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## ***An imaging system for microbial corrosion analysis***

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# An imaging system for microbial corrosion analysis

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**Abstract**—This paper describes a viable and self-contained imaging system able to assess and quantify the effects of microbial corrosion on metals surface. The proposed image processing uses Scanning Electron Microscope micrographs to analyze bacteria attachment on sample surface and to estimate the degradation degree of the material. After a preliminary brightness and contrast normalization, which refines the image taken by the operator, the software is able to identify dark spots on the clear metal surface. These are then attributed to singly attached bacteria or to larger clusters, which are the most dangerous ones, as they could overlay corrosion pits. After that, the degradation of the material is evaluated through the quantification of microbial attachment on the surface and through dimensional distribution of bacteria clusters.

**Index Terms**—Microbial Corrosion, Image Processing, Implants, Scanning Electron Microscopy

## I. INTRODUCTION

Prosthesis implantation is a surgical operation in which the choice of material is an issue of paramount importance. The interaction between a material (especially in case of metal alloys) and the body fluids can lead to corrosion phenomena that, in the most serious cases, can cause the failure of the implant. A particularly severe degradation is due to microbial corrosion, in which bacteria catalyze metal oxidation, significantly increasing the corrosion rate.

Since corrosion is an electrochemical reaction, electrochemical techniques are generally the most effective approach in the study of material-environment interaction and of corrosion mechanisms [1]. In the field of biomaterials, Electrochemical Impedance Spectroscopy (EIS), monitoring of open circuit potential, polarization curves and other traditional electrochemical measurements are routinely employed to assess the corrosion behavior of different metals and alloys when in contact with electrolytic solutions that simulate the aggressiveness of body fluids. However, most of them cannot be considered as stand-alone techniques, as they

must be coupled to each other or to other chemicophysical characterization techniques such as optical and electron microscopy, in order to investigate the mechanism and the morphology of the corrosion reactions occurring on the metallic surface. From this considerations, the necessity to develop easy-to-use solutions able to quantitative evaluate the extent of the degradation phenomenon occurred on the material surface arises.

Many approaches related to imaging analysis when microbial corrosion (MIC) is concerned are possible. In particular, these methods can be classified taking into account the technique used for image collection or the main purpose of the investigation (corrosion assessment or biofilm growth). When surface morphology is the main concern, often stereomicroscope, reflected light microscopy and SEM (Scanning Electron Microscopy) are used [2]–[5]. In order to have a more accurate 3D reconstruction of the surface, white light interferometry or Fourier Transform Profilometry can be employed, so as to detect pit formation (pitting is a form of localized corrosion) [6], [7]. Main advantages of these techniques are related to the easy way in which corrosion pits can be detected and quantified (e.g. through parameters such as the pit density); drawbacks are related to the detection of biofilm and attached bacteria, as magnification is too low. On the other hand, to have the best morphological characterization of the biofilm, CLSM (Confocal Laser Scanning Microscope) is generally used [8]–[13]. This technique allows to have quantitative information about biofilm coverage and also its 3D structure, providing important information to researchers in biology. Main limitation is that sample preparation requires staining techniques, that are not always available in material science laboratories.

Aim of this research work is to develop a simple and fast imaging system able to quantitatively estimate the microbial corrosion effects and the biofilm growth on metallic specimens

subjected to MIC laboratory tests. Imaging based systems have been employed in several fields [14]–[16] mainly thanks to their low cost and non invasiveness [17]–[19].

The proposed approach is based on the processing of surface micrographs collected by means of a Scanning Electron Microscope (SEM), which provides a sufficient resolution at high magnification to examine bacteria attachment, even in case of low biofilm coverage. Electron microscopy is a traditional characterization technique routinely used in all material science laboratories, so it does not require additional equipment or skills. The system is based on the open-source openCV library, which can count many applications in the identification of specific objects present in images [20], [21].

After a preliminary processing of the image, which is described in section II, the software is able to identify bacteria aggregates that are present on the sample (section III). Obtained results are finally presented and discussed in section IV.

## II. IMAGE PROCESSING

### A. Image acquisition

To assess the microbial corrosion effects and the bacteria growth onto the metal, surface images are collected by means of a Field Emission Scanning Electron Microscope (FESEM - Supra 40 Zeiss). Fig. 1 shows, as an example, a FESEM image of a metallic specimen after a microbial corrosion test carried out in an electrolytic solution containing bacteria inoculum; after 200 h of exposure to the MIC test, a flat surface on which some bacteria agglomerates, highlighted as dark areas, can be observed. The image magnification is selected by the operator and automatically computed by the microscope; magnification can range from few nanometers per pixel to several tens of micrometers per pixel, while the image dimension can be  $2000 \times 1500$  pixels. Since the expected microorganisms have dimensions in the range of  $2 \mu\text{m}$  to  $15 \mu\text{m}$ , a resolution of the order of  $0.2 \mu\text{m}/\text{pixel}$  can be conveniently used; therefore, the analyzed area is of the order of  $400 \mu\text{m} \times 300 \mu\text{m}$ . This value has been used in all the images processed in this paper.

### B. Image brightness and contrast normalization

In addition to magnification and pixel size, the operator selects also the image brightness and contrast in order to obtain the best visualization of bacteria aggregates and localized corrosion attacks (pits). Setting these parameters is a critical operation, which makes any automatic image processing difficult to be applied.

In addition, the image brightness can change over the scanned surface because of the different topography and roughness and as a consequence of charging effects occurring when the electron beam hits non conductive species (bacteria aggregates). The low voltages used by the field emission microscope and the efficiency of its detector greatly help to reduce the charging effects making them almost negligible. However, brightness can still significantly change due to the different surface morphology in the different areas (i. e. different positions) of the scanned image [19].

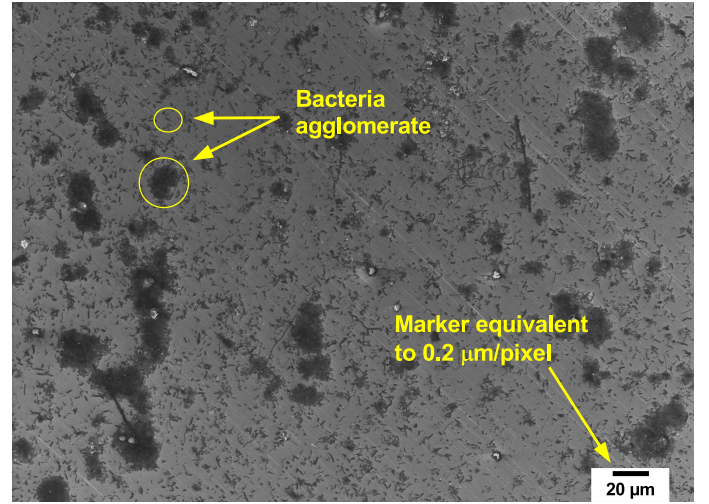


Fig. 1. FESEM image of a metallic surface affected by microbial corrosion.

Therefore, an image normalization has to be performed in order to obtain FESEM images whose contrast and brightness are reasonably independent from the operator; this processing involves three steps:

- 1) Estimation of brightness changes due to the different morphology of the surface area;
- 2) Correction of brightness changes in the different positions;
- 3) Normalization of the image brightness to eliminate the operator setting effect.

The first step requires identifying a brightness plane whose value is described by the following equation:

$$B = B_0 + \alpha_{BX}x + \alpha_{BY}y \quad (1)$$

where  $B$  is the brightness of the metallic surface area not interested by microbial corrosion,  $B_0$  is the average brightness value,  $\alpha_{BX}$  and  $\alpha_{BY}$  are the brightness changes with position coefficients, and  $x$  and  $y$  are the position coordinates.

Since the samples have an average brightness which is almost constant in the absence of corrosion, the identification of the brightness average value could be easily performed if one is able to find on the image the areas not affected by corrosion, i.e. in the case of the described images, areas not affected by bacteria growth. Such a manual selection can be avoided in most cases since the area covered by the bacteria is usually limited with respect to the total scanned metallic surface, so that the image brightness as a function of the coordinates can be easily estimated by solving in the Least Square sense the matrix equation:

$$\mathbf{B} = \mathbf{U} \cdot \lambda \quad (2)$$

where  $\mathbf{B}$  is the column vector of all pixels brightness,  $\mathbf{U}$  is  $n_{Pix} \times 3$  matrix in which the row is  $[1, x, y]$  and accounts for the position of each line in the  $\mathbf{B}$  vector and

$\lambda = [B_0, \alpha_{BX}, \alpha_{BY}]'$  is the vector containing the sensitivities whose estimation is referred to as  $\hat{\lambda}$ .

Of course the actual estimated brightness values depend on the way the images are acquired, i.e. on the FESEM operator; however, tests performed on several images with average brightness in the range of 140 to 220 out of a maximum value of 255 led to position coefficients whose maximum value is of the order of  $9 \cdot 10^{-3}$  on x axis and of  $3 \cdot 10^{-2}$  on the y axis. This leads to maximum brightness changes of the order of 40 units out of 255 units.

The second pre-processing step, once  $\hat{\lambda}$  has been estimated, can be simply obtained by computing the corrected brightness  $B_c$  as:

$$B_c = \frac{B}{1 + \alpha_{BX}x + \alpha_{BY}y} \quad (3)$$

obtaining therefore an image whose brightness is no longer affected by brightness changes in the different surface areas.

The third step can be obtained simply by changing image contrast by modifying the brightness values, so that they cover all the available gray levels. In order to avoid a normalization affected by the presence of few abnormal white and black spots, the image brightness  $B_{min}$  and  $B_{max}$  can be obtained by:

$$B_{min} = \sum_{pixelnumber} P_{xy} : B_p < B_{min}; kP_{tot} \quad (4)$$

$$B_{max} = \sum_{pixelnumber} P_{xy} : B_p > B_{max}; kP_{tot} \quad (5)$$

where  $P_{xy}$  is the generic pixel,  $P_{tot}$  is the number of pixels in the image and  $k$  is the coefficient usually in the range of 5% to 10%, which is used to define the number of pixels shown as black and white.

Once  $B_{min}$  and  $B_{max}$  have been estimated, the image brightness can be normalized as:

$$B_s = \begin{cases} 0, & \text{if } B \leq B_{min} \\ (B - B_{min}) \frac{M}{B_{max} - B_{min}}, & \text{if } B_{min} < B < B_{max} \\ Max_B, & \text{if } B \geq B_{max} \end{cases} \quad (6)$$

where  $Max_B$  is the maximum brightness value of the image. Fig. 2 shows four different examples of normalized images collected on metallic specimens which have undergone to the MIC test in different solutions. The normalized images appear to have similar brightness and contrast even though the bacteria proliferation and coverage appears to be quite different.

### III. BACTERIA IDENTIFICATION

On FESEM images collected on metallic samples exposed to MIC test, bacteria appear as dark areas whose dimension ranges from few squared micrometers for aggregate of few bacteria to several hundreds of micrometers for large bacteria aggregate.

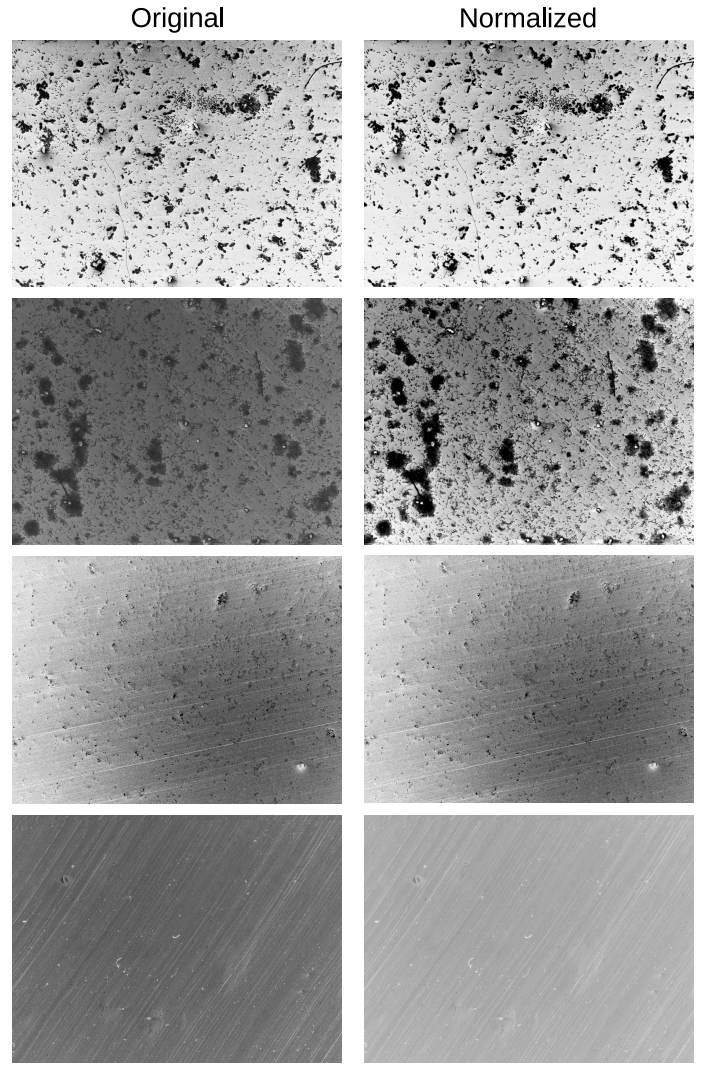


Fig. 2. Example of results of the normalization process on images with different brightness, contrast, and bacteria coverage.

The identification of the bacteria aggregates has been performed by using the OpenCV opensource environment [22], [23] by using the `findContour` function set to identify dark spots. This function returns a list of contours which can be easily shown on the FESEM image as it is possible to observe in Fig. 3. The figure shows normalized and identified images collected on four metallic specimens immersed in different solutions. Two of the images (Artificial seawater and Stankey solution) refer to samples immersed in solutions containing the inoculum and therefore rich of bacteria, the other two (Abiotic artificial seawater and Abiotic Stankey solution) refer to samples immersed in solutions having the same chemical composition but without the presence of bacteria.

In the identified images, as an example, a green contour has been used to flag aggregates with size of less than  $200 \mu\text{m}^2$  while a red contour has been used to flag bigger aggregates with size above  $200 \mu\text{m}^2$ .

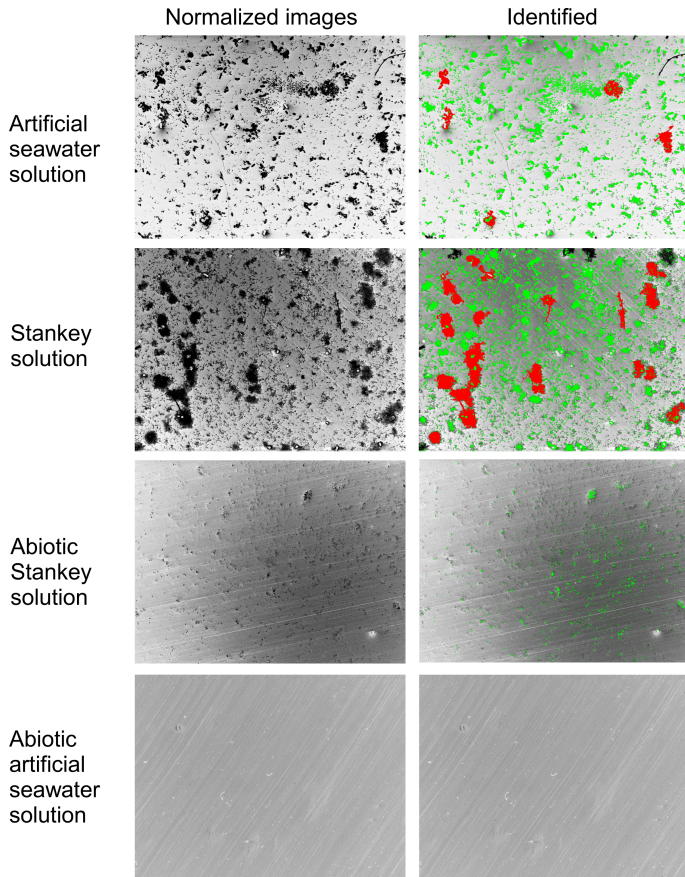


Fig. 3. Example of results of the identification process where in green are shown bacteria clusters with size below  $200 \mu\text{m}^2$  and in red cluster of bacteria above such value. Four examples are shown: in two cases samples were immersed in solution containing bacteria, and in the other two without bacteria. The corrosion difference is immediately visible.

#### IV. RESULTS

After a microbial corrosion test, FESEM characterization provides morphological information about biofilm growth and corrosion attacks onto the specimen surface.

Depending on the material and on the electrolytic solution employed in the MIC test, a different amount of microorganisms singly attached or agglomerated in big clusters can be found. So important information can be gathered by means of an imaging analysis able to quantitatively estimate the amount of bacteria grown on the metallic surface as a function of the aggressiveness of the MIC test; therefore the imaging approach could be very useful to compare the behaviors of different metals and alloys exposed to MIC and to put in evidence also the different aggressiveness of bacteria inocula.

In order to validate the proposed image processing, two standard microbial corrosion tests have been carried out, by immersing stainless steel specimens (AISI304, composition in Table I) in two different electrolytic solutions. AISI 304 Stainless Steel is an alloy basically employed in biomedical applications, such as orthopedic implants. The specimens have been immersed in two electrolytic solutions, the Stankey

medium and artificial seawater (solutions composition are reported in Table I). The former is a classic growth medium for bacteria, containing mainly sulphates and chlorides ions; the latter has a more complex composition simulating marine environment (also a favorable solution for microorganisms proliferation) and has been chosen in order to obtain a different corrosion morphology affecting the metallic surface after the test and compare it through the image analysis. Salt marsh sediments have been used as inoculum for the two solutions (it was added in the concentration of 10 vol%).

TABLE I  
MATERIAL AND ELECTROLYTIC SOLUTION COMPOSITION

Material	Chemical Composition [wt%]
AISI 304	Cr 18, Ni 10, Mn 2, Si 1, C 0.7
Electrolytic Solution	Concentration [g/L]
Stankey Medium	$K_2HPO_4$ 0.50, $NH_4Cl$ 1.00, $Na_2SO_4$ 1.00, $CaCl_2 \cdot 2H_2O$ 0.100, $MgSO_4 \cdot 7H_2O$ 2.00, $NaCl$ 24.53, $CH_3COONa$ 3.00
Artificial Seawater	$NaCl$ 24.53, $MgCl_2$ 5.20, $Na_2SO_4$ 4.09, $CaCl_2$ 1.16, $KCl$ 0.695, $NaHCO_3$ 0.201, $KBr$ 0.101, $H_3BO_3$ 0.027, $SrCl_2$ 0.025, $NaF$ 0.003, $CH_3COONa$ 3.00

Results from dimensional distribution of microorganism are shown in Fig. 4 and 5 as histograms for the samples immersed in artificial seawater and Stankey medium respectively. Data have been divided into two plots, one collecting single bacteria or small aggregates (having area up to  $20 \mu\text{m}^2$ ) and the other one grouping larger ones (until  $3000 \mu\text{m}^2$ ). Different information can be gathered arranging data in this way, as smaller objects provide indications about bacteria attachment, while bigger ones can be considered the most dangerous as far as microbial corrosion is concerned. Actually, they could generate preferential sites for pitting formation, as they create areas of differential aeration or where corrosive compounds (such as chlorides ions) can accumulate or pH changes can occur; so identifying and quantifying them is an important point to assess the microbial corrosion behavior of the metallic specimen. In the histograms, aggregates size is plotted as a function of covered surface percentage, in order to keep in the same graph the contribution arising from single bacteria (high in number but with very small dimensions) and larger aggregates.

As can be seen in Fig. 3, the software is able to correctly identify almost all microorganisms present on the sample. Main limitations are due to agglomerates which lie near the edges of the image, as their contour can't be determined by the `findContour` algorithm. In virtually all other cases, bacteria are identified and labeled according to their dimension. Thus, analyzing data from the histograms, it is possible to recognize the different morphologies of the two samples. Immersion in artificial seawater lead to a condition in which bacteria are singularly and randomly attached on the surface with only few bigger agglomerates, which can be recognized in the histogram

TABLE II  
RESULTS OBTAINED FOR SAMPLES IMMERSSED IN ARTIFICIAL SEAWATER, IN STANKEY SOLUTION AND IN ABIOTIC STANKEY SOLUTION.

Electrolytic Solution	Number of Small aggl.	Total Surface of small aggl. ( $\mu\text{m}^2$ )	Average surface ( $\mu\text{m}^2$ )	Percentage of covered surface	Number of Large aggl.	Total Surface of large aggl. ( $\mu\text{m}^2$ )	Average surface ( $\mu\text{m}^2$ )	Percentage of covered surface
Artificial seawater	6660	11000	1.6	8.7	5	1640	380	1.3
Stankey	16400	15700	0.9	12.5	16	8000	500	6.3
Abiotic Stankey	20300	3800	4.5	2.9	-	-	-	-

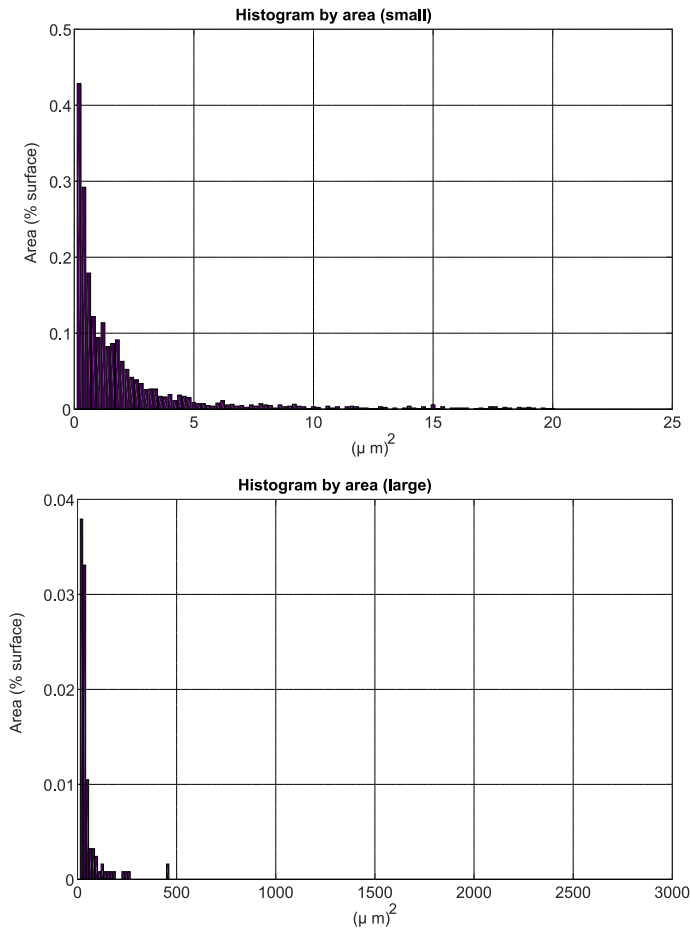


Fig. 4. Histograms showing dimensional distribution (expressed as agglomerate area) as a function of the covered surface percentage for the samples immersed in artificial seawater.

ranging until  $3000 \mu\text{m}^2$ . On the other hand, immersion in Stankey solution induced the formation of a higher number of bigger aggregates (area above  $20 \mu\text{m}^2$ ), which can also be quantified through their density:

$$\text{aggregates density} = \frac{\text{number of aggregates}}{\text{surface area}} \quad (7)$$

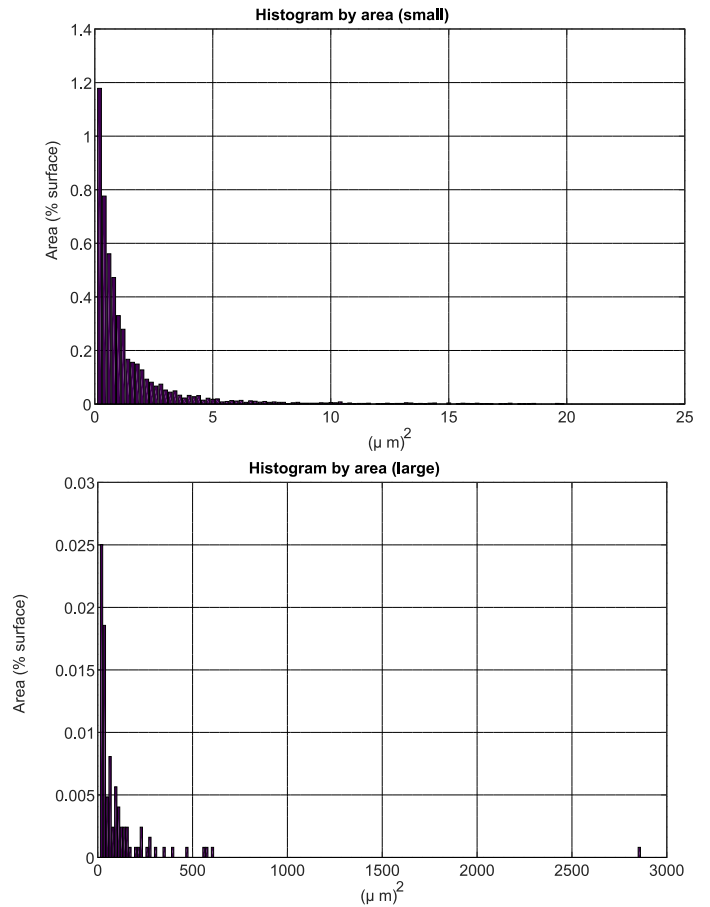


Fig. 5. Histograms showing dimensional distribution (expressed as agglomerate area) as a function of the covered surface percentage for the samples immersed in the Stankey solution.

This parameter has been found equal to  $41.7 \text{ mm}^{-2}$  for sample immersed in artificial seawater and equal to  $133.3 \text{ mm}^{-2}$  for the sample immersed in Stankey solution, providing an important additional clue for the material characterization. Such values have been calculated dividing the total number of bigger aggregates (reported in Table II) by the micrograph area (expressed as  $\text{mm}^{-2}$ )

Finally, another useful parameter that can be derived from the image processing is the biofilm coverage, defined as:

$$\text{biofilm coverage} = \frac{\text{pixel covered by bacteria}}{\text{total number of pixel}} \cdot 100 \quad (8)$$

Its value has been calculated for the two samples, resulting 10.0 % for the test in artificial seawater and 18.8 % for the test in Stankey medium. Table II summarizes further data that can be gained from the processing, such as: number of aggregates (distinguished according to the size), their total surface, the average surface of aggregates and the percentage of covered surface (from which the biofilm coverage can be calculated summing the percentage of covered surface deriving from small and from large aggregates).

As a further analysis, analogous corrosion tests have been carried out in sterile conditions (using the same electrolytic solutions and material). In this case of course no microorganisms were present, but information about pitting corrosion could be gained through the image processing. Same magnification was used to collect micrographs and results from contour identification are presented in Fig. 3. In the case of Stankey solution, the sample revealed initial hints of degradation, properly detected and classified by the software. On the other hand, for the artificial seawater sample no signs of corrosion could be found, as clearly visible from the FESEM micrograph.

## V. CONCLUSIONS

A novel imaging system has been proposed for bacteria identification after microbial corrosion test. The processing, based on OpenCV environment, is able to quantify microbial attachment on specimen surface, providing to researchers important information about material degradation, such as biofilm coverage, aggregates density and bacteria dimensional distribution. Results obtained from the software have been presented, showing the effectiveness in bacteria identification and the main information that can be derived from the processing.

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