

Spray Freeze-Drying for Inhalable L-leucine, Mannitol-based Microparticles: The Impact of Process Variables, L-leucine, and Crystallinity on Aerosolization Properties

Lorena Pasero¹ (lorena.pasero@polito.it), Adamo Sulpizi², Tomaso Guidi², Roberto Pisano¹ (roberto.pisano@polito.it)

¹ Department of Applied Science and Technology, Politecnico di Torino, 24 corso Duca degli Abruzzi, 10129 Torino, Italy
² Chiesi Farmaceutici S.p.A, R&D Department, Largo F. Belloli 11/A, 43122 Parma, Italy

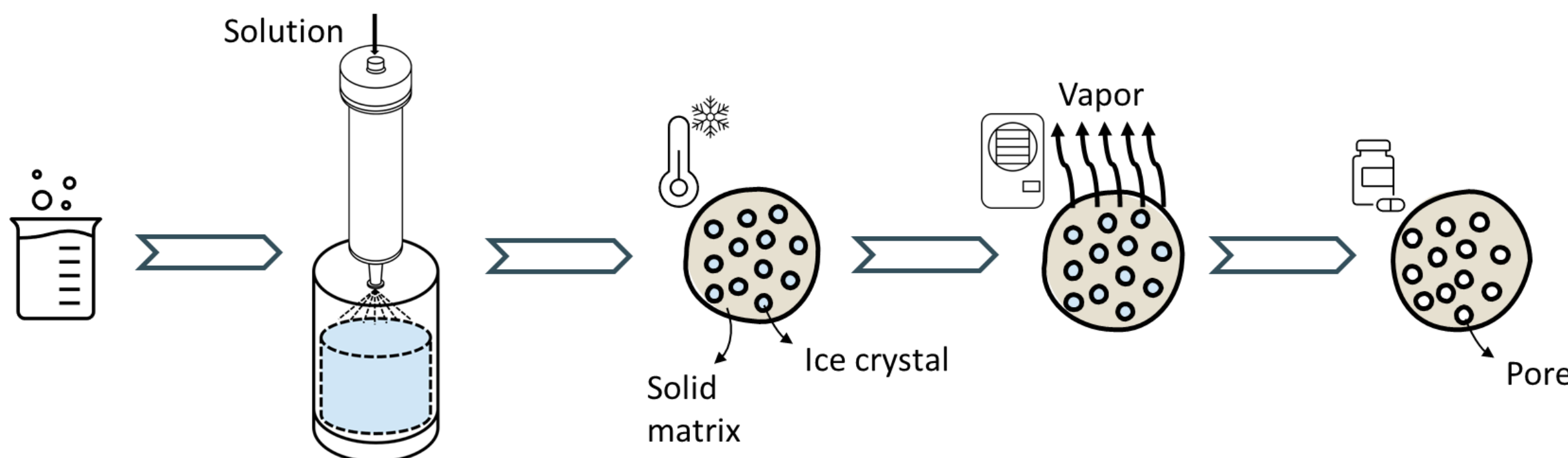


Politecnico di Torino



Introduction

Spray freeze-drying (SFD) has emerged as a cutting-edge technology for the manufacturing of temperature-sensitive pharmaceuticals. SFD involves the atomisation of a solution into droplets, which are instantaneously frozen into a cryogenic liquid (e.g., N₂) and then dried under vacuum [1]. The formation and subsequent sublimation of ice crystals provide the MPs with a porous structure, constituted by an excipient-based matrix embedding the drug [2]. The high porosity of such MPs reduces their mass density, thus making them extremely aerodynamically performant [3]. Therefore, SFD represents a promising approach to producing drugs to be administered through dry powder inhalers, which require excellent aerodynamic properties to deposit in the target site of the lung and exert their action [4].

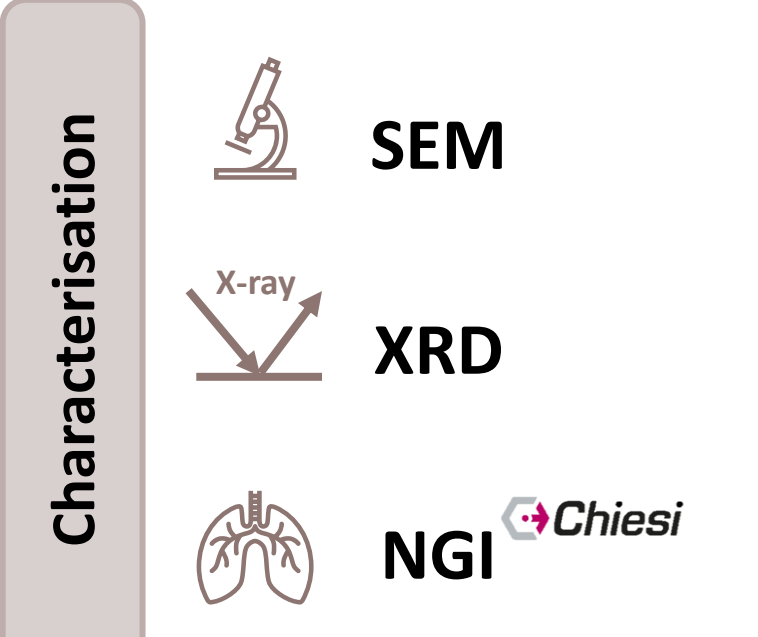


Owing to their large surface area, spray freeze-dried MPs are exposed to inter-particle cohesiveness which can affect their flowability [5]. Moreover, these powders are extremely sensitive to humidity-induced deterioration due to their great hygroscopicity. In this study, LL was employed to increase the MPs' flowability, assessing the LL optimal content to reach the highest FPF. In addition, the relationship between the crystallinity of mannitol and LL MPs and their aerodynamic behaviour was uncovered, providing further information about the mechanism of action of this amino acid.

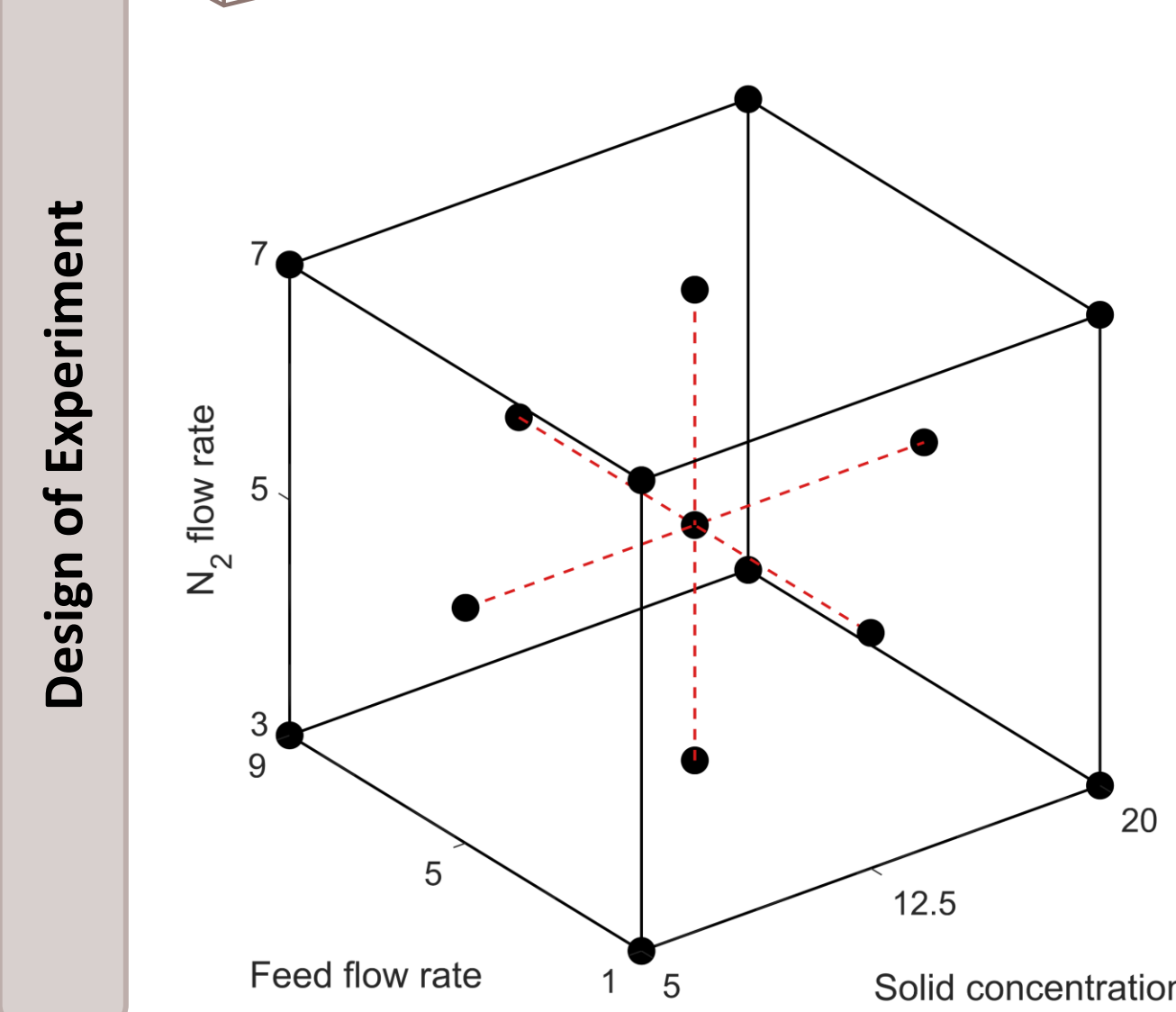
Methods

Formulation	Solid (% w/v)	Mannitol (% w/w _{dw})	SS (% w/w _{dw})	LL (% w/w _{dw})
F2	5, 12.5, 20	99	1	-
F3	5	94	1	5
F4	5	89	1	10
F5	5	79	1	20

- Primary drying: 10 °C, 20 Pa
- Secondary drying: 20 °C, 20 Pa, 5 h



Design of Experiment



Response variables:

- y₁ = geometric diameter = d_g
- y₂ = aerodynamic diameter = d_{ae}

Model:

$$y_u = b_0 + \sum_{i=1}^k b_i x_i + \sum_{i,j=1}^k b_{ij} x_i x_j + \sum_{i=1}^k b_{ii} x_i^2 \quad u = 1,2$$

Level	Concentration (% w/v)	N ₂ flow rate (NL/min)	Feed flow rate (mL/min)
-1	5	3	1
0	12.5	5	5
1	20	7	9
Variable	x ₁	x ₂	x ₃

Mannitol + SS MPs

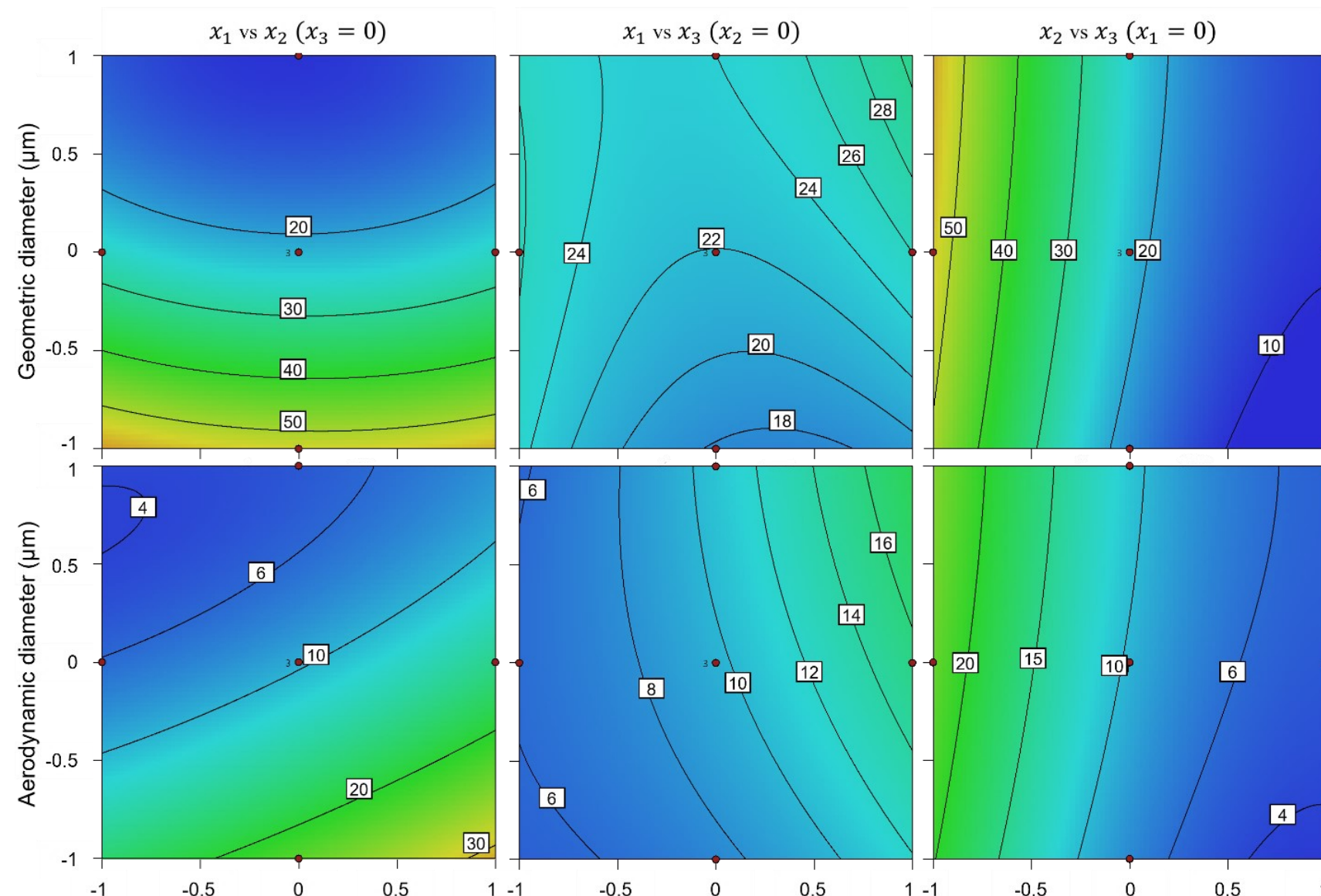
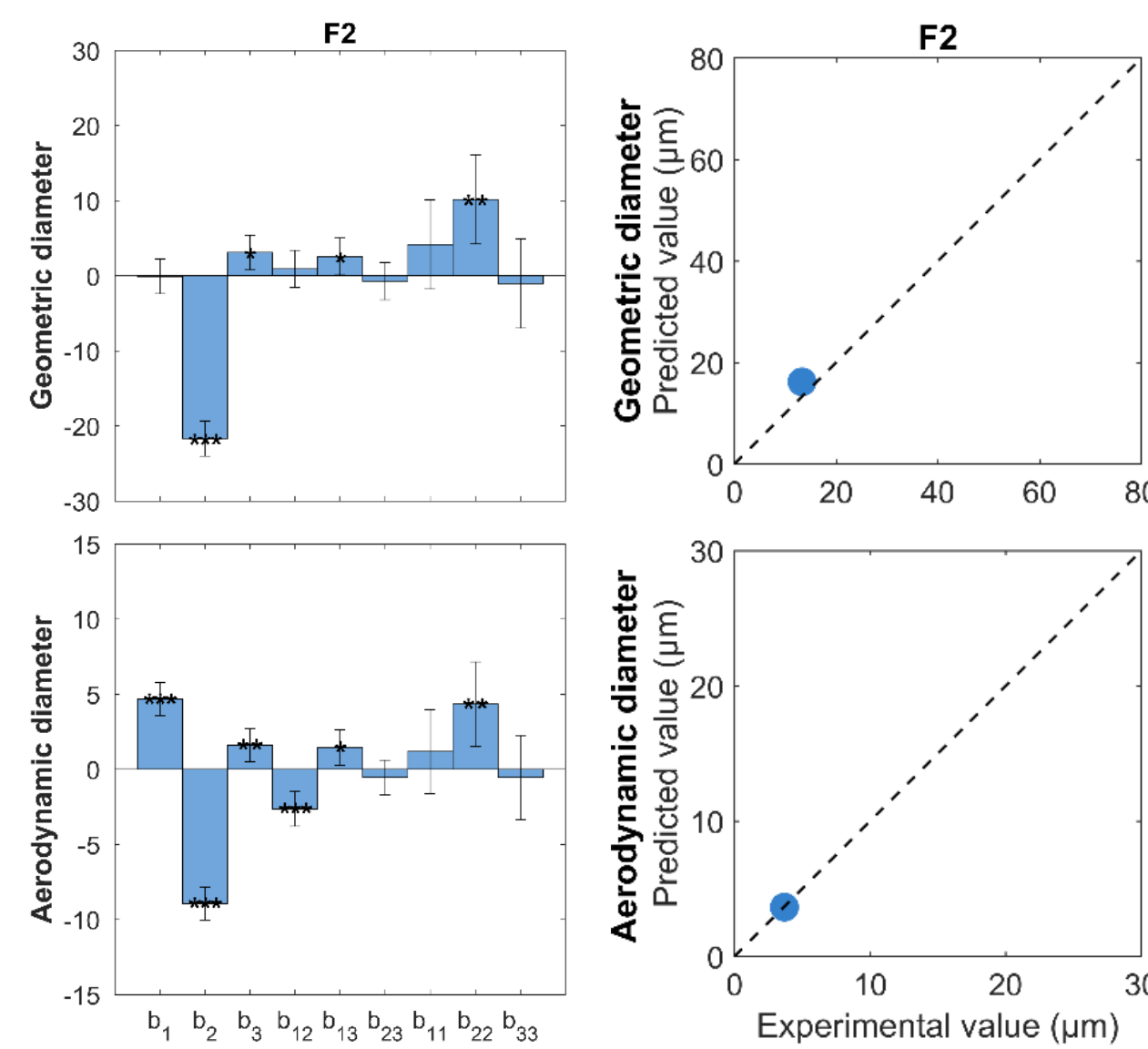
Face centered design

The following model was obtained:

- y₁ = 21.9 - 0.09x₁ - 21.7x₂ + 3.1x₃ + 0.9x₁x₂ + 2.6x₁x₃ - 0.7x₂x₃ + 4.2x₁² + 10.2x₂² - 1.04x₃²
- y₂ = 9.6 + 4.7x₁ - 8.9x₂ + 1.6x₃ - 2.6x₂ + 1.4x₁x₃ - 0.55x₂x₃ + 1.2x₁² + 4.3x₂² - 0.56x₃²

The model was validated:

- excellent agreement between experimental and predicted values;
- model predicting the size of MPs in the given domain.



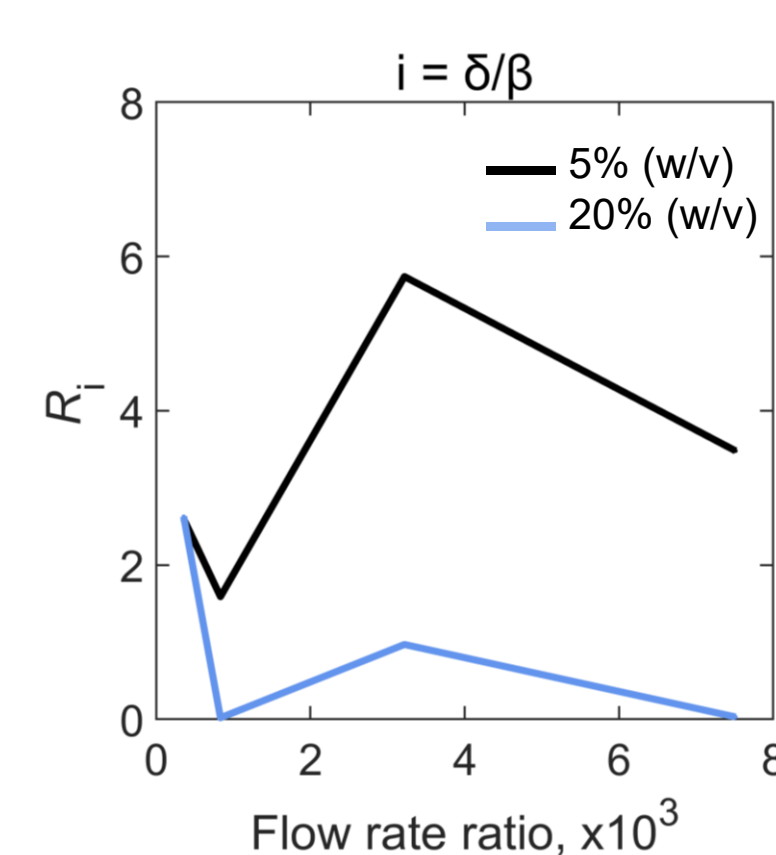
Cristallinity

- Prevalence of β- and δ-mannitol
- Relationship between size and polymorphism

Which is the cause?

- Only small droplets with nucleation time lower than their falling time could have nucleated in the vapour phase;
- the lower driving force of vapour N₂ could have slowed the freezing process, guaranteeing enough time for the formation of β-mannitol (Ostwald rule [6]);
- bigger droplets nucleated in the liquid phase at a higher freezing rate, promoting the formation of metastable δ-mannitol.

↑ size and ↑ R_{δ/β} at x₂ = 3 NL/min
↓ size and ↓ R_{δ/β} at x₂ = 7 NL/min



Which factors were significant?

- Geometric diameter:
 - ↑ x₂ → ↓ d_g (p < 0.001)
 - ↑ x₃ → ↑ d_g (p < 0.05)
- Aerodynamic diameter:
 - ↑ x₁ → ↑ d_{ae} (p < 0.001)
 - ↑ x₂ → ↓ d_{ae} (p < 0.001)
 - ↑ x₃ → ↑ d_{ae} (p < 0.01)

Which was the best combination?

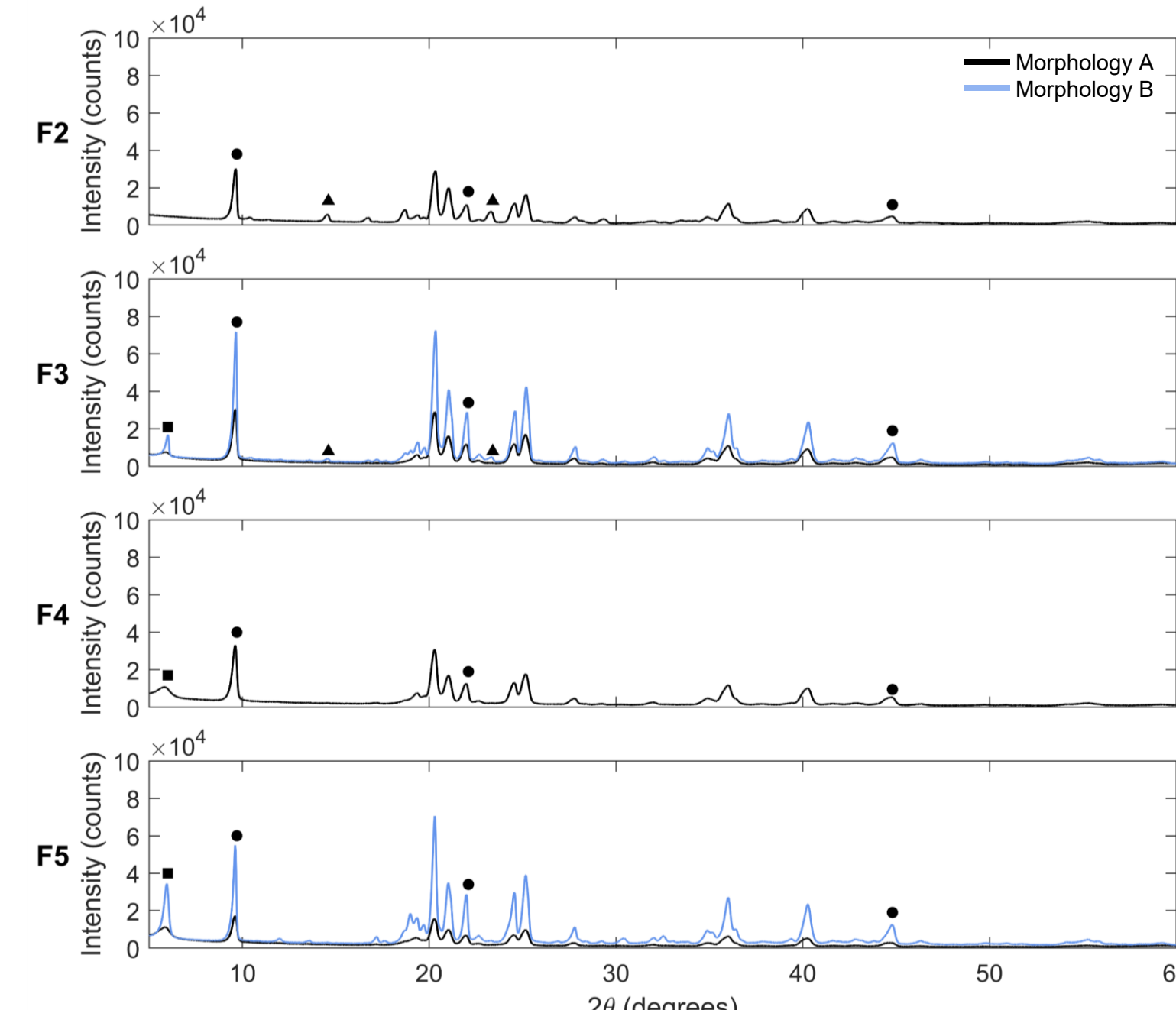
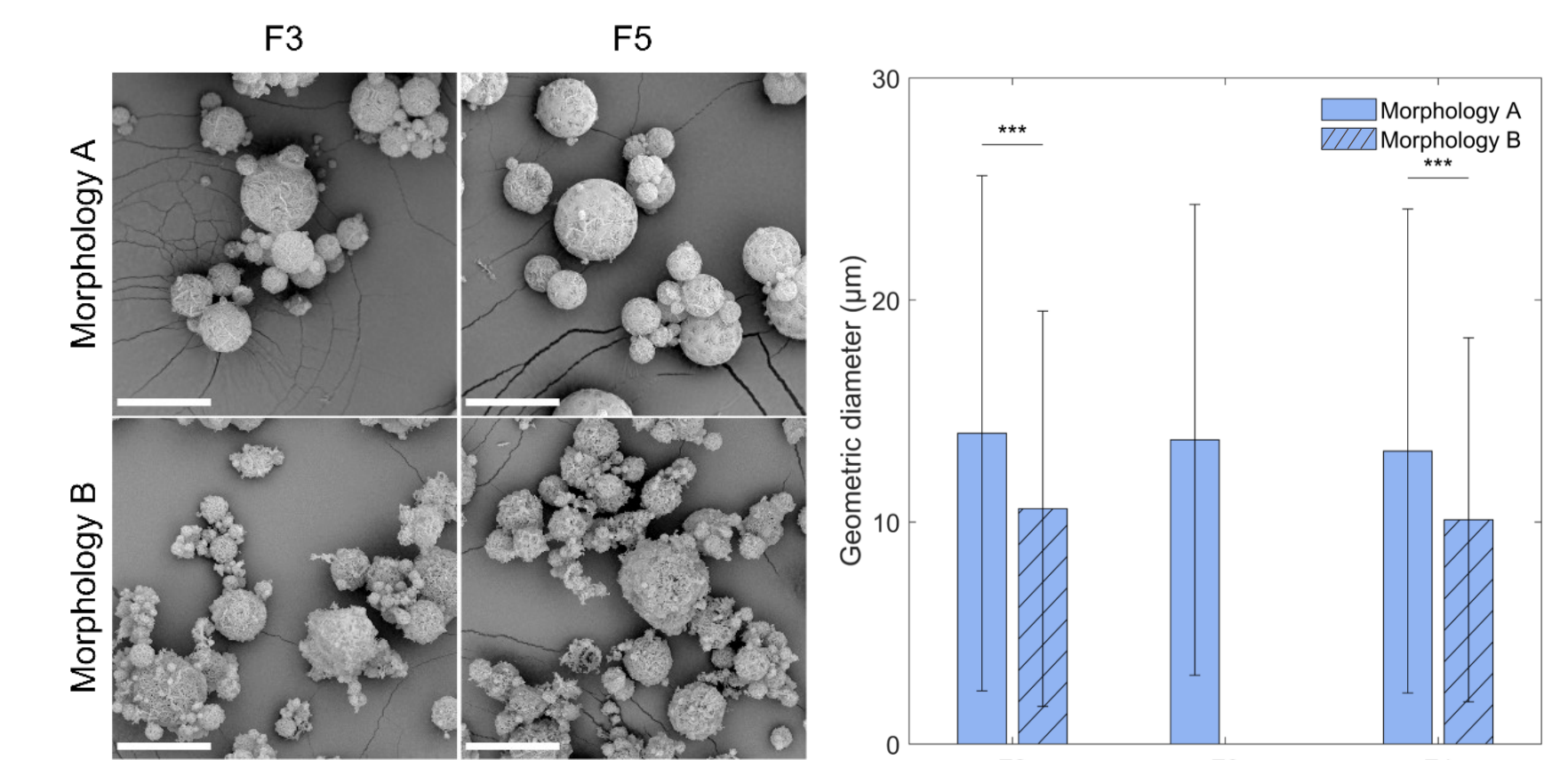
- x₁ = 5% (w/v)
 - x₂ = 7 NL/min
 - x₃ = 9 ml/min
- to obtain ↓ d_{ae} and avoid nozzle clogging.

Mannitol + SS + LL MPs

Morphology and size

Formation of two morphologies:

- Morphology A:**
 - well-defined shape
 - smooth surface
 - higher size
 - lower cohesion
- Morphology B:**
 - less-defined shape
 - rough surface
 - lower size
 - higher cohesion



Cristallinity

Presence of:
 (■) crystalline LL
 (▲) β-mannitol
 (●) δ-mannitol

- Morphology A:** ↑ R_{δ/β} and ↓ crystallinity
- Morphology B:** ↓ R_{δ/β} ↑ crystallinity

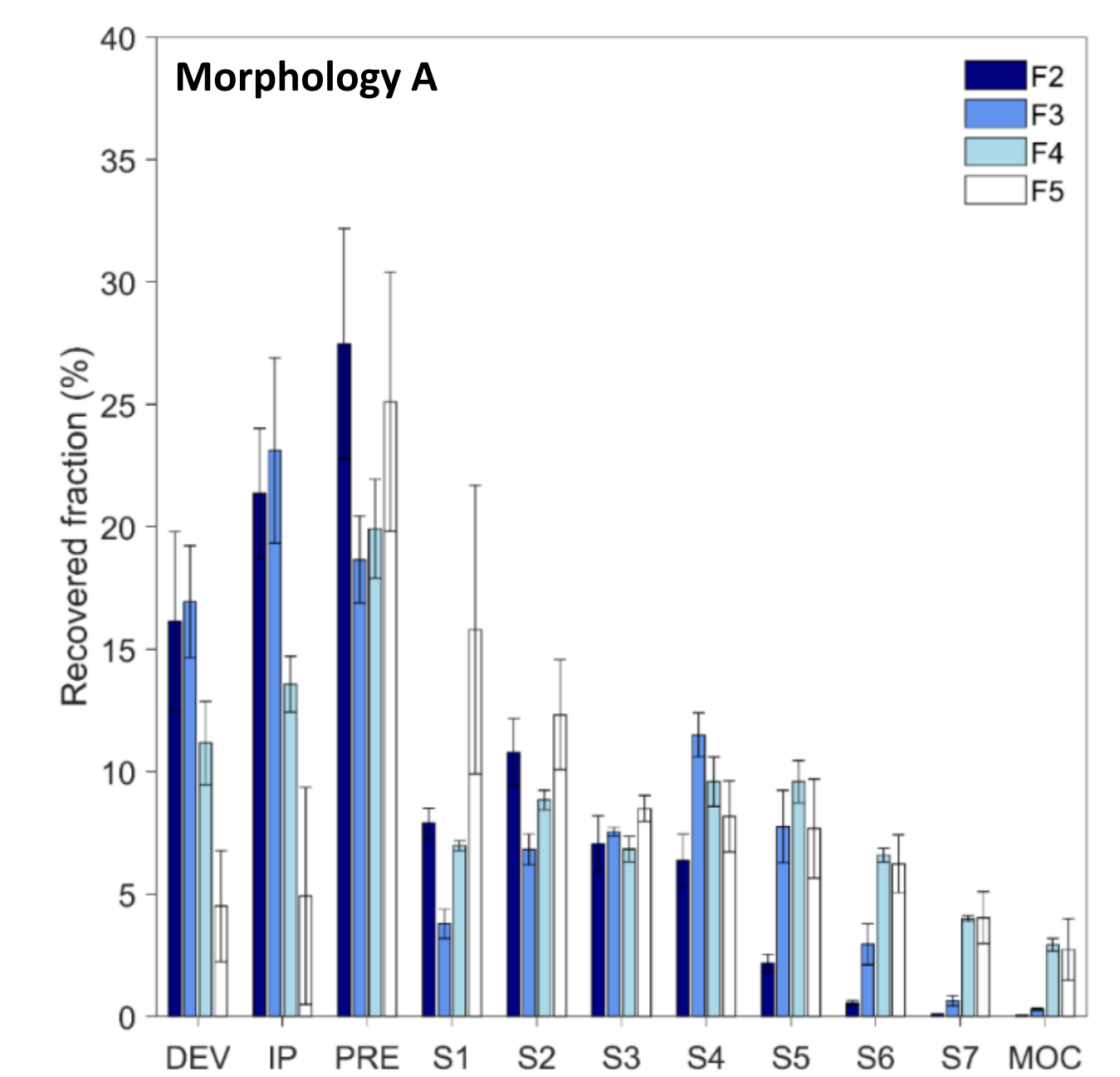
Which is the cause of morphology B?

- Humidity-induced recrystallization of mannitol and LL.

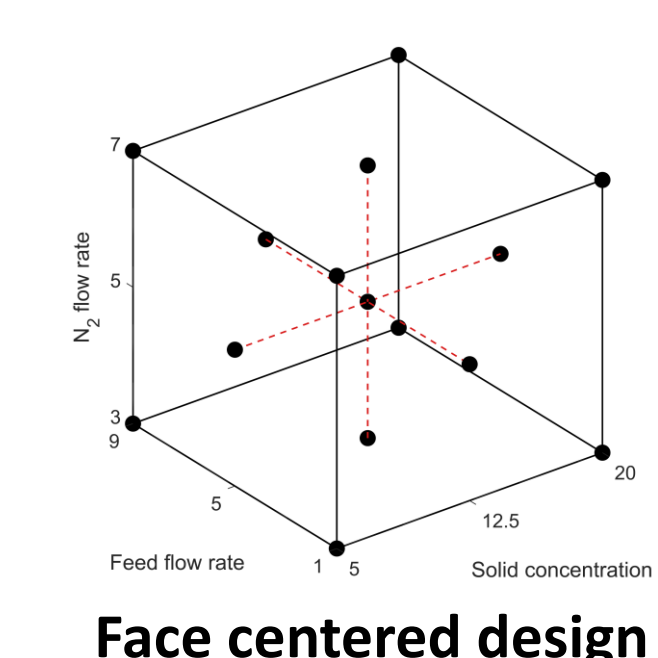
In vitro drug deposition

- Morphology A:** ↑ FPF and ↓ MMAD
- Morphology B:** ↓ FPF and ↑ MMAD
- Best condition: morphology A with 10% (w/w_{dw}) LL

Morphology	EF (%)	FPF (%)	MMAD (µm)	GSD (-)	
F2	-	83.8 ± 3.6	23.6 ± 4.9	4.5 ± 0.2	2.2 ± 0.1
F3	A	83.1 ± 2.3	39.8 ± 3.7	2.4 ± 0.1	2.4 ± 0.1
F3	B	92.1 ± 0.9	21.4 ± 4.1	4.8 ± 0.3	2.1 ± 0.1
F4	A	88.8 ± 1.7	48 ± 3	2.0 ± 0.1	3.2 ± 0.1
F5	A	95.5 ± 2.3	43.2 ± 7.5	3.5 ± 1.1	4.3 ± 0.7
F5	B	93.2 ± 1.1	15.8 ± 4.3	5.1 ± 0.6	2.4 ± 0.1



Conclusions



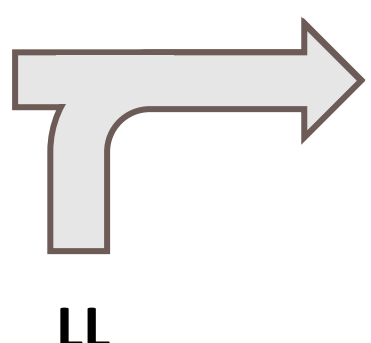
- Influence of the process variables on MPs size
- Relationship between MPs size, nucleation time, and polymorphism

↑ solid concentration → ↑ d_{ae}
 ↑ N₂ flow rate → ↓ d_{ae}
 ↑ feed flow rate → ↑ d_{ae}

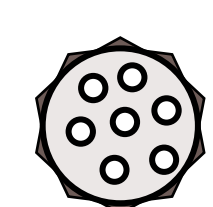
↑ size → ↑ δ-mannitol
 ↓ size → ↑ β-mannitol

Optimal condition:

- 5% (w/w) solid
- 7 NL/min
- 9 ml/min

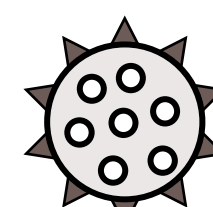


Morphology A



- ↓ crystallinity
 - Smooth surface
 - ↓ cohesiveness
- ⇒ ↑ FPF
 ↓ MMAD

Morphology B



- ↑ crystallinity
 - Rough surface
 - ↑ cohesiveness
- ⇒ ↓ FPF
 ↑ MMAD

References

- T. Duong et al., "A Pathway From Porous Particle Technology Toward Tailoring Aerogels for Pulmonary Drug Administration," *Frontiers in Bioengineering and Biotechnology*, vol. 9, 2021. doi: 10.3389/fbioe.2021.671381.
- S. Wanning, R. Süverkrüp, and A. Lamprecht, "Pharmaceutical spray freeze drying," *Int J Pharm*, vol. 488, no. 1–2, pp. 136–153, Jul. 2015, doi: 10.1016/j.ijpharm.2015.04.053.
- S. M. D'Addio, J. G. Y. Chan, P. C. L. Kwok, R. K. Prud'Homme, and H. K. Chan, "Constant size, variable density aerosol particles by ultrasonic spray freeze drying," *Int J Pharm*, vol. 427, no. 2, pp. 185–191, May 2012, doi: 10.1016/j.ijpharm.2012.01.048.
- D. A. Vishali, J. Monisha, S. K. Sivakamasundari, J. A. Moses, and C. Anandharamakrishnan, "Spray freeze drying: Emerging applications in drug delivery," *Journal of Controlled Release*, vol. 300, pp. 93–101, Apr. 2019, doi: 10.1016/j.jconrel.2019.02.044.
- H. Otake, T. Okuda, D. Hira, H. Kojima, Y. Shimada, and H. Okamoto, "Inhalable Spray-Freeze-Dried Powder with L-Leucine that Delivers Particles Independent of Inspiratory Flow Pattern and Inhalation Device," *Pharm Res*, vol. 33, no. 4, pp. 922–931, Apr. 2016, doi: 10.1007/s11095-015-1838-4.
- Y. Y. Lee, J. X. Wu, M. Yang, P. M. Young, F. Van Den Berg, and J. Rantanen, "Particle size dependence of polymorphism in spray-dried mannitol," in *European Journal of Pharmaceutical Sciences*, Sep. 2011, pp. 41–48. doi: 10.1016/j.ejps.2011.06.002.

