Feasibility Assessment of Acid Gas Injection in an Iranian Offshore Aquifer

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Feasibility Assessment of Acid Gas Injection in Saline Aquifers: A Case Study of an Iranian Offshore

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Abstract: This study examines the feasibility of Acid Gas Injection (AGI) in a saline aquifer offshore, Iran. Reservoir properties, geomechanical aspects, caprock integrity, and gas plume dynamics are investigated. The Surmeh formation was proposed as a suitable candidate for AGI due to the presence of upper dolomite and lower carbonate in the rock sections. Geomechanical analysis reveals a pore pressure of 3800 psi and a fracture pressure of 6100 psi. Caprock integrity, particularly the Hith formation, is crucial for both containment and long-term stability. Seismic mapping indicates caprock thickness variations that affect containment effectiveness. Capillary trapping is found to play a significant role in short-term gas trapping and plume distribution. Numerical simulations demonstrate how heterogeneous rock properties influence capillary trapping and gas plume motion.

Approximately 2 TCF (Trillion Cubic Feet) of acid gas are projected to be injected into the Surmeh formation. The recommended injection rate is 180 MMSCFD (Million standard cubic feet per day), based on acid gas content and gas in place of the source of injection. The tight nature of the Surmeh formation limits injectivity, with a maximum achievable rate of 7 MMSCFD for permeability of 1 mD (millidarcy). However, higher porosity (12%) and permeability of 100 mD enable more efficient injection without fracturing the formation. This study provides valuable insights into the feasibility of AGI in saline aquifers, emphasizing reservoir characterization, geomechanics, caprock integrity, and rock properties. The findings contribute to the implementation of environmentally sustainable acid gas disposal at offshore reservoirs.

Keywords: Acid Gas Injection (AGI), Saline aquifers, Capillary pressure, Reservoir characterization, Caprock integrity, Capillary pressure.

1. Introduction

In accordance with the scenarios outlined by the Intergovernmental Panel on Climate Change (IPCC), to limit global warming to 1.5 °C, it is necessary to achieve global net-zero CO_2 emissions by approximately 2050. This goal requires significant reductions in all human-caused emissions and balancing any remaining unavoidable anthropogenic emissions with equivalent carbon removal measures. The aim is to minimize emissions as

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much as possible and offset any residual emissions through effective carbon removal strategies[1].

Carbon capture and storage (CCS) represents a crucial technology with the aim of tackling greenhouse gas emissions and reducing the effects of climate change. CCS technologies offer the means to achieve both carbon dioxide removal and emissions reduction. These technologies start by isolating carbon dioxide through a capture process. Subsequently, the captured carbon dioxide is conditioned, transported and finally stored in geological formations. As a result, CCS technologies effectively reduce carbon dioxide emissions at specific sources or extract carbon dioxide directly from the atmosphere [2].

Acid gas injection operations serve as the commercial equivalent to some aspects of geological CO_2 storage. Acid gas, comprising H_2S and CO_2, together with small amounts of hydrocarbon gases originating from petroleum production or processing, make up the acid gas composition. The primary objective of acid gas injection operations is the disposal of H_2S. However, substantial amounts of CO_2 are injected simultaneously due to the economic impossibility of separating the two gases [3].

The primary method of acid gas injection involves the injection of a stream consisting mainly of H_2S and CO_2, obtained from the sweetening plant. This stream is compressed and piped to an injection well, where it is directed downward into a subsurface formation typically intended for disposal. The general injection scheme, which includes the sweetening plant and associated processes, can be represented by a block diagram (Figure 1).
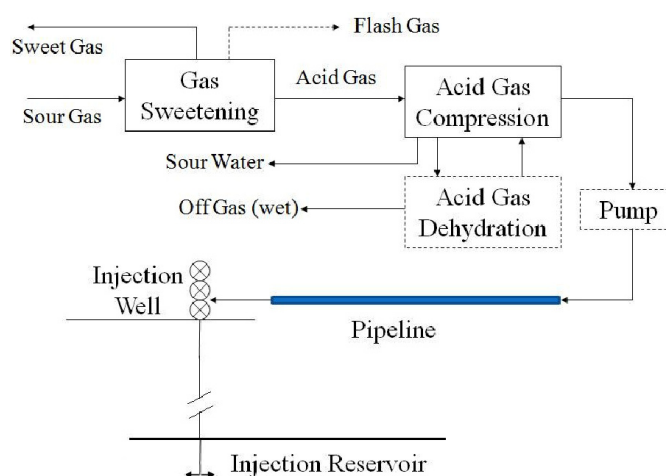

Figure 1. Representation of AGI by a block diagram [4].

Regulatory agencies in Western Canada are currently granting approval for several parameters related to acid gas injection. These include the maximum allowed fraction of H_2S, the maximum injection pressure, and rate at the wellhead, as well as the maximum injection volume. Acid gas injection operations are currently conducted in 51 distinct formations, located at 44 different sites across the Alberta Basin in the provinces of Alberta and British Columbia. The injection of acid gas occurs in various types of formations at different sites. Specifically, it takes place in deep saline formations at 27 sites, depleted oil and/or gas reservoirs at 19 sites, and in the underlying water leg of depleted oil and gas reservoirs at 4 sites. These different types of formations act as places for the injection process. Of all the sites, 29 rely on carbonates as the primary reservoir formation for acid gas injection. Conversely, the remaining 21 sites predominantly use quartz-rich sandstones as the dominant reservoir formation. In most cases, shales act as caprocks, serving as an upper confining unit for injection zones. However, in the remaining injection zones, narrow

limestones, evaporites, and anhydrites are responsible for the confinement of the injected substances and their effective containment [5].

Numerous sedimentary regions worldwide possess varying degrees of suitability for CO_2 storage. In general, geological storage sites must have the following characteristics to be suitable for CO_2 storage:

- Sufficient capacity and injectivity to accommodate the CO_2 being injected.
- An effective sealing caprock, or confining unit, to prevent CO_2 leakage.
- A geologically stable environment that ensures the long-term integrity of the storage site, minimizing the risk of any potential compromise [6].

In this study, the focus is on assessing the feasibility of AGI in saline aquifers located in an Iranian offshore reservoir. A comprehensive survey was carried out, incorporating geological data, drilling data, petrophysical and geophysical information, as well as geomechanical data. Next, a simulation study was conducted using commercial software to determine optimal reservoir properties and maximum injection rate while ensuring that the bottom-hole pressure (BHP) remained below the fracture pressure threshold. The primary objective is to maintain the integrity of the storage site and prevent any risks associated with exceeding the fracture pressure.

2. Case Study

The proposed concept involves injecting the sour gas into the reservoir with the produced gas, which has a high H_2S content of around 40,000 ppm. To make the gas suitable for commercial use, a sweetening process is necessary to remove unwanted components. However, the gas containing H_2S, CO_2, and CH_4, which is not economically viable, must be disposed of properly. The disposal process ensures proper handling of gases that are not economically feasible for commercial use, while maintaining safety and environmental standards.

The study area selected for this research is located in the Persian Gulf, approximately 100 km from the Iranian shoreline and 120 km from the city of Bushehr. The geological features of this region include basal forces and salt diapirism, which have contributed to the formation of an asymmetrical dome structure with an east-west (E-W) trend [7] (see Figure 2). Figure 3 represents the stratigraphic column of the field.

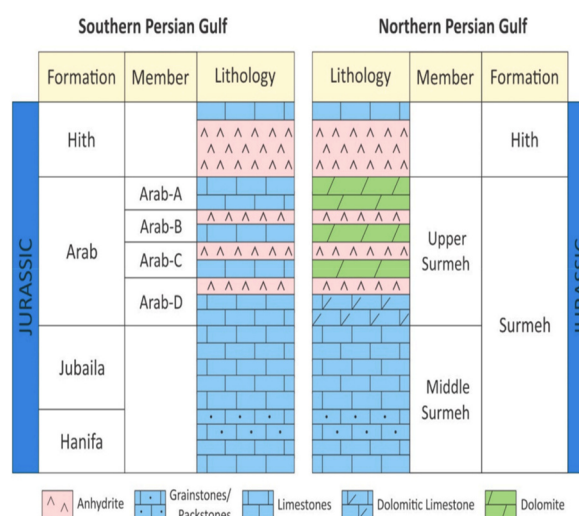**Figure 2.** Injection site location

Figure 3. Upper Jurassic stratigraphy of the Surmeh reservoir and its equivalent in the southern part of the Persian Gulf (Sfidari et al., 2021)

Three geological formations have been extensively studied for the purpose of storing CO_2: oil and gas reservoirs, deep saline formations, and non-mineable coal beds. In all three cases, the process of geological storage of CO_2 involves its injection in dense form into a subsurface rock formation. Rock formations that are porous and capable of holding, or that have previously held, fluids as natural gas, oil or brine, such as depleted oil and gas reservoirs, are considered suitable options for CO_2 storage. Suitable storage formations for CO_2 can be found in both onshore and offshore sedimentary basins. These basins are large-scale natural depressions in the earth's crust that are filled with sediments[8].

A thorough site characterization of a deep saline aquifer is crucial to assess its safety and long-term viability for effective geological storage of CO_2. Various techniques, including core analysis, well-logging analysis, and geological modeling, are employed to gather essential information and to form the basis for evaluation. Detailed analysis of sediment cores and interpretation of well-log data are valuable for understanding the vertical and lateral heterogeneity caused by changes in depositional environments within the framework of sequence stratigraphy. However, these data primarily provide information about geology and petrology near the wells. To reduce uncertainties in site characterization, it is essential to effectively incorporate seismic data, which greatly assist in building a geological model describing the reservoir architecture away from the wells. Using seismic data, a more complete understanding of the site can be achieved, extending beyond the immediate vicinity of the wells [9].

The Surmeh formation, as shown in Figure 3, was selected as a candidate for studying the feasibility of AGI. The upper part of the formation is primarily composed of dolomite, whereas the lower part consists mainly of carbonate rocks. The Surmeh formation has a total thickness of approximately 800 m, with both the upper and lower sections measuring around 400 m each. Based on the available log data (Figure 4), the porosity in the upper part of the formation is relatively higher (approximately 6%) and less permeable than the lower part (2.5%), where mud losses have been recorded due to the higher permeability.
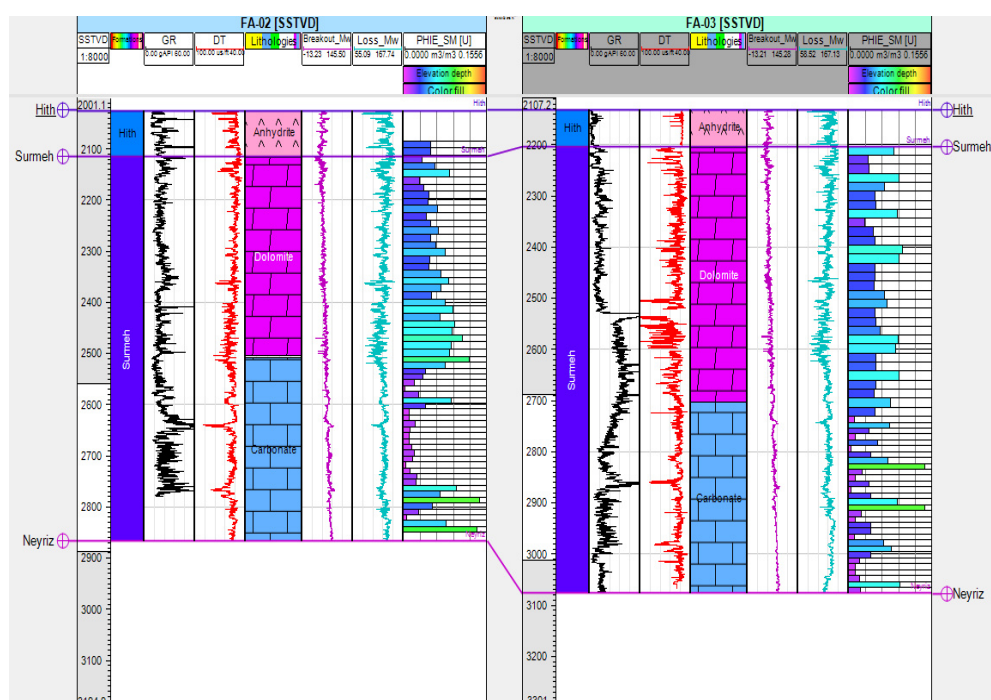
Figure 4. Schematic representation of Surmeh formation in two wells (FA_02 and FA_03 of the injection field)

As previously mentioned, due to the lack of available permeability data, an empirical correlation was used to incorporate permeability into the simulator for both carbonate and dolomite formations. Based on this correlation, the average permeability for the upper part of the Surmeh formation was estimated to be approximately 0.1 mD (Figure 5). The Arab formation in the Iranian zone corresponds to the Surmeh formation. It is noteworthy to be noticed that the Arab formation is fully saturated with water, commonly referred to as brine, which has a salinity level of 200,000 ppm. This high salinity is a feature of the formation's water content.
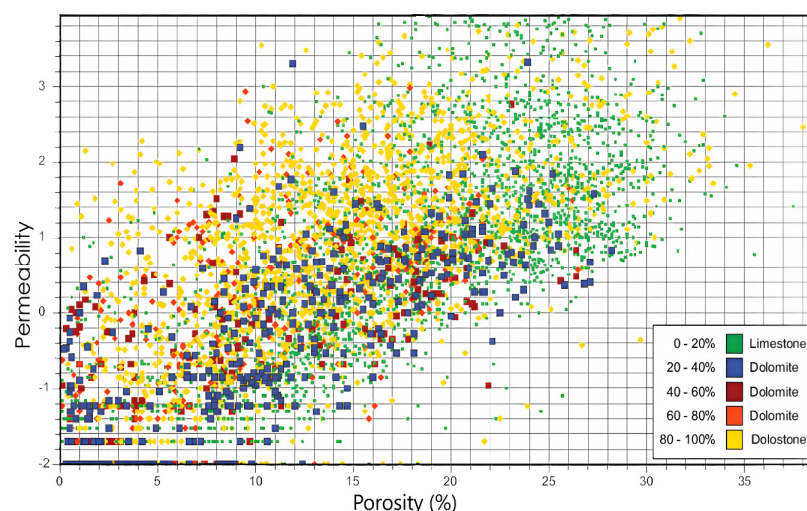

Figure 5. Porosity-permeability relation for the Arab formation (Surmeh)

During well logging in the Surmeh formation, recorded temperatures ranged from 75 to100°C. To validate these measurements, they were cross-checked with temperature survey data obtained from the Kangan-Upper Dalan reservoir, resulting in an estimated temperature in the Surmeh formation of approximately 80°C. Figure 4 illustrates the potential use of the Surmeh formation as a storage reservoir. The positioning of the Hith formation as the caprock in this context is crucial to the study. The caprock acts as a

barrier, preventing the migration and leakage of stored fluids, such as CO_2, from the reservoir. This highlights the importance of the Hith formation in ensuring the integrity and containment of the Surmeh formation as a proper storage site.

3. Geomechanical Study

When CO_2 is injected and stored in an underground geological structure, pore pressure buildup is inevitable. This change in pore pressure redistributes stress and induces poroelastic response in both caprock and target formation. In some cases, this can lead to geomechanical hazards, such as leakage of the injected CO_2, uplift of the surface, and induced seismic activity. These issues are significant environmental concerns during CCS projects. It is also important to consider the integrity of the well, as the injected CO_2 could potentially leak through any well component intended to serve as the expected flow path. Uncontrolled release of the injected fluid can shorten well life and increase the risk of CO_2 leaks. Therefore, establishing an optimal CCS design that takes into account geomechanical hazards is critical to ensure environmentally safe implementation of the design and gain public acceptance.

A comprehensive geomechanical study has been conducted in the field using both 1D and 3D approaches, incorporating all available data. This includes information such as formation tops, drilling and completion reports, location maps, graphic well logs, final geological reports, compressional and shear slowness data, open hole logs (including measurements such as gamma ray, density, neutron porosity, and resistivity), static formation pressure data from MDT/XPT (Modular formation Dynamics Tester/ Express Pressure Tool) tools especially in the reservoir section, caliper logs, and core data. These different data sources were used to perform a detailed analysis of the geomechanical properties and behavior of the field [7,10–14].

The geomechanical model was employed to estimate pore and fracture pressures in the Surmeh formation (Figure 6). Drilling data were used to estimate the pore pressure specifically in this formation. Drilling data indicates that the lower portion of the Surmeh formation exhibits higher permeability, as evidenced by the mud loss data. However, the porosity in this area is very low, suggesting a higher degree of fracturing. Using the Geomechanical Earth Modeling (GEM) approach, the estimated pore pressure in the Surmeh formation is 3800 psi, while the estimated fracture pressure is 6100 psi.
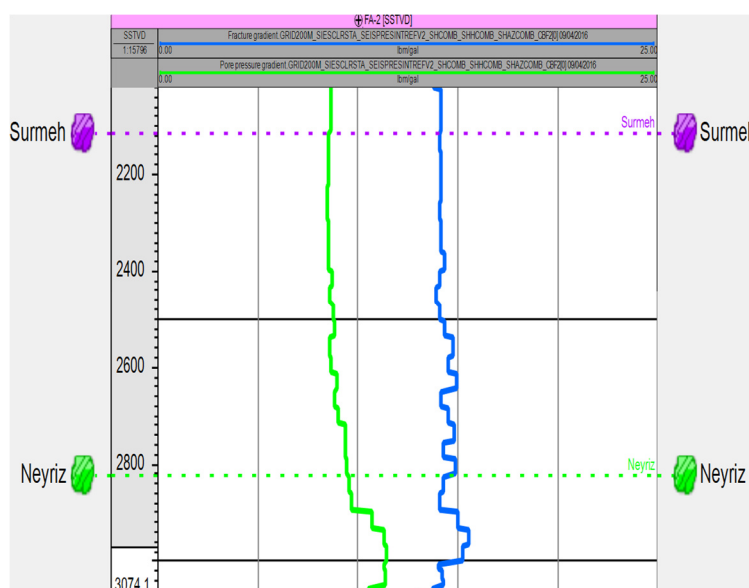

Figure 6. Estimated pore and fracture pressures in FA-02

Seismic thickness mapping revealed that the integrity of the Hith Formation caprock varies at the desired location for injection well drilling (Figure 7). The thickness ranges from approximately 60 m to almost 110 m. This information is essential in assessing the effectiveness of the caprock as a barrier for containing the injected fluids within the Surmeh formation. Variations in caprock thickness can affect the overall containment and long-term stability of the CO_2 storage operation.
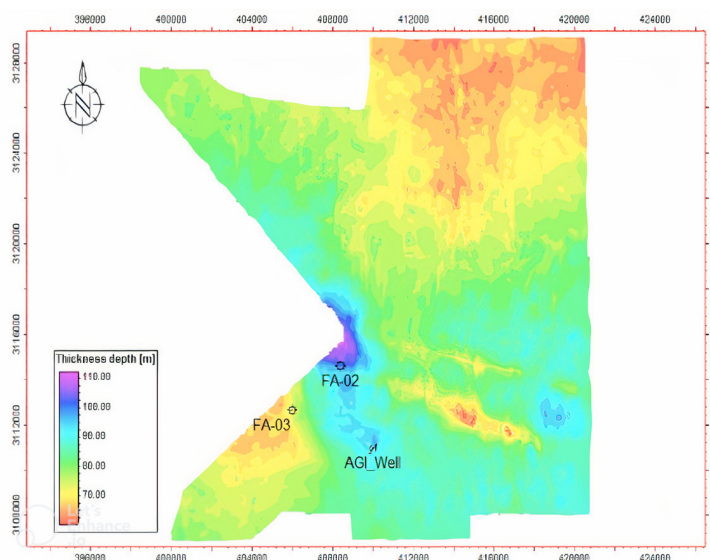

Figure 7. Thickness map of the Hith formation (Surmeh Cap rock)

Table 1 shows the properties that predominantly constitute the static model.

Table 1. Static properties of Surmeh formation in upper and lower parts.

Property	Description
Reservoir Formation	Surmeh Formation (corresponds to Arab Formation)
Porosity (%)	Range: 2.5 - 6
Permeability (mD)	Range: 0.1 - 100
Formation Water	Fully saturated with brine (220,000 ppm salinity)
Temperature (°C)	80
Pore Pressure (psi)	3,800
Fracture Pressure (psi)	6,100

4. Dynamic Modelling

The Plan is to transfer a sour gas that has been extracted from one field (source field) and injected into the Surmeh Formation. As mentioned above, acid gas consists mainly of H_2S and CO_2. According to the PVT (Standard Pressure Volume Temerature) data, the H_2S concentration in the source field is 39,000 ppm, while the CO_2 concentration is 68,000 ppm (see Table 2).

Considering the extraction of all H2S, CO2, and a small portion of methane during the acid gas processing, approximately 12% of the injected gas is acid gas. Based on the MDP (Master Development plan) of source field, the gas estimated in place is 22 TCF of sour gas. Consequently, roughly 2 TCF of acid gas will be injected into the Surmeh formation, considering a recovery factor of 75% for dry gas reservoirs.

[15] expresses the capillary pressure as a function of saturation (S) using the following equation [1]:

$$P_c = P_e \left(\frac{S_w - S_{wi}}{1 - S_{wi}}\right)^{-\frac{1}{\lambda}}$$ [1]

where S is water saturation, P_e is the minimum pressure required for the gas to enter the pores of the rock and λ is a fit parameter known as the pore size distribution index. For this formation, λ was 0.55.

Table 2. injected Fluid composition.

Components	Reservoir Fluid Composition (Dry Basis)	
	Mole %	Mass %
Nitrogen	11.797	15.657
CO_2	6.795	14.167
H_2S	3.908	6.309
Methane	73.861	56.139
Ethane	2.330	3.319
Propane	0.526	1.100

The relative permeability curves employed in this study were based on the relationships of [16], specifically using the equations [2] - [4] proposed by [17]:

$$K_{rw} = (S_w^*)^{N_w}$$ [2]

$$S_w^* = \frac{S_w - S_{wi}}{1 - S_{wi}}$$ [3]

$$K_{rg} = K_{rg}(S_{wi})(1 - S_w^*)^2(1 - (S_w^*)^{N_{gas}})$$ [4]

The variables N_w and N_{gas} are used as fitting parameters, known as the Corey exponents for water and gas, respectively. The following values are used, $N_w = 5$ and $N_{gas} = 4$.

Table 3 shows a summary of the saturation function data.

Upper Surmeh	Porosity (%)	Irreducible Water Saturation Swi (%)	Gas Saturation Sgc (%)	Minimum pressure Pe (psi)	Gas relative permeability Krg
	6	22	4	660	0.65

Table 3. Saturation function data for acid gas simulation

Figure 8 andFigure **9** show the relative permeability and capillary pressure curves, respectively.
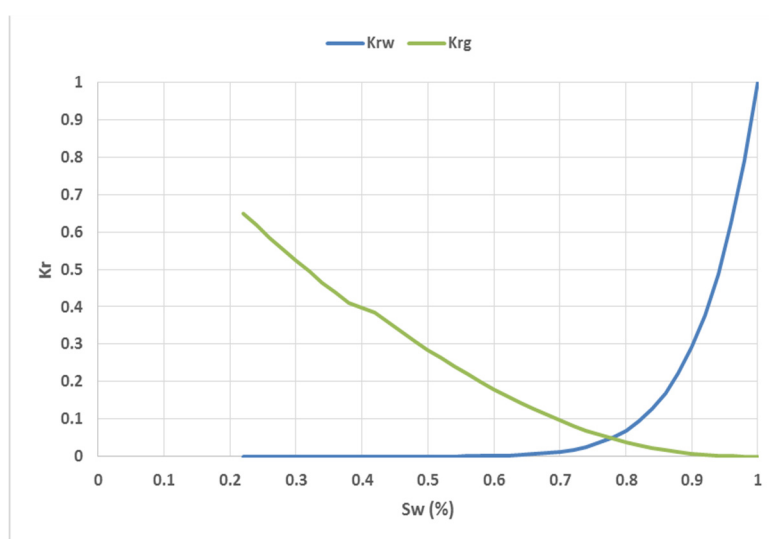
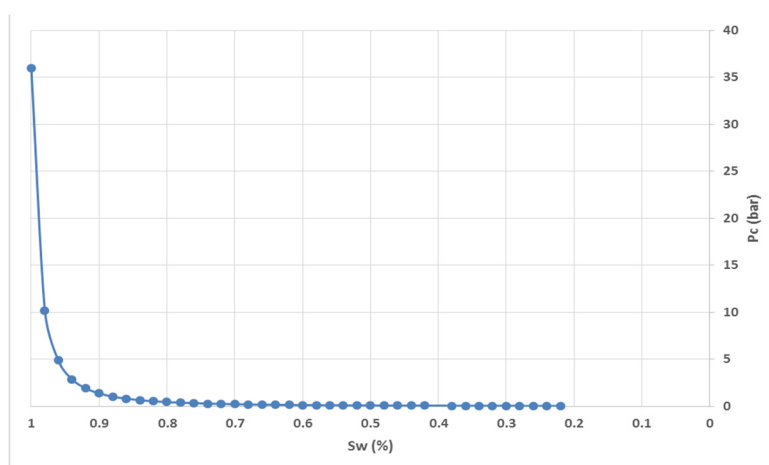**Figure 8.** Relative permeability curves used for water and gas

Figure 9. Capillary pressure of water and gas

In the context of gas injection scenarios for storage purposes, a comprehensive understanding of the four identified trapping mechanisms is crucial: structural, capillary, solubility, and mineralization trapping. These mechanisms are responsible for the prolonged storage of injected gas within geological formations. The relative contribution of each trapping mechanism during the storage period may vary. In the short term, capillary trapping serves as the primary mechanism for gas retention in porous media. This phenomenon can be attributed to the hysteresis of relative permeability and capillary pressure. Capillary trapping plays a pivotal role in the initial stages of CO_2 storage, effectively containing a significant portion of the gas plume within the formation. Furthermore, capillary pinning is expected to occur because of contrasting constitutive relations among different rock types, such as variations in capillary pressure (P_c), relative permeability (K_r), and irreducible water saturation (S_{wi}), which are commonly observed in natural rock formations. Consequently, capillary trapping exerts a considerable influence on the spatial distribution of gas plumes within the reservoirs [18].

The capillary trapping mechanism can be clarified as follows: once the gas is injected into a saline aquifer, the significant density and viscosity disparities between the gas and water phases cause the gas plume to ascend towards the highest accessible formation top until it encounters an impermeable cap rock layer. In this phase a drainage process takes place, during which the non-wet gaseous phase displaces the wet phase, which in this case is saline water. Upon completion of the injection, the saline water reabsorbs into the

formation, displacing the gas plume in a process like imbibition. During this process, the 261
continuous gas plume fragments, forming bubbles which subsequently become trapped 262
within the formation [19]. 263

To simulate the trapping mechanism from a conceptual perspective, a simple struc- 264
ture with dimensions of 100x100x10 m is used. To replicate this phenomenon in a simula- 265
tion, two methods are employed: one assumes homogeneous rock properties with a single 266
rock type, employing single relative permeability and capillary pressure curves without 267
incorporating a hysteresis process; the other considers heterogeneous rock properties with 268
two different rock types, defining the single relative permeability and capillary pressure 269
curves, and hypothesizing a hysteresis process for both curves. 270

Figures 10 and Figure **11** show vertical cross-sections of gas saturation for the homo- 271
geneous and heterogeneous cases after the injection is stopped. Once injected, the CO_2 272
disperses both horizontally and vertically. The buoyancy force drives the upward move- 273
ment of the gas plume. The relative strengths of the viscous and gravitational forces de- 274
termine the shape of the plume uniformly. In a homogeneous reservoir, the gas plume 275
rapidly rises to the top of the reservoir and then expands laterally at a slower rate. 276
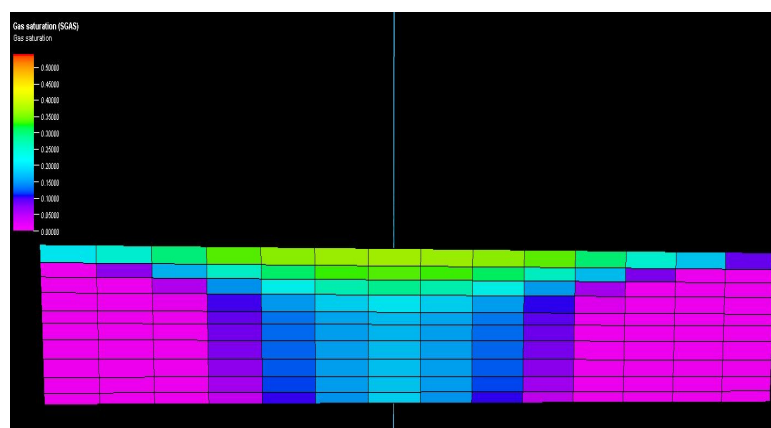

277
Figure 10. Vertical cross section of reservoirs showing gas saturation in homogenous model. 278

279
Figure 11. Vertical cross section of reservoirs showing gas saturation in heterogenous model. 280

On the other hand, under heterogeneous conditions, the gas plume exhibits distinct 281
characteristics. It becomes twice as wide as in the homogeneous case, and only a portion 282
of the gas manages to reach the top of the reservoir. Furthermore, the distribution of gas 283
saturation within the reservoir becomes non-uniform. These results highlight the signifi- 284
cant influence of capillary curves and hysteresis processes associated with different rock 285
types, which fundamentally dictate trapping mechanisms and consequently shape and 286
govern gas plume dynamics. 287

Extensive research has been carried out on CCS, resulting in the availability of relevant data. In this case, the acid gas composition is primarily composed of CO_2 rather than H_2S. Therefore, only the CO_2 composition was considered for simulation purposes. The solubility of CO_2 in water was determined thanks to the experimental data from [20]. Figure 12 shows the behavior of CO_2 in presence of water, when increasing the pressure. This information serves as a crucial input for the simulation processes.
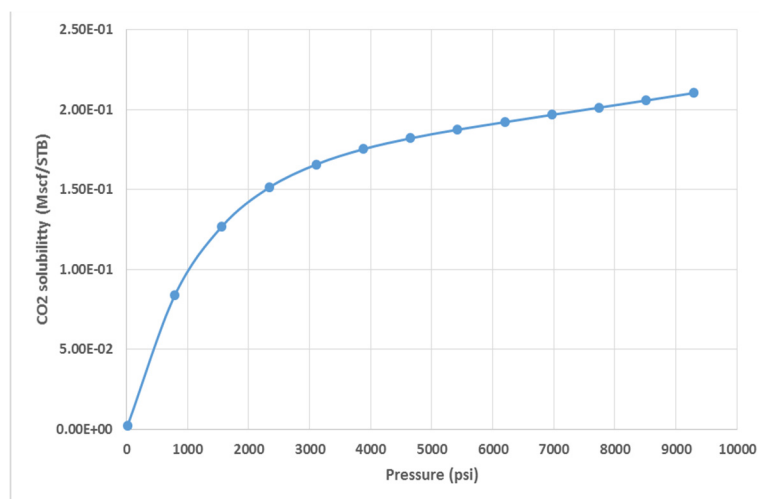

Figure 12. CO_2 solubility in water (T = 80 oC).

An important remark is that both CO_2 and H_2S exhibit similar phase equilibria. The critical points, as regards CO_2, are at T=31.1°C and P=1070 psi , and for H_2S at T=100.2°C and P=1300 psi [21]. It is worth noting that CO_2 can form hydrates at temperatures up to 10°C and H_2S above 30°C, even if there is no free water. Due to the lack of experimental data, a decision was taken to consider and simulate only one component, which led to the selection of CO_2. This choice was due to the extensive research that was conducted on the behavior and properties of CO_2 in various studies [15,18,20].

In the project plan, about 2 TCF of acid gas should be injected from the source field into the Surmeh formation. The source field consists of three production platforms, each with a capacity of 500 MMSCFD. Therefore, the total daily production from these platforms will be 1500 MMSCFD, with 12% of this gas being acid. This results in an average daily acid gas rate of 180 MMSCFD.

Considering this acid gas rate and taking into account factors such as fluctuating flow rates, it is estimated that it will take about 30-35 years to inject the 2 TCF of acid gas into the formation. Simulation models were run over a 100-year lifetime to observe pressure changes throughout the injection period. This long simulation period allows a complete understanding of pressure dynamics throughout the injection process.

5. Results

Two different cases were examined, respectively with a permeability of 1 and 0,1 mD. In the following, the results obtained are highlighted.

Permeability = 1 mD: To assess the gas injection capacity into the reservoir, a sensitivity analysis was conducted based on empirical correlations. These correlations show that for a porosity of 6%, the typical range of permeability falls between 0.1 and 1 mD (see Figure 5). By varying the permeability values within this range, the study aimed to determine the maximum gas injection volume that could be accommodated by the reservoir.

When performing dynamic modeling, it is important to consider the following factors:

- Vertical permeability: in all models, the vertical permeability (Ky) was assumed to be 0.1 times the horizontal permeability (KH).

- Chemical reactions: none of the simulation cases accounted for chemical reactions between rock and fluids. It is necessary to conduct precise laboratory tests to evaluate such reactions accurately.

- Wellbore diameter: in all simulation cases it was assumed equal to19 cm, which is the default value in the software.

- Simulation area: all simulations were conducted in the upper Surmeh reservoir, due to its higher porosity value, which is a crucial factor for accurate modeling.

Based on the estimations made, the fracture pressure of the rock is 6100 psi, which is equivalent to 420 bars. Therefore, it is important to note that as the pressure approaches 400 bar (with a safety margin of 20 bars), the risk of rock fracturing arises.

In the first model, the horizontal permeability was 1 mD, and the vertical one was 0.1 mD. Figure 13 shows the trend of the bottom-hole pressure with an injection rate of 200,000 Sm^3/d or 7 MMSCFD.
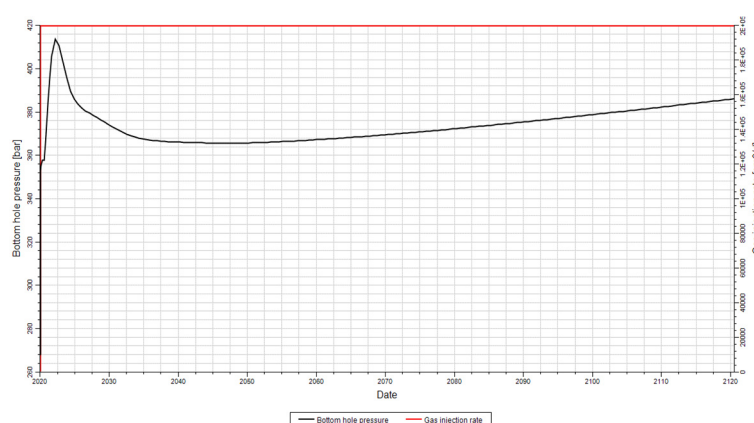

Figure 13. Trend of the Well bottom-hole pressure according to a gas injection rate of 200,000 Sm^3/d

Figure 13 depicts the well's bottom-hole pressure profile for the injector in the implemented model. The pressure initially started at 270 bar and reached a maximum of 386 bar over the 100-year injection period. This rapid pressure rise is influenced by the compressibility of both fluids and rock, and it depends on the type of well and the length of completion.

As more gas is injected, the bottom-hole pressure gradually increases due to the increasing gas saturation. The initial sharp spike in the pressure is caused by the low relative permeability of the gas at the beginning of the injection process [22].

The pressure changes at the well affect the pressure at the cap rock, which is crucial for maintaining the seal integrity. The slow, long-term increase in the well's bottom-hole pressure shown in Figure 13 is a result of the net accumulation of fluid within the formation.

It was anticipated that injecting a flow rate of 300,000 Sm^3/d could potentially cause fractures in the formation, given its low permeability. The upper limit for this permeability is determined to be 200,000 Sm^3/d.

As observed, the reservoir permeability is quite low to accommodate an injection rate of 180 MMSCFD. Therefore, achieving high well injectivity becomes crucial to enable the injection of such a flow rate. However, it is important to ensure that the bottom-hole pressure does not exceed 400 bar to maintain the integrity of the reservoir.

According to the information provided, the initial bottom-hole pressure of 270 bar increases to 386 bar over a span of 100 years. Considering the typical Vertical Flow Performance curves, the corresponding pressure drop along the wellbore would be estimated to be between 1000-1300 psi (68-90 bar). Consequently, the resulting wellhead pressure is projected to be within the range of 352-475 bars.

Permeability = 0.1 mD: Based on the simulation results for a permeability of 1 mD, it can be concluded that lower permeability tends to result in reduced injectivity. In this case, to prevent formation damage, the simulation suggests that an injection rate of 20,000 Sm^3/d (0.7 MMSCF/D) can be safely maintained. This rate is determined to be within the acceptable range for injection without causing the formation to fracture or break.

6. Ideal Reservoir

In 2003, Gas Liquids Engineering (GLE) conducted a conceptual design study to explore the possibility of acid gas injection at Kharg Island. Recently, in the current year, GLE has initiated a Front-End Engineering Design (FEED) study for an acid gas injection facility to replace the existing sulfur plant [23]. These studies are crucial for evaluating the feasibility and potential of implementing acid gas injection in the reservoirs at Kharg Island and further assessing the suitability of the reservoirs for this process.

Based on the available data from the formation core and logs [23], the primary reservoir properties in the Dhruma Zone are as follows:

- Depth range: 4016-4150 m

- Net Pay: Approximately 125 m

- Typical porosity of the injection zone: 15-23%

- Typical permeability of the injection zone: 10-600 mD

- Proposed injection rate: 80 MMSCFD

As can be seen, the permeability of the reservoir is relatively high, and the formation is highly porous (15% porosity). By considering a permeability of 100 mD and a porosity of 6%, the results are shown Figure 14. It should be emphasized that a porosity of 6% in rocks generally results in a permeability of less than 100 mD: this value is only considered for the conceptual feasibility of acid gas injection. Under such conditions, the bottom-hole pressure increases over time. After 25 years it reaches 400 bars, which coincides with the fracture pressure. Throughout this period, the injection rate remains constant at 2,000,000 Sm3/d (70 MMSCFD).

In addition to black oil simulation, a compositional model was used to validate the results. This involved considering two components, CO_2 and H_2O. By comparing the results obtained from the compositional model with those from the black oil simulation, the accuracy and reliability of the findings were assessed. This approach allows a more comprehensive understanding of the behavior and performance of the reservoir during acid gas injection.
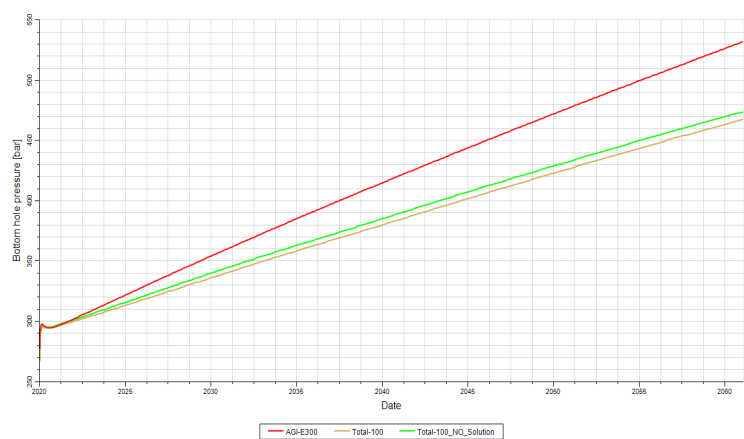
Figure 14. BHP for 3 cases (Red: Compositional; Green: Black oil without CO_2 Solubility; Brown: Black oil with CO_2 solubility).

Figure 14 shows two black oil cases and one compositional case: it is evident that the injection rate involves a high risk of fracturing and, over a period of 30 years of injection, there is a possibility of formation breakdown. In the black oil cases, two scenarios were examined, and the descriptions of both of them are shown in Table 4.

Table 4. Characteristics of the 3 cases examined (Porosity: 6%, Permeability = 100 mD).

Simulator	Differences	Time of BHP limit (400 bars)
Black Oil	CO_2 solubility was considered	24 years
Black Oil	CO_2 solubility was ignored	23 years
Compositional	interaction between various hydrocarbon phases considered	17 years

Figure 15 and Figure **16** display the gas saturation distribution within the reservoir: they show how the gas fluid rapidly moves upward, driven by the buoyancy force, and accumulates in the crest of the structure. As a result, a gas cap is formed over the water zone in the formation. The visualization of gas saturation provides insights into the spatial distribution and movement of gas within the reservoir.
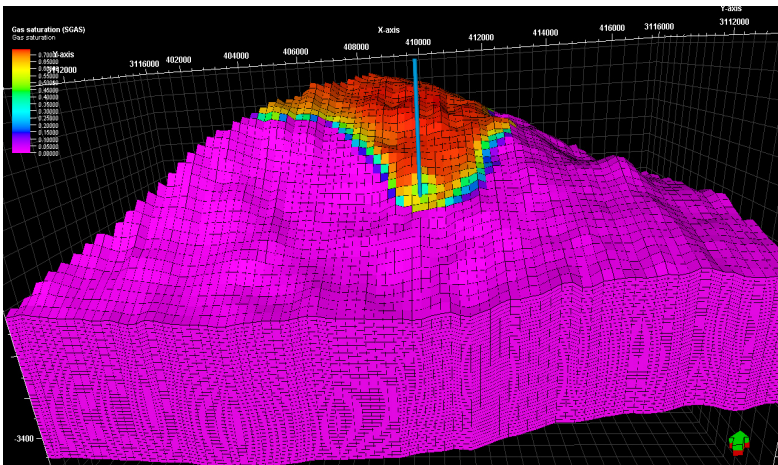

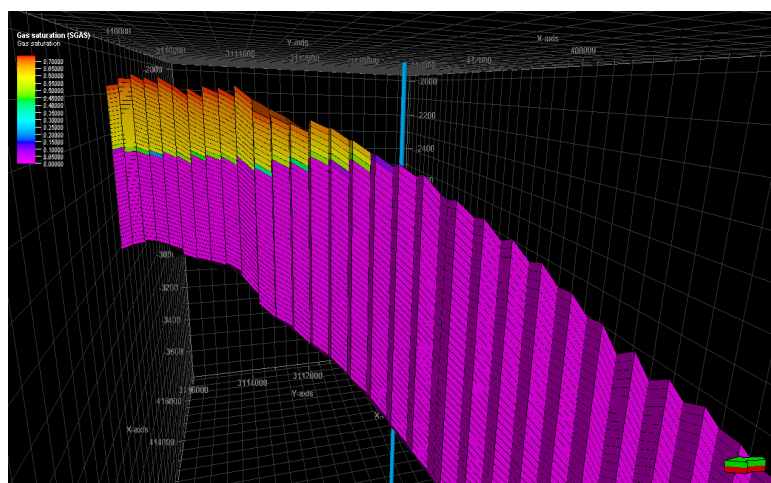**Figure 15.** Illustration of gas saturation after 50 years.

Figure 16. Secondary gas cap after acid gas injection.

The formation's porosity of 6% and permeability of 100 mD restrict the injection rate, allowing a maximum injection of 70 MMSCFD. To maintain the pressure of the reservoir and reduce the pressure of the bottom-hole three injection wells are needed. The low porosity of the reservoir significantly affects the increase in reservoir pressure and BHP.

Figure 17 shows a comparison between the bottom-hole pressures obtained for two different cases, based on the porosity values of 6% and 12%. The Arab formation (Surmeh), depicted in Figure 5, exhibits a porosity range of 10-15% and a permeability range of 1-100 mD, highlighting its high heterogeneity. Considering a porosity of 12%, higher than 6%, BHP is expected to differ between the two cases.
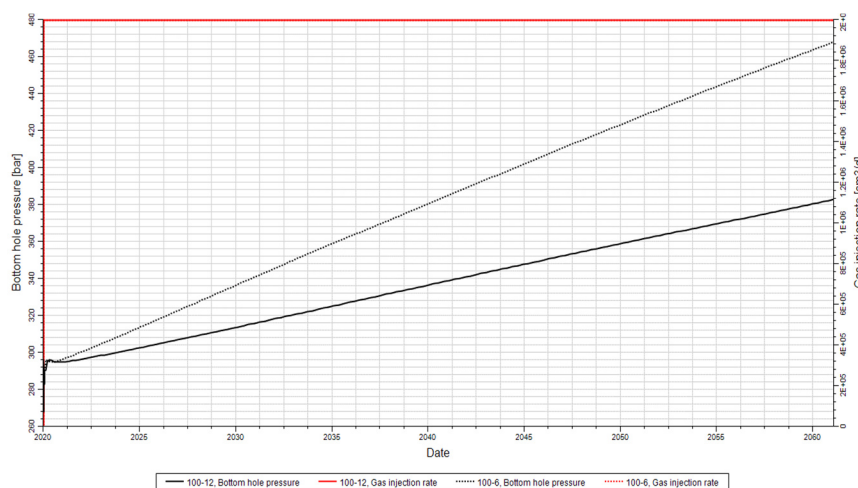

Figure 17. BHP for the same case (Surmeh formation) with porosity values of 6% and 12% and permeability of 100 md

Based on the information provided, the porosity of 12% resulted in a reduction of the bottom-hole pressure over a period of 50 years of injection. In this case, the BHP does not exceed 400 bars, indicating that higher flow rates, potentially exceeding 2,000,000 Sm^3/d, could be injected.

On the contrary, for the compositional case, the BHP reaches 400 bar after approximately 35 years. This suggests that the compositional characteristics of the reservoir have a different impact on the pressure behavior than the case of higher porosity mentioned earlier (Figure 18).
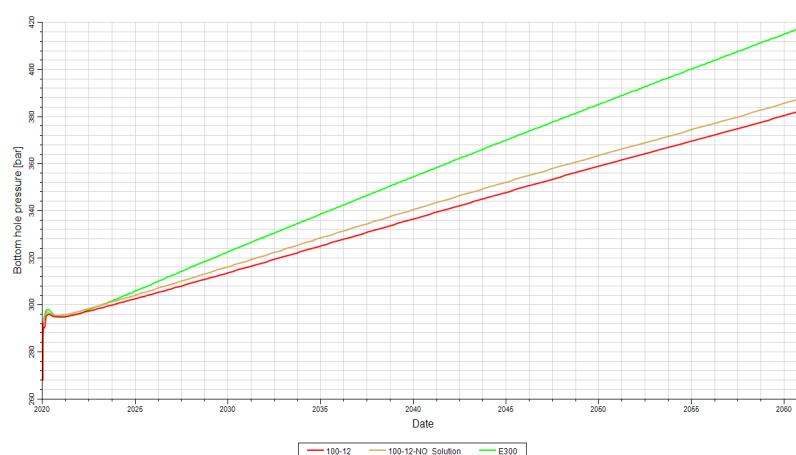
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Figure 18. BHP for 3 cases with an injection rate of 70 MMSCFD and porosity of 12%. (Red: 438
Compositional; Green: Black oil without CO_2 Solubility; Brown: Black oil with CO_2 solubility). 439

7. Main uncertainties

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Based on the simulation results and analysis, several crucial reservoir properties and 441
factors have been identified as main uncertainties for the Surmeh formation. These in- 442
clude: 443

1. Porosity and permeability: they play a significant role in controlling fluid 444
flow and pressure behavior. Accurate analysis of these properties is crucial 445
and can be obtained through coring and laboratory testing. 446
2. Formation temperature: it affects the solubility of gas in water and impacts 447
the behavior of fluids within the reservoir. Understanding the formation 448
temperature is important for accurate modeling and prediction. 449
3. Injectivity and fall-off test: conducting injectivity tests and fall-off tests pro- 450
vide valuable information about the reservoir's ability to accept injected flu- 451
ids and the behavior of pressure response. These tests help to determine the 452
formation injectivity and generate Vertical Flow Performance (VFP) curves, 453
necessary for compressor design and calculation of well-head pressure. 454
4. Relative permeability and capillary pressure curves: understanding relative 455
permeability and capillary pressure relationships is essential for accurate res- 456
ervoir modeling and simulation. These curves provide insights on multi- 457
phase flow behavior and fluid displacement within the reservoir. 458
5. Fracture pressure: Determining the fracture pressure of the reservoir is im- 459
portant for well design and drilling operations. It helps to ensure that the 460
pressure exerted during the injection or production operations does not ex- 461
ceed the integrity of the reservoir. 462
6. Cap rock integrity: it acts as a seal for the reservoir and is crucial to prevent 463
fluid migration and to maintain the reservoir pressure. Analyzing the cap 464
rock's integrity helps assess the risk of potential leaks or breaches. 465
7. Gas solubility in water: the temperature of the reservoir influences the solu- 466
bility of gas in water. Understanding the gas solubility is vital for accurately 467
modeling gas-water interactions and predicting fluid behavior during injec- 468
tion and production processes. 469

Addressing and reducing these uncertainties through comprehensive analysis and 470
testing will improve the understanding of the Surmeh formation and enhance decision- 471
making in reservoir management and development strategies. 472

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8. Conclusions

The Surmeh formation is characterized by its tightness and exhibits very low porosity and permeability, showing a remarkably low reservoir quality. With a permeability of only 1 mD, the maximum achievable injection rate is limited to 7 MMSCFD, while the total required injection rate is significantly higher, equal to180 MMSCFD. The nature of this formation restricts the injection of gas at high flow rates, highlighting the importance of considering structures with higher porosity and permeability values for a more efficient injection. An ideal scenario would involve porosity of 12% and permeability of 100 mD, allowing high flow gas injection without fracturing the formation (reaching 400 bars). To achieve this, it would be necessary to have two injection wells, each with a capacity of 90 MMSCFD.

Reservoir rock typing plays a crucial role in understanding capillary trapping phenomena. In this scientific paper, two simulation models are analyzed. The first mode explores the homogeneous scenario, where the gas plume quickly reaches the top of the reservoir and spreads laterally. The second mode focuses on the heterogeneous case, revealing a wider plume with non-uniform gas saturation distribution, as only a fraction of the gas reaches the top. These findings underscore the significant impact of diverse rock types and hysteresis in capillary curves on trapping processes and the dynamic behavior of the gas plume.

In summary, the Surmeh formation's reservoir properties, including its low porosity and permeability, pose challenges for gas injection operations. Structures with higher porosity and permeability are better suited for efficient injection: a porosity of 12% and permeability of 100 mD are considered ideal for achieving high flow rates without fracturing the formation.

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