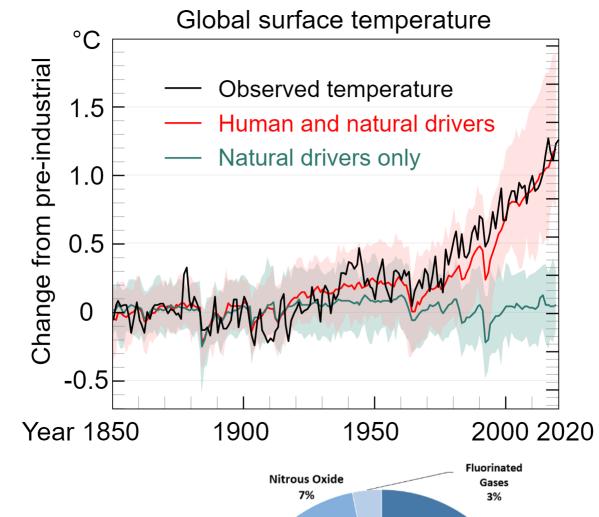


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### Research context and motivation

• Starting from 1900 global temperature started to increase, mainly because of human influence: emission of greenhouse gases (g.h.g.) and deforestation

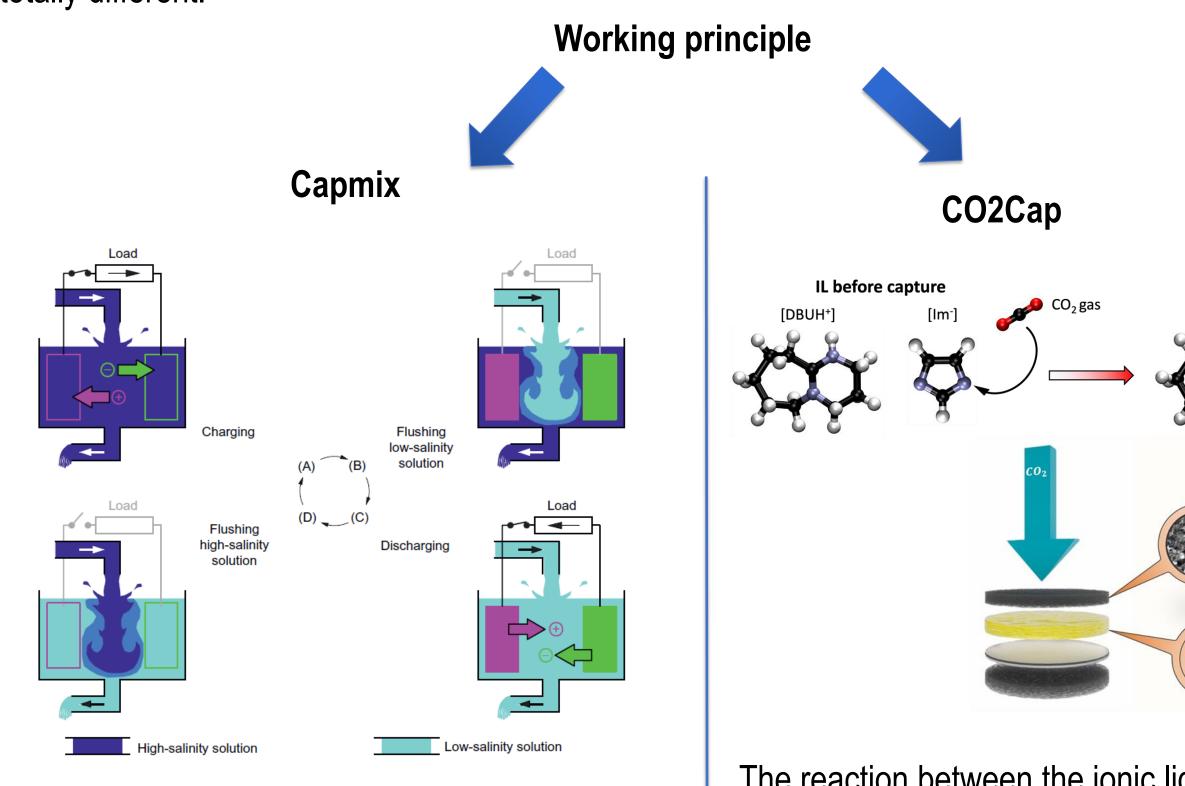




- Greenhouse gas composition: the most abundant gas among g.h.g. is Carbon dioxide (CO<sub>2</sub>), coming mainly from oil and coal
- CO<sub>2</sub> is **responsible for 60% of global warming** due to human activity

## Addressed research questions/problems

• The goal is to harvest energy from CO<sub>2</sub> capture adapting the Capmix technique, but the mechanism is totally different.



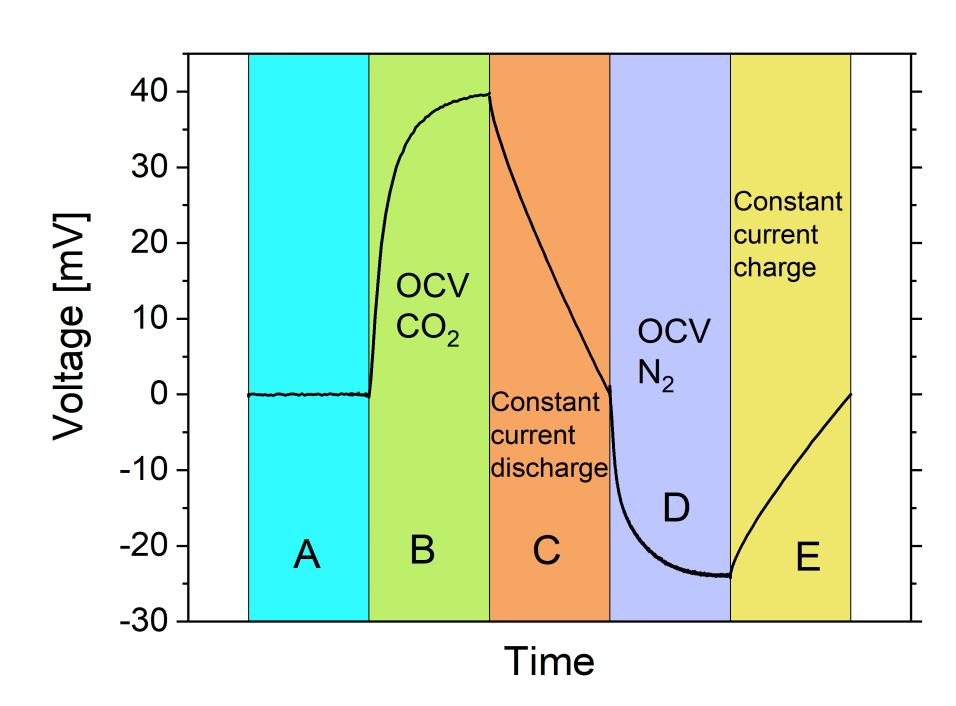
Technology exploited in blue energy field, based on **EDL enlargement** 

The reaction between the ionic liquid and CO<sub>2</sub> happens only on one electrode, producing a junction across which a voltage difference is created

# Adopted methodology

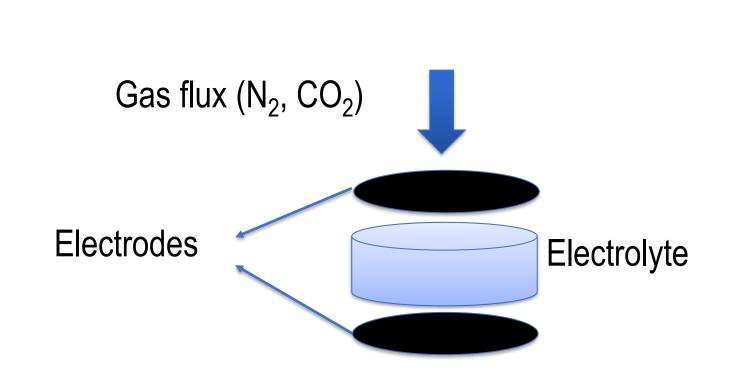
**Procedure:** inspired from Capmix, but avoiding the polarization of the device used to store charges at the electrodes interfaces.

- A: 5' short circuit
- B: OCV + CO<sub>2</sub> flush (50 ml/min)
- C: Constant current discharge (energy recovery)
- D: 15' OCV +  $N_2$  flush (50 ml/min) (regeneration of the electrolyte)
- E: Constant current charge (energy recovery)



### Results

 Idea to improve CO2Cap performances is to exploit lonic liquids as electrolyte inside the harvesting device. As in Capmix technology, we substitute high and low concentration solutions with fluxes of CO<sub>2</sub> and

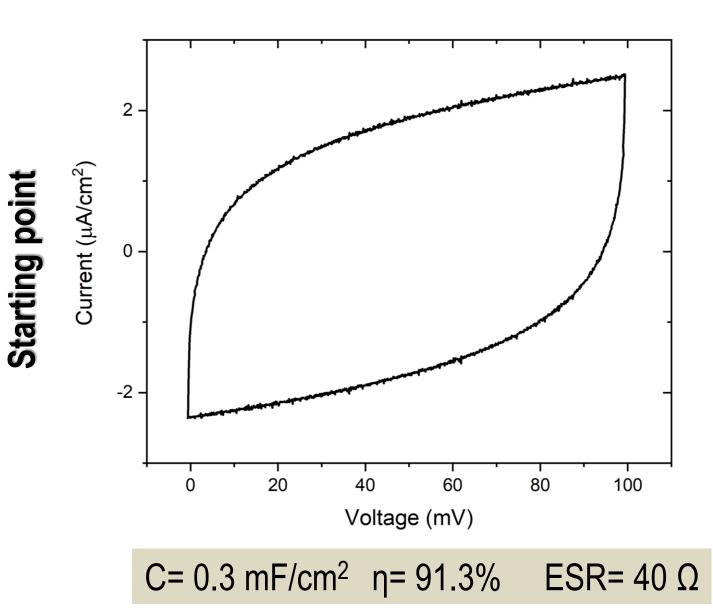




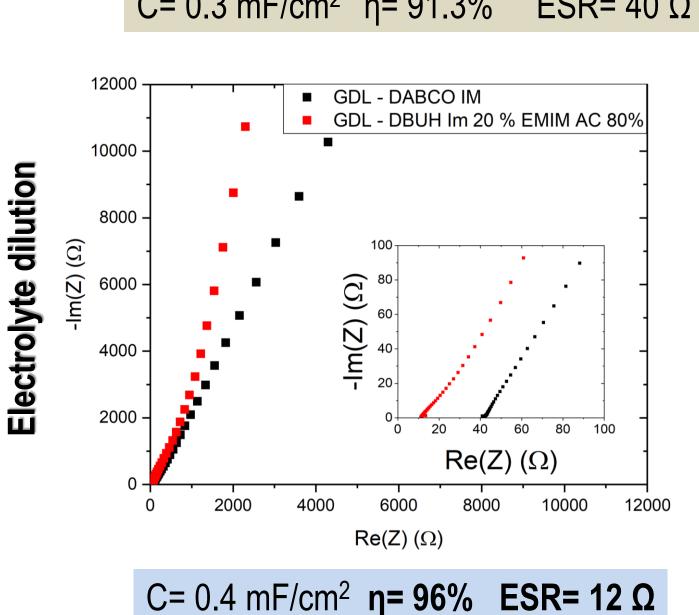
**Electrodes:** GDL **Electrolyte:** 20% Dabco Im 80% Emim AC

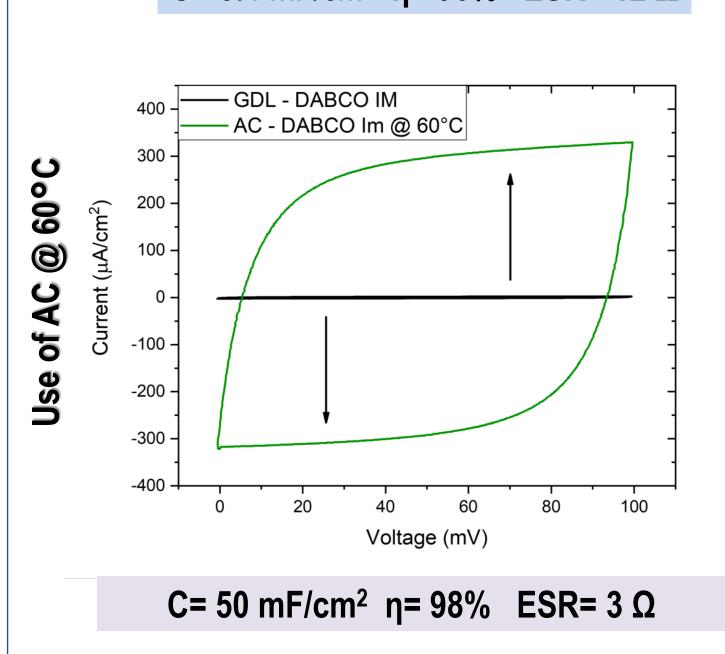
3 tested configurations

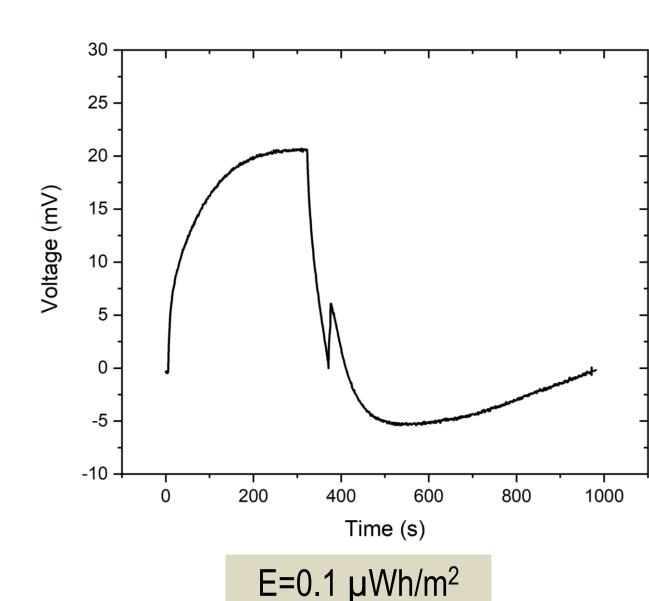
**Electrodes:** Act. Carbons **Electrolyte:** 20% Dabco Im 80% Emim AC



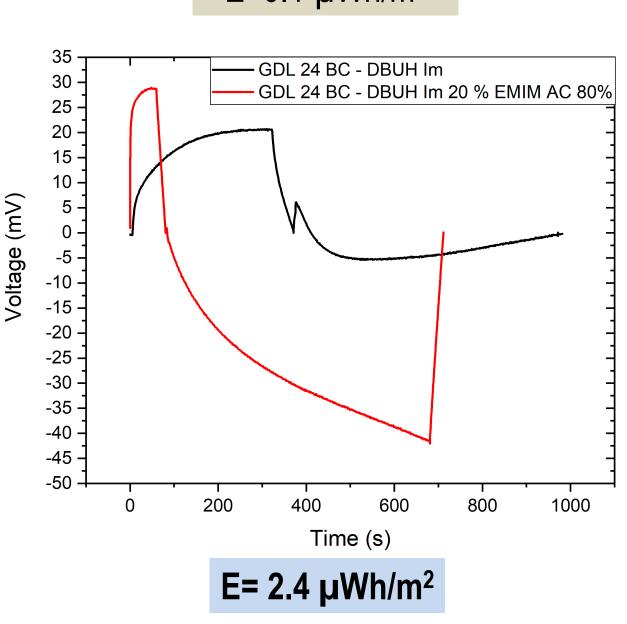
**EC** characterization

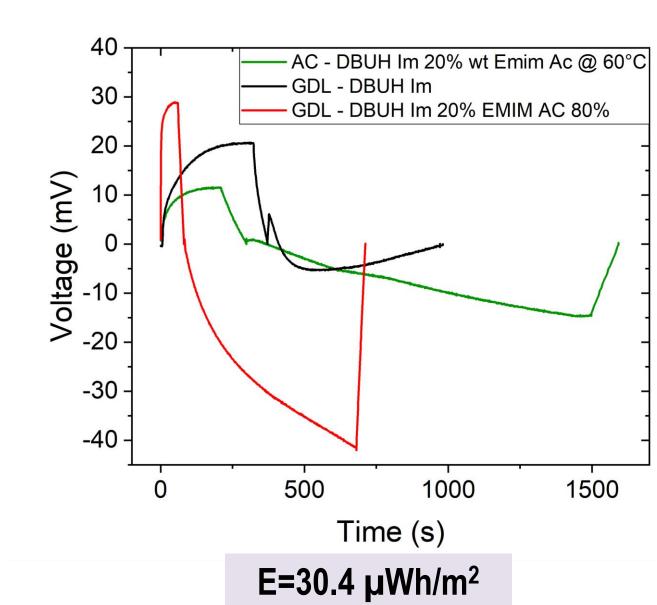






**Capmix performances** 





### **Future work**

- Use of new ionic liquids more selective for CO<sub>2</sub> capture
- Improve ionic mobility, reducing ion pairing by polar aprotic solvent, such as Propylene carbonate
- Enhance conducibility of the electrolyte by inserting a supporting salt
- Increase the voltage rise due to the adsorption of CO<sub>2</sub> by exploiting functionalized electrodes, able to autonomously accumulate specific charges at their surface

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